

IUCLID

Data Set

Existing Chemical : ID: 68526-84-1 CAS No. : 68526-84-1

EINECS Name : Alcohols, C8-10-iso-, C9-rich

: 271-233-5 EC No.

: Alcohols, C8-10-iso-, C9-rich TSCA Name IUPAC Name : Alcohols C8-C10, Iso-, C9 rich

Producer related part

Company : ExxonMobil Biomedical Sciences Inc.

Creation date : 29.09.2004

Substance related part

Company : ExxonMobil Biomedical Sciences Inc.

Creation date : 29.09.2004

Status

Memo : Prepared for EMCC - US HPV

Printing date : 06.04.2006

Revision date

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Material Safety Dataset, Risk Assessment, Directive 67/548/EEC, SIDS

ld 68526-84-1

Date

1.0.1 APPLICANT AND COMPANY INFORMATION

1.0.2 LOCATION OF PRODUCTION SITE, IMPORTER OR FORMULATOR

1.0.3 IDENTITY OF RECIPIENTS

1.0.4 DETAILS ON CATEGORY/TEMPLATE

Comment: see free text

Remark : The Alkyl Alcohols C6 to C13 Category is a family of saturated alcohols

that are produced from olefins by the hydroformylation or "oxo" process. Hydroformylation is the reaction of an olefin with carbon monoxide and hydrogen to produce an aldehyde, and its subsequent hydrogenation to the

alcohol.

The number of carbon atoms in the category members ranges from 6 to 13. Category members contain predominantly branched alkyl groups. Each substance consists of an isomeric mixture of saturated primary alcohols of high purity, and the following basic structure: CH3-R-CH2-OH, where R is a branched isomeric structure.

The justification for the Alkyl Alcohols C6 to C13 Category is that the members have:

- · similar chemical structures,
- · similar physico-chemical properties,
- · comparable environmental fate,
- · the same mode of action.

In general, aliphatic alcohol toxicity occurs by non-polar narcosis (Lipnick et al., 1985). The mode of action is disruption of biological membrane function (van Wezel and Opperhuizen, 1995). Metabolic pathways, through which alcohols are metabolized, are likely to include similar reactions for all category members and result in structurally similar metabolites (carboxylic acids).

The data demonstrate that the category is valid for a screening-level hazard assessment for the Category and its members. One can assess the untested endpoints by extrapolation between and among the category members.

20.03.2006

1.1.0 SUBSTANCE IDENTIFICATION

IUPAC Name : Smiles Code :

Molecular formula : C9H20O Molecular weight : 144.26

Petrol class

Flag : Critical study for SIDS endpoint

11.01.2005

1. General Information

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1.1.1 GENERAL SUBSTANCE INFORMATION

Purity type

Substance type : organic Physical status : liquid

Purity : Colour : Odour :

Remark: CAS Registry Number, Name, and General Structure for Members of the

Alkyl Alcohols C6 to C13 Category and Analogue Substances:

CAS RN: 85566-14-9

IUPAC Name: Alcohols, C7-11-branched and linear

R Length (C number): C7 to C11

Structure of R: Linear

Category Member: No (analogue substance used for supporting

information)

CAS RN: 68526-84-1

IUPAC Name: Alcohols C8-C10-iso, C9 rich

R Length (C number): C7

Structure of R: Branched, primarily methyl branched (isomeric structures)

Category Member: Yes

06.04.2006

1.1.2 SPECTRA

1.2 SYNONYMS AND TRADENAMES

Exxal(r) 9

12.03.2004

Isononanol

09.03.2004

Isononyl alcohol

09.03.2004

1.3 IMPURITIES

Purity : typical for marketed substance

CAS-No : 68526-84-1 **EC-No** : 271-233-5

EINECS-Name : Alcohols, C8-10-iso-, C9-rich

Molecular formula : C9H20O Value : = 99.6 % w/w

Remark : Commercial product typically consists of dimethyl-1-heptanols and methyl-

1-octanols.

06.01.2005

1. General Information

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1.4 ADDITIVES

Purity type : typical for marketed substance

CAS-No : EC-No :

EINECS-Name Molecular formula Value

Function of additive :

Remark : No additives present

11.10.2005

1.5 TOTAL QUANTITY

1.6.1 LABELLING

1.6.2 CLASSIFICATION

1.6.3 PACKAGING

1.7 USE PATTERN

Type of use : industrial

Category : Chemical industry: used in synthesis

Remark: Primarily used as a chemical intermediate in the production of esters:

phthalates, acetates, and adipates. Also frequently used in the paint

industry.

06.01.2005

1.7.1 DETAILED USE PATTERN

1.7.2 METHODS OF MANUFACTURE

1.8 REGULATORY MEASURES

1.8.1 OCCUPATIONAL EXPOSURE LIMIT VALUES

1.8.2 ACCEPTABLE RESIDUES LEVELS

1.8.3 WATER POLLUTION

Id 68526-84-1 1. General Information Date 06.04.2006 1.8.4 MAJOR ACCIDENT HAZARDS 1.8.5 AIR POLLUTION 1.8.6 LISTINGS E.G. CHEMICAL INVENTORIES 1.9.1 DEGRADATION/TRANSFORMATION PRODUCTS 1.9.2 COMPONENTS 1.10 SOURCE OF EXPOSURE 1.11 ADDITIONAL REMARKS 1.12 LAST LITERATURE SEARCH 1.13 REVIEWS

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2.1 MELTING POINT

Value : = -54 °C Decomposition : no, at °C

Sublimation: no

Method : other: ASTM D97

Year

GLP : no Test substance :

Test substance : CAS No. 68526-84-1; Alcohols, C8-C10-iso, C9 rich

Reliability : (2) valid with restrictions

Although the original data were not retrieved and reviewed for quality, they

were developed following acceptable test methods and therefore

considered reliable.

Flag : Critical study for SIDS endpoint

06.04.2006 (1) (16)

2.2 BOILING POINT

Value : $= 202 - 219 \, ^{\circ}\text{C}$ at 1013 hPa

Decomposition

Method : other: ASTM D1078/01

Year

GLP : no data

Test substance :

Test substance : CAS No. 68526-84-1; Alcohols, C8-C10-iso, C9 rich

Reliability : (2) valid with restrictions

Although the original data were not retrieved and reviewed for quality, they

were developed following acceptable test methods and therefore

considered reliable.

Flag : Critical study for SIDS endpoint

06.04.2006 (16)

2.3 DENSITY

Type : density

Value : .832 g/cm³ at 20 °C

Method : other: ASTM D4052/86 equivalent

Year : 1998 GLP : no data

Test substance :

Method : The density was measured by taking the weight of 10 mls of the test

substance at 20 deg C.

Test substance : CAS No. 68526-84-1; Alcohols, C8-C10-iso, C9 rich

Reliability : (1) valid without restriction

06.04.2006 (11)

2.3.1 GRANULOMETRY

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2.4 VAPOUR PRESSURE

Value : = .054 hPa at 25 °C

Decomposition

Method : other (calculated)

Year

GLP : no data

Test substance

Method : Vapor pressure calculation by MPBPWIN ver. 1.40 using calculation

method of Grain.

Remark: EPIWIN is used and advocated by the US EPA for chemical property

estimation.

Test substance : CAS No. 68526-84-1; Alcohols, C8-C10-iso, C9 rich

Reliability : (2) valid with restrictions

The value was calculated based on chemical structure as modeled by EPIWIN. This robust summary has a reliability rating of 2 because the data

are calculated and not measured.

Flag : Critical study for SIDS endpoint

06.04.2006 (20)

2.5 PARTITION COEFFICIENT

Partition coefficient :

Log pow : = 3.8 - 4.3 at 30 °C

pH value

Method : OECD Guide-line 117 "Partition Coefficient (n-octanol/water), HPLC

Method"

Year : 1998 GLP : yes Test substance :

Remark : Test Type: N-Octanol/Water Partition Coefficient (HPLC method)

Result : The test substance eluted as several groups. The three major components

C8, C9, C10 alcohols had Log Pow values of 3.8, 4.2, and 4.3 respectively.

The retention time for the 3 major components were 6.91, 8.42, and 8.96

minutes.

All values were measured using High Performance Liquid Chromatography

HPLC).

Test condition: The test substance was evaluated as a solution in HPLC grade methanol.

Six reference compounds were also evaluated in a standard combined reference solution (2-butanone, acetophenone, naphthalene, biphenyl, n-butylbenzene, and 4,4-DDT) in 75% methanol and 25% distilled water.

The pH of the solution was 5.4.

Two customized alcohol reference solutions were also prepared containing five of the ten alcohol compounds (1-hexanol, 1-heptanol, 1-octanol, 1-

nonanol, 1-decanol, 1-undecanol, 1-dodecanol, 1-tridecanol, 1-tetradecanol, 1-pentadecanol) in 87.5% methanol and 12.5% distilled

water. The pH of both solutions was 7.3.

The pH of the evaluated solutions was the same as the reference solution it

was evaluated against.

The test substance was analyzed against a Standard Log Pow Reference Compound Solution and a customized Alcohol Reference Compound Solution. Only the peaks detected by refractive index (RI) were reported.

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: CAS No. 68526-84-1; Alcohols, C8-C10-iso, C9 rich

Test substance Reliability : (1) valid without restriction : Critical study for SIDS endpoint Flag

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2.6.1 SOLUBILITY IN DIFFERENT MEDIA

Solubility in : Water

Value $= 90.4 \text{ mg/l at } 25 ^{\circ}\text{C}$

pH value

at °C concentration

Temperature effects

Examine different pol.

pKa at 25 °C

Description

Stable

Deg. product

Method other: calculated

Year 1998 no data **GLP**

Test substance

: Water solubility calculated using WSKOWWIN ver 1.36 based on Kow Method

correlation method of Meylan and Howard.

Remark : EPIWIN is used and advocated by the US EPA for chemical property

Test substance : CAS No. 68526-84-1; Alcohols, C8-C10-iso, C9 rich

Reliability (2) valid with restrictions

The value was calculated based on chemical structure as modeled by EPIWIN. This robust summary has a reliability rating of 2 because the data

are calculated and not measured.

Flag : Critical study for SIDS endpoint

06.04.2006 (5)

2.6.2 SURFACE TENSION

2.7 FLASH POINT

= 90 °C Value Type closed cup

other: PMCC ASTM D93 Method

Year

GLP no data

Test substance

Test substance : Analogue substance: Isononyl alcohol (CAS 27458-94-2)

(2) valid with restrictions Reliability

Although the original data were not retrieved and reviewed for quality, they

were developed following acceptable test methods and therefore

considered to be reliable.

06.04.2006 (16)

AUTO FLAMMABILITY 2.8

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2.9 FLAMMABILITY

2.10 EXPLOSIVE PROPERTIES

2.11 OXIDIZING PROPERTIES

2.12 DISSOCIATION CONSTANT

Acid-base constant : = 15.935 at 25°C

Method : other: calculated

Year

GLP : no data

Test substance :

Method : pKa calculation by SPARC 2003 using a Linux calculation engine.

Remark : SPARC On-line calculator can be accessed at

http://ibmlc2.chem.uga.edu/sparc/index.cfm

Test substance : CAS No. 68526-84-1; Alcohols, C8-C10-iso, C9 rich

Reliability : (2) valid with restrictions

The value was calculated based on chemical structure as modeled by SPARC. This robust summary has a reliability rating of 2 because the data

are calculated and not measured.

06.04.2006 (17)

2.13 VISCOSITY

Value : ca. 17 - at 20 °C

Result

Method : other: ASTM D445

Year

GLP : no data

Test substance : other TS: Isononyl alcohol (CAS 27458-94-2)

Remark : Value measured in cSt

Test substance : Analogue substance: Isononyl alcohol (CAS 27458-94-2)

Reliability : (2) valid with restrictions

Although the original data were not retrieved and reviewed for quality, they

were developed following acceptable test methods and therefore

considered to be reliable.

26.10.2005 (16)

2.14 ADDITIONAL REMARKS

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3.1.1 PHOTODEGRADATION

Type : water

Light source

Light spectrum: nm

Relative intensity : based on intensity of sunlight

Deg. product :

Method : other (calculated): Technical Discussion

Year : GLP : Test substance :

Remark: These data represent a key study for characterising the potential of

substances in the Alkyl Alcohols C6 to C13 Category to undergo direct

photodegradation.

Result: Photolysis as a Function of Molecular Structure

The direct photolysis of an organic molecule occurs when it absorbs sufficient light energy to result in a structural transformation (Harris, 1982). The reaction process is initiated when light energy in a specific wavelength range elevates a molecule to an electronically excited state. However, the excited state is competitive with various deactivation processes that can result in the return of the molecule to a non excited state.

The absorption of light in the ultra violet (UV)-visible range, 110-750 nm, can result in the electronic excitation of an organic molecule. Light in this range contains energy of the same order of magnitude as covalent bond dissociation energies (Harris, 1982). Higher wavelengths (e.g. infrared) result only in vibrational and rotational transitions, which do not tend to produce structural changes to a molecule.

The stratospheric ozone layer prevents UV light of less than 290 nm from reaching the earth's surface. Therefore, only light at wavelengths between 290 and 750 nm can result in photochemical transformations in the environment (Harris, 1982). Although the absorption of UV light in the 290-750 nm range is necessary, it is not always sufficient for a chemical to undergo photochemical degradation. Energy may be re-emitted from an excited molecule by mechanisms other than chemical transformation, resulting in no change to the parent molecule.

A conservative approach to estimating a photochemical degradation rate is to assume that degradation will occur in proportion to the amount of light wavelengths >290 nm absorbed by the molecule (Zepp and Cline, 1977).

Substances in the Oxo Alcohols C6 to C13 Category contain molecules that are oxygenated aliphatic compounds which will absorb UV light below 220 nm (Boethling and Mackay, 2000) and will not undergo direct photolysis. Therefore, this fate process will not contribute to a measurable degradative removal of chemical components in this category from the environment.

References

Boethling, R.S., Mackay, D. (2000). Handbook of Property Estimation Methods for Chemicals. CRC Press, Boca Raton, FL, USA.

Harris, J. C. 1982. "Rate of Aqueous Photolysis," Chapter 8 in: W. J. Lyman, W. F. Reehl, and D. H. Rosenblatt, eds., Handbook of Chemical

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Property Estimation Methods, McGraw-Hill Book Company, New York,

USA.

Zepp, R. G. and D. M. Cline. 1977. Rates of Direct Photolysis in the

Aqueous Environment, Environ. Sci. Technol., 11:359-366.

Test substance : Alkyl Alcohols C6 to C13 Category members

Flag : Critical study for SIDS endpoint

03.04.2006 (15)

INDIRECT PHOTOLYSIS

Sensitizer : OH

Conc. of sensitizer : 1500000 molecule/cm³

Rate constant : = $.000000000139469 \text{ cm}^3/(\text{molecule*sec})$

Degradation: % after

Deg. product

Method : other (calculated): Calculated values using AOPWIN version 1.89, a

subroutine of the computer program EPIWIN version 3.04

Year : 1999
GLP : no
Test substance :

Result: Atmospheric Oxidation Potential

In the environment, organic chemicals emitted into the troposphere are degraded by several important transformation processes. The dominant transformation process for most compounds is the daylight reaction with hydroxyl (OH-) radicals (Atkinson, 1988, 1989). The rate at which an organic compound reacts with OH- radicals is a direct measure of its atmospheric persistence (Meylan and Howard, 1993).

AOPWIN estimates the rate constant for the atmospheric, gas-phase reaction between photochemically produced hydroxyl radicals and organic chemicals. The rate constants estimated by the program are then used to calculate atmospheric half-lives for organic compounds based upon average atmospheric concentrations of hydroxyl radicals.

Since the reactions only take place in the presence of sunlight, the atmospheric half-lives are normalized for a 12-hour day.

Calculated* OH- Rate Constant half-life (hrs) (cm3/molecule-sec)

9.2 13.95 E-12

References:

Atkinson, R. 1988. Estimation of gas-phase hydroxyl radical rate constants for organic chemicals. Environ. Toxicol. Chem. 7:435-442.

Atkinson, R. 1989. Kinetics and mechanisms of the gas-phase reactions of the hydroxyl radical with organic compounds. J. Phys. Chem. Ref. Data Monograph No. 1, Amer. Inst. Physics & Amer. Chem. Soc., NY.

Meylan, W.M. and P.H. Howard. 1993. Computer estimation of the atmospheric gas-phase reaction rate of organic compounds with hydroxyl

radicals and ozone. Chemosphere 12:2293-2299.

Test condition : Indirect photodegradation, or atmospheric oxidation potential, is based on

the structure-activity relationship methods developed by R. Atkinson.

Temperature: 25°C

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Sensitizer: OH radical

Concentration of Sensitizer: 1.5 E6 OH radicals/cm3 CAS No. 68526-84-1; Alcohols, C8-C10-iso, C9 rich

: (2) valid with restrictions Reliability

> The results include calculated data based on chemical structure as modeled by AOPWIN. The data represent a potential atmospheric half-life range for the test substance. This robust summary represents a "key

study" for atmospheric half-life range based on constituent data.

Critical study for SIDS endpoint Flag

06.04.2006 (21)

3.1.2 STABILITY IN WATER

Test substance

Type abiotic t1/2 pH4 at °C t1/2 pH7 at °C t1/2 pH9 at °C

Deg. product

Method other: Technical Discussion

Year

GLP

Test substance

Remark These data represent a key study for characterising the potential of

substances in the Alkyl Alcohols C6 to C13 Category to undergo

hvdrolvsis.

Result : Hydolysis as a Function of Molecular Structure

> Hydrolysis of an organic molecule occurs when a molecule (R-X) reacts with water (H2O) to form a new carbon-oxygen bond after the carbon-X bond is cleaved (Gould, 1959; Harris, 1982). Mechanistically, this reaction is referred to as a nucleophilic substitution reaction, where X is the leaving group being replaced by the incoming nucleophilic oxygen from the water molecule.

> Chemicals that are susceptible to hydrolysis contain functional groups that can be displaced by a nucleophilic substitution reaction. Substances that have the potential to hydrolyze include alkyl halides, amides, carbamates, carboxylic acid esters and lactones, epoxides, phosphate esters, and sulfonic acid esters (Neely, 1985). The lack of a leaving group renders a compound resistant to hydrolysis.

Alkyl alcohols are resistant to hydrolysis because they lack a functional group that is hydrolytically reactive (Harris, 1982).

References:

Gould, E.S. (1959). Mechanism and Structure in Organic Chemistry, Holt, Reinhart and Winston, New York, NY, USA.

Harris, J.C. (1982). "Rate of Hydrolysis," Chapter 7 in: W.J. Lyman, W.F. Reehl, and D.H. Rosenblatt, eds., Handbook of Chemical Property Estimation Methods, McGraw-Hill Book Company, New York, NY, USA.

Neely, W.B. (1985). Hydrolysis. In: W.B. Neely and G.E. Blau, eds., Environmental Exposure from Chemicals. Vol. 1, pp. 157-173. CRC Press, Boca Raton, FL. USA.

Conclusion Hydrolysis will not contribute to the removal of Alkyl Alcohols from the

environment.

Flag Critical study for SIDS endpoint

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20.03.2006 (14)

3.1.3 STABILITY IN SOIL

3.2.1 MONITORING DATA

3.2.2 FIELD STUDIES

3.3.1 TRANSPORT BETWEEN ENVIRONMENTAL COMPARTMENTS

Type : fugacity model level III

Media: other: air - water - soil - sedimentAir: % (Fugacity Model Level I)Water: % (Fugacity Model Level I)Soil: % (Fugacity Model Level I)Biota: % (Fugacity Model Level II/III)Soil: % (Fugacity Model Level II/III)

Method : other: Calculation according Mackay, Level III

Year : 2003

Method: The EQC Level III model is a steady state model that is useful for

determining how the medium of release affects environmental fate. Level III fugacity allows non-equilibrium conditions to exist between connected media as steady state, and illustrate important transport and transformation

processes.

Physicochemical input values for the model were calculated using the EPIWIN Estimation v 3.04 program. Measured input values were also used where available and obtained from the EPIWIN database. Distribution data from the equilibrium model provide basic information on the potential partitioning behavior of chemicals between selected environmental compartments (i.e., air, water, soil, and sediment).

Input values used:

Molecular mass = 144.26 g/mol Water solubility = 90.4 mg/L Vapour pressure = 5.4 Pa

log Kow = 4.05

Melting point = $-54 \deg C$

Degradation half-lives:

Air - 9.2 hrs Water - 120 hrs Soil - 720 hrs Sediment - 7200 hrs

This model was run assuming 100% discharge to water.

Result : Air - 0.8%

Water - 79.5% Soil - 0.03% Sediment - 19.7%

Test substance : CAS No. 68526-84-1; Alcohols, C8-C10-iso, C9 rich

Reliability : (2) valid with restrictions

This robust summary has a reliability rating of 2 because the data are

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calculated and not measured.

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Type : fugacity model level I

Media : other: air - biota - sediment(s) - soil - water

Air : % (Fugacity Model Level I)

Water : % (Fugacity Model Level I)

Soil : % (Fugacity Model Level I)

Biota : % (Fugacity Model Level II/III)

Soil : % (Fugacity Model Level II/III)

Method : other: Calculation according Mackay, Level I

Year : 2003

Method : The EQC Level I is a steady state, equilibrium model that utilizes the input

of basic chemical properties including molecular weight, vapor pressure, and water solubility to calculate distribution within a standardized regional

environment.

Physicochemical input values for the model were calculated using the EPIWIN Estimation v 3.04 program. Measured input values were also used where available and obtained from the EPIWIN database. Distribution data from the equilibrium model provide basic information on the potential partitioning behavior of chemicals between selected environmental compartments (i.e., air, water, soil, sediment, suspended sediment, biota).

Input values used:

Molecular mass = 144.26 g/mol Water solubility = 90.4 mg/L Vapour pressure = 5.4 Pa

log Kow = 4.05

Melting point = -54 deg C

Remark: Results were calculated using measured value for log Kow

Result : Air - 19.9%

Water - 7.2% Soil - 71.3% Sediment - 1.6%

Suspended Sed - 0.04%

Biota - < 0.01%

Test substance : CAS No. 68526-84-1; Alcohols, C8-C10-iso, C9 rich

Reliability : (2) valid with restrictions

This robust summary has a reliability rating of 2 because the data are

calculated and not measured.

26.10.2005 (19)

Type : volatility
Media : water - air

Air : % (Fugacity Model Level I)
Water : % (Fugacity Model Level I)
Soil : % (Fugacity Model Level I)
Biota : % (Fugacity Model Level II/III)
Soil : % (Fugacity Model Level II/III)

Method : other: calculation

Year : 2005

Method : Henry's Law Constants (HLC) are based on vapor pressure and water

solubility values, and molecular weights. HLC values were calculated

using equations found in Mackay, 2003.

Vapor pressure = 5.4 Pa Water solubility = 90.4 mg/l Molecular mass = 144.26 g/mol

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Result : HLC (at 25 deg C) = 8.62 Pa-m3/mole

Test substance : CAS No. 68526-84-1; Alcohols, C8-C10-iso, C9 rich

Reliability : (2) valid with restrictions

This robust summary has a reliability rating of (2) because the data are

calculated and not measured.

Flag : Critical study for SIDS endpoint

26.10.2005 (19)

3.3.2 DISTRIBUTION

Media : other: Koc

Method : other (calculation)

Year : 2005

Method : The soil adsorption coefficient (Koc) was calculated by PCKOCWIN ver.

1.66, using equations developed by Lyman (1980).

PCKOCWIN is a subroutine of EPIWIN. EPIWIN is used and advocated by the USEPA for chemical property estimation. Koc provides an indication of the extent to which a chemical partitions between solid and solution phases

in soil, or between water and sediment in aquatic ecosystems.

Result : Koc = 43.71

log Koc = 1.64

Test substance: CAS No. 68526-84-1; Alcohols, C8-C10-iso, C9 rich

Reliability : (2) valid with restrictions

This robust summary has a reliability rating of 2 because the data are calculated and not measured. The value was calculated based on

chemical structure as modeled by EPIWIN.

26.10.2005 (18) (22)

3.4 MODE OF DEGRADATION IN ACTUAL USE

3.5 BIODEGRADATION

3.6 BOD5, COD OR BOD5/COD RATIO

3.7 BIOACCUMULATION

Species: Oncorhynchus mykiss (Fish, fresh water)

Exposure period : at °C

Concentration

BCF : 15

Elimination

Method : OECD Guide-line 305

Year : 1998 GLP : yes Test substance :

Remark: Duration: 16 days uptake, 10 days depuration

The duration of the uptake phase was shorter than the guideline recommended duration of 28 days. The duration of the uptake phase was determined based on the concentration of the test substance reaching a steady state in the tissue of the fish. Samples of the fish were analyzed at

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regular intervals. To determine the concentration of the test substance in the fish tissue, under steady-state conditions, one-tail T-tests for the comparison of the means at the sampling times were performed. No difference in mortality or growth was observed between the treatments and the control after 16 days of exposure, or at the end of the study (26

Result Calculated Bioconcentration Factor (BCF):

> 0.75 mg/L = 14.80.15 mg/L = 15.5

Mean BCF for the study = 15.2

Test condition

: A 100 mg/L stock solution was prepared by adding the appropriate amount of the test substance to 18 liters of dilution water. The stock solution was mixed for 24 hours prior to its use in the test. It was then pumped from the mixing vessel via a glass tube, silicon tubing and varistaltic pump and delivered to the test chambers via a 2 liter proportional diluter. Two concentrations were prepared for the test, 0.75 mg/L and 0.15 mg/L. A dilution water control was also prepared. A new stock solution was prepared in this manner every 4 days and mixed for 24 hours prior to use in the test.

One test chamber was prepared for each treatment and the control. At the start of the test each test chamber contained 45 fish. Fish were observed daily for mortality and/or abnormal behavior or appearance. Water and fish samples were removed periodically during both the uptake and depuration phases of the study. Samples were analyzed by gas chromatography.

Mean measured concentrations of the test substance in water:

0.75 mg/L = 0.91 mg/L (s.d.= 0.15)0.15 mg/L = 0.16 mg/L (s.d.= 0.02)

The test substance was not detected in the control during the study or in either treatment during the depuration phase.

Mean measured concentrations of the test substance in fish tissue:

0.75 mg/L = 13.8 mg/kg0.15 mg/L = 2.5 mg/kg

Fish samples were also removed at the beginning and end of the uptake phase and analyzed for lipid content. The mean percent lipid content was 3.53%.

Test temperature was 14.6 Deg C., Lighting was 16 hours light: 8 hours dark with 438 to 612 Lux during full daylight periods.

Dissolved Oxygen ranged from 8.7 to 9.3 mg/L. The pH was ranged from 7.1 to 7.3 during the study. Fish were fed at a rate of 2% of their bodyweight per day during the study.

Fish Mean Weight was approximately = 0.367g at the start of the test. Test

Loading = 0.47 g of fish/L.

Test substance Read across from: Alcohols, C9-11-iso-, C10-rich (CAS. 68526-85-2) The test substance is expected to have a low potential to bioaccumulate based on read across from the C10 alcohol. Value based on wet weight

Reliability (1) valid without restriction Critical study for SIDS endpoint Flag

data.

06.04.2006 (12)

ADDITIONAL REMARKS 3.8

Conclusion

Date

4.1 ACUTE/PROLONGED TOXICITY TO FISH

Type : semistatic

Species: Oncorhynchus mykiss (Fish, fresh water)

Exposure period : 96 hour(s)
Unit : mg/l

LC50 : = 10.1 measured/nominal

Limit test : no Analytical monitoring : yes

Method : OECD Guide-line 203 "Fish, Acute Toxicity Test"

Year : 1995 GLP : yes Test substance :

Method : Bionomial Method

Result : LC50 = 10.1mg/L (Cl 7.3 to 14.1), based upon measured concentrations of

mean of old and new samples.

Analytical method used was GC-FID

LL50 = 11.2 mg/L (CI 7.5 to 16.6), based upon nominal loading levels.

Nominal Con	c. Measured Conc.	%	6 Mortality	@	96	hr.
Control	Below detection	0		_		
0.7 mg/L	1.7 mg/L	0				
1.5 mg/L	1.9 mg/L	0				
3.3 mg/L	3.9 mg/L	0				
7.5 mg/L	7.3 mg/L	0				
16.6 mg/L	14.1 mg/L	100)			

Dissolved oxygen levels dropped below 60% of saturation in some of the treatments on Days 1 through 4 of the test. Since no mortality occurred in these treatments, the deviations are not believed to have affected the outcome of the study.

Test condition

Individual Water Accomodated Fractions (WAF's) were prepared for each test treatment. The test substance was added volumetrically, via a syringe, to 19L of dilution water in a 20L glass carboy. The solutions were mixed for 24 hours at a vortex of </= 10% of the total depth. The test solutions were pumped from each mixing vessel into three replicates of 4.5L in 4.0L glass aspirator bottles (no headspace). Five fish were added to each test replicate and the replicates sealed. Daily renewals were performed by removing ~80% of the test solution through the port at the bottom and refilling with fresh solution.

Test temperature was 15.0 Deg C., Lighting was 16 hours light: 8 hours dark with 572 to 573 Lux during full daylight periods.

Dissolved Oxygen at initiation ranged from 8.4 to 9.0 mg/L and from 4.8 to 6.3 mg/L in "old" solutions prior to renewals. The pH was ranged from 6.8 to 8.5 during the study. Fish were not fed during the study.

Fish Mean Wt.= 0.361g. Mean Total length = 3.8cm, Test Loading = 0.40 g of fish/L.

CAS No. 68526-84-1; Alcohols, C8-C10-iso, C9 rich

Conclusion : Test substance is considered to have moderate acute toxicity.

Reliability : (1) valid without restriction
Flag : Critical study for SIDS endpoint

06.04.2006 (11)

Type :

Test substance

Date

Species: other: freshwater fish

Exposure period : 96 hour(s)
Unit : mg/l

LC50 : = 7.6 calculated

Method : other: ECOSAR Computer Model

Year : 2005

GLP : Test substance :

Test condition : Log Kow (octanol/water partition coefficient) values and a chemical

structure are needed to calculate aquatic toxicity using the ECOSAR model. The Kow calculation is performed by KOWWIN based on an atom/fragment contribution method of Meylan and Howard (1), which is a subroutine in the EPIWIN computer model (2). KOWWIN also has a

database of experimental Kow values (EXPKOW.DB).

The ECOSAR program was run using a C9 alcohol with a Kow of 3.22.

1. Meylan, W. and P. Howard. 1995. Atom/fragment contribution method for estimating octanol-water partition coefficients. J. Pharm. Sci. 84:83-92.

2. Meylan, M., SRC 1994-1999. KOWWIN is contained in the computer program EPIWIN. 1999. Estimation Program Interface for Windows, version 3.04. Syracuse Research Corporation, Syracuse, NY, USA.

Test substance : CAS No. 68526-84-1; Alcohols, C8-C10-iso, C9 rich

Conclusion : Based on the calculated Kow value, the C9 alcohol is expected to have an

acute 96-hour LC50 of 7.6 mg/L and a Chronic Value of 1.2 mg/L

Reliability : (2) valid with restrictions

The value was calculated based on chemical structure as modeled by EPIWIN. This robust summary has a reliability rating of 2 because the data

are calculated and not measured.

06.04.2006 (3)

4.2 ACUTE TOXICITY TO AQUATIC INVERTEBRATES

Type : static

Species : Daphnia magna (Crustacea)

Exposure period : 48 hour(s)
Unit : mg/l

EC50 : = 4.9 measured/nominal

Limit Test : no Analytical monitoring : yes

Method : OECD Guide-line 202

Year : 1996 GLP : yes Test substance :

Method : Probit procedure of SAS (Finney, 1971)Remark : Test Type: Daphnid Acute Toxicity Test

Result : 48-hour EC50 = 4.9 mg/L (Cl 4.5 - 5.4), based upon measured

concentrations of mean of old and new samples.

Analytical method used was Total Organic Carbon (TOC).

Nominal Conc.	Measured Conc	% Immobilization @ 24 hr.
Control	0 0	
1.56 mg/L	0.80 mg/L	0
3.12 mg/L	1.82 mg/L	0
6.25 mg/L	3.05 mg/L	0
12.5 mg/L	4.39 mg/L	40

Date

25.0 mg/L 6.14 mg/L 85 **Test condition**: Individual Water Accomodated Fraction

Individual Water Accomodated Fractions (WAF's) were prepared for each test treatment. The test substance was added to 2.0L of dilution water in a 2L glass aspirator bottle. The solutions were mixed for 25 hours at a vortex of </= 20% of the total depth. The test solutions were removed through the outlet at the bottom of each mixing vessel into four replicates of 140 mL in 125 mL glass erlenmeyer flasks (no headspace). Five daphnids were added to each test replicate and the replicates sealed. The test was performed under static conditions with no aeration.

Test temperature was 21.4 Deg C., Lighting was 16 hours light: 8 hours dark with 638 to 639 Lux during full daylight periods. Dissolved oxygen ranged from 7.3 to 8.2 mg/L during the study. The pH was ranged from 7.7 to 8.4 during the study.

Organisms were supplied by in-house cultures. Age = <24 hours old, from

13 and 16-day old parents.

Test substance : CAS No. 68526-84-1; Alcohols, C8-C10-iso, C9 rich

Conclusion: Test substance is considered to have moderate acute toxicity.

Reliability : (1) valid without restriction

Code 1, Reliable without Restrictions Critical study for SIDS endpoint

06.04.2006 (10)

Type :

Flag

Species : Daphnia sp. (Crustacea)

Exposure period : 48 hour(s)
Unit : mg/l

EC50 : = 8.9 calculated

Method : other: ECOSAR Computer Model

Year : 2005

GLP :

Test substance :

Test condition: Log Kow (octanol/water partition coefficient) values and a chemical

structure are needed to calculate aquatic toxicity using the ECOSAR model. The Kow calculation is performed by KOWWIN based on an atom/fragment contribution method of Meylan and Howard (1), which is a subroutine in the EPIWIN computer model (2). KOWWIN also has a

database of experimental Kow values (EXPKOW.DB).

The ECOSAR program was run using a C9 alcohol with a Kow of 3.22.

1. Meylan, W. and P. Howard. 1995. Atom/fragment contribution method for estimating octanol-water partition coefficients. J. Pharm. Sci. 84:83-92.

2. Meylan, M., SRC 1994-1999. KOWWIN is contained in the computer program EPIWIN. 1999. Estimation Program Interface for Windows, version 3.04. Syracuse Research Corporation. Syracuse. NY. USA.

Test substance Conclusion

: CAS No. 68526-84-1; Alcohols, C8-C10-iso, C9 rich

: Based on the calculated Kow value, the C9 alcohol is expected to have an

acute 48-hour EC50 of 8.9 mg/L and a Chronic Value of 0.8 mg/L.

Reliability : (2) valid with restrictions

The value was calculated based on chemical structure as modeled by EPIWIN. This robust summary has a reliability rating of 2 because the data

are calculated and not measured.

06.04.2006 (3)

4.3 TOXICITY TO AQUATIC PLANTS E.G. ALGAE

Species : other algae: Pseudokirchneriella subcapitata

Date

Endpoint

Exposure period : 96 hour(s)
Unit : mg/l

EC50: = $\overline{6}$ calculatedChV: = 1.207 calculated

Limit test : no Analytical monitoring : yes

Method : other: ECOSAR Computer Model

Year : 1999 GLP : no Test substance :

Test condition: Log Kow (octanol/water partition coefficient) values and a chemical

structure are needed to calculate aquatic toxicity using the ECOSAR model. The Kow calculation is performed by KOWWIN based on an atom/fragment contribution method of Meylan and Howard, which is a subroutine in the EPIWIN computer model. KOWWIN also has a database

of experimental Kow values (EXPKOW.DB).

The ECOSAR program was run using a C9 alcohol with a Kow of 3.22.

Test substance : CAS No. 68526-84-1; Alcohols, C8-C10-iso, C9 rich

Conclusion : Based on the calculated Kow value, the C9 alcohol is expected to have an

acute 96-hour EC50 of 6.0 mg/L and a Chronic Value of 1.207 mg/L.

Reliability : (2) valid with restrictions

The value was calculated based on chemical structure as modeled by EPIWIN. This robust summary has a reliability rating of 2 because the data

are calculated and not measured.

Flag : Critical study for SIDS endpoint

06.04.2006 (3)

Species: other algae: green alga

Endpoint

Exposure period : 96 hour(s)
Unit : mg/l

Method : other: ECOSAR Computer Model

Year : 2005

GLP :

Test substance :

Test condition: Log Kow (octanol/water partition coefficient) values and a chemical

structure are needed to calculate aquatic toxicity using the ECOSAR model. The Kow calculation is performed by KOWWIN based on an atom/fragment contribution method of Meylan and Howard (1), which is a subroutine in the EPIWIN computer model (2). KOWWIN also has a

database of experimental Kow values (EXPKOW.DB).

The ECOSAR program was run using a C9 alcohol with a Kow of 3.22.

1. Meylan, W. and P. Howard. 1995. Atom/fragment contribution method for estimating octanol-water partition coefficients. J. Pharm. Sci. 84:83-92.

2. Meylan, M., SRC 1994-1999. KOWWIN is contained in the computer program EPIWIN. 1999. Estimation Program Interface for Windows, version 3.04. Syracuse Research Corporation, Syracuse, NY, USA.

Test substance

CAS No. 68526-84-1; Alcohols, C8-C10-iso, C9 rich

Conclusion: Based on the calculated Kow value, the C9 alcohol is expected to have an

acute 96-hour EC50 of 6.0 mg/L and a Chronic Value of 1.2 mg/L.

Reliability : (2) valid with restrictions

The value was calculated based on chemical structure as modeled by EPIWIN. This robust summary has a reliability rating of 2 because the data

are calculated and not measured.

06.04.2006 (3)

Date

4.4 TOXICITY TO MICROORGANISMS E.G. BACTERIA

4.5.1 CHRONIC TOXICITY TO FISH

Species: other: Freshwater Fish (calculated toxicity values are not species specific)

Endpoint : other: LC50
Exposure period : 30 day(s)
Unit : mg/l

ChV : = 1.196 calculated Method : other: calculated

Year : 2005

GLP

Test substance

Test condition: Log Kow (octanol/water partition coefficient) values and a chemical

structure are needed to calculate aquatic toxicity using the ECOSAR model. The Kow calculation is performed by KOWWIN based on an atom/fragment contribution method of Meylan and Howard, which is a subroutine in the EPIWIN computer model. KOWWIN also has a database

of experimental Kow values (EXPKOW.DB).

The ECOSAR program was run using a C9 alcohol with a Kow of 3.22.

Test substance : CAS No. 68526-84-1; Alcohols, C8-C10-iso, C9 rich

Conclusion: Based on the calculated Kow value, the C9 alcohol is expected to have a

30-day Chronic Value of 1.196 mg/L.

Reliability : (2) valid with restrictions

The value was calculated based on chemical structure as modeled by ECOSAR. This robust summary has a reliability rating of 2 because the

data are calculated and not measured.

Flag : Critical study for SIDS endpoint

06.04.2006 (3)

4.5.2 CHRONIC TOXICITY TO AQUATIC INVERTEBRATES

Species : Daphnia sp. (Crustacea)

Endpoint : mortality
Exposure period : 16 day(s)
Unit : mg/l

EC50 : = .778 calculated

Method : other: ECOSAR Computer Model

Year : 2005

GLP :

Test substance :

Test condition: Log Kow (octanol/water partition coefficient) values and a chemical

structure are needed to calculate aquatic toxicity using the ECOSAR model. The Kow calculation is performed by KOWWIN based on an atom/fragment contribution method of Meylan and Howard, which is a subroutine in the EPIWIN computer model. KOWWIN also has a database

of experimental Kow values (EXPKOW.DB).

The ECOSAR program was run using a C9 alcohol with a Kow of 3.22.

Test substance : CAS No. 68526-84-1; Alcohols, C8-C10-iso, C9 rich

Conclusion: Based on the calculated Kow value, the C9 alcohol is expected yo have a

16-day EC50 of 0.778 mg/L.

Reliability: (2) valid with restrictions

4. Ecotoxicity Id 68526-84-1

Date 06.04.2006

The value was calculated based on chemical structure as modeled by EPIWIN. This robust summary has a reliability rating of 2 because the data

are calculated and not measured.

Flag : Critical study for SIDS endpoint

06.04.2006 (3)

4.6.1 TOXICITY TO SEDIMENT DWELLING ORGANISMS

4.6.2 TOXICITY TO TERRESTRIAL PLANTS

4.6.3 TOXICITY TO SOIL DWELLING ORGANISMS

Type :

Species: other: Earthworm

Endpoint :

 Exposure period
 : 16 day(s)

 Unit
 : mg/kg soil dw

 LC50
 : = 373.58 calc

LC50 : = 373.58 calculated

Method : other: ECOSAR Computer Model

Year : 2005

GLP

Test substance :

Test condition: Log Kow (octanol/water partition coefficient) values and a chemical

structure are needed to calculate acute terrestrial toxicity using the ECOSAR model. The Kow calculation is performed by KOWWIN based on an atom/fragment contribution method of Meylan and Howard, which is a subroutine in the EPIWIN computer model. KOWWIN also has a database

of experimental Kow values (EXPKOW.DB).

The ECOSAR program was run using a C9 alcohol with a Kow of 3.22.

Test substance : CAS No. 68526-84-1; Alcohols, C8-C10-iso, C9 rich

Conclusion : Based on the calculated Kow value, the C9 alcohol is expected yo have a

16-day EC50 of 373.58 mg/kg soil.

Reliability : (2) valid with restrictions

The value was calculated based on chemical structure as modeled by EPIWIN. This robust summary has a reliability rating of 2 because the data

are calculated and not measured.

Flag : Critical study for SIDS endpoint

06.04.2006 (3)

4.6.4 TOX. TO OTHER NON MAMM. TERR. SPECIES

4.7 BIOLOGICAL EFFECTS MONITORING

4.8 BIOTRANSFORMATION AND KINETICS

4.9 ADDITIONAL REMARKS

Date

5.0 TOXICOKINETICS, METABOLISM AND DISTRIBUTION

5.1.1 ACUTE ORAL TOXICITY

Type : LD50

Value : = 2979 mg/kg bw

Species : rat

Strain : Sprague-Dawley

Sex : male Number of animals : 5

Vehicle : other: None

Doses : 34.6, 120, 417, 1450, 5000, or 10,000 mg/kg bw

Method: other: NAYear: 1968GLP: noTest substance: other TS

Remark: No deaths occurred in the 34.6, 120, 417, and 1450 mg/kg groups

throughout the study. Two of the five animals in the 5000 mg/kg group died within 24 hours and all of the animals in the 10,000 mg/kg group died

within 24 hours. Depression, labored respiration and evidence of

excessive urination and/or diarrhea were observed at the 5,000 and 10,000 mg/kg dose levels. These signs of toxicity were observed within one hour of administration. At necropsy, abscessed lungs, dark red lungs and a dark zone between the renal cortex and medulla were observed in animals from

the 5,000 and 10,000 mg/kg dose levels. Route of administration: Gastric Intubation Frequency of Treatment: Single Exposure

Dose/Concentration Levels: 34.6, 120, 417, 1450, 5000 or 10,000 mg/kg

Control group and Treatment: None

Result : LD50 = 2979 mg/kg

Test condition : After a three to four hour fasting period, groups of 5 rats (approximately

252-295 grams) received the undiluted test material at doses of 34.6, 120, 417, 1450, 5000 or 10,000 mg/kg body weight. Observations were recorded immediately after dosing; at one, four and 24 hours; and once

daily for a total of 14 days.

Test substance : CAS No. 68526-84-1; Alcohols, C8-C10-iso, C9 rich

Conclusion: Under conditions of this study, Alcohols, C8-10 iso, C9 rich have a low

order of acute oral toxicity in rats.

Reliability : (2) valid with restrictions

Pre-GLP. From peer reviewed literature.

Flag : Critical study for SIDS endpoint

03.02.2005 (8) (27)

5.1.2 ACUTE INHALATION TOXICITY

Species : rat

Strain : Sprague-Dawley

Sex : male Number of animals : 10

Vehicle: other: noneDoses: 3630 ppmExposure time: 6 hour(s)Method: other

Date

Year : 1968 **GLP** : no

Test substance : other TS: isononyl alcohol (CAS No. 68526-84-1)

Remark: There were no deaths during the exposure, but 2/10 rats died during the

first 14 hours post exposure. Systemic effects related to central nervous system depression were seen and these included inactivity, shallow, or labored respiration and prostration. Local irritation effects were seen, but

quickly returned to normal at the termination of exposure.

Result : LC50 > 3630 ppm (approximately 21.7 mg/l)

Test condition: Observations were made for mortality, signs of toxic effect, and body wight

changes during a 14-day post-exposure observation period. Feed and water were freely available during the post-exposure holding period. The experimental atmosphere was generated by metering a measured amount of the test substance with a Harvard infusion pump into a positive pressure spray nozzle situated within the chamber. Chamber concentrations were not determined. The nominal concentration was determined by the ratio of the weight of the test substance aerosolized to the total chamber airflow

per unit time.

After exposure, the animals were removed from the chamber and group

housed by species.

Conclusion: Under the conditions of this study, isononyl alcohol has a low order of

acute inhalation toxicity in rats.

Reliability : (2) valid with restrictions

Pre-GLP, vapor concentration not analyzed. From peer reviewed literature.

Flag : Critical study for SIDS endpoint

16.03.2006 (7) (27)

Type : LC50

Value : > 3630 ppm

Species: mouseStrain: SwissSex: maleNumber of animals: 10

Vehicle: other: noneDoses: 3630 ppmExposure time: 6 hour(s)Method: otherYear: 1968GLP: no

Test substance : other TS: isononyl alcohol (CAS No. 68526-84-1)

Remark : There were no deaths during the exposure, but 1/10 mice died during the

first 14 hours post exposure. Systemic effects related to central nervous system depression were seen and these included inactivity, shallow, or labored respiration and prostration. Local irritation effects were seen, but

quickly returned to normal at the termination of exposure.

Result : LC50 > 3630 ppm (approximately 21.7 mg/l)

Test condition : Observations were made for mortality, signs of toxic effect, and body wight

changes during a 14-day post-exposure observation period. Feed and water were freely available during the post-exposure holding period. The experimental atmosphere was generated by metering a measured amount of the test substance with a Harvard infusion pump into a positive pressure spray nozzle situated within the chamber. Chamber concentrations were not determined. The nominal concentration was determined by the ratio of the weight of the test substance aerosolized to the total chamber airflow

per unit time.

After exposure, the animals were removed from the chamber and group

housed by species.

Conclusion: Under the conditions of this study, isononyl alcohol has a low order of

acute inhalation toxicity in mice.

Reliability : (2) valid with restrictions

Date

Pre-GLP, vapor concentration not analyzed. From peer reviewed literature.

Flag : Critical study for SIDS endpoint

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16.03.2006 (7) (27)

 Type
 : LC50

 Value
 : > 3630 ppm

 Species
 : guinea pig

 Strain
 : Hartley

 Sex
 : male

 Number of animals
 : 10

Vehicle : other: none
Doses : 3630 ppm
Exposure time : 6 hour(s)
Method : other
Year : 1968

GLP

Test substance: other TS: isononyl alcohol (CAS No. 68526-84-1)

Remark: There were no deaths during the exposure. No guinea pigs died during the

study. Systemic effects related to central nervous system depression were seen and these included inactivity, shallow, or labored respiration and prostration. Local irritation effects were seen, but quickly returned to

normal at the termination of exposure.

Result : LC50 > 3630 ppm (approximately 21.7 mg/l)

Test condition: Observations were made for mortality, signs of toxic effect, and body wight

changes during a 14-day post-exposure observation period. Feed and water were freely available during the post-exposure holding period. The experimental atmosphere was generated by metering a measured amount of the test substance with a Harvard infusion pump into a positive pressure spray nozzle situated within the chamber. Chamber concentrations were not determined. The nominal concentration was determined by the ratio of the weight of the test substance aerosolized to the total chamber airflow

per unit time.

After exposure, the animals were removed from the chamber and group

housed by species.

Conclusion : Under the conditions of this study, isononyl alcohol has a low order of

acute inhalation toxicity guinea pigs.

Reliability : (2) valid with restrictions

Pre-GLP, vapor concentration not analyzed. From peer reviewed literature.

Flag : Critical study for SIDS endpoint

16.03.2006 (7) (27)

5.1.3 ACUTE DERMAL TOXICITY

Type : LD50

Value : > 3160 mg/kg bw

Species : rabbit

Strain : New Zealand white Sex : male/female

Number of animals : 4

Vehicle : other: none

Doses : 50, 200, 794, and 3160 mg/kg bw

Method : other Year : 1968 GLP : no

Test substance: as prescribed by 1.1 - 1.4

Remark: No deaths were observed at any timepoint in this study. No evidence of

systemic toxicity was observed. Dose-related moderate to severe skin irritation was produced. For all of the doses tested, no compound-related

Date

alterations were observed at necropsy. Route of administration: Dermal

Frequency of Treatment: Single Exposure

Dose/Concentration Levels: 50, 200, 794 and 3160 mg/kg

Result : LD50 > 3160 mg/kg of body weight.

Test condition : A single application of the test material was made to four groups of four

rabbits (2.0 to 2.8 kg) at doses of 50, 200, 794 and 3160 mg/kg. The material was applied to abraded abdominal skin under occlusive dressing. Observations were recorded immediately following application; at one, four

and 24 hours; and once daily thereafter for a total of 14 days.

Conclusion: Under the conditions of this study, Alcohols, C8-10 iso, C9 rich has a low

order of acute dermal toxicity in rats.

Reliability : (2) valid with restrictions

Pre-GLP. From peer reviewed literature.

Flag : Critical study for SIDS endpoint

03.02.2005 (8) (27)

5.1.4 ACUTE TOXICITY, OTHER ROUTES

5.2.1 SKIN IRRITATION

Species : rabbit

Concentration

Exposure : Occlusive Exposure time : 24 hour(s)

Number of animals : 16

Vehicle : other: none

PDII

Result: moderately irritating

Classification

Method : other: similar to OECD TG 404

Year : 1968 **GLP** : no

Test substance : other TS: (CAS No. 68526-84-1)

Remark : A single 24-hour application of Alcohols, C8-10 iso-, C9 rich to intact rabbit

skin produced marked irritation. Skin irritation consisted of slight to severe erythema, slight to moderate edema and atonia and slight to severe

desquamation persisted to termination on day 14.

Result: Markedly irritating.

Test condition : Alcohols, C8-10 iso-, C9 rich was applied to the clipped, abdominal skin of

4 groups of 4 rabbits. Each application consisted of 0.05, 0.2, 0.794, or 3.16 g/kg test material under an occlusive dressing which was left in place for 24 hours. Measurements were made at 1 hour and at 24, 48 and 72

hours after dressing removal and continued daily for 14 days.

Conclusion : Under the conditions of the study, Alcohols, C8-10 iso-, C9 rich produced

marked irritation to rabbit skin.

Reliability : (2) valid with restrictions

Pre-GLP.

Flag : Critical study for SIDS endpoint

04.01.2005 (9) (27)

5.2.2 EYE IRRITATION

Species : rabbit
Concentration : .1 undiluted

Dose :

Date

Exposure time :

Comment :

Number of animals : 6

Vehicle : none

Result: moderately irritating

Classification

Method : other: similar to OECD TG 405

Year : 1968
GLP : no
Test substance : other TS

Remark: Mean scores at 24, 48, and 72 hours were 3, 2.83, 2 for conjunctival

redness, 1.33, 0.17, 0 for chemosis, 1.67, 1.67, 0.67 for discharge, 1, 0.83,

0.17 for iridial irritation, and 1.17, 1 and 0.83 for corneal opacity,

respectively.

Maximum Group Draize score = 33 (out of 110)

The test substance produced marked irritation in the eyes of rabbits which

generally cleared by 10 days.

Test condition: Animals were individually housed in stainless steel cages, with adequate

food and water.

The test material was administered as a single instillation of 0.1 ml into the lower conjunctival sac of the left eye of each animal. The upper and lower lids were gently held together for approximately 1 second to prevent loss of

the material. The untreated eye served as the control.

The eyes of each animal were examined 1, 4, 24, 48, and 72 hours, and 4, 7 and 10 days after administration. At each interval the treated and control eyes were examined and scored for ocular reactions according to the

Draize Standard Eye Irritation Grading Scale.

Body weights were determined at study initiation and termination.

Test substance : CAS No. 68526-84-1; Alcohols, C8-C10-iso, C9 rich

Reliability : (2) valid with restrictions

Pre-GLP. From peer reviewed literature.

04.01.2005 (6) (27)

Species : rabbit
Concentration : .1 undiluted

Dose

Exposure time : Comment : Number of animals : 6 Vehicle : none

Result: moderately irritating

Classification

Method : other: OECD TG 405; EEC 84/449, section B.5

Year : 1993 GLP : yes Test substance : other TS

Remark: Mean scores at 24, 48, and 72 hours were 1.8, 1.3, 1.2 for conjunctival

redness, 1.3, 1.2, 0.67 for chemosis, 1.3, 0.5, 0.17 for discharge, 0.83, 0.67, 0.33 for iridial irritation, and 0.83, 0.83 and 0.83 for corneal opacity,

respectively.

Maximum Group Draize score = 23.2 (out of 110)

The test substance produced significant conjunctival irritation in the eyes of

rabbits which generally cleared by 7 days.

Test condition: Animals were individually housed in stainless steel cages, with adequate

food and water.

Date

The test material was administered as a single instillation of 0.1 ml into the lower conjunctival sac of the right eye of each animal. The upper and lower lids were gently held together for approximately 1 second to prevent loss of the material. The untreated eye served as the control.

The eyes of each animal were examined 1, 24, 48, and 72 hours, and 7 days after administration. At each interval the treated and control eyes were examined and scored for ocular reactions according to the Draize

Standard Eye Irritation Grading Scale.

Test substance : CAS No. 68526-84-1; Alcohols, C8-C10-iso, C9 rich

Reliability : (1) valid without restriction
Flag : Critical study for SIDS endpoint

09.03.2004 (26)

5.3 SENSITIZATION

Type : other: Human Maximization (as described by Epstein and Kligman)

Species: human

Number of animals

Vehicle: petrolatumResult: not sensitizingClassification: not sensitizingMethod: other: not specified

Year

GLP : no data Test substance : other TS

Remark: The test substance was prepared and administered as a 1%

solution of 1-hexanol in petrolatum.

Test substance : Analog substance: 1-hexanol (CAS 111-27-3)

Reliability : (4) not assignable

This robust summary has a reliability rating of 4 because the data were not

retrieved and reviewed for quality.

11.10.2005 (24)

5.4 REPEATED DOSE TOXICITY

Type : Sub-chronic

Species: ratSex: maleStrain: Wistar

Route of admin. : other: Oral gavage

Exposure period : 14 days **Frequency of treatm.** : once/day

Post exposure period

Doses : 144 mg/kg bw

Control group : yes, concurrent vehicle

NOAEL : >= 144 mg/kg

Method: otherYear: 1983GLP: no dataTest substance: other TS

Method: Mean values compared to controls by Student's t-test.

Remark: Isononanol did not significantly change body weight gain, liver to body

weight ratio, or testis to body weight ratio when compared to vehicle controls. Isononanol did not induce any changes in glycogen vacuolation or fat vacuolation. The levels of cholesterol and triglyceride were not

ld 68526-84-1 5. Toxicity

Date

significantly different between animals treated with isononanol and vehicle controls. There was a slight induction of palmitoyl CoA oxidase activity. However, the activity of other peroxisome-associated enzymes was not affected and overall peroxisome number was not effected. No

hyperlipidemia was observed.

Type of Study: 14-Day Repeated Dose

No. of animals/sex/dose: 5/treatment, 10/control: 1mmol/kg/day of

isononanol (144 mg/kg/day) Vehicle: Polyethylene glycol 300

NOAEL >= 144 mg/kg/day Result

After acclimation for 1 week, five animals received 1mmol/kg/day (144 **Test condition**

> mg/kg/day) of the test substance by oral gavage and ten animals received only the vehicle, PEG 300, daily for 14 days. Animals were sacrificed after 14 days by halothane overdose and blood was withdrawn by cardiac puncture and analyzed for plasma cholesterol and triglycerides. The liver was removed for histopathological analysis, analysis of catalase, and CN-

insensitive palmitoyl CoA oxidation. Testicular weight was also

determined.

Test substance Isononanol

Under the conditions of this study, isononanol has a low order of sub-acute Conclusion

toxicity in male rats for the endpoints studied.

Reliability (2) valid with restrictions

From peer reviewed literature. Screening study with limited examinations

but scientifically valid and well documented.

16.03.2006 (25)

GENETIC TOXICITY 'IN VITRO'

5.6 **GENETIC TOXICITY 'IN VIVO'**

Type Cytogenetic assay

Species Sex Strain

Route of admin. gavage

Exposure period

Doses 2.26-12.8 mg/kg (1/5 LD50)

Result negative

Method

Year 1987

GLP

Test substance

Method Each 10 rats were investigated, 1 ml test article/rat was orally administered

by gavage as a 40% suspension in water. Preparation of bone marrow 48 h

after administration and microscopic analysis of 50 cells per animal.

Number of cells investigated was too few to correspond to scientific and

regulatory requirements.

Original reference in Russian.

Result 3% of cells with aberration were observed, control group 0% of cells with

aberrations.

Test substance Analogue substance: CAS No. 28473-21-4; nonanol

(2) valid with restrictions Reliability

Documentation insufficient for complete assessment since reference is in a

foreign language. However, reference is from a peer reviewed journal.

06.04.2006 (2)

ld 68526-84-1 5. Toxicity

Date

5.7 CARCINOGENICITY

5.8.1 TOXICITY TO FERTILITY

5.8.2 DEVELOPMENTAL TOXICITY/TERATOGENICITY

Species : rat Sex female

Sprague-Dawley Strain Route of admin. : inhalation

Exposure period Gestation days 1-19

: 7 hrs/dav Frequency of treatm.

: Animals were sacrificed on day 20 of gestation **Duration of test**

Doses 150 mg/m3 (Saturated vapors) Control group other: 15 sham-exposed rats

NOAEL maternal tox. $= 150 \text{ mg/m}^3$

Method other Year 1989 no data **GLP** Test substance other TS

MANOVA, ANOVA, Kruskal-Wallis test Method

No treatment-related effects were observed in dams. There were no Remark

statistically significant differences in maternal weight gain, feed

consumption, and water intake between the control and treated groups. In addition, no signs of fetal toxicity were observed. There were no statistically significant differences between the mean number of corpora lutea and resorptions, the sex ratio, and the mean fetal weights between

the control and treated groups.

No. of animals/sex/dose: 15 dams/dose

Result NOAEL = 150 mg/m3

Test condition Throughout the study, all animals were housed under standard

> environmental conditions and allowed free access to food and water except when the pregnant females were in the exposure chamber. Following mating, sperm-positive females were placed in cages and weighed. Dams were weighed daily for the first week of exposure and weekly thereafter. Exposures were conducted in Hinners-type chambers. The purity of the test substance was 99% as measured by gas chromatography. A

constant flow of the test substance was mixed with a known volume of heat compressed air, resulting in instantaneous vaporization of the test substance, which then flowed into the chamber. The concentration of the test substance was monitored continuously and recorded every hour. Calibration checks were completed daily. Exposure concentrations were verified on a weekly basis using a secondary method of analysis. The highest concentration of vapor that could be generated was 3500 mg/m3. Dams were exposed from days 1-19 of gestation. On day 20, dams were sacrificed by CO2 asphyxiation, and the uterus and ovaries were removed and examined for corpa lutea, implantations, resorption sites, and live fetuses. Fetuses were removed and examined for external malformations,

sexed, weighed, and examined for visceral or skeletal defects.

Test substance Analog substance: 1-Nonanol

Under the conditions of this study, exposure of pregnant rats to saturated Conclusion

vapors of 1-Nonanol does not induce maternal or fetal toxicity.

Reliability (2) valid with restrictions

2 - Reliable with restrictions - Similar to guideline study; only one exposure

level.

ld 68526-84-1 5. Toxicity

Date

11.10.2005 (23)

Species : rat female Sex Strain Wistar

Route of admin. : other: Oral gavage Exposure period : Gestation days 6-15

Frequency of treatm.

Duration of test : Animals sacrificed on gestation day 20

Doses 144, 720, 1440 mg/kg/day (equivalent to 1, 5, and 10 mmol/kg/day)

Control group yes

NOAEL maternal tox. = 144 mg/kg bw

Method OECD Guide-line 414 "Teratogenicity"

Year 1989 **GLP** yes other TS **Test substance**

Method Dunnett's test, Fisher's exact test

At the lowest dose level, no maternal toxicity was observed. There were an Remark

increased number of fetuses with hydroureter. However, the significance of this endpoint as an indicator of marginal developmental toxicity is questionable. At both the 144 and 720 mg/kg/day dose levels, there were no effects on the following parameters: uterine weight, conception rate, mean number of corpora lutea and implantation sites, pre- and postimplantation loss, number of resorptions, and viable fetuses. At the 720 mg/kg/day level, the following signs of maternal toxicity were observed reduced food consumption, reduced body weight, unsteady gait, and reddish nasal discharge. Fetal effects included a slightly reduced mean fetal body weight and an increased number of fetuses with hydroureter. Signs of maternal toxicity at the 1440 mg/kg/day level included reduced rood consumption and mean body weight, severe clinical symptoms like abdominal or lateral position, and unsteady gait. In addition, 7 of the animals found dead by gestation day 11 and the remaining 3 were sacrificed in moribund condition by gestation day 10. At necropsy, all animals had light brown-gray discoloration of the liver and some had evidence of lung edema and petechiae in the lungs. Because of the death of all dams within the high dose group, no data were available to assess uterus weight, reproduction parameters, or fetal effects.

No. of animals/sex/dose: 10/dose

Vehicle: Aqueous emulsion in 0.005% Cremophor EL

Control group and Treatment:

Control Group 1: Doubly distilled water

Control Group 2: Doubly distilled water with 0.005%

Cremophor EL

NOAEL = 144 mg/kg/day (Maternal and Fetal) Result

The study was conducted according to OECD 414 guidelines except that **Test condition**

10 animals instead of the recommended 20 per group were employed. Isononylalcohol 1 or Isononylalcohol 2 were administered to rats (10/dose) on days 6 through 15 of gestation at doses of 144, 720, or 1440 mg/kg/day (equivalent to 1, 5, and 10 mmol/kg/day). A standard dose volume of 5 ml/kg was used. Control group 1 was dosed with doubly distilled water. Control group 2 was dosed with emulsifier (doubly distilled water with 0.005% Cremophor EL). The state of health of the animals was monitored daily and food consumption and body weights of the animals were recorded regularly. Females were sacrificed on gestation day 20. Fetuses

were removed and evaluated for sex, weight, and any external, soft tissue,

or skeletal findings.

Analog substance: Isononylalcohol 1 (CAS No. 68515-81-1) Test substance

Conclusion When administered by oral gavage under the conditions of this study,

Isononylalcohol 1 causes embryo/fetal toxicity at doses that induce overt maternal toxicity. In addition, Isononylalcohol 1 does not alter reproductive

parameters at doses that are not maternally toxic.

5. Toxicity ld 68526-84-1

Date 06.04.2006

Reliability : (2) valid with restrictions

2 - Reliable with restrictions - Only 10 animals instead of the recommended

20 per group (OECD 414) were employed.

11.10.2005 (4)

Species: ratSex: femaleStrain: Wistar

Route of admin. : other: oral gavage Exposure period : Gestation days 6-15

Frequency of treatm. : daily

Duration of test : Animals were sacrificed on gestation day 20

Doses : 144, 720, 1440 mg/kg/day (equivalent to 1, 5, and 10 mmol/kg/day)

Control group : yes

NOAEL maternal tox. : = 144 - mg/kg bw

Method : OECD Guide-line 414 "Teratogenicity"

Year : 1989
GLP : yes
Test substance : other TS

Method : Dunnett's test, Fisher's exact test

Remark: At the lowest dose level, no maternal or fetal toxicity was observed. In

addition, there were no changes in reproductive parameters. At the 720 mg/kg/day level, signs of maternal toxicity included unsteady gait, piloerection, salivation, and reduced body weight gain and food consumption. There was also an increased frequency of fetuses with hydroureter at this level. At this level, there were no significant changes in reproductive parameters. Although there was an increased number of late resorptions, this number was within the range of biologic variation, was not

dose-dependent, and was therefore considered incidental.

At the highest dose level, dams exhibited marked decreases in weight gain and food consumption, and displayed severe clinical symptoms, including unsteady gait, apathy, and abdominal or lateral position. One animal was found dead on gestation day 9 and two other dams were sacrificed in moribund condition on gestation days 8 and 109. At necropsy, light browngray discoloration of the liver, lung edema, and petechiae in the lungs, heart, or bladder were observed. Fetuses from the high dose group had markedly reduced mean fetal body weight, increased frequency of hydroureter, and a higher frequency of fetuses with skeletal variations and retardations. At the highest dose, there were no changes in fertility parameters.

No. of animals: 10/group Control Group and Treatment:

Control Group 1: Doubly distilled water

Control Group 2: Doubly distilled water with 0.005%

Cremophor EL

Result : NOAEL = 144 mg/kg/day

Test condition: The study was conducted according to OECD 414 guidelines except that

10 animals instead of the recommended 20 per group were employed. Isononylalcohol 1 or Isononylalcohol 2 were administered to rats (10/dose) on days 6 through 15 of gestation at doses of 144, 720, or 1440 mg/kg/day (equivalent to 1, 5, and 10 mmol/kg/day). A standard dose volume of 5 ml/kg was used. Control group 1 was dosed with doubly distilled water. Control group 2 was dosed with emulsifier (doubly distilled water with 0.005% Cremophor EL). The state of health of the animals was monitored daily and food consumption and body weights of the animals were

recorded regularly. Females were sacrificed on gestation day 20. Fetuses were removed and evaluated for sex, weight, and any external, soft tissue,

or skeletal findings.

Test substance : Analog substance: Isononylalcohol 2 (CAS No. 68515-81-1)

Conclusion: When administered by oral gavage under the conditions of this study,

Date

Isononylalcohol 2 causes embryo/fetal toxicity at doses that induce overt maternal toxicity. In addition, Isononyl alcohol 2 does not alter fertility

parameters at doses that are not maternally toxic.

Reliability : (2) valid with restrictions

2 - Reliable with restrictions - Only 10 animals instead of the recommended

20 per group (OECD 414) were employed.

11.10.2005 (4)

5.8.3 TOXICITY TO REPRODUCTION, OTHER STUDIES

Remark : Low potential for reproductive toxicity. Adequate data exists for

developmental and subchronic studies that show no effect on reproductive

organs.

02.10.2003

5.9 SPECIFIC INVESTIGATIONS

5.10 EXPOSURE EXPERIENCE

5.11 ADDITIONAL REMARKS

6. Analyt. Meth. for Detection and Identification	68526-84-1 06.04.2006	
6.1 ANALYTICAL METHODS		
6.2 DETECTION AND IDENTIFICATION		
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7. Eff	. Against Target Org. and Into	ended Uses	Id Date	68526-84-1	
7.1	FUNCTION				
7.2	EFFECTS ON ORGANISMS TO BE CONT	FROLLED			
7.3	ORGANISMS TO BE PROTECTED				
7.4	USER				
7.5	RESISTANCE				
		35 / 39			

8. M	eas. Nec. to Prot. Man, Animals, Environment	Id Date	68526-84-1
8.1	METHODS HANDLING AND STORING		
8.2	FIRE GUIDANCE		
8.3	EMERGENCY MEASURES		
8.4	POSSIB. OF RENDERING SUBST. HARMLESS		
8.5	WASTE MANAGEMENT		
8.6	SIDE-EFFECTS DETECTION		
8.7	SUBSTANCE REGISTERED AS DANGEROUS FOR GROUND WATER		
8.8	REACTIVITY TOWARDS CONTAINER MATERIAL		
	36 / 39		

9. References Id 68526-84-1

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9. References Id 68526-84-1 Date 06.04.2006

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10. Summary and Evaluation		68526-84-1
	Date	
10.1 END POINT SUMMARY		
10.2 HAZARD SUMMARY		
10.3 RISK ASSESSMENT		
3	39 / 39	

IUCLID

Data Set

 Existing Chemical
 : ID: 28473-21-4

 CAS No.
 : 28473-21-4

 EINECS Name
 : nonan-1-ol

 EC No.
 : 249-048-6

 TSCA Name
 : Nonanol

 Molecular Formula
 : C9H20O

Producer related part

Company: ExxonMobil Biomedical Sciences Inc.

Creation date : 30.09.2004

Substance related part

Company: ExxonMobil Biomedical Sciences Inc.

Creation date : 30.09.2004

Status

Memo : Prepared for EMCC - US HPV

Printing date : 06.04.2006

Revision date

Date of last update : 06.04.2006

Number of pages : 29

Chapter (profile) : Chapter: 1, 2, 3, 4, 5, 6, 7, 8, 10 Reliability (profile) : Reliability: without reliability, 1, 2, 3, 4

Flags (profile) : Flags: without flag, confidential, non confidential, WGK (DE), TA-Luft (DE),

Material Safety Dataset, Risk Assessment, Directive 67/548/EEC, SIDS

1.0.1 APPLICANT AND COMPANY INFORMATION

1.0.2 LOCATION OF PRODUCTION SITE, IMPORTER OR FORMULATOR

1.0.3 IDENTITY OF RECIPIENTS

1.0.4 DETAILS ON CATEGORY/TEMPLATE

Comment: see free text

Remark : The Alkyl Alcohols C6 to C13 Category is a family of saturated alcohols

that are produced from olefins by the hydroformylation or "oxo" process. Hydroformylation is the reaction of an olefin with carbon monoxide and hydrogen to produce an aldehyde, and its subsequent hydrogenation to the

alcohol.

The number of carbon atoms in the category members ranges from 6 to 13. Category members contain predominantly branched alkyl groups. Each substance consists of an isomeric mixture of saturated primary alcohols of high purity, and the following basic structure: CH3-R-CH2-OH, where R is a branched isomeric structure.

The justification for the Alkyl Alcohols C6 to C13 Category is that the members have:

- · similar chemical structures,
- · similar physico-chemical properties,
- · comparable environmental fate,
- · the same mode of action.

In general, aliphatic alcohol toxicity occurs by non-polar narcosis (Lipnick et al., 1985). The mode of action is disruption of biological membrane function (van Wezel and Opperhuizen, 1995). Metabolic pathways, through which alcohols are metabolized, are likely to include similar reactions for all category members and result in structurally similar metabolites (carboxylic acids).

The data demonstrate that the category is valid for a screening-level hazard assessment for the Category and its members. One can assess the untested endpoints by extrapolation between and among the category members.

20.03.2006

1.1.0 SUBSTANCE IDENTIFICATION

IUPAC Name : Smiles Code :

Molecular formula : C9H20O Molecular weight : 144.26

Petrol class

Flag : Critical study for SIDS endpoint

05.04.2006

1. General Information

ld 28473-21-4

Date

1.1.1 GENERAL SUBSTANCE INFORMATION

Purity type

Substance type : organic Physical status : liquid

Purity :
Colour :
Odour :

Remark: CAS Registry Number, Name, and General Structure for Members of the

Alkyl Alcohols C6 to C13 Category and Analogue Substances:

CAS RN: 143-08-8; 28473-21-4 IUPAC Name: Nonan-1-ol R Length (C number): C9 Structure of R: Linear Category Member: Yes

05.04.2006

1.1.2 SPECTRA

1.2 SYNONYMS AND TRADENAMES

Nonanol

30.09.2004

Nonyl alcohol

30.09.2004

1.3 IMPURITIES

Purity : typical for marketed substance

CAS-No : 28473-21-4 **EC-No** : 249-048-6

EINECS-Name

Remark: Commercial product typically consists of approximately 80% 3,5,5-

trimethylhexanol.

05.04.2006

1.4 ADDITIVES

Purity type : typical for marketed substance

CAS-No :

EINECS-Name : Molecular formula :

Value :

Function of additive

1. General Information

ld 28473-21-4

Date

Remark : No additives present. 05.04.2006

1.5 TOTAL QUANTITY

1.6.1 LABELLING

1.6.2 CLASSIFICATION

1.6.3 PACKAGING

1.7 USE PATTERN

Type of use : industrial

Category : Chemical industry: used in synthesis

Remark: Primarily used as a chemical intermediate in the production of esters, and

in fragrances.

05.04.2006

1.7.1 DETAILED USE PATTERN

1.7.2 METHODS OF MANUFACTURE

1.8 REGULATORY MEASURES

1.8.1 OCCUPATIONAL EXPOSURE LIMIT VALUES

1.8.2 ACCEPTABLE RESIDUES LEVELS

1.8.3 WATER POLLUTION

1.8.4 MAJOR ACCIDENT HAZARDS

1.8.5 AIR POLLUTION

1.8.6 LISTINGS E.G. CHEMICAL INVENTORIES

1. Ge	neral Information	ld	28473-21-4	
		Date	06.04.2006	
1.9.1	DEGRADATION/TRANSFORMATION PRODUCTS			
1.9.2	COMPONENTS			
1.10	SOURCE OF EXPOSURE			
1.11	ADDITIONAL REMARKS			
1.12	LAST LITERATURE SEARCH			
1.13	REVIEWS			

5/29

ld 28473-21-4

Date

2.1 MELTING POINT

Value = -18.7 °C

Sublimation

Method other: calculated

Year 2003 **GLP** no data

Test substance

Method Calculated values using MPBPWIN version 1.40, a subroutine of the

computer program EPIWIN version 3.04

Melting Point estimations performed by MPBPWIN are based on the **Test condition**

average result of the calculation methods of K. Joback and Gold and Ogle.

Joback's Method is described in Joback, K.G. 1982. A Unified Approach to Physical Property Estimation Using Multivariate Statistical Techniques. In The Properties of Gases and Liquids. Fourth Edition. 1987. R.C. Reid, J.M.

Prausnitz and B.E. Poling, Eds.

The Gold and Ogle Method simply uses the formula

Tm = 0.5839Tb, where Tm is the melting point in Kelvin and Tb is the

boiling point in Kelvin.

CAS No. 28473-21-4; nonan-1-ol **Test substance**

(2) valid with restrictions Reliability

The result is a calculated value based on the chemical structure and

represents a potential melting point for the substance with the CAS number

listed under test substance.

: Critical study for SIDS endpoint Flag

05.04.2006 (15)

BOILING POINT 2.2

Value = 192 - 204 °C at 1013 hPa

Decomposition

Method other: ASTM D1078/01

Year : 2003 **GLP** no data

Test substance

Test substance CAS No. 28473-21-4: nonan-1-ol

(2) valid with restrictions Reliability

Although the original data were not retrieved and reviewed for quality, they

were developed following acceptable test methods and therefore

considered reliable.

: Critical study for SIDS endpoint Flag

05.04.2006 (8)

2.3 **DENSITY**

Type density

Value = .832 g/cm3 at 20 °C Method other: D1078/01

Year 2003 **GLP** no data

Test substance

ld 28473-21-4

Date

: CAS No. 28473-21-4; nonan-1-ol Test substance

: (2) valid with restrictions Reliability

Although the original data were not retrieved and reviewed for quality, they

were developed following acceptable test methods and therefore

considered reliable.

Critical study for SIDS endpoint Flag

05.04.2006 (8)

2.3.1 GRANULOMETRY

2.4 **VAPOUR PRESSURE**

: = .4 hPa at 25 °CValue

Decomposition

Method other (calculated)

Year 2003 **GLP** no data

Test substance

Method Vapor pressure calculation by MPBPWIN ver. 1.40 using calculation

method of Grain.

Remark EPIWIN is used and advocated by the US EPA for chemical property

estimation.

Test substance : CAS No. 28473-21-4; nonan-1-ol

(2) valid with restrictions Reliability

> The value was calculated based on chemical structure as modeled by EPIWIN. This robust summary has a reliability rating of 2 because the data

are calculated and not measured.

Flag Critical study for SIDS endpoint

05.04.2006 (13)

2.5 **PARTITION COEFFICIENT**

Partition coefficient : octanol-water Log pow = 3.2 - 4.9 at 30 °C

pH value

Method OECD Guide-line 117 "Partition Coefficient (n-octanol/water), HPLC

Method"

Year 1998 **GLP** yes

Test substance

Remark Test Type: N-Octanol/Water Partition Coefficient (HPLC method)

The test substance eluted as several groups. The five major components Result

C6, C7, C8, C9, C10 alcohols had Log Pow values of 3.2, 3.7, 4.4, 4.5 and

4.9 respectively.

The retention time for the 5 major components were 6.55, 8.68, 13.3, 13.8,

and 19.0 minutes.

All values were measured using High Performance Liquid Chromatography

(HPLC).

Test condition The test substance was evaluated as a solution in HPLC grade methanol.

> Six reference compounds were also evaluated in a standard combined reference solution (2-butanone, acetophenone, naphthalene, biphenyl, nbutylbenzene, and 4,4-DDT) in 75% methanol and 25% distilled water.

The pH of the solution was 5.4.

Date

Two customized alcohol reference solutions were also prepared containing five of the ten alcohol compounds (1-hexanol, 1-heptanol, 1-octanol, 1-nonanol, 1-decanol, 1-undecanol, 1-dodecanol, 1-tridecanol, 1-tridecanol, 1-tetradecanol, 1-pentadecanol) in 87.5% methanol and 12.5% distilled water. The pH of both solutions was 7.3.

The pH of the evaluated solutions was the same as the reference solution it was evaluated against.

The test substance was analyzed against a Standard Log Pow Reference Compound Solution and a customized Alcohol Reference Compound Solution. Only the peaks detected by refractive index (RI) were reported.

Test substance : CAS No. 28473-21-4; nonanol Reliability : (1) valid without restriction : Critical study for SIDS endpoint

05.04.2006 (4)

2.6.1 SOLUBILITY IN DIFFERENT MEDIA

Solubility in : Water

Value : = 128 mg/l at 20 $^{\circ}$ C

pH value

concentration : at °C

Temperature effects

Examine different pol.

pKa : at 25 °C

Description

Stable

Deg. product

beg. product .

Method : other: Slow stir

Year : 1998 GLP : no data

Test substance :

Method : Slow-stir water solubility vessels consisted of glass aspirator bottles with

capacities of 4 to 12 L. The spigot at the bottom of the vessel was fitted with short length of Tefzel tubing and a #13 glass stopper. The bottle was rinsed with a mixture of 1:1 methylene chloride: acetone followed by 2,2,4-trimethyl pentane. The bottle was then air dried in a laboratory fume hood

and rinsed three times with dilution water.

A glass stir bar was added to the bottle and then the bottle was filled with the appropriate amount of water. The test substance was added to the bottle at a loading of 100 mg/L and the solution stirred quiescently with little

or no visible vortex on a magnetic stir-plate.

Mixing was stopped one hour prior to sampling. Samples were removed through the outlet at the bottom of the vessel. To avoid losses and to prevent contamination, the samples were analyzed immediately by either CC MSD or CC FID. Samples were taken as David 1.3.7 and 31.

GC-MSD or GC-FID. Samples were taken on Days 1, 3, 7, and 21.

Test substance : CAS No. 28473-21-4; nonanol : (1) valid without restriction : Critical study for SIDS endpoint

05.04.2006 (10)

2.6.2 SURFACE TENSION

ld 28473-21-4

Date

2.7 FLASH POINT

Value : $= 86.1 \, ^{\circ}\text{C}$ Type : closed cup

Method : other: PMCC ASTM D93

Year : 2003

GLP :

Test substance : CAS No. 28473-21-4; nonanol **Reliability** : (2) valid with restrictions

Although the original data were not retrieved and reviewed for quality, they

were developed following acceptable test methods and therefore

considered to be reliable.

Flag : Critical study for SIDS endpoint

05.04.2006 (8)

2.8 AUTO FLAMMABILITY

2.9 FLAMMABILITY

2.10 EXPLOSIVE PROPERTIES

2.11 OXIDIZING PROPERTIES

2.12 DISSOCIATION CONSTANT

Acid-base constant : = 15.5 at 25°C Method : other: calculated

Year : 2003

GLP : Test substance :

Method: pKa calculation by SPARC 2003 using a Linux calculation engine.

Test substance : CAS No. 28473-21-4; nonanol Reliability : (2) valid with restrictions

The value was calculated based on chemical structure as modeled by SPARC. This robust summary has a reliability rating of 2 because the data

are calculated and not measured.

05.04.2006 (9) (18)

2.13 VISCOSITY

Value : $= 18 - at 20 \, ^{\circ}C$

Result

Method : other: ASTM D445

Year

GLP : no data **Test substance** : other TS

Remark : Value measured in cSt

9/29

ld 28473-21-4 **Date** 06.04.2006

Test substance : CAReliability : (2)

: CAS No. 28473-21-4; nonanol: (2) valid with restrictions

Although the original data were not retrieved and reviewed for quality, they

were developed following acceptable test methods and therefore

considered to be reliable.

Flag : Critical study for SIDS endpoint

05.04.2006 (8)

2.14 ADDITIONAL REMARKS

Date

3.1.1 PHOTODEGRADATION

Type : water

Light source

Light spectrum: nm

Relative intensity : based on intensity of sunlight

Deg. product

Method : other (calculated): Technical Discussion

Year : GLP :

Test substance

Remark: These data represent a key study for characterising the potential of

substances in the Alkyl Alcohols C6 to C13 Category to undergo direct

photodegradation.

Result: Photolysis as a Function of Molecular Structure

The direct photolysis of an organic molecule occurs when it absorbs sufficient light energy to result in a structural transformation (Harris, 1982). The reaction process is initiated when light energy in a specific wavelength range elevates a molecule to an electronically excited state. However, the excited state is competitive with various deactivation processes that can result in the return of the molecule to a non excited state.

The absorption of light in the ultra violet (UV)-visible range, 110-750 nm, can result in the electronic excitation of an organic molecule. Light in this range contains energy of the same order of magnitude as covalent bond dissociation energies (Harris, 1982). Higher wavelengths (e.g. infrared) result only in vibrational and rotational transitions, which do not tend to produce structural changes to a molecule.

The stratospheric ozone layer prevents UV light of less than 290 nm from reaching the earth's surface. Therefore, only light at wavelengths between 290 and 750 nm can result in photochemical transformations in the environment (Harris, 1982). Although the absorption of UV light in the 290-750 nm range is necessary, it is not always sufficient for a chemical to undergo photochemical degradation. Energy may be re-emitted from an excited molecule by mechanisms other than chemical transformation, resulting in no change to the parent molecule.

A conservative approach to estimating a photochemical degradation rate is to assume that degradation will occur in proportion to the amount of light wavelengths >290 nm absorbed by the molecule (Zepp and Cline, 1977).

Substances in the Oxo Alcohols C6 to C13 Category contain molecules that are oxygenated aliphatic compounds which will absorb UV light below 220 nm (Boethling and Mackay, 2000) and will not undergo direct photolysis. Therefore, this fate process will not contribute to a measurable degradative removal of chemical components in this category from the environment.

References

Boethling, R.S., Mackay, D. (2000). Handbook of Property Estimation Methods for Chemicals. CRC Press, Boca Raton, FL, USA.

Harris, J. C. 1982. "Rate of Aqueous Photolysis," Chapter 8 in: W. J. Lyman, W. F. Reehl, and D. H. Rosenblatt, eds., Handbook of Chemical

Date

Property Estimation Methods, McGraw-Hill Book Company, New York,

Zepp, R. G. and D. M. Cline. 1977. Rates of Direct Photolysis in the

Aqueous Environment, Environ. Sci. Technol., 11:359-366.

Test substance : Alkyl Alcohols C6 to C13 Category members

Flag : Critical study for SIDS endpoint

03.04.2006 (7)

Type :

Light source : Sun light Light spectrum : nm

Relative intensity : based on intensity of sunlight

INDIRECT PHOTOLYSIS

Sensitizer : OH

Conc. of sensitizer : 1500000 molecule/cm³

Rate constant : = .0000000000139553 cm³/(molecule*sec)

Degradation: % after

Deg. product :

Method : other (calculated): Calculated values using AOPWIN version 1.89, a

subroutine of the computer program EPIWIN version 3.04

Year : 1999

GLP : Test substance :

Result: Atmospheric Oxidation Potential

In the environment, organic chemicals emitted into the troposphere are degraded by several important transformation processes. The dominant transformation process for most compounds is the daylight reaction with hydroxyl (OH-) radicals (Atkinson, 1988, 1989). The rate at which an organic compound reacts with OH- radicals is a direct measure of its atmospheric persistence (Meylan and Howard, 1993).

AOPWIN estimates the rate constant for the atmospheric, gas-phase reaction between photochemically produced hydroxyl radicals and organic chemicals. The rate constants estimated by the program are then used to calculate atmospheric half-lives for organic compounds based upon average atmospheric concentrations of hydroxyl radicals.

Since the reactions only take place in the presence of sunlight, the atmospheric half-lives are normalized for a 12-hour day.

Calculated* OH- Rate Constant half-life (hrs) (cm3/molecule-sec)

9.2 13.96 E-12

References:

Atkinson, R. 1988. Estimation of gas-phase hydroxyl radical rate constants for organic chemicals. Environ. Toxicol. Chem. 7:435-442.

Atkinson, R. 1989. Kinetics and mechanisms of the gas-phase reactions of the hydroxyl radical with organic compounds. J. Phys. Chem. Ref. Data Monograph No. 1, Amer. Inst. Physics & Amer. Chem. Soc., NY.

Meylan, W.M. and P.H. Howard. 1993. Computer estimation of the atmospheric gas-phase reaction rate of organic compounds with hydroxyl radicals and ozone. Chemosphere 12:2293-2299.

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Test condition : Indirect photodegradation, or atmospheric oxidation potential, is based on

the structure-activity relationship methods developed by R. Atkinson.

Temperature: 25°C Sensitizer: OH radical

Concentration of Sensitizer: 1.5 E6 OH radicals/cm3

Test substance Reliability : CAS No. 28473-21-4; nonanol : (2) valid with restrictions

The results include calculated data based on chemical structure as modeled by AOPWIN. The data represent a potential atmospheric half-life range for the test substance. This robust summary represents a "key

study" for atmospheric half-life range based on constituent data.

Flag : Critical study for SIDS endpoint

05.04.2006 (14)

3.1.2 STABILITY IN WATER

 Type
 : abiotic

 t1/2 pH4
 : at °C

 t1/2 pH7
 : at °C

 t1/2 pH9
 : at °C

Deg. product

Method : other: Technical Discussion

Year : GLP : Test substance :

Remark: These data represent a key study for characterising the potential of

substances in the Alkyl Alcohols C6 to C13 Category to undergo

hydrolysis.

Result : Hydolysis as a Function of Molecular Structure

Hydrolysis of an organic molecule occurs when a molecule (R-X) reacts with water (H2O) to form a new carbon-oxygen bond after the carbon-X bond is cleaved (Gould, 1959; Harris, 1982). Mechanistically, this reaction is referred to as a nucleophilic substitution reaction, where X is the leaving group being replaced by the incoming nucleophilic oxygen from the water molecule.

Chemicals that are susceptible to hydrolysis contain functional groups that can be displaced by a nucleophilic substitution reaction. Substances that have the potential to hydrolyze include alkyl halides, amides, carbamates, carboxylic acid esters and lactones, epoxides, phosphate esters, and sulfonic acid esters (Neely, 1985). The lack of a leaving group renders a compound resistant to hydrolysis.

Alkyl alcohols are resistant to hydrolysis because they lack a functional group that is hydrolytically reactive (Harris, 1982).

References:

Gould, E.S. (1959). Mechanism and Structure in Organic Chemistry, Holt, Reinhart and Winston, New York, NY, USA.

Harris, J.C. (1982). "Rate of Hydrolysis," Chapter 7 in: W.J. Lyman, W.F. Reehl, and D.H. Rosenblatt, eds., Handbook of Chemical Property Estimation Methods, McGraw-Hill Book Company, New York, NY, USA.

Neely, W.B. (1985). Hydrolysis. In: W.B. Neely and G.E. Blau, eds., Environmental Exposure from Chemicals. Vol. 1, pp. 157-173. CRC Press,

ld 28473-21-4

Date

Boca Raton, FL, USA.

Conclusion: Hydrolysis will not contribute to the removal of Alkyl Alcohols from the

environment.

Flag : Critical study for SIDS endpoint

20.03.2006 (6)

3.1.3 STABILITY IN SOIL

3.2.1 MONITORING DATA

3.2.2 FIELD STUDIES

3.3.1 TRANSPORT BETWEEN ENVIRONMENTAL COMPARTMENTS

Type : fugacity model level III

Media: other: air - water - soil - sedimentAir: % (Fugacity Model Level I)Water: % (Fugacity Model Level I)Soil: % (Fugacity Model Level I)Biota: % (Fugacity Model Level II/III)Soil: % (Fugacity Model Level II/III)

Method : other: Calculation according Mackay, Level III

Year : 2003

Method : The EQC Level III model is a steady state model that is useful for

determining how the medium of release affects environmental fate. Level III fugacity allows non-equilibrium conditions to exist between connected media as steady state, and illustrate important transport and transformation

processes.

Physicochemical input values for the model were calculated using the EPIWIN Estimation v 3.04 program. Measured input values were also used where available and obtained from the EPIWIN database. Distribution data from the equilibrium model provide basic information on the potential partitioning behavior of chemicals between selected environmental compartments (i.e., air, water, soil, and sediment).

Input values used:

Molecular mass = 144.26 g/mol Water solubility = 128 mg/L Vapour pressure = 40.0 Pa

log Kow = 3.8

Melting point = -18.65 deg C

Degradation half-lives:

Air - 12.1 hrs Water - 360 hrs Soil - 720 hrs Sediment - 7200 hrs

This model was run assuming 100% discharge to water.

Result : Air - 2.1%

Water - 86.0% Soil - 0.05%

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Sediment - 11.9%

Test substance : CAS No. 28473-21-4; nonanol **Reliability** : (2) valid with restrictions

This robust summary has a reliability rating of 2 because the data are

calculated and not measured.

Flag : Critical study for SIDS endpoint

05.04.2006 (12)

Type : fugacity model level I

Media : other: air - biota - sediment(s) - soil - water

Air : % (Fugacity Model Level I)
Water : % (Fugacity Model Level I)
Soil : % (Fugacity Model Level I)
Biota : % (Fugacity Model Level II/III)
Soil : % (Fugacity Model Level II/III)

Method : other: Calculation according Mackay, Level I

Year : 2003

Method : The EQC Level I is a steady state, equilibrium model that utilizes the input

of basic chemical properties including molecular weight, vapor pressure, and water solubility to calculate distribution within a standardized regional

environment.

Physicochemical input values for the model were calculated using the EPIWIN Estimation v 3.04 program. Measured input values were also used where available and obtained from the EPIWIN database. Distribution data from the equilibrium model provide basic information on the potential partitioning behavior of chemicals between selected environmental

compartments (i.e., air, water, soil, sediment, suspended sediment, biota).

Input values used:

Molecular mass = 144.26 g/mol Water solubility = 128 mg/L Vapour pressure = 40.0 Pa

log Kow = 3.8

Melting point = -18.65 deg C

Result : Soil - 57.5%

Air - 6.3% Water - 35.3% Sediment - 0.8%

Suspended Sed - 0.02%

Biota - < 0.01%

Test substance : CAS No. 28473-21-4; nonanol **Reliability** : (2) valid with restrictions

This robust summary has a reliability rating of 2 because the data are

calculated and not measured.

Flag : Critical study for SIDS endpoint

05.04.2006 (12)

3.3.2 DISTRIBUTION

Media : other: Koc Method : other (calculation)

Year : 2003

Method : The soil adsorption coefficient (Koc) was calculated by PCKOCWIN ver.

1.66, using equations developed by Lyman (1980).

PCKOCWIN is a subroutine of EPIWIN. EPIWIN is used and advocated by the USEPA for chemical property estimation. Koc provides an indication of the extent to which a chemical partitions between solid and solution phases

ld 28473-21-4 Date 06.04.2006

in soil, or between water and sediment in aquatic ecosystems.

Result : Koc = 52.14

log Koc = 1.72

: CAS No. 28473-21-4; nonanol Test substance Reliability : (2) valid with restrictions

> This robust summary has a reliability rating of 2 because the data are calculated and not measured. The value was calculated based on

chemical structure as modeled by EPIWIN.

Flag : Critical study for SIDS endpoint

05.04.2006 (11)(16)

MODE OF DEGRADATION IN ACTUAL USE

3.5 **BIODEGRADATION**

Type aerobic

Inoculum activated sludge, domestic

Contact time

Degradation $= 83.7 (\pm) \%$ after 28 day(s)

: readily biodegradable Result

Deg. product

OECD Guide-line 301 F "Ready Biodegradability: Manometric Method

Respirometry Test"

Year 1998 **GLP** yes

Test substance

Remark Test Type: Manometric Respirometry Test

Exposure Period: 28 days

Result Test material was readily biodegradable. Half-life was reached by day 6. By

day 28, 83.7% degradation of the test material was observed. 10%

biodegradation was achieved by day 2.

By day 14, >60% biodegradation of positive control was observed, which met the guideline requirement. No excursions from the protocol were

noted.

Biodegradation was based on oxygen consumption and the theoretical oxygen demand of the test material as calculated using results of an

elemental analysis of the test material.

Mean % Degradation % Degradation* (day 28)

Sample (day 28)

Test Material 85.8, 83.7, 81.883.7 Na Benzoate 87.1, 85.4 86.2

* replicate data

Test condition Non acclimated activated sludge and test medium were combined prior to

> test material addition. Test medium consisted of glass distilled water and mineral salts (Phosphate buffer, Ferric chloride, Magnesium sulfate,

Calcium chloride).

Test vessels were 1L glass flasks placed in a waterbath and electronically

monitored for oxygen consumption.

Test material was tested in triplicate, controls and blanks were tested in

duplicate.

Test material concentration was approximately 53 mg/L. Sodium benzoate

(positive control) concentration was 50.6mg/L.

Test temperature was 22 +/- 1 Deg C.

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All test vessels were stirred constantly for 28 days using magnetic stir bars

and plates.

Test substance : CAS No. 28473-21-4; nonanol

Conclusion: Test substance is considered readily biodegradable.

Reliability : (1) valid without restriction
Flag : Critical study for SIDS endpoint

05.04.2006 (5)

3.6 BOD5, COD OR BOD5/COD RATIO

3.7 BIOACCUMULATION

3.8 ADDITIONAL REMARKS

4. Ecotoxicity Id 28473-21-4

Date

4.1 ACUTE/PROLONGED TOXICITY TO FISH

Type : other: calculated
Species : other: frfeshwater fish

Exposure period : 96 hour(s)
Unit : mg/l

LC50 : = 6.4 calculated

Method : other: ECOSAR Computer Model

Year : 2005

GLP

Test substance :

Test condition: Log Kow (octanol/water partition coefficient) values and a chemical

structure are needed to calculate aquatic toxicity using the ECOSAR model. The Kow calculation is performed by KOWWIN based on an atom/fragment contribution method of Meylan and Howard (1), which is a subroutine in the EPIWIN computer model (2). KOWWIN also has a

database of experimental Kow values (EXPKOW.DB).

The ECOSAR program was run using a C9 alcohol with a Kow of 3.30.

1. Meylan, W. and P. Howard. 1995. Atom/fragment contribution method for estimating octanol-water partition coefficients. J. Pharm. Sci. 84:83-92.

2. Meylan, M., SRC 1994-1999. KOWWIN is contained in the computer program EPIWIN. 1999. Estimation Program Interface for Windows, version 3.04. Syracuse Research Corporation, Syracuse, NY, USA.

Test substance Conclusion

: CAS No. 28473-21-4; nonanol

: Based on the calculated Kow value, the C9 alcohol is expected to have an

acute 96-hour LC50 of 6.4 mg/L and a Chronic Value of 1.0 mg/L.

Reliability : (2) valid with restrictions

The value was calculated based on chemical structure as modeled by EPIWIN. This robust summary has a reliability rating of 2 because the data

are calculated and not measured.

Flag : Critical study for SIDS endpoint

05.04.2006 (2)

4.2 ACUTE TOXICITY TO AQUATIC INVERTEBRATES

Туре

Species : Daphnia sp. (Crustacea)

Exposure period : 48 hour(s)
Unit : mg/l

EC50 : = 7.5 calculated

Method : other: ECOSAR Computer Model

Year : 2005

GLP :

Test substance :

Test condition: Log Kow (octanol/water partition coefficient) values and a chemical

structure are needed to calculate aquatic toxicity using the ECOSAR model. The Kow calculation is performed by KOWWIN based on an atom/fragment contribution method of Meylan and Howard (1), which is a subroutine in the EPIWIN computer model (2). KOWWIN also has a

database of experimental Kow values (EXPKOW.DB).

The ECOSAR program was run using a C9 alcohol with a Kow of 3.30.

ld 28473-21-4 4. Ecotoxicity

Date

1. Meylan, W. and P. Howard. 1995. Atom/fragment contribution method for estimating octanol-water partition coefficients. J. Pharm. Sci. 84:83-92.

2. Meylan, M., SRC 1994-1999. KOWWIN is contained in the computer program EPIWIN. 1999. Estimation Program Interface for Windows, version 3.04. Syracuse Research Corporation, Syracuse, NY, USA.

Test substance

CAS No. 28473-21-4: nonanol

Conclusion Based on the calculated Kow value, the C9 alcohol is expected to have an

acute 48-hour EC50 of 7.5 mg/L and a Chronic Value of 0.7 mg/L.

Reliability : (2) valid with restrictions

The value was calculated based on chemical structure as modeled by EPIWIN. This robust summary has a reliability rating of 2 because the data

are calculated and not measured.

Flag

: Critical study for SIDS endpoint

05.04.2006 (2)

TOXICITY TO AQUATIC PLANTS E.G. ALGAE

other algae: green alga **Species**

Endpoint biomass **Exposure period** 96 hour(s) Unit mg/l

EC50 = 5.1 calculated

Method other: ECOSAR Computer Model

Year 2005

GLP Test substance

Test condition Log Kow (octanol/water partition coefficient) values and a chemical

> structure are needed to calculate aquatic toxicity using the ECOSAR model. The Kow calculation is performed by KOWWIN based on an atom/fragment contribution method of Meylan and Howard (1), which is a subroutine in the EPIWIN computer model (2). KOWWIN also has a

database of experimental Kow values (EXPKOW.DB).

The ECOSAR program was run using a C9 alcohol with a Kow of 3.30.

1. Meylan, W. and P. Howard. 1995. Atom/fragment contribution method for estimating octanol-water partition coefficients. J. Pharm. Sci. 84:83-92.

2. Meylan, M., SRC 1994-1999. KOWWIN is contained in the computer program EPIWIN. 1999. Estimation Program Interface for Windows, version 3.04. Syracuse Research Corporation, Syracuse, NY, USA.

Test substance CAS No. 28473-21-4; nonanol

Conclusion Based on the calculated Kow value, the C9 alcohol is expected to have an

acute 96-hour EC50 of 5.1 mg/L and a Chronic Value of 1.1 mg/L.

Reliability (2) valid with restrictions

The value was calculated based on chemical structure as modeled by EPIWIN. This robust summary has a reliability rating of 2 because the data

are calculated and not measured.

Critical study for SIDS endpoint Flag

05.04.2006 (2)

TOXICITY TO MICROORGANISMS E.G. BACTERIA

4. Ecotoxicity Id 28473-21-4

Date

4.5.1 CHRONIC TOXICITY TO FISH

Species: other: Freshwater Fish (calculated toxicity values are not species specific)

Endpoint : other: LC50
Exposure period : 30 day(s)
Unit : mg/l

ChV : = 1 calculated

Method : other: calculated

Year : 2005

GLP : Test substance :

Test condition: Log Kow (octanol/water partition coefficient) values and a chemical

structure are needed to calculate aquatic toxicity using the ECOSAR model. The Kow calculation is performed by KOWWIN based on an atom/fragment contribution method of Meylan and Howard, which is a subroutine in the EPIWIN computer model. KOWWIN also has a database

of experimental Kow values (EXPKOW.DB).

The ECOSAR program was run using a C9 alcohol with a Kow of 3.30.

Test substance: CAS No. 28473-21-4; nonanol

Conclusion : Based on the calculated Kow value, the C9 alcohol is expected to have a

30-day Chronic Value of 1.0 mg/L

Reliability : (2) valid with restrictions

The value was calculated based on chemical structure as modeled by ECOSAR. This robust summary has a reliability rating of 2 because the

data are calculated and not measured.

Flag : Critical study for SIDS endpoint

05.04.2006 (2)

4.5.2 CHRONIC TOXICITY TO AQUATIC INVERTEBRATES

Species : Daphnia sp. (Crustacea)

Endpoint : mortality
Exposure period : 16 day(s)
Unit : mg/l

EC50 : = .7 calculated

Method : other: ECOSAR Computer Model

Year : 2005

GLP : Test substance :

Test condition : Log Kow (octanol/water partition coefficient) values and a chemical

structure are needed to calculate aquatic toxicity using the ECOSAR model. The Kow calculation is performed by KOWWIN based on an atom/fragment contribution method of Meylan and Howard, which is a subroutine in the EPIWIN computer model. KOWWIN also has a database

of experimental Kow values (EXPKOW.DB).

The ECOSAR program was run using a C9 alcohol with a Kow of 3.30.

Test substance: CAS No. 28473-21-4; nonanol

Conclusion: Based on the calculated Kow value, the C9 alcohol is expected yo have a

16-day EC50 of 0.7 mg/L.

Reliability : (2) valid with restrictions

The value was calculated based on chemical structure as modeled by EPIWIN. This robust summary has a reliability rating of 2 because the data

are calculated and not measured.

Flag : Critical study for SIDS endpoint

05.04.2006 (2)

4. Ecotoxicity Id 28473-21-4

Date 06.04.2006

4.6.1 TOXICITY TO SEDIMENT DWELLING ORGANISMS

4.6.2 TOXICITY TO TERRESTRIAL PLANTS

4.6.3 TOXICITY TO SOIL DWELLING ORGANISMS

Type :

Species: other: Earthworm

Endpoint : mortality
Exposure period : 14 day(s)
Unit : mg/kg soil dw
LC50 : = 353 calculated

Method : other: ECOSAR Computer Model

Year : 2005

GLP : Test substance :

Test condition: Log Kow (octanol/water partition coefficient) values and a chemical

structure are needed to calculate acute terrestrial toxicity using the ECOSAR model. The Kow calculation is performed by KOWWIN based on an atom/fragment contribution method of Meylan and Howard, which is a subroutine in the EPIWIN computer model. KOWWIN also has a database

of experimental Kow values (EXPKOW.DB).

The ECOSAR program was run using a C9 alcohol with a Kow of 3.30.

Test substance: CAS No. 28473-21-4; nonanol

Conclusion: Based on the calculated Kow value, the C7 alcohol is expected to have a

14-day EC50 of 353 mg/kg soil.

Reliability : (2) valid with restrictions

The value was calculated based on chemical structure as modeled by EPIWIN. This robust summary has a reliability rating of 2 because the data

are calculated and not measured.

Flag : Critical study for SIDS endpoint

05.04.2006 (2)

4.6.4 TOX. TO OTHER NON MAMM, TERR, SPECIES

4.7 BIOLOGICAL EFFECTS MONITORING

4.8 BIOTRANSFORMATION AND KINETICS

4.9 ADDITIONAL REMARKS

5. Toxicity Id 28473-21-4

Date

5.0 TOXICOKINETICS, METABOLISM AND DISTRIBUTION

5.1.1 ACUTE ORAL TOXICITY

5.1.2 ACUTE INHALATION TOXICITY

5.1.3 ACUTE DERMAL TOXICITY

5.1.4 ACUTE TOXICITY, OTHER ROUTES

5.2.1 SKIN IRRITATION

5.2.2 EYE IRRITATION

5.3 SENSITIZATION

5.4 REPEATED DOSE TOXICITY

Type : Sub-chronic

Species : I Sex : Strain :

Route of admin. : inhalation
Exposure period : 2 months
Frequency of treatm. : 2 h/d
Post exposure period : none

Doses : 0.2; 0.6; 0.8 mg/l control group : no data specified

Method

Year : 1964

GLP : Test substance :

Method : Rats were exposed to concentrations of 0.2; 0.6 and 0.8 mg/l (33, 99 and

136 ppm) for 2h/day for 2 months.

Result : Small amounts of deformed or degenerate glial elements diffusely

scattered in the cerebral cortex and subcortex were observed.

Test substance: CAS No. 28473-21-4; nonanol

Reliability : (4) not assignable

Secondary literature, documentation insufficient for assessment. Can be

used to support weight of evidence.

06.04.2006 (3)

5.5 GENETIC TOXICITY 'IN VITRO'

5. Toxicity ld 28473-21-4
Date 06.04.2006

5.6 GENETIC TOXICITY 'IN VIVO'

Type : Cytogenetic assay

Species : rat

Sex

Strain

Route of admin. : gavage

Exposure period

Doses : 2.26-12.8 mg/kg (1/5 LD50)

Result : negative

Method :

Year : 1987

GLP : Test substance :

Method : Each 10 rats were investigated, 1 ml test article/rat was orally administered

by gavage as a 40% suspension in water. Preparation of bone marrow 48 h

after administration and microscopic analysis of 50 cells per animal.

Number of cells investigated was too few to correspond to scientific and

regulatory requirements.

Original reference in Russian.

Result: 3% of cells with aberration were observed, control group 0% of cells with

aberrations.

Test substance : CAS No. 28473-21-4; nonanol **Reliability** : (2) valid with restrictions

Documentation insufficient for complete assessment since reference is in a

foreign language. However, reference is from a peer reviewed journal.

06.04.2006 (1)

5.7 CARCINOGENICITY

5.8.1 TOXICITY TO FERTILITY

5.8.2 DEVELOPMENTAL TOXICITY/TERATOGENICITY

Species : rat Sex : female

Strain : Sprague-Dawley Route of admin. : inhalation

Exposure period: Gestation days 1-19

Frequency of treatm. : 7 hrs/day

Duration of test : Animals were sacrificed on day 20 of gestation

Doses : 150 mg/m3 (Saturated vapors)
Control group : other: 15 sham-exposed rats

NOAEL maternal tox. : = 150 mg/m^3 NOAEL teratogen. : = $150 \cdot \text{mg/m}^3$

Method

Year : 1987 **GLP** :

Test substance :

Method: MANOVA, ANOVA, Kruskal-Wallis test

Remark: No treatment-related effects were observed in dams. There were no

statistically significant differences in maternal weight gain, feed

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5. Toxicity ld 28473-21-4

Date

consumption, and water intake between the control and treated groups. In addition, no signs of fetal toxicity were observed. There were no statistically significant differences between the mean number of corpora lutea and resorptions, the sex ratio, and the mean fetal weights between the control and treated groups.

No. of animals/sex/dose: 15 dams/dose

Test condition: Throughout the study, all animals were housed under standard

environmental conditions and allowed free access to food and water except when the pregnant females were in the exposure chamber. Following mating, sperm-positive females were placed in cages and weighed. Dams were weighed daily for the first week of exposure and weekly thereafter. Exposures were conducted in Hinners-type chambers. The purity of the test substance was 99% as measured by gas chromatography. A constant flow of the test substance was mixed with a known volume of heat compressed air, resulting in instantaneous vaporization of the test substance, which then flowed into the chamber. The concentration of the test substance was monitored continuously and recorded every hour. Calibration checks were completed daily. Exposure concentrations were verified on a weekly basis using a secondary method of analysis. The highest concentration of vapor that could be generated was 3500 mg/m3. Dams were exposed from days 1-19 of gestation. On day 20, dams were sacrificed by CO2 asphyxiation, and the uterus and ovaries were removed and examined for corpa lutea, implantations, resorption sites, and live fetuses. Fetuses were removed and examined for external malformations,

sexed, weighed, and examined for visceral or skeletal defects.

Test substance : CAS No. 28473-21-4; nonanol

Conclusion : Under the conditions of this study, exposure of pregnant rats to saturated

vapors of 1-Nonanol does not induce maternal or fetal toxicity.

Reliability : (2) valid with restrictions

Similar to guideline study; only one exposure level.

Flag : Critical study for SIDS endpoint

06.04.2006 (17)

5.8.3 TOXICITY TO REPRODUCTION, OTHER STUDIES

5.9 SPECIFIC INVESTIGATIONS

5.10 EXPOSURE EXPERIENCE

5.11 ADDITIONAL REMARKS

6. Analyt. Meth. for Detection and Identification Id 28473-21-4 Date 06.04.2006	
6.1 ANALYTICAL METHODS	
6.2 DETECTION AND IDENTIFICATION	
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7. Eff	. Against Target Org. and Intended Uses	Id Date	28473-21-4	
7.1	FUNCTION			
7.2	EFFECTS ON ORGANISMS TO BE CONTROLLED			
7.3	ORGANISMS TO BE PROTECTED			
7.4	USER			
7.5	RESISTANCE			
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8. M	eas. Nec. to Prot. Man, Animals, Environment	ld Date	28473-21-4
8.1	METHODS HANDLING AND STORING		
8.2	FIRE GUIDANCE		
8.3	EMERGENCY MEASURES		
8.4	POSSIB. OF RENDERING SUBST. HARMLESS		
8.5	WASTE MANAGEMENT		
8.6	SIDE-EFFECTS DETECTION		
8.7	SUBSTANCE REGISTERED AS DANGEROUS FOR GROUND WATER		
8.8	REACTIVITY TOWARDS CONTAINER MATERIAL		
	27 / 29		

9. References Id 28473-21-4

Date

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10. S	ummary and Evaluation		ld	28473-21-4	
	•		Date		
10.1	END POINT SUMMARY				
10.2	HAZARD SUMMARY				
1012					
10.3	RISK ASSESSMENT				
		29 / 29			

IUCLID

Data Set

Existing Chemical : ID: 68526-79-4 **CAS No.** : 68526-79-4

EINECS Name : Hexanol, branched and linear

EC No. : 271-227-2

TSCA Name : Hexanol, branched and linear

Molecular Formula : Unspecified

Producer related part

Company: ExxonMobil Biomedical Sciences Inc.

Creation date : 18.09.2001

Substance related part

Company: ExxonMobil Biomedical Sciences Inc.

Creation date : 18.09.2001

Status

Memo : Prepared for EMCC - US HPV

Printing date : 04.04.2006

Revision date

Date of last update : 04.04.2006

Number of pages : 36

Chapter (profile) : Chapter: 1, 2, 3, 4, 5, 6, 7, 8, 10 Reliability (profile) : Reliability: without reliability, 1, 2, 3, 4

Flags (profile) : Flags: without flag, confidential, non confidential, WGK (DE), TA-Luft (DE),

Material Safety Dataset, Risk Assessment, Directive 67/548/EEC, SIDS

1.0.1 APPLICANT AND COMPANY INFORMATION

1.0.2 LOCATION OF PRODUCTION SITE, IMPORTER OR FORMULATOR

1.0.3 IDENTITY OF RECIPIENTS

1.0.4 DETAILS ON CATEGORY/TEMPLATE

Comment: see free text

Remark : The Alkyl Alcohols C6 to C13 Category is a family of saturated alcohols

that are produced from olefins by the hydroformylation or "oxo" process. Hydroformylation is the reaction of an olefin with carbon monoxide and hydrogen to produce an aldehyde, and its subsequent hydrogenation to the

alcohol.

The number of carbon atoms in the category members ranges from 6 to 13. Category members contain predominantly branched alkyl groups. Each substance consists of an isomeric mixture of saturated primary alcohols of high purity, and the following basic structure: CH3-R-CH2-OH, where R is a branched isomeric structure.

The justification for the Alkyl Alcohols C6 to C13 Category is that the members have:

· similar chemical structures,

- · similar physico-chemical properties,
- · comparable environmental fate,
- · the same mode of action.

In general, aliphatic alcohol toxicity occurs by non-polar narcosis (Lipnick et al., 1985). The mode of action is disruption of biological membrane function (van Wezel and Opperhuizen, 1995). Metabolic pathways, through which alcohols are metabolized, are likely to include similar reactions for all category members and result in structurally similar metabolites (carboxylic acids).

The data demonstrate that the category is valid for a screening-level hazard assessment for the Category and its members. One can assess the untested endpoints by extrapolation between and among the category members.

20.03.2006

1.1.0 SUBSTANCE IDENTIFICATION

IUPAC Name : Smiles Code :

Molecular formula : C6H14O Molecular weight : 102.18

Petrol class

Flag : Critical study for SIDS endpoint

03.04.2006

1. General Information

ld 68526-79-4 **Date** 04.04.2006

1.1.1 GENERAL SUBSTANCE INFORMATION

Purity type :

Substance type : organic Physical status : liquid

Purity : Colour : Odour :

Remark: CAS Registry Number, Name, and General Structure for Members of the

Alkyl Alcohols C6 to C13 Category and Analogue Substances:

CAS RN: 111-27-3 IUPAC Name: hexan-1-ol R Length (C number): C6 Structure of R: Linear

Category Member: No (analogue substance used for supporting

information)

CAS RN: 68526-79-4

IUPAC Name: Hexanol, branched and linear

R Length (C number): C6

Structure of R: Branched, primarily methyl branched (isomeric structures)

Category Member: Yes

04.04.2006

1.1.2 SPECTRA

1.2 SYNONYMS AND TRADENAMES

Exxal 6

17.09.2004

hexyl alcohol

17.09.2004

isohexanol

17.09.2004

1.3 IMPURITIES

Purity : typical for marketed substance

CAS-No : 68526-79-4 **EC-No** : 271-227-2

EINECS-Name : Hexanol, branched and linear

Molecular formula : C6H14OValue : = 98 % w/w

Remark : Commercial product typically consists of three isomers with the following

composition: 44% 1-hexanol, 53% methyl pentanols, and 3% 2-

ethylbutanol.

03.04.2006

1. General Information

ld 68526-79-4 **Date** 04.04.2006

1.4 ADDITIVES

Purity type : typical for marketed substance

CAS-No :

EC-No
EINECS-Name
Molecular formula
Value

Function of additive :

Remark: No additives present.

03.04.2006

1.5 TOTAL QUANTITY

1.6.1 LABELLING

1.6.2 CLASSIFICATION

1.6.3 PACKAGING

1.7 USE PATTERN

Type of use : industrial

Category : Chemical industry: used in synthesis

Remark: Primarily used as a chemical intermediate in the production of esters:

phthalates and acetates.

03.04.2006

1.7.1 DETAILED USE PATTERN

1.7.2 METHODS OF MANUFACTURE

1.8 REGULATORY MEASURES

1.8.1 OCCUPATIONAL EXPOSURE LIMIT VALUES

1.8.2 ACCEPTABLE RESIDUES LEVELS

1.8.3 WATER POLLUTION

Id 68526-79-4 1. General Information **Date** 04.04.2006 1.8.4 MAJOR ACCIDENT HAZARDS 1.8.5 AIR POLLUTION 1.8.6 LISTINGS E.G. CHEMICAL INVENTORIES 1.9.1 DEGRADATION/TRANSFORMATION PRODUCTS 1.9.2 COMPONENTS 1.10 SOURCE OF EXPOSURE 1.11 ADDITIONAL REMARKS 1.12 LAST LITERATURE SEARCH 1.13 REVIEWS

ld 68526-79-4

Date

2.1 MELTING POINT

Value = -49.3 °C

Sublimation

Method other: calculated

Year 2003 **GLP** no **Test substance**

Method Calculated values using MPBPWIN version 1.40, a subroutine of the

computer program EPIWIN version 3.04

Melting Point estimations performed by MPBPWIN are based on the **Test condition**

average result of the calculation methods of K. Joback and Gold and Ogle.

Joback's Method is described in Joback, K.G. 1982. A Unified Approach to Physical Property Estimation Using Multivariate Statistical Techniques. In The Properties of Gases and Liquids. Fourth Edition. 1987. R.C. Reid, J.M.

Prausnitz and B.E. Poling, Eds.

The Gold and Ogle Method simply uses the formula

Tm = 0.5839Tb, where Tm is the melting point in Kelvin and Tb is the

boiling point in Kelvin.

CAS No. 68526-79-4; Hexanol, branched and linear **Test substance**

(2) valid with restrictions Reliability

The result is a calculated value based on the chemical structure and represents a potential melting point for the substance with the CAS number

listed under test substance.

: Critical study for SIDS endpoint

Flag 03.04.2006 (20)

BOILING POINT 2.2

Value = 152 - 163 °C at 1013 hPa

Decomposition

Method other: D1078/01

Year : 2003 **GLP** no data

Test substance

Test substance CAS No. 68526-79-4; Hexanol, branched and linear

(2) valid with restrictions Reliability

Although the original data were not retrieved and reviewed for quality, they

were developed following acceptable test methods and therefore

considered reliable.

: Critical study for SIDS endpoint Flag

03.04.2006 (11)

2.3 **DENSITY**

Type density

Value = .827 g/cm3 at 20 °C

Method other: ASTM D4052/86 equivalent

Year 2003 **GLP** no data

Test substance

ld 68526-79-4 **Date** 04.04.2006

Test substance: CAS No. 68526-79-4; Hexanol, branched and linear

Reliability : (2) valid with restrictions

Although the original data were not retrieved and reviewed for quality, they

were developed following acceptable test methods and therefore

considered reliable.

Flag : Critical study for SIDS endpoint

03.04.2006 (11)

2.3.1 GRANULOMETRY

2.4 VAPOUR PRESSURE

Value : = 2.56 hPa at 25 °C

Decomposition

Method

Year : 2003 GLP : no data

Test substance :

Method : Vapor pressure calculation by MPBPWIN ver. 1.40 using calculation

method of Grain. Result supplied by the experimental database in

EPIWIN.

Remark: EPIWIN is used and advocated by the US EPA for chemical property

estimation.

Test substance: CAS No. 68526-79-4; Hexanol, branched and linear

Reliability : (2) valid with restrictions

The result is a calculated value based on the chemical structure and

represents a potential melting point for the substance with the CAS number

listed under test substance.

Flag : Critical study for SIDS endpoint

04.04.2006 (17)

2.5 PARTITION COEFFICIENT

Partition coefficient : octanol-water Log pow : = 1.8 - 2 at °C

pH value

Method : other (calculated)

Year : 2003

GLP

Test substance

Method : Calculated values using KOWWIN version 1.65, a subroutine of the

computer program EPIWIN version 3.04

Test condition : Octanol / Water Partition Coefficient estimations performed by KOWWIN

are based on an atom/fragment contribution method of W. Meylan and P. Howard in "Atom/fragment contribution method for estimating octanol-water

partition coefficients". 1995. J. Pharm. Sci. 84:83-92.

Test substance: CAS No. 68526-79-4; Hexanol, branched and linear

Reliability : (2) valid with restrictions

The result is a calculated value based on the chemical structure and represents a potential melting point for the substance with the CAS number

listed under test substance.

Flag : Critical study for SIDS endpoint

03.04.2006 (19)

ld 68526-79-4 **Date** 04.04.2006

2.6.1 SOLUBILITY IN DIFFERENT MEDIA

Solubility in : Water

Value : = 10340 - 11950 mg/l at 25 °C

pH value

concentration : at °C

Temperature effects

Examine different pol.

pKa : at 25 °C

Description

Stable

Deg. product

Method : other: calculated

Year : 2003 GLP : no data

Test substance

Method : Calculated values using WSKOWWIN version 1.36, a subroutine of the

computer program EPIWIN version 3.04

Test condition: Water Solubility estimations performed by WSKOWWIN are based on a

Kow correlation method described by W. Meylan, P. Howard and R. Boethling in "Improved method for estimating water solubility from octanol/water partition coefficient". Environ. Toxicol. Chem. 15:100-106.

1995.

Test substance: CAS No. 68526-79-4; Hexanol, branched and linear

Reliability : (2) valid with restrictions

The result is a calculated value based on the chemical structure and

represents a potential melting point for the substance with the CAS number

listed under test substance.

Flag : Critical study for SIDS endpoint

03.04.2006 (20)

2.6.2 SURFACE TENSION

2.7 FLASH POINT

Value : $= 60.5 \, ^{\circ}\text{C}$ Type : closed cup

Method : other: TCC ASTM D56

Year : 2003 GLP : no data

Test substance :

Test substance: CAS No. 68526-79-4; Hexanol, branched and linear

Reliability : (2) valid with restrictions

Although the original data were not retrieved and reviewed for quality, they

were developed following acceptable test methods and therefore

considered to be reliable.

Flag : Critical study for SIDS endpoint

03.04.2006 (11)

2.8 AUTO FLAMMABILITY

ld 68526-79-4 **Date** 04.04.2006

2.9 FLAMMABILITY

2.10 EXPLOSIVE PROPERTIES

2.11 OXIDIZING PROPERTIES

2.12 DISSOCIATION CONSTANT

Acid-base constant : = 15.50 at 25°C

Method : other: calculated

Year

GLP : no data

Test substance :

Method : pKa calculation by SPARC 2003 using a Linux calculation engine.

Remark: SPARC On-line calculator can be accessed at http://ibmlc2.chem.uga.edu/sparc/index.cfm

Test substance: CAS No. 68526-79-4; Hexanol, branched and linear.

Reliability : (2) valid with restrictions

The value was calculated based on chemical structure as modeled by SPARC. This robust summary has a reliability rating of 2 because the data

are calculated and not measured.

03.04.2006 (14)

2.13 VISCOSITY

Value : = $6.8 - at 20 \, ^{\circ}C$

Result

Method : other: ASTM D445

Year : 2003 GLP : no data

Test substance :

Remark : Value measured in cSt

Test substance: CAS No. 68526-79-4; Hexanol, branched and linear

Reliability : (2) valid with restrictions

Although the original data were not retrieved and reviewed for quality, they

were developed following acceptable test methods and therefore

considered to be reliable.

03.04.2006 (11)

2.14 ADDITIONAL REMARKS

ld 68526-79-4

Date

3.1.1 PHOTODEGRADATION

Type : water

Light source :

Light spectrum: nm

Relative intensity : based on intensity of sunlight

Deg. product

Method : other (calculated): Technical Discussion

Year : GLP :

Test substance

Remark: These data represent a key study for characterising the potential of

substances in the Alkyl Alcohols C6 to C13 Category to undergo direct

photodegradation.

Result: Photolysis as a Function of Molecular Structure

The direct photolysis of an organic molecule occurs when it absorbs sufficient light energy to result in a structural transformation (Harris, 1982). The reaction process is initiated when light energy in a specific wavelength range elevates a molecule to an electronically excited state. However, the excited state is competitive with various deactivation processes that can result in the return of the molecule to a non excited state.

The absorption of light in the ultra violet (UV)-visible range, 110-750 nm, can result in the electronic excitation of an organic molecule. Light in this range contains energy of the same order of magnitude as covalent bond dissociation energies (Harris, 1982). Higher wavelengths (e.g. infrared) result only in vibrational and rotational transitions, which do not tend to produce structural changes to a molecule.

The stratospheric ozone layer prevents UV light of less than 290 nm from reaching the earth's surface. Therefore, only light at wavelengths between 290 and 750 nm can result in photochemical transformations in the environment (Harris, 1982). Although the absorption of UV light in the 290-750 nm range is necessary, it is not always sufficient for a chemical to undergo photochemical degradation. Energy may be re-emitted from an excited molecule by mechanisms other than chemical transformation, resulting in no change to the parent molecule.

A conservative approach to estimating a photochemical degradation rate is to assume that degradation will occur in proportion to the amount of light wavelengths >290 nm absorbed by the molecule (Zepp and Cline, 1977).

Substances in the Oxo Alcohols C6 to C13 Category contain molecules that are oxygenated aliphatic compounds which will absorb UV light below 220 nm (Boethling and Mackay, 2000) and will not undergo direct photolysis. Therefore, this fate process will not contribute to a measurable degradative removal of chemical components in this category from the environment.

References

Boethling, R.S., Mackay, D. (2000). Handbook of Property Estimation Methods for Chemicals. CRC Press, Boca Raton, FL, USA.

Harris, J. C. 1982. "Rate of Aqueous Photolysis," Chapter 8 in: W. J. Lyman, W. F. Reehl, and D. H. Rosenblatt, eds., Handbook of Chemical

ld 68526-79-4

Date

Property Estimation Methods, McGraw-Hill Book Company, New York,

USA.

Zepp, R. G. and D. M. Cline. 1977. Rates of Direct Photolysis in the

Aqueous Environment, Environ. Sci. Technol., 11:359-366.

Test substance : Alkyl Alcohols C6 to C13 Category members

Flag : Critical study for SIDS endpoint

03.04.2006 (10)

Type

Light source : Sun light Light spectrum : nm

Relative intensity: based on intensity of sunlight

INDIRECT PHOTOLYSIS

Sensitizer : OH

Conc. of sensitizer : 1500000 molecule/cm³

Rate constant : = .000000000099923 cm³/(molecule*sec)

Degradation: % after

Deg. product :

Method: other (calculated): Calculated values using AOPWIN version 1.89, a

subroutine of the computer program EPIWIN version 3.04

Year

GLP :

Test substance : other TS

Result: Atmospheric Oxidation Potential

In the environment, organic chemicals emitted into the troposphere are degraded by several important transformation processes. The dominant transformation process for most compounds is the daylight reaction with hydroxyl (OH-) radicals (Atkinson, 1988, 1989). The rate at which an organic compound reacts with OH- radicals is a direct measure of its atmospheric persistence (Meylan and Howard, 1993).

AOPWIN estimates the rate constant for the atmospheric, gas-phase reaction between photochemically produced hydroxyl radicals and organic chemicals. The rate constants estimated by the program are then used to calculate atmospheric half-lives for organic compounds based upon average atmospheric concentrations of hydroxyl radicals.

Since the reactions only take place in the presence of sunlight, the atmospheric half-lives are normalized for a 12-hour day.

Calculated* OH- Rate Constant half-life (hrs) (cm3/molecule-sec)

12.8 9.99 E-12

References:

Atkinson, R. 1988. Estimation of gas-phase hydroxyl radical rate constants for organic chemicals. Environ. Toxicol. Chem. 7:435-442.

Atkinson, R. 1989. Kinetics and mechanisms of the gas-phase reactions of the hydroxyl radical with organic compounds. J. Phys. Chem. Ref. Data Monograph No. 1, Amer. Inst. Physics & Amer. Chem. Soc., NY.

Meylan, W.M. and P.H. Howard. 1993. Computer estimation of the atmospheric gas-phase reaction rate of organic compounds with hydroxyl radicals and ozone. Chemosphere 12:2293-2299.

ld 68526-79-4

Date

Test condition: Indirect photodegradation, or atmospheric oxidation potential, is based on

the structure-activity relationship methods developed by R. Atkinson.

Temperature: 25°C Sensitizer: OH radical

Concentration of Sensitizer: 1.5 E6 OH radicals/cm3 CAS No. 68526-79-4: Hexanol. branched and linear

Reliability : (2) valid with restrictions

The results include calculated data based on chemical structure as

modeled by AOPWIN. The data represent a potential atmospheric half-life

for the test substance.

Flag : Critical study for SIDS endpoint

17.09.2004 (18)

3.1.2 STABILITY IN WATER

Test substance

 Type
 : abiotic

 t1/2 pH4
 : at °C

 t1/2 pH7
 : at °C

 t1/2 pH9
 : at °C

Deg. product

Method : other: Technical Discussion

Year : GLP :

Test substance :

Remark: These data represent a key study for characterising the potential of

substances in the Alkyl Alcohols C6 to C13 Category to undergo

hydrolysis.

Result : Hydolysis as a Function of Molecular Structure

Hydrolysis of an organic molecule occurs when a molecule (R-X) reacts with water (H2O) to form a new carbon-oxygen bond after the carbon-X bond is cleaved (Gould, 1959; Harris, 1982). Mechanistically, this reaction is referred to as a nucleophilic substitution reaction, where X is the leaving group being replaced by the incoming nucleophilic oxygen from the water molecule.

Chemicals that are susceptible to hydrolysis contain functional groups that can be displaced by a nucleophilic substitution reaction. Substances that have the potential to hydrolyze include alkyl halides, amides, carbamates, carboxylic acid esters and lactones, epoxides, phosphate esters, and sulfonic acid esters (Neely, 1985). The lack of a leaving group renders a compound resistant to hydrolysis.

Alkyl alcohols are resistant to hydrolysis because they lack a functional group that is hydrolytically reactive (Harris, 1982).

References:

Gould, E.S. (1959). Mechanism and Structure in Organic Chemistry, Holt, Reinhart and Winston, New York, NY, USA.

Harris, J.C. (1982). "Rate of Hydrolysis," Chapter 7 in: W.J. Lyman, W.F. Reehl, and D.H. Rosenblatt, eds., Handbook of Chemical Property Estimation Methods, McGraw-Hill Book Company, New York, NY, USA.

Neely, W.B. (1985). Hydrolysis. In: W.B. Neely and G.E. Blau, eds., Environmental Exposure from Chemicals. Vol. 1, pp. 157-173. CRC Press, Boca Raton, FL, USA.

ld 68526-79-4 **Date** 04.04.2006

Conclusion: Hydrolysis will not contribute to the removal of Alkyl Alcohols from the

environment.

Flag : Critical study for SIDS endpoint

20.03.2006 (9)

3.1.3 STABILITY IN SOIL

3.2.1 MONITORING DATA

3.2.2 FIELD STUDIES

3.3.1 TRANSPORT BETWEEN ENVIRONMENTAL COMPARTMENTS

Type : fugacity model level III

Media: other: air - water - soil - sedimentAir: % (Fugacity Model Level I)Water: % (Fugacity Model Level I)Soil: % (Fugacity Model Level I)Biota: % (Fugacity Model Level II/III)Soil: % (Fugacity Model Level II/III)

Method : other: Calculation according Mackay, Level III

Year : 2003

Method : The EQC Level III model is a steady state model that is useful for

determining how the medium of release affects environmental fate. Level III fugacity allows non-equilibrium conditions to exist between connected media as steady state, and illustrate important transport and transformation

processes.

Physicochemical input values for the model were calculated using the EPIWIN Estimation v 3.04 program. Measured input values were also used where available and obtained from the EPIWIN database. Distribution data from the equilibrium model provide basic information on the potential partitioning behavior of chemicals between selected environmental

compartments (i.e., air, water, soil, and sediment).

Input values used:

Molecular mass = 102.18 g/mol Water solubility = 11950 mg/L Vapour pressure = 256 Pa

log Kow = 1.75

Melting point = -49 deg C

Degradation half-lives:

Air - 12.9 hrs Water - 120 hrs Soil - 720 hrs Sediment - 7200 hrs

This model was run assuming 100% discharge to water.

Result : Air - 0.3%

Water - 99.4% Soil - 0.02% Sediment - 0.28%

ld 68526-79-4

Date

Test substance : CAS No. 68526-79-4; Hexanol, branched and linear

Reliability : (2) valid with restrictions

This robust summary has a reliability rating of 2 because the data are

calculated and not measured.

Flag : Critical study for SIDS endpoint

04.04.2006 (16)

Type : fugacity model level I

Media : other: air - biota - sediment(s) - soil - water

Air : % (Fugacity Model Level I)
Water : % (Fugacity Model Level I)
Soil : % (Fugacity Model Level I)
Biota : % (Fugacity Model Level II/III)
Soil : % (Fugacity Model Level II/III)

Method

Year : 1997

Method : The EQC Level I is a steady state, equilibrium model that utilizes the input

of basic chemical properties including molecular weight, vapor pressure, and water solubility to calculate distribution within a standardized regional

environment.

Physicochemical input values for the model were calculated using the EPIWIN Estimation v 3.04 program. Measured input values were also used where available and obtained from the EPIWIN database. Distribution data from the equilibrium model provide basic information on the potential partitioning behavior of chemicals between selected environmental

compartments (i.e., air, water, soil, sediment, suspended sediment, biota).

Input values used:

Molecular mass = 102.18 g/mol Water solubility = 11950 mg/L Vapour pressure = 256 Pa

log Kow = 1.75

Melting point = -49 deg C

Result : Air - 29.5%

Water - 67.0 % Soil - 3.3% Sediment - 0.1%

Suspended Sed - < 0.01%

Biota - < 0.01%

Test substance: CAS No. 68526-79-4; Hexanol, branched and linear

Reliability : (2) valid with restrictions

This robust summary has a reliability rating of 2 because the data are

calculated and not measured.

Flag : Critical study for SIDS endpoint

03.04.2006 (16)

3.3.2 DISTRIBUTION

Media : other: Koc

Method : other (calculation)

Year : 2003

Method : The soil adsorption coefficient (Koc) was calculated by PCKOCWIN ver.

1.66, using equations developed by Lyman (1980).

PCKOCWIN is a subroutine of EPIWIN. EPIWIN is used and advocated by the USEPA for chemical property estimation. Koc provides an indication of the extent to which a chemical partitions between solid and solution phases

ld 68526-79-4 **Date** 04.04.2006

in soil, or between water and sediment in aquatic ecosystems.

Result : Koc = 7.30

log Koc = 0.86

Test substance: CAS No. 68526-79-4; Hexanol, branched and linear

Reliability : (2) valid with restrictions

This robust summary has a reliability rating of 2 because the data are calculated and not measured. The value was calculated based on

chemical structure as modeled by EPIWIN.

03.04.2006 (15) (21)

3.4 MODE OF DEGRADATION IN ACTUAL USE

3.5 BIODEGRADATION

3.6 BOD5, COD OR BOD5/COD RATIO

3.7 BIOACCUMULATION

3.8 ADDITIONAL REMARKS

4. Ecotoxicity Id 68526-79-4

Date

4.1 ACUTE/PROLONGED TOXICITY TO FISH

Type : flow through

Species: Pimephales promelas (Fish, fresh water)

Exposure period : 96 hour(s)
Unit : mg/l

LC50 : = 97.7 measured/nominal

Limit test

Analytical monitoring: yes

Method : other: no data

Year : 1980 GLP : no data

Test substance :

Method : Trimmed Spearman Karber Method

Result : 96 hour LC50 = 97.7 mg/L (95% CI 89.7 to 106) based upon measured

values

Analytical method used was Gas-Liquid Chromatography.

Measured	Fish Total
Conc. (mg/L)	Mortality (@96 hrs)
Control	0
26.7	0
49.2	0
90.6	20
170.0	50
261.5	50

^{* 50} fish added at test initiation

Test condition: Treatment solutions were prepared by diluting a 3720mg/L stock solution.

Nominal hexanol treatment levels were 41, 68, 113, 189, 315mg/L, which

measured 26.7, 49.2, 90.6, 170.0, and 261.5mg/L, respectively.

Control/dilution water was EPA Duluth laboratory water.

Fifty fish were tested per treatment, divided into two replicates. Treatment volume = 6.3L. Test parameters were as follows: temperature=26.2 Deg C; dissolved oxygen = 6.2mg/L; pH = 7.6; fish age = 28 days old; fish mean wt = 0.117g; fish mean length = 19.7mm; fish loading = 0.464g/L/day. Organism supplier was U.S. EPA Environmental Research Lab, Duluth,

MN, USA.

Test substance: CAS No. 68526-79-4; Hexanol, branched and linear

Reliability : (1) valid without restriction

04.04.2006 (2)

Type :

Species: other: freshwater fish

Exposure period : 96 hour(s)
Unit : mg/l

LC50 : = 130.3 calculated

Method : other: ECOSAR Computer Model

Year : 2005

GLP :

Test substance :

Test condition: Log Kow (octanol/water partition coefficient) values and a chemical

structure are needed to calculate aquatic toxicity using the ECOSAR model. The Kow calculation is performed by KOWWIN based on an atom/fragment contribution method of Meylan and Howard (1), which is a subroutine in the EPIWIN computer model (2). KOWWIN also has a

database of experimental Kow values (EXPKOW.DB).

The ECOSAR program was run using a C6 alcohol with a Kow of 1.75.

1. Meylan, W. and P. Howard. 1995. Atom/fragment contribution method for estimating octanol-water partition coefficients. J. Pharm. Sci. 84:83-92.

2. Meylan, M., SRC 1994-1999. KOWWIN is contained in the computer program EPIWIN. 1999. Estimation Program Interface for Windows, version 3.04. Syracuse Research Corporation, Syracuse, NY, USA.

Test substance Conclusion

: CAS No. 68526-79-4; Hexanol, branched and linear

: Based on the calculated Kow value, the C6 alcohol is expected to have an

acute 96-hour LC50 of 130 mg/L and a Chronic Value of 16.1 mg/L.

Reliability : (2) valid with restrictions

The value was calculated based on chemical structure as modeled by EPIWIN. This robust summary has a reliability rating of 2 because the data

are calculated and not measured.

04.04.2006 (3)

4.2 ACUTE TOXICITY TO AQUATIC INVERTEBRATES

Type :

Species : Daphnia sp. (Crustacea)

Exposure period : 48 hour(s)
Unit : mg/l

EC50 : = 137 calculated

Method : other: ECOSAR Computer Model

Year : 2005

GLP

Test substance

Test condition : Log Kow (octanol/water partition coefficient) values and a chemical

structure are needed to calculate aquatic toxicity using the ECOSAR model. The Kow calculation is performed by KOWWIN based on an atom/fragment contribution method of Meylan and Howard (1), which is a subroutine in the EPIWIN computer model (2). KOWWIN also has a

database of experimental Kow values (EXPKOW.DB).

The ECOSAR program was run using a C6 alcohol with a Kow of 1.75.

1. Meylan, W. and P. Howard. 1995. Atom/fragment contribution method for estimating octanol-water partition coefficients. J. Pharm. Sci. 84:83-92.

2. Meylan, M., SRC 1994-1999. KOWWIN is contained in the computer program EPIWIN. 1999. Estimation Program Interface for Windows, version 3.04. Syracuse Research Corporation, Syracuse, NY, USA.

Test substance Conclusion

: CAS No. 68526-79-4; Hexanol, branched and linear

: Based on the calculated Kow value, the C6 alcohol is expected to have an

acute 48-hour EC50 of 137 mg/L and a Chronic Value of 6.3 mg/L.

Reliability : (2) valid with restrictions

The value was calculated based on chemical structure as modeled by EPIWIN. This robust summary has a reliability rating of 2 because the data

are calculated and not measured.

03.04.2006 (4)

4.3 TOXICITY TO AQUATIC PLANTS E.G. ALGAE

Species: other algae: Pseudokirchneriella subcapitata

ld 68526-79-4 4. Ecotoxicity Date 04.04.2006

Endpoint biomass **Exposure period** 72 hour(s)

Unit mg/l

NOEC = 5 measured/nominal = 89 measured/nominal **EC50**

Limit test **Analytical monitoring** ves

Method OECD Guide-line 201 "Algae, Growth Inhibition Test"

Year 2003 **GLP** yes **Test substance**

Remark Statistical Procedure - Proc regression procedure of SAS, Anova

procedure of SAS for NOEC

72 hour EC50b=89 mg/L (biomass) Result

72 hour EC50gr=159 mg/L (growth rate)

72 hour NOECb=5 mg/L (biomass) 72 hour NOECgr=35 mg/L (growth rate)

	Mean Cell		
Nominal	Biomass -	72 hr Conc 72 hr	
Conc. (mg/L)	(% Inhibition)	(cells/ml)	
Control	n/a	9.3 x105	
6	5.2	8.9 x105	
14	13	7.8 x105	
34	16	7.8 x105	
82	48	4.3 x105	
196	94	6.1 x104	

n/a - Not applicable

Test condition

Individual test treatment solutions were prepared as Water Accommodated Fractions (WAFs). Test material was added to algal media in 2.0L aspirator bottles. The mixing vessels were sealed with Teflon-covered stoppers and mixed on magnetic stir plates with teflon coated stir bars for 23 hours at room temperature. After mixing the solutions were allowed to settle for one hour and the WAF was removed from the bottom of the mixing vessel via the port and used for testing. Test vessels were 125ml glass Erlenmeyer flasks with approximately 60 ml of treatment solution and inoculated with algae. Test vessels were sealed with foam stoppers. Samples were taken daily for cell counts. Four replicates were prepared for each treatment level. The initial algal concentration was 1.0 x 104 cells/ml. All test replicates were placed on a shaker table at 109 oscillations per minute during the study. Biomass was calculated as the area under the growth curve.

Nominal loading levels were 6, 14, 34, 82, and 196 mg/L

Test temperature was 23.0 Deg. C. Lighting was continuous at 7400 to 8200 Lux. The pH was 7.3 to 7.4 at test initiation and ranged from 6.9 to 7.3 at test termination.

Test treatments were analyzed by GC-FID. Measured values on Day 0 were 5.0, 12.7, 34.7, 83.4 and 182 mg/L. The test material was not detected in the control.

Test substance Conclusion

CAS No. 68526-79-4; Hexanol, branched and linear

Test substance is considered to have moderate acute toxicity to aquatic

(1) valid without restriction Reliability Critical study for SIDS endpoint Flag

03.04.2006 (8)

Species: other algae: green alga

Endpoint

Exposure period : 96 hour(s)
Unit : mg/l

EC50 : = 84 calculated

Method : other: ECOSAR Computer Model

Year : 2005

GLP

Test substance

Test condition: Log Kow (octanol/water partition coefficient) values and a chemical

structure are needed to calculate aquatic toxicity using the ECOSAR model. The Kow calculation is performed by KOWWIN based on an atom/fragment contribution method of Meylan and Howard (1), which is a subroutine in the EPIWIN computer model (2). KOWWIN also has a

database of experimental Kow values (EXPKOW.DB).

The ECOSAR program was run using a C6 alcohol with a Kow of 1.75.

1. Meylan, W. and P. Howard. 1995. Atom/fragment contribution method for estimating octanol-water partition coefficients. J. Pharm. Sci. 84:83-92.

2. Meylan, M., SRC 1994-1999. KOWWIN is contained in the computer program EPIWIN. 1999. Estimation Program Interface for Windows, version 3.04. Syracuse Research Corporation, Syracuse, NY, USA.

Test substance: CAS No. 68526-79-4; Hexanol, branched and linear

Conclusion: Based on the calculated Kow value, the C6 alcohol is expected to have an

acute 96-hour EC50 of 84 mg/L and a Chronic Value of 7.3 mg/L.

Reliability : (2) valid with restrictions

The value was calculated based on chemical structure as modeled by EPIWIN. This robust summary has a reliability rating of 2 because the data

are calculated and not measured.

03.04.2006 (4)

4.4 TOXICITY TO MICROORGANISMS E.G. BACTERIA

4.5.1 CHRONIC TOXICITY TO FISH

Species: other: Freshwater Fish (calculated toxicity values are not species specific)

Endpoint : other: LC50
Exposure period : 30 day(s)
Unit : mg/l

ChV : = 16.1 calculated Method : other: calculated

Year : 2005

GLP

Test substance :

Test condition: Log Kow (octanol/water partition coefficient) values and a chemical

structure are needed to calculate aquatic toxicity using the ECOSAR model. The Kow calculation is performed by KOWWIN based on an atom/fragment contribution method of Meylan and Howard, which is a subroutine in the EPIWIN computer model. KOWWIN also has a database

of experimental Kow values (EXPKOW.DB).

The ECOSAR program was run using a C6 alcohol with a Kow of 1.75.

Test substance: CAS No. 68526-79-4; Hexanol, branched and linear

Conclusion: Based on the calculated Kow value, the C6 alcohol is expected to have a

30-day Chronic Value of 16.1 mg/L.

Reliability : (2) valid with restrictions

The value was calculated based on chemical structure as modeled by ECOSAR. This robust summary has a reliability rating of 2 because the

data are calculated and not measured.

Flag : Critical study for SIDS endpoint

03.04.2006 (3)

4.5.2 CHRONIC TOXICITY TO AQUATIC INVERTEBRATES

Species : Daphnia magna (Crustacea)

Endpoint : mortality
Exposure period : 16 day(s)
Unit : mg/l

EC50 : = 6.3 calculated

Method : other: ECOSAR Computer Model

Year : 2005

GLP : Test substance :

Test condition: Log Kow (octanol/water partition coefficient) values and a chemical

structure are needed to calculate aquatic toxicity using the ECOSAR model. The Kow calculation is performed by KOWWIN based on an atom/fragment contribution method of Meylan and Howard, which is a subroutine in the EPIWIN computer model. KOWWIN also has a database

of experimental Kow values (EXPKOW.DB).

The ECOSAR program was run using a C6 alcohol with a Kow of 1.75.

Test substance: CAS No. 68526-79-4; Hexanol, branched and linear

Conclusion : Based on the calculated Kow value, the C6 alcohol is expected yo have a

16-day EC50 of 6.3 mg/L.

Reliability : (2) valid with restrictions

The value was calculated based on chemical structure as modeled by EPIWIN. This robust summary has a reliability rating of 2 because the data

are calculated and not measured.

Flag : Critical study for SIDS endpoint

03.04.2006 (3)

4.6.1 TOXICITY TO SEDIMENT DWELLING ORGANISMS

4.6.2 TOXICITY TO TERRESTRIAL PLANTS

4.6.3 TOXICITY TO SOIL DWELLING ORGANISMS

Type :

Species: other: Earthworm

Endpoint : mortality
Exposure period : 14 day(s)
Unit : mg/kg soil dw
LC50 : = 750.5 calculated

Method : other: ECOSAR Computer Model

Year : 2005

GLP :

Test substance :

Test condition : Log Kow (octanol/water partition coefficient) values and a chemical

structure are needed to calculate acute terrestrial toxicity using the ECOSAR model. The Kow calculation is performed by KOWWIN based on an atom/fragment contribution method of Meylan and Howard, which is a subroutine in the EPIWIN computer model. KOWWIN also has a database

of experimental Kow values (EXPKOW.DB).

The ECOSAR program was run using a C6 alcohol with a Kow of 1.75.

Test substance

CAS No. 68526-79-4; Hexanol, branched and linear

Conclusion

: Based on the calculated Kow value, the C6 alcohol is expected to have a

14-day EC50 of 750.5 mg/kg soil.

Reliability : (2) valid with restrictions

The value was calculated based on chemical structure as modeled by EPIWIN. This robust summary has a reliability rating of 2 because the data

are calculated and not measured.

Flag

: Critical study for SIDS endpoint

04.04.2006 (3)

4.6.4 TOX. TO OTHER NON MAMM. TERR. SPECIES

4.7 BIOLOGICAL EFFECTS MONITORING

4.8 BIOTRANSFORMATION AND KINETICS

4.9 ADDITIONAL REMARKS

5.0 TOXICOKINETICS, METABOLISM AND DISTRIBUTION

5.1.1 ACUTE ORAL TOXICITY

Type : LD50

Value : = 3670 mg/kg bw

Species : rat

Strain : Sprague-Dawley

Sex : male Number of animals : 5

Vehicle : other: None

Doses :

Method: otherYear: 1960GLP: noTest substance:

Remark: None of the animals died in the 26, 82, 259, and 820 mg/kg dose groups.

One of the animals in the 2591 mg/kg group died within 24 hours of dosing. All animals in the 8200 mg/kg group died within 4 hours following dose administration. Treatment resulted in depression (i.e. inactivity, depressed righting reflexes, ataxia) and labored respiration. These signs had an early onset and recovery was complete by the second day after dosing. Gross necropsy on the animals that died showed congested kidneys. Also, animals that died during the first hour after administration showed evidence

of gastrointestinal irritation.

Route of administration: Gastric Intubation Frequency of Treatment: Single exposure

Dose/Concentration Levels: 26, 82, 259, 820, 2591, 8200 mg/kg

Control group and Treatment: None

Result : LD50= 3670 mg/kg

Test condition : After a three to four hour fasting period, groups of 5 rats received the test

material at dose levels of 26, 82, 259, 820, 2591, and 8200 mg/kg of body weight. The results were converted to weight units by means of the specific gravity. Observations for signs of toxicity were made immediately and at one and 24 hours after compound administration and daily for a period of 7days. Gross necropsy examinations were performed on all animals that

died or were killed.

Test substance: CAS No. 68526-79-4; Hexanol, branched and linear

Conclusion: Under the conditions of this study, Hexanol, branched and linear has a low

order of acute oral toxicity in rats.

Reliability : (1) valid without restriction
Flag : Critical study for SIDS endpoint

03.04.2006 (13)

5.1.2 ACUTE INHALATION TOXICITY

Type : LC50

Value : > 1060 ppm

Species : rat
Strain : Wistar
Sex : male
Number of animals : 10
Vehicle : other: NA

Doses :

Exposure time : 6 hour(s)

Method: otherYear: 1960GLP: noTest substance:

Remark: No deaths were seen during or after the exposure period. Thirty minutes

after exposure, slow, deep respiration was observed. After 90 minutes of exposure, the rats exhibited gasping, labored respiration, lacrimation and nasal discharge. These signs persisted until the termination of exposure. Gross necropsy results indicate that the test material produced slight lung congestion in all animals. All other tissues and organs were unremarkable.

Route of administration: Inhalation

Frequency of Treatment: Single 6 hour exposure

Dose/Concentration Levels: 1060 ppm Control group and Treatment: None

Result : LC50 > 1060 ppm for rats.

Test condition: Rats received a single, 6-hour exposure to the test material in air.

Exposures were at atmospheres nearly saturated with vapors of the alcohol (1060 ppm). The exposure was conducted in a 500-liter stainless steel inhalation chamber equipped at the inlet with a device for generating a near-saturated vapor of the test material. Vapor was generated by using two separate fritted disk glass bubblers, connected in parallel, each containing 200 ml. Of the test material. Air flow through each bubbler was 18 liters/minute, so the total flow through the chamber was 36 liters/min. Actual chamber concentration was not measured during the exposure. The theoretical chamber concentration was calculated to be 1060 ppm based upon the amount of test material that vaporized and the rate of air flow. During exposure, all animals were observed for gross signs of toxicity at 30-minute intervals. Gross necropsies were performed on animals 24 hours

after exposure.

Test substance: CAS No. 68526-79-4; Hexanol, branched and linear

Conclusion: Under the conditions of this study, Hexanol, branched and linear has a low

order of acute inhalation toxicity in rats.

Reliability : (2) valid with restrictions

No analysis of exposure atmosphere.

Flag : Critical study for SIDS endpoint

03.04.2006 (13)

Type : LC50

Value : > 1060 ppm

Species: mouseStrain: SwissSex: maleNumber of animals: 10

Vehicle : other: NA

Doses

Exposure time : 6 hour(s)

Method : other

Year : 1960

GLP : no

Test substance :

Remark: No deaths were seen during or after the exposure period. Thirty minutes

after exposure, slow, deep respiration was observed. After 90 minutes of exposure, the mice exhibited gasping, labored respiration, lacrimation and nasal discharge. These signs persisted until the termination of exposure. Gross necropsy results indicate that the test material produced slight lung congestion in all animals. All other tissues and organs were unremarkable.

Route of administration: Inhalation

Frequency of Treatment: Single 6 hour exposure

Dose/Concentration Levels: 1060 ppm Control group and Treatment: None

Result : LC50 > 1060 ppm for mice.

Test condition : Mice received a single, 6-hour exposure to the test material in air.

Exposures were at atmospheres nearly saturated with vapors of the alcohol (1060 ppm). The exposure was conducted in a 500-liter stainless steel inhalation chamber equipped at the inlet with a device for generating a near-saturated vapor of the test material. Vapor was generated by using two separate fritted disk glass bubblers, connected in parallel, each containing 200 ml. Of the test material. Air flow through each bubbler was 18 liters/minute, so the total flow through the chamber was 36 liters/min. Actual chamber concentration was not measured during the exposure. The theoretical chamber concentration was calculated to be 1060 ppm based upon the amount of test material that vaporized and the rate of air flow. During exposure, all animals were observed for gross signs of toxicity at 30-minute intervals. Gross necropsies were performed on animals 24 hours

after exposure.

Test substance: CAS No. 68526-79-4; Hexanol, branched and linear

Conclusion: Under the conditions of this study, Hexanol, branched and linear has a low

order of acute inhalation toxicity in mice.

Reliability : (2) valid with restrictions

No analysis of exposure atmosphere.

Flag : Critical study for SIDS endpoint

03.04.2006 (13)

Type : LC50
Value : > 1060 ppm
Species : guinea pig

Strain : other: English short hair

Sex : male : 10 Vehicle : other: NA

Doses

Exposure time : 6 hour(s)

Method : other

Year : 1960

GLP : no

Test substance :

Remark: No deaths were seen during or after the exposure period. Thirty minutes

after exposure, slow, deep respiration was observed. After 90 minutes of exposure, the guinea pigs exhibited gasping, labored respiration, lacrimation and nasal discharge. These signs persisted until the termination of exposure. Gross necropsy results indicate that the test material produced slight lung congestion in all animals. All other tissues

and organs were unremarkable. Route of administration: Inhalation

Frequency of Treatment: Single 6 hour exposure

Dose/Concentration Levels: 1060 ppm Control group and Treatment: None

Result : LC50 > 1060 ppm for guinea pigs. **Test condition** : Guinea pigs received a single, 6-h

Guinea pigs received a single, 6-hour exposure to the test material in air. Exposures were at atmospheres nearly saturated with vapors of the alcohol (1060 ppm). The exposure was conducted in a 500-liter stainless steel inhalation chamber equipped at the inlet with a device for generating a near-saturated vapor of the test material. Vapor was generated by using two separate fritted disk glass bubblers, connected in parallel, each containing 200 ml. Of the test material. Air flow through each bubbler was 18 liters/minute, so the total flow through the chamber was 36 liters/min. Actual chamber concentration was not measured during the exposure. The theoretical chamber concentration was calculated to be 1060 ppm based upon the amount of test material that vaporized and the rate of air flow. During exposure, all animals were observed for gross signs of toxicity at 30-minute intervals. Gross necropsies were performed on animals 24 hours

after exposure.

Test substance: CAS No. 68526-79-4; Hexanol, branched and linear

Conclusion: Under the conditions of this study, Hexanol, branched and linear has a low

order of acute inhalation toxicity in guinea pigs.

Reliability : (2) valid with restrictions

No analysis of exposure atmosphere.

Flag : Critical study for SIDS endpoint

03.04.2006 (13)

5.1.3 ACUTE DERMAL TOXICITY

Type : LD50

Value : > 2600 mg/kg bw

Species: rabbitStrain: other: AlbinoSex: male/female

Number of animals : 4

Vehicle : other: None

Doses

Method : other
Year : 1960
GLP : no
Test substance :

Remark: Route of administration: Dermal Application

Frequency of Treatment: Single exposure

Dose/Concentration Levels: 82, 259, 820, and 2600 mg/kg

Control group and Treatment: None

There were no mortalities at any dosage level tested. The LD50 in albino rabbits is greater than the highest dose tested (approx. 2.6 g/kg body weight). Signs of toxicity included labored respiration and central nervous system depression. All animals recovered within 4-48 hours after the exposure period began. Moderate erythema and edema were observed.

Result : LD50 > 2600 mg/kg

Test condition: A single dermal application of the test material was made to four groups of

four rabbits at doses of 82, 259, 820, and 2600 mg/kg. The results were converted to weight units by means of the specific gravity. The test material

was applied to intact abdominal skin and covered with an occlusive

covering for 24 hours. Observations for signs of toxicity were made at one, four and 24 hours after compound administration and thereafter for a total of 7 days. Gross necropsies were performed on all animals at the end of

the observation period.

Test substance: CAS No. 68526-79-4; Hexanol, branched and linear

Conclusion: Under conditions of this study, Hexanol, branched and linear has a low

order of acute dermal toxicity in rabbits.

Reliability : (1) valid without restriction
Flag : Critical study for SIDS endpoint

03.04.2006 (13)

5.1.4 ACUTE TOXICITY, OTHER ROUTES

5.2.1 SKIN IRRITATION

Species : rabbit

Concentration

Exposure : Occlusive **Exposure time** : 24 hour(s)

Number of animals : 16

Vehicle : other: none

PDII

Result: moderately irritating

Classification

Method : other: similar to OECD TG 404

Year : 1960

GLP : Test substance :

Remark: A single 24-hour application of Hexanol, branched and linear to intact rabbit

skin produced moderate irritation. Skin irritation consisted of slight to moderate erythema, moderate edema, atonia, and desquamation persisted

to termination on day 14.

Test condition: Hexanol, branched and linear was applied to the clipped, abdominal skin of

4 groups of 4 rabbits. Each application consisted of 0.10, 0.316, 1.00, or 3.16 g/kg test material under an occlusive dressing which was left in place for 24 hours. Measurements were made at 1 hour and at 24, 48 and 72

hours after dressing removal and continued daily for 7 days.

Test substance: CAS No. 68526-79-4; Hexanol, branched and linear

Conclusion: Under the conditions of the study, Hexanol, branched and linear produced

moderate irritation to rabbit skin.

Reliability : (2) valid with restrictions

Pre-GLP

Flag : Critical study for SIDS endpoint

03.04.2006 (12) (22)

5.2.2 EYE IRRITATION

Species : rabbit
Concentration : .1 undiluted

Dose :

Exposure time :
Comment :
Number of animals : 6
Vehicle :

Result : highly irritating

Classification

Method : other: similar to OECD TG 405

Year :

Test substance :

Result: Median Scores were:

24 hr - 31 72 hr - 23 7 days - 0

Hexyl alcohol caused severe eye irritation involving all eye structures and was based on persistent iritis, corneal opacity, and in two animals, corneal

vascularization.

Test condition: A single application of 0.1 ml of undiluted alcohol was made into the

conjunctival sac of the left eye of each rabbit. The untreated eye served as the control. The treated eye was held closed for 1 second, and eyes were not washed. Observations for irritancy and possible systemic toxicity were made at 1, 4, and 24 hours and at 2, 3, 4, and 7 days. At each interval the treated and control eyes were examined and scored for ocular reactions

according to the Draize Standard Eye Irritation Grading Scale.

Test substance: CAS No. 68526-79-4; Hexanol, branched and linear

5. Toxicity Id 68526-79-4

Date

Reliability : (2) valid with restrictions

Pre GLP

Flag : Critical study for SIDS endpoint

03.04.2006 (22)

5.3 SENSITIZATION

Type : Guinea pig maximization test

Species : guinea pig

Number of animals

Vehicle

Result : not sensitizing

Classification

Method: other: Magnusson, B. Kligman, A.M. J. Invest. Dermatol., 52, 1969.

Year : 1969

GLP :

Test substance :

Test substance : Analogue substance CAS No. 111-27-3; 1-Hexanol

Reliability : (1) valid without restriction

Peer reviewed publication.

Flag : Critical study for SIDS endpoint

03.04.2006 (5)

Type : other: Human Maximization (as described by Epstein and Kligman)

Species : human

Number of animals

Vehicle

Result : not sensitizing
Classification : not sensitizing
Method : other: not specified

Year :

GLP :

Test substance :

Remark: The test substance was prepared and administered as a 1%

solution of 1-hexanol in petrolatum.

Test substance : CAS No. 111-27-3, 1-hexanol **Conclusion** : Not likely to be a skin sensitizer.

Reliability : (2) valid with restrictions

Peer reviewed publication.

Flag : Critical study for SIDS endpoint

03.04.2006 (5)

5.4 REPEATED DOSE TOXICITY

Type :

Species: rabbitSex: male/femaleStrain: other: AlbinoRoute of admin.: dermalExposure period: 10 daysFrequency of treatm.: once/day

Post exposure period

Doses : 0 g/kg, 0.4 g/kg and 2.0 g/kg **Control group** : other: Isopropyl alcohol

NOAEL : = 2000 mg/kg

Method : other

ld 68526-79-4 5. Toxicity

Date

1961 Year **GLP** no **Test substance**

Remark Duration of test: 12 Days

> There was no evidence of systemic toxicity at either dose of the test substance. Histopathological findings were unremarkable. Repeated application of the test material to the skin of albino rabbits at both dose levels produced moderate to marked degree of irritation. A slight to marked degree of edema was observed in two low-dose animals and three high-dose animals following one or more of the first three applications. Also, the exposed skin of two high-dose animals showed necrosis.

Result NOAEL for systemic toxicity = 2.0 g/kg

Undiluted control and test materials were applied to intact skin of the **Test condition**

animals. Materials were applied once daily for a total of ten applications with a one-day rest period between the third and fourth and eighth and

ninth applications. The exposed skin area of each animal was

approximately 10% of the total body surface at the 0.4 g/kg dosage level and approximately 40% of the total body surface at the 2.0 g/kg dosage level. After the first application, exposed skin was covered by rubber dental damming. In subsequent applications, loose gauze and adhesive tape were used to cover the exposed area since the authors felt that the damming itself may have induced some irritation. Each exposure period lasted approximately 18-24 hours. Animals were observed daily throughout the study and body weights were recorded prior to each

application and at study termination.

Clinical hematology and urinalysis were performed at the beginning of the study and 24 hours after the final application of test material. Animals were sacrificed 48 hours after the tenth application and brain, liver, kidney, and blood samples were taken. In addition, samples of brain, thyroid, lung, heart, liver, kidneys, adrenals, skin, and bone marrow were preserved.

Test substance Analogue substance - CAS No. 111-27-3; 1-Hexanol

Conclusion Under the conditions of this study, Hexanol, branched and linear can

produce moderate skin irritation following repeated dermal exposures. However, the test material did not produce any evidence of systemic

toxicity under the conditions of this study.

Reliability (1) valid without restriction Flag Critical study for SIDS endpoint

03.04.2006 (7)

5.5 **GENETIC TOXICITY 'IN VITRO'**

Ames test Type

System of testing Salmonella typhimurium

Test concentration 8, 40, 200, 1000, and 5000 micrograms/plate

Cycotoxic concentr.

Metabolic activation with and without

Result negative

Method OECD Guide-line 471

Year 1983 **GLP** yes Test substance

Test substance Analogue substance - CAS No. 111-27-3, 1-hexanol

(1) valid without restriction Reliability Flag Critical study for SIDS endpoint

03.04.2006 (6)

Ames test Type

System of testing : Salmonella typhimurium

Test concentration: 6.25, 25, 100, 400, and 1600 micrograms/plate

Cycotoxic concentr. :

Metabolic activation: with and without

Result : negative

Method : OECD Guide-line 471

Test substance

Test substance: Analogue substance - CAS No. 111-27-3, 1-hexanol

Reliability : (1) valid without restriction
Flag : Critical study for SIDS endpoint

03.04.2006 (6)

5.6 GENETIC TOXICITY 'IN VIVO'

5.7 CARCINOGENICITY

5.8.1 TOXICITY TO FERTILITY

5.8.2 DEVELOPMENTAL TOXICITY/TERATOGENICITY

Species : rat Sex : female

Strain : Sprague-Dawley
Route of admin. : inhalation

Exposure period : 10 days

Exposure period : 19 days **Frequency of treatm**. : 7 hrs/day

Duration of test : Gestation days 1-19

Doses : 3500 mg/m3 (Saturated vapors)
Control group : other: 15 sham-exposed rats

NOAEL maternal tox. : $> 3500 - \text{mg/m}^3$

Method: otherYear: 1989GLP: no dataTest substance: other TS

Remark: No. of animals/sex/dose: 15 dams/treatment

Statistical methods: MANOVA, ANOVA, Kruskal-Wallis test

The test substance was administered by inhalation to reflect the route of exposure found in industry. However, due to the low volatility of the alcohols, concentrations sufficient to induce maternal toxicity could not be achieved. There were no significant fetal malformations associated with inhalation of 1-hexanol by the dam. There was a slight but statistically significant increase in resorptions (1.3 vs. 0.4 per litter for controls). However, this resorption mean was still in the range seen in historical

controls.

Result : NOAEL > 3500 mg/m3

Test condition: Throughout the study, all animals were housed under standard

environmental conditions and allowed free access to food and water except when the pregnant females were in the exposure chamber. Following mating, sperm-positive females were placed in cages and weighed. Dams were weighed daily for the first week of exposure and weekly thereafter. Exposures were conducted in Hinners-type chambers. The purity of the

test substance was ³ 99% as measured by gas chromatography. A constant flow of the test substance was mixed with a known volume of heat compressed air, resulting in instantaneous vaporization of the test substance, which then flowed into the chamber. The concentration of the test substance was monitored continuously and recorded every hour. Calibration checks were completed daily. Exposure concentrations were verified on a weekly basis using a secondary method of analysis. The highest concentration of vapor that could be generated was 3500 mg/m3. Dams were exposed from days 1-19 of gestation. On day 20, dams were sacrificed by CO2 asphyxiation, and the uterus and ovaries were removed and examined for corpa lutea, implantations, resorption sites, and live fetuses. Fetuses were removed and examined for external malformations, sexed, weighed, and examined for visceral or skeletal defects.

Test substance

: 1-Hexanol

Conclusion

: Inhalation of saturated vapors of 1-hexanol is not maternally toxic or

teratogenic in rats.

Reliability

(2) valid with restrictions2 - Reliable with restrictions.

28.11.2001

(1)

5.8.3 TOXICITY TO REPRODUCTION, OTHER STUDIES

5.9 SPECIFIC INVESTIGATIONS

5.10 EXPOSURE EXPERIENCE

5.11 ADDITIONAL REMARKS

6. Analyt. Meth. for Detection and Identification	ld 68526-79-4 Date 04.04.2006
6.1 ANALYTICAL METHODS	
6.2 DETECTION AND IDENTIFICATION	
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7. Eff	. Against Target Org. and Int	ended Uses		68526-79-4	
7.1	FUNCTION				
7.2	EFFECTS ON ORGANISMS TO BE CON	TROLLED			
7.3	ORGANISMS TO BE PROTECTED				
7.4	USER				
7.5	RESISTANCE				
		32 / 36			
	7.1 7.2 7.3 7.4	7.1 FUNCTION 7.2 EFFECTS ON ORGANISMS TO BE CON 7.3 ORGANISMS TO BE PROTECTED 7.4 USER	 7.1 FUNCTION 7.2 EFFECTS ON ORGANISMS TO BE CONTROLLED 7.3 ORGANISMS TO BE PROTECTED 7.4 USER 	7.1 FUNCTION 7.2 EFFECTS ON ORGANISMS TO BE CONTROLLED 7.3 ORGANISMS TO BE PROTECTED 7.4 USER 7.5 RESISTANCE	7.1 FUNCTION 7.2 EFFECTS ON ORGANISMS TO BE CONTROLLED 7.3 ORGANISMS TO BE PROTECTED 7.4 USER 7.5 RESISTANCE

Id 68526-79-4 8. Meas. Nec. to Prot. Man, Animals, Environment **Date** 04.04.2006 8.1 METHODS HANDLING AND STORING 8.2 FIRE GUIDANCE **EMERGENCY MEASURES** 8.3 8.4 POSSIB. OF RENDERING SUBST. HARMLESS **WASTE MANAGEMENT** 8.6 **SIDE-EFFECTS DETECTION** 8.7 SUBSTANCE REGISTERED AS DANGEROUS FOR GROUND WATER 8.8 REACTIVITY TOWARDS CONTAINER MATERIAL

9. References Id 68526-79-4 Date 04.04.2006

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ld 68526-79-4 9. References **Date** 04.04.2006 (19)Meylan, M., SRC 1994-1999. KOWWIN is contained in the computer program EPIWIN. 1999. Estimation Program Interface for Windows, version 3.04. Syracuse Research Corporation, Syracuse, NY, USA. Meylan, M., SRC 1994-1999. WSKOWWIN is contained in the computer program EPIWIN. (20)1999. Estimation Program Interface for Windows, version 3.04. Syracuse Research Corporation, Syracuse, NY, USA. Meylan, W., Howard, P.H. and R.S. Boethling. (1992). Molecular Topolgy/Fragment (21)Contribution Method for Predicting Soil Sorption Coefficients. Environ. Sci. Technol. 26:1560-7. (22)Scala RA and Burtis EG (1973). Acute toxicity of a homologous series of branched chain primary alcohols. Am Ind Hyg Assoc J 34:493-499.

10. Summary and Evaluation		04.04.2006	
10.1 END POINT SUMMARY			
10.2 HAZARD SUMMARY			
10.3 RISK ASSESSMENT			
	36 / 36		

IUCLID

Data Set

Existing Chemical : ID: 68526-83-0 **CAS No.** : 68526-83-0

EINECS Name : Alcohols, C7-9-iso-, C8-rich

EC No. : 271-231-4

TSCA Name : Alcohols, C7-9-iso-, C8-rich

Producer related part

Company: ExxonMobil Biomedical Sciences Inc.

Creation date : 18.09.2001

Substance related part

Company: ExxonMobil Biomedical Sciences Inc.

Creation date : 18.09.2001

Status

Memo : Prepared for EMCC - US HPV

Printing date : 05.04.2006

Revision date :

Date of last update : 05.04.2006

Number of pages : 43

Chapter (profile) : Chapter: 1, 2, 3, 4, 5, 6, 7, 8, 10 Reliability (profile) : Reliability: without reliability, 1, 2, 3, 4

Flags (profile) : Flags: without flag, confidential, non confidential, WGK (DE), TA-Luft (DE),

Material Safety Dataset, Risk Assessment, Directive 67/548/EEC, SIDS

ld 68526-83-0

Date

1.0.1 APPLICANT AND COMPANY INFORMATION

1.0.2 LOCATION OF PRODUCTION SITE, IMPORTER OR FORMULATOR

1.0.3 IDENTITY OF RECIPIENTS

1.0.4 DETAILS ON CATEGORY/TEMPLATE

Comment: see free text

Remark : The Alkyl Alcohols C6 to C13 Category is a family of saturated alcohols

that are produced from olefins by the hydroformylation or "oxo" process. Hydroformylation is the reaction of an olefin with carbon monoxide and hydrogen to produce an aldehyde, and its subsequent hydrogenation to the

alcohol.

The number of carbon atoms in the category members ranges from 6 to 13. Category members contain predominantly branched alkyl groups. Each substance consists of an isomeric mixture of saturated primary alcohols of high purity, and the following basic structure: CH3-R-CH2-OH, where R is a branched isomeric structure.

The justification for the Alkyl Alcohols C6 to C13 Category is that the members have:

- · similar chemical structures,
- · similar physico-chemical properties,
- · comparable environmental fate,
- · the same mode of action.

In general, aliphatic alcohol toxicity occurs by non-polar narcosis (Lipnick et al., 1985). The mode of action is disruption of biological membrane function (van Wezel and Opperhuizen, 1995). Metabolic pathways, through which alcohols are metabolized, are likely to include similar reactions for all category members and result in structurally similar metabolites (carboxylic acids).

The data demonstrate that the category is valid for a screening-level hazard assessment for the Category and its members. One can assess the untested endpoints by extrapolation between and among the category members.

20.03.2006

1.1.0 SUBSTANCE IDENTIFICATION

IUPAC Name : Smiles Code :

Molecular formula : C8H18O Molecular weight : 130.23

Petrol class

Flag : Critical study for SIDS endpoint

04.04.2006

1. General Information

ld 68526-83-0 **Date** 05.04.2006

1.1.1 GENERAL SUBSTANCE INFORMATION

Purity type

Substance type : organic Physical status : liquid

Purity : Colour : Odour :

Remark: CAS Registry Number, Name, and General Structure for Members of the

Alkyl Alcohols C6 to C13 Category and Analogue Substances:

CAS RN: 104-76-7

IUPAC Name: 2-ethylhexan-1-ol R Length (C number): C8

Structure of R: Branched and Linear

Category Member: No (analogue substance used for supporting

information)

CAS RN: 111-87-5 IUPAC Name: octan-1-ol R Length (C number): C8 Structure of R: Linear

Category Member: No (analogue substance used for supporting

information)

CAS RN: 68526-83-0

IUPAC Name: Alcohols, C7-C9-iso, C8 rich

R Length (C number): C8

Structure of R: Branched, primarily methyl branched (isomeric structures)

Category Member: Yes

04.04.2006

1.1.2 SPECTRA

1.2 SYNONYMS AND TRADENAMES

Exxal 8

21.09.2004

isooctanol

21.09.2004

isooctyl alcohol

21.09.2004

1.3 IMPURITIES

Purity: typical for marketed substance

CAS-No : 68526-83-0 **EC-No** : 271-231-4

EINECS-Name : Alcohols, C7-9-iso-, C8-rich

Molecular formula : C8H18O

1. General Information

ld 68526-83-0 **Date** 05.04.2006

Value : = 99 % w/w

Remark : Commercial product typically consists of methyl-1-heptanols and/or

dimethyl-1-hexanols.

04.04.2006

1.4 ADDITIVES

Purity type : typical for marketed substance

CAS-No :

EINECS-Name Molecular formula Value

Function of additive :

Remark: No additives present.

04.04.2006

1.5 TOTAL QUANTITY

1.6.1 LABELLING

1.6.2 CLASSIFICATION

1.6.3 PACKAGING

1.7 USE PATTERN

Type of use : industrial

Category : Chemical industry: used in synthesis

Remark: Priciple use of these alcohols is in the preparation of plasticizers, mainly

dioctyl phthalate, but also esters of adipic, sebacic, azelmic, and trimellitic acids. Also used as a solvent for fats, oils, and waxes, as well as various

rubber formulations and resins.

04.04.2006

1.7.1 DETAILED USE PATTERN

1.7.2 METHODS OF MANUFACTURE

1.8 REGULATORY MEASURES

1.8.1 OCCUPATIONAL EXPOSURE LIMIT VALUES

1. General Information Date 05.04.2006 1.8.2 ACCEPTABLE RESIDUES LEVELS 1.8.3 WATER POLLUTION 1.8.4 MAJOR ACCIDENT HAZARDS 1.8.5 AIR POLLUTION 1.8.6 LISTINGS E.G. CHEMICAL INVENTORIES 1.9.1 DEGRADATION/TRANSFORMATION PRODUCTS 1.9.2 COMPONENTS 1.10 SOURCE OF EXPOSURE 1.11 ADDITIONAL REMARKS 1.12 LAST LITERATURE SEARCH 1.13 REVIEWS

Id 68526-83-0

ld 68526-83-0

Date

2.1 MELTING POINT

Value : = -65 °C

Sublimation

Method : other: ASTM D97

Year :

GLP : no data

Test substance :

Test substance : CAS No. 68526-83-0; Alcohols, C7-9-iso, C8-rich

Reliability : (2) valid with restrictions

Although the original data were not retrieved and reviewed for quality, they

were developed following acceptable test methods and therefore

considered reliable.

Flag : Critical study for SIDS endpoint

04.04.2006 (12)

2.2 BOILING POINT

Value : $= 185 - 193 \, ^{\circ}\text{C}$ at 1013 hPa

Decomposition

Method: other: D1078/01

Year

GLP : no data

Test substance :

Test substance : CAS No. 68526-83-0; Alcohols, C7-9-iso, C8-rich

Reliability : (2) valid with restrictions

Although the original data were not retrieved and reviewed for quality, they

were developed following acceptable test methods and therefore

considered reliable.

Flag : Critical study for SIDS endpoint

04.04.2006 (12)

2.3 DENSITY

Type : density

Value : = .831 g/cm³ at 20 °C

Method : other: ASTM D4052/86 equivalent

Year

GLP : no data Test substance : other TS

Test substance : CAS No. 68526-83-0; Alcohols, C7-9-iso, C8-rich

Reliability : (4) not assignable

Although the original data were not retrieved and reviewed for quality, they

were developed following acceptable test methods and therefore

considered reliable.

Flag : Critical study for SIDS endpoint

21.09.2004 (12)

2.3.1 GRANULOMETRY

ld 68526-83-0 **Date** 05.04.2006

2.4 VAPOUR PRESSURE

Value : = 2.59 hPa at 25 °C

Decomposition : Method : Year :

GLP : no data

Test substance

Test substance : CAS No. 68526-83-0; Alcohols, C7-9-iso, C8-rich

Reliability : (2) valid with restrictions

Although the original data were not retrieved and reviewed for quality, they

were developed following acceptable test methods and therefore

considered reliable. Value was provided by the experimental database of

the EPIWIN program.

Flag : Critical study for SIDS endpoint

04.04.2006 (20)

2.5 PARTITION COEFFICIENT

Partition coefficient :

Log pow : = 2.9 - 3.4 at 30 °C

pH value

Method : OECD Guide-line 117 "Partition Coefficient (n-octanol/water), HPLC

Method"

Year : 1998 GLP : yes Test substance :

Remark : Test Type: N-Octanol/Water Partition Coefficient (HPLC method)

Result: The test substance eluted as several groups. The three major components

C7, C8, C9 alcohols had Log Pow values of 2.9, 3.0, and 3.4 respectively.

The retention time for the 3 major components were 5.72, 6.03, and 7.28

minutes.

All values were measured using High Performance Liquid Chromatography

(HPLC).

Test condition: The test substance was evaluated as a solution in HPLC grade methanol.

Six reference compounds were also evaluated in a standard combined reference solution (2-butanone, acetophenone, naphthalene, biphenyl, n-butylbenzene, and 4,4-DDT) in 75% methanol and 25% distilled water.

The pH of the solution was 5.4.

Two customized alcohol reference solutions were also prepared containing five of the ten alcohol compounds (1-hexanol, 1-heptanol, 1-octanol, 1-

nonanol, 1-decanol, 1-undecanol, 1-dodecanol, 1-tridecanol, 1-tetradecanol, 1-pentadecanol) in 87.5% methanol and 12.5% distilled

water. The pH of both solutions was 7.3.

The pH of the evaluated solutions was the same as the reference solution it

was evaluated against.

The test substance was analyzed against a Standard Log Pow Reference Compound Solution and a customized Alcohol Reference Compound Solution. Only the peaks detected by refractive index (RI) were reported.

Test substance : CAS No. 68526-83-0; Alcohols, C7-9-iso, C8-rich

Reliability : (1) valid without restriction
Flag : Critical study for SIDS endpoint

ld 68526-83-0 Date 05.04.2006

04.04.2006 (7)

2.6.1 SOLUBILITY IN DIFFERENT MEDIA

Solubility in Water

Value = 1379 - 1485 mg/l at 25 °C

pH value

at °C concentration

Temperature effects

Examine different pol.

at 25 °C pKa

Description

Test substance

Stable

Deg. product

Method other: calculated

Year

GLP

Calculated values using WSKOWWIN version 1.36, a subroutine of the Method

computer program EPIWIN version 3.04

Test condition Water Solubility estimations performed by WSKOWWIN are based on a

> Kow correlation method described by W. Meylan, P. Howard and R. Boethling in "Improved method for estimating water solubility from octanol/water partition coefficient". Environ. Toxicol. Chem. 15:100-106.

1995.

Test substance : CAS No. 68526-83-0; Alcohols, C7-9-iso, C8-rich

(2) valid with restrictions Reliability

The result is a calculated value based on the chemical structure and

represents a potential melting point for the substance with the CAS number

listed under test substance.

: Critical study for SIDS endpoint Flag

04.04.2006 (22)

2.6.2 SURFACE TENSION

2.7 FLASH POINT

= 60 °C **Value** Type closed cup

Method other: TCC ASTM D56

Year

GLP : no data Test substance other TS

Test substance : CAS No. 68526-83-0; Alcohols, C7-9-iso, C8-rich

(4) not assignable Reliability

Although the original data were not retrieved and reviewed for quality, they

were developed following acceptable test methods and therefore

considered to be reliable.

: Critical study for SIDS endpoint Flag

21.09.2004 (12)

2.8 AUTO FLAMMABILITY

ld 68526-83-0 **Date** 05.04.2006

2.9 FLAMMABILITY

2.10 EXPLOSIVE PROPERTIES

2.11 OXIDIZING PROPERTIES

2.12 DISSOCIATION CONSTANT

Acid-base constant : = 15.50 at 25°C

Method: pKa calculation by SPARC 2003 using a Linux calculation engine.

Remark: SPARC On-line calculator can be accessed at

http://ibmlc2.chem.uga.edu/sparc/index.cfm

Test substance: CAS No. 70914-20-4; Alcohols, C6-8 branched

Reliability : (2) valid with restrictions

The value was calculated based on chemical structure as modeled by SPARC. This robust summary has a reliability rating of 2 because the data

are calculated and not measured.

04.04.2006 (17)

2.13 VISCOSITY

Value : $= 13 - at 20 \, ^{\circ}C$

Result

Method : other: ASTM D445

Year

GLP : no data Test substance : other TS

Remark : Value measured in cSt

Test substance : CAS No. 68526-83-0; Alcohols, C7-9-iso, C8-rich

Reliability : (4) not assignable

Although the original data were not retrieved and reviewed for quality, they

were developed following acceptable test methods and therefore

considered to be reliable.

21.09.2004 (12)

2.14 ADDITIONAL REMARKS

ld 68526-83-0

Date

3.1.1 PHOTODEGRADATION

Type : water

Light source :

Light spectrum: nm

Relative intensity : based on intensity of sunlight

Deg. product

Method : other (calculated): Technical Discussion

Year : GLP : Test substance :

Remark: These data represent a key study for characterising the potential of

substances in the Alkyl Alcohols C6 to C13 Category to undergo direct

photodegradation.

Result: Photolysis as a Function of Molecular Structure

The direct photolysis of an organic molecule occurs when it absorbs sufficient light energy to result in a structural transformation (Harris, 1982). The reaction process is initiated when light energy in a specific wavelength range elevates a molecule to an electronically excited state. However, the excited state is competitive with various deactivation processes that can result in the return of the molecule to a non excited state.

The absorption of light in the ultra violet (UV)-visible range, 110-750 nm, can result in the electronic excitation of an organic molecule. Light in this range contains energy of the same order of magnitude as covalent bond dissociation energies (Harris, 1982). Higher wavelengths (e.g. infrared) result only in vibrational and rotational transitions, which do not tend to produce structural changes to a molecule.

The stratospheric ozone layer prevents UV light of less than 290 nm from reaching the earth's surface. Therefore, only light at wavelengths between 290 and 750 nm can result in photochemical transformations in the environment (Harris, 1982). Although the absorption of UV light in the 290-750 nm range is necessary, it is not always sufficient for a chemical to undergo photochemical degradation. Energy may be re-emitted from an excited molecule by mechanisms other than chemical transformation, resulting in no change to the parent molecule.

A conservative approach to estimating a photochemical degradation rate is to assume that degradation will occur in proportion to the amount of light wavelengths >290 nm absorbed by the molecule (Zepp and Cline, 1977).

Substances in the Oxo Alcohols C6 to C13 Category contain molecules that are oxygenated aliphatic compounds which will absorb UV light below 220 nm (Boethling and Mackay, 2000) and will not undergo direct photolysis. Therefore, this fate process will not contribute to a measurable degradative removal of chemical components in this category from the environment.

References

Boethling, R.S., Mackay, D. (2000). Handbook of Property Estimation Methods for Chemicals. CRC Press, Boca Raton, FL, USA.

Harris, J. C. 1982. "Rate of Aqueous Photolysis," Chapter 8 in: W. J. Lyman, W. F. Reehl, and D. H. Rosenblatt, eds., Handbook of Chemical

ld 68526-83-0

Date

Property Estimation Methods, McGraw-Hill Book Company, New York,

USA.

Zepp, R. G. and D. M. Cline. 1977. Rates of Direct Photolysis in the

Aqueous Environment, Environ. Sci. Technol., 11:359-366.

Test substance : Alkyl Alcohols C6 to C13 Category members

Flag : Critical study for SIDS endpoint

03.04.2006 (11)

Type

Light source : Sun light Light spectrum : nm

Relative intensity: based on intensity of sunlight

INDIRECT PHOTOLYSIS

Sensitizer : OH Conc. of sensitizer : 1500000

Rate constant : = .0000000000125338 cm³/(molecule*sec)

Degradation: % after

Deg. product :

Method : other (calculated): Calculated values using AOPWIN version 1.89, a

subroutine of the computer program EPIWIN version 3.04

Year : GLP : Test substance :

Result : Atmospheric Oxidation Potential

In the environment, organic chemicals emitted into the troposphere are degraded by several important transformation processes. The dominant transformation process for most compounds is the daylight reaction with hydroxyl (OH-) radicals (Atkinson, 1988, 1989). The rate at which an organic compound reacts with OH- radicals is a direct measure of its atmospheric persistence (Meylan and Howard, 1993).

AOPWIN estimates the rate constant for the atmospheric, gas-phase reaction between photochemically produced hydroxyl radicals and organic chemicals. The rate constants estimated by the program are then used to calculate atmospheric half-lives for organic compounds based upon average atmospheric concentrations of hydroxyl radicals.

Since the reactions only take place in the presence of sunlight, the atmospheric half-lives are normalized for a 12-hour day.

Calculated* OH- Rate Constant half-life (hrs) (cm3/molecule-sec)

10.2 12.53 E-12

References:

Atkinson, R. 1988. Estimation of gas-phase hydroxyl radical rate constants for organic chemicals. Environ. Toxicol. Chem. 7:435-442.

Atkinson, R. 1989. Kinetics and mechanisms of the gas-phase reactions of the hydroxyl radical with organic compounds. J. Phys. Chem. Ref. Data Monograph No. 1, Amer. Inst. Physics & Amer. Chem. Soc., NY.

Meylan, W.M. and P.H. Howard. 1993. Computer estimation of the atmospheric gas-phase reaction rate of organic compounds with hydroxyl radicals and ozone. Chemosphere 12:2293-2299.

ld 68526-83-0

Date

Test condition: Indirect photodegradation, or atmospheric oxidation potential, is based on

the structure-activity relationship methods developed by R. Atkinson.

Temperature: 25°C Sensitizer: OH radical

Concentration of Sensitizer: 1.5 E6 OH radicals/cm3 CAS No. 68526-83-0; Alcohols, C7-9-iso, C8-rich

Test substance : CAS No. 68526-83-0; A

Reliability : (2) valid with restrictions
The results include calculated data based on chemical structure as

modeled by AOPWIN. The data represent a potential atmospheric half-life

for the test substance.

Flag : Critical study for SIDS endpoint

04.04.2006 (21)

3.1.2 STABILITY IN WATER

 Type
 : abiotic

 t1/2 pH4
 : at °C

 t1/2 pH7
 : at °C

 t1/2 pH9
 : at °C

Deg. product

Method : other: Technical Discussion

Year :

GLP : Test substance :

Remark: These data represent a key study for characterising the potential of

substances in the Alkyl Alcohols C6 to C13 Category to undergo

hydrolysis.

Result : Hydolysis as a Function of Molecular Structure

Hydrolysis of an organic molecule occurs when a molecule (R-X) reacts with water (H2O) to form a new carbon-oxygen bond after the carbon-X bond is cleaved (Gould, 1959; Harris, 1982). Mechanistically, this reaction is referred to as a nucleophilic substitution reaction, where X is the leaving group being replaced by the incoming nucleophilic oxygen from the water molecule.

Chemicals that are susceptible to hydrolysis contain functional groups that can be displaced by a nucleophilic substitution reaction. Substances that have the potential to hydrolyze include alkyl halides, amides, carbamates, carboxylic acid esters and lactones, epoxides, phosphate esters, and sulfonic acid esters (Neely, 1985). The lack of a leaving group renders a compound resistant to hydrolysis.

Alkyl alcohols are resistant to hydrolysis because they lack a functional group that is hydrolytically reactive (Harris, 1982).

References:

Gould, E.S. (1959). Mechanism and Structure in Organic Chemistry, Holt, Reinhart and Winston, New York, NY, USA.

Harris, J.C. (1982). "Rate of Hydrolysis," Chapter 7 in: W.J. Lyman, W.F. Reehl, and D.H. Rosenblatt, eds., Handbook of Chemical Property Estimation Methods, McGraw-Hill Book Company, New York, NY, USA.

Neely, W.B. (1985). Hydrolysis. In: W.B. Neely and G.E. Blau, eds., Environmental Exposure from Chemicals. Vol. 1, pp. 157-173. CRC Press, Boca Raton, FL, USA.

ld 68526-83-0 **Date** 05.04.2006

Conclusion: Hydrolysis will not contribute to the removal of Alkyl Alcohols from the

environment.

Flag : Critical study for SIDS endpoint

20.03.2006 (10)

3.1.3 STABILITY IN SOIL

3.2.1 MONITORING DATA

3.2.2 FIELD STUDIES

3.3.1 TRANSPORT BETWEEN ENVIRONMENTAL COMPARTMENTS

Type : fugacity model level III

Media: other: air - water - soil - sedimentAir: % (Fugacity Model Level I)Water: % (Fugacity Model Level I)Soil: % (Fugacity Model Level I)Biota: % (Fugacity Model Level II/III)Soil: % (Fugacity Model Level II/III)

Method : other: Calculation according Mackay, Level III

Year : 2003

Method : The EQC Level III model is a steady state model that is useful for

determining how the medium of release affects environmental fate. Level III fugacity allows non-equilibrium conditions to exist between connected media as steady state, and illustrate important transport and transformation

processes.

Physicochemical input values for the model were calculated using the EPIWIN Estimation v 3.04 program. Measured input values were also used where available and obtained from the EPIWIN database. Distribution data from the equilibrium model provide basic information on the potential partitioning behavior of chemicals between selected environmental

compartments (i.e., air, water, soil, and sediment).

Input values used:

Molecular mass = 130.23 g/mol Water solubility = 1379 mg/L Vapour pressure = 2.59 Pa

log Kow = 3.20

Melting point = -65 deg C

Degradation half-lives:

Air - 10.2 hrs Water - 120 hrs Soil - 720 hrs Sediment - 7200 hrs

This model was run assuming 100% discharge to water.

Result : Air - 0.03%

Water - 96.6% Soil - 0.03% Sediment - 3.3%

ld 68526-83-0

Date

Test substance : CAS No. 68526-83-0; Alcohols, C7-9-iso, C8-rich

Reliability : (2) valid with restrictions

This robust summary has a reliability rating of 2 because the data are

calculated and not measured.

Flag : Critical study for SIDS endpoint

04.04.2006 (19)

Type : fugacity model level I

Media : other: air - biota - sediment(s) - soil - water

Air : % (Fugacity Model Level I)

Water : % (Fugacity Model Level I)

Soil : % (Fugacity Model Level I)

Biota : % (Fugacity Model Level II/III)

Soil : % (Fugacity Model Level II/III)

Method : other: Calculation according Mackay, Level I

Year : 2003

Method : The EQC Level I is a steady state, equilibrium model that utilizes the input

of basic chemical properties including molecular weight, vapor pressure, and water solubility to calculate distribution within a standardized regional

environment.

Physicochemical input values for the model were calculated using the EPIWIN Estimation v 3.04 program. Measured input values were also used where available and obtained from the EPIWIN database. Distribution data from the equilibrium model provide basic information on the potential partitioning behavior of chemicals between selected environmental

compartments (i.e., air, water, soil, sediment, suspended sediment, biota).

Input values used:

Molecular mass = 130.23 g/mol Water solubility = 1379 mg/L Vapour pressure = 2.59 Pa

log Kow = 3.20

Melting point = -65 deg C

Result : Air - 2.0%

Water - 40.2 % Soil - 56.5% Sediment - 1.3% Suspended Sed - 0.04%

Biota - < 0.01%

Test substance : CAS No. 68526-83-0; Alcohols, C7-9-iso, C8-rich

Reliability : (2) valid with restrictions

This robust summary has a reliability rating of 2 because the data are

calculated and not measured.

Flag : Critical study for SIDS endpoint

05.04.2006 (19)

3.3.2 DISTRIBUTION

Media : other: Koc Method : other (calculation)

Year : 2003

Method : The soil adsorption coefficient (Koc) was calculated by PCKOCWIN ver.

1.66, using equations developed by Lyman (1980).

PCKOCWIN is a subroutine of EPIWIN. EPIWIN is used and advocated by the USEPA for chemical property estimation. Koc provides an indication of the extent to which a chemical partitions between solid and solution phases

ld 68526-83-0 **Date** 05.04.2006

in soil, or between water and sediment in aquatic ecosystems.

Result : Koc = 23.7

log Koc = 1.37

Test substance : CAS No. 68526-83-0; Alcohols, C7-9-iso, C8-rich

Reliability : (2) valid with restrictions

This robust summary has a reliability rating of 2 because the data are calculated and not measured. The value was calculated based on

chemical structure as modeled by EPIWIN.

Flag : Critical study for SIDS endpoint

05.04.2006 (18) (23)

3.4 MODE OF DEGRADATION IN ACTUAL USE

3.5 BIODEGRADATION

Type : aerobic

Inoculum : activated sludge, domestic

Contact time : 28 day(s)

Degradation : = 82 (±) % after 28 day(s)

Result
Deg. product

Method : OECD Guide-line 301 F "Ready Biodegradability: Manometric

Respirometry Test"

Year : 1997 **GLP** : yes

Test substance :

Remark: Test Type: Manometric Respirometry Test

Result: Test material was readily biodegradable. Half-life was reached by day 11.

By day 28, 82% degradation of the test material was observed. 10%

biodegradation was achieved on day 3.

By day 14, >60% biodegradation of positive control was observed, which met the guideline requirement. No excursions from the protocol were

noted.

Biodegradation was based on oxygen consumption and the theoretical oxygen demand of the test material as calculated using results of an

elemental analysis of the test material.

% Degradation* Mean % Degradation

 Sample
 (day 28)
 (day 28)

 Test Material
 84.7, 77.1, 84.0
 82.0

 Na Benzoate
 91.3, 81.3
 86.3

* replicate data

Test condition : Non acclimated activated sludge and test medium were combined prior to

test material addition. Test medium consisted of glass distilled water and mineral salts (Phosphate buffer, Ferric chloride, Magnesium sulfate,

Calcium chloride).

Test vessels were 1L glass flasks placed in a waterbath and electronically

monitored for oxygen consumption.

Test material was tested in triplicate, controls and blanks were tested in

duplicate.

Test material concentration was approximately 51 mg/L. Sodium benzoate

(positive control) concentration was 44mg/L.

Test temperature was 22 +/- 1 Deg C.

All test vessels were stirred constantly for 28 days using magnetic stir bars

and plates.

Test substance : CAS No. 68526-83-0; Alcohols, C7-9-iso, C8-rich

Id 68526-83-0

Date

Conclusion : Test substance is readily biodegradable.

Reliability : (1) valid without restriction

05.04.2006 (8)

3.6 BOD5, COD OR BOD5/COD RATIO

3.7 BIOACCUMULATION

3.8 ADDITIONAL REMARKS

4. Ecotoxicity Id 68526-83-0

Date

4.1 ACUTE/PROLONGED TOXICITY TO FISH

Type : flow through

Species: Pimephales promelas (Fish, fresh water)

Exposure period : 96 hour(s)
Unit : mg/l

LC50 : = 14 measured/nominal

Limit test : no Analytical monitoring : yes

Method : other: no data

Year : 1986 GLP : no data

Test substance

Method : Trimmed Spearman Karber Method

Result: 96 hour LC50 = 14.0 mg/L (95% CI 13.6 to 14.5) based upon measured

values

Analytical method used was Gas-Liquid Chromatography.

Measured	Fish Total
Conc. (mg/L)	Mortality (@96 hrs)*
Control	0
8.8	0
10.7	1
12.7	2
16.5	20
20.4	20

^{* 20} fish added at test initiation

Test condition: Treatment solutions were prepared by diluting a 275mg/L stock solution.

Nominal octanol treatment levels were 8.6, 10.8, 13.5, 16.9, 21.1mg/L, which measured 8.8, 10.7, 12.7, 16.5, and 20.4mg/L, respectively.

Control/dilution water was EPA Duluth laboratory water.

Twenty fish were tested per treatment. Treatment volume = 2.0L. Test parameters were as follows: temperature=25.3 Deg C; dissolved oxygen = 7.1mg/L; pH = 7.7; fish age = 28 days old; fish mean wt = 0.075g; fish

mean length = 16.5mm; fish loading = 0.75g/L/day.

Organism supplier was U.S. EPA Environmental Research Lab, Duluth,

MN, USA.

Test substance : CAS No. 68526-83-0; Alcohols, C7-9-iso, C8-rich

Reliability : (1) valid without restriction
Flag : Critical study for SIDS endpoint

05.04.2006 (13)

Type : other: calculated

Species : other: freshwater fish

Exposure period : 96 hour(s)
Unit : mg/l

LC50 : = 19.9 calculated

Method : other: ECOSAR Computer Model

Year : 2005 GLP :

GLP : Test substance :

Test condition: Log Kow (octanol/water partition coefficient) values and a chemical

structure are needed to calculate aquatic toxicity using the ECOSAR model. The Kow calculation is performed by KOWWIN based on an atom/fragment contribution method of Meylan and Howard (1), which is a

4. Ecotoxicity Id 68526-83-0

Date

subroutine in the EPIWIN computer model (2). KOWWIN also has a database of experimental Kow values (EXPKOW.DB).

The ECOSAR program was run using a C8 alcohol with a Kow of 2.73.

1. Meylan, W. and P. Howard. 1995. Atom/fragment contribution method for estimating octanol-water partition coefficients. J. Pharm. Sci. 84:83-92.

2. Meylan, M., SRC 1994-1999. KOWWIN is contained in the computer program EPIWIN. 1999. Estimation Program Interface for Windows, version 3.04. Syracuse Research Corporation, Syracuse, NY, USA.

Test substance : CAS No. 68526-83-0; Alcohols, C7-9-iso, C8-rich

Conclusion: Based on the calculated Kow value, the C8 alcohol is expected to have an

acute 96-hour LC50 of 19.9 mg/L and a Chronic Value of 2.9 mg/L.

Reliability : (2) valid with restrictions

The value was calculated based on chemical structure as modeled by EPIWIN. This robust summary has a reliability rating of 2 because the data

are calculated and not measured.

05.04.2006 (4)

4.2 ACUTE TOXICITY TO AQUATIC INVERTEBRATES

Type : static

Species : Daphnia magna (Crustacea)

 Exposure period
 : 48 hour(s)

 Unit
 : mg/l

 LC50
 : = 31.8

 Limit Test
 : no

 Analytical monitoring
 : no

Method : other: US EPA 660/3-75-009

Year : 1980 GLP : no data

Test substance :

Method : Spearman-Karber (Finney, D.J., 1971)

Remark: Test Type: Methods for Acute Toxicity Tests with Fish, Macroinvertebrates

and Amphibians

Result : 48-hour LC50 = 31.8 mg/L (Cl 26.5 - 38.2) as Total Carbon, based upon

nominal concentrations.

Nominal Conc. % Mortality @ 48 hr.

Control 0 10 mg/L 10 18 mg/L 20 32 mg/L 25 56 mg/L 95 100 mg/L 100 180 mg/L 100

Test condition

Individual treatments were prepared by adding varying amounts of test material directly to 250 mL of dilution water in glass beakers. Nominal test concentrations were 10, 18, 32, 56, 100 and 180 mg/L. Four replicates were prepared for each treatment and control. Five daphnids per replicate chamber. Test placed in a temperature-controlled waterbath at 20.5 to 21.0 Deg. C. The test was performed under static conditions.

Lighting was 16 hours light: 8 hours dark. Dissolved oxygen ranged from 8.6 to 9.6 mg/L during the study. The pH was ranged from 7.8 to 8.4 during the study. Dilution water hardness was 240 mg/L as CaCO3, alkalinity was 145 mg/L as CaCO3, and conductivity was 600 µmhos/cm. Organisms were supplied by in-house cultures. Age = <20 hours old.

4. Ecotoxicity Id 68526-83-0

Date

Test substance : CAS No. 68526-83-0; Alcohols, C7-9-iso, C8-rich

Conclusion: Test substance is considered to have moderate acute toxicity.

Reliability : (2) valid with restrictions

Analytical verification not performed, quality assurance unknown.

Flag : Critical study for SIDS endpoint

05.04.2006 (26)

Type

Species : Daphnia sp. (Crustacea)

Exposure period : 48 hour(s)
Unit : mg/l

EC50 : = 22.4 calculated

Method : other: ECOSAR Computer Model

Year : 2005

GLP :

Test condition : Log Kow (octanol/water partition coefficient) values and a chemical

structure are needed to calculate aquatic toxicity using the ECOSAR model. The Kow calculation is performed by KOWWIN based on an atom/fragment contribution method of Meylan and Howard (1), which is a subroutine in the EPIWIN computer model (2). KOWWIN also has a

database of experimental Kow values (EXPKOW.DB).

The ECOSAR program was run using a C8 alcohol with a Kow of 2.73.

1. Meylan, W. and P. Howard. 1995. Atom/fragment contribution method for estimating octanol-water partition coefficients. J. Pharm. Sci. 84:83-92.

2. Meylan, M., SRC 1994-1999. KOWWIN is contained in the computer program EPIWIN. 1999. Estimation Program Interface for Windows, version 3.04. Syracuse Research Corporation, Syracuse, NY, USA.

Test substance : CAS No. 68526-83-0; Alcohols, C7-9-iso, C8-rich

Conclusion : Based on the calculated Kow value, the C8 alcohol is expected to have an

acute 48-hour EC50 of 22.4 mg/L and a Chronic Value of 1.6 mg/L.

Reliability : (2) valid with restrictions

The value was calculated based on chemical structure as modeled by EPIWIN. This robust summary has a reliability rating of 2 because the data

are calculated and not measured.

05.04.2006 (4)

4.3 TOXICITY TO AQUATIC PLANTS E.G. ALGAE

Species: other algae: green alga

Endpoint : biomass
Exposure period : 96 hour(s)
Unit : mg/l

EC50 : = 14.6 calculated

Method : other: ECOSAR Computer Model

Year : 2005

GLP : Test substance :

Test condition : Log Kow (octanol/water partition coefficient) values and a chemical

structure are needed to calculate aquatic toxicity using the ECOSAR model. The Kow calculation is performed by KOWWIN based on an atom/fragment contribution method of Meylan and Howard (1), which is a subroutine in the EPIWIN computer model (2). KOWWIN also has a

database of experimental Kow values (EXPKOW.DB).

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Date 05.04.2006

The ECOSAR program was run using a C8 alcohol with a Kow of 2.73.

1. Meylan, W. and P. Howard. 1995. Atom/fragment contribution method for estimating octanol-water partition coefficients. J. Pharm. Sci. 84:83-92.

2. Meylan, M., SRC 1994-1999. KOWWIN is contained in the computer program EPIWIN. 1999. Estimation Program Interface for Windows, version 3.04. Syracuse Research Corporation, Syracuse, NY, USA.

Test substance: CAS No. 68526-83-0; Alcohols, C7-9-iso, C8-rich

Conclusion : Based on the calculated Kow value, the C8 alcohol is expected to have an

acute 96-hour EC50 of 14.6 mg/L and a Chronic Value of 2.2 mg/L.

Reliability : (2) valid with restrictions

The value was calculated based on chemical structure as modeled by EPIWIN. This robust summary has a reliability rating of 2 because the data

are calculated and not measured.

Flag : Critical study for SIDS endpoint

05.04.2006 (4)

4.4 TOXICITY TO MICROORGANISMS E.G. BACTERIA

4.5.1 CHRONIC TOXICITY TO FISH

Species: other: Freshwater Fish (calculated toxicity values are not species specific)

Endpoint : other: LC50
Exposure period : 30 day(s)
Unit : mg/l

ChV : = 2.9 calculated

Method : other: calculated

Year : 2005

GLP

Test substance :

Test condition: Log Kow (octanol/water partition coefficient) values and a chemical

structure are needed to calculate aquatic toxicity using the ECOSAR model. The Kow calculation is performed by KOWWIN based on an atom/fragment contribution method of Meylan and Howard, which is a subroutine in the EPIWIN computer model. KOWWIN also has a database

of experimental Kow values (EXPKOW.DB).

The ECOSAR program was run using a C8 alcohol with a Kow of 2.73.

Test substance: CAS No. 68526-83-0; Alcohols, C7-9-iso, C8-rich

Conclusion : Based on the calculated Kow value, the C8 alcohol is expected to have a

30-day Chronic Value of 2.9 mg/L

Reliability : (2) valid with restrictions

The value was calculated based on chemical structure as modeled by ECOSAR. This robust summary has a reliability rating of 2 because the

data are calculated and not measured.

Flag : Critical study for SIDS endpoint

05.04.2006 (4)

4.5.2 CHRONIC TOXICITY TO AQUATIC INVERTEBRATES

Species: Daphnia sp. (Crustacea)

Endpoint

Exposure period : 16 day(s)
Unit : mg/l

EC50 : = 1.6 calculated

4. Ecotoxicity Id 68526-83-0
Date 05.04.2006

Method : other: ECOSAR Computer Model

Year : 2005

GLP :

Test substance :

Test condition: Log Kow (octanol/water partition coefficient) values and a chemical

structure are needed to calculate aquatic toxicity using the ECOSAR model. The Kow calculation is performed by KOWWIN based on an atom/fragment contribution method of Meylan and Howard, which is a subroutine in the EPIWIN computer model. KOWWIN also has a database

of experimental Kow values (EXPKOW.DB).

The ECOSAR program was run using a C8 alcohol with a Kow of 2.73.

Test substance : CAS No. 68526-83-0; Alcohols, C7-9-iso, C8-rich

Conclusion : Based on the calculated Kow value, the C8 alcohol is expected yo have a

16-day EC50 of 1.6 mg/L.

Reliability : (2) valid with restrictions

The value was calculated based on chemical structure as modeled by EPIWIN. This robust summary has a reliability rating of 2 because the data

are calculated and not measured.

Flag : Critical study for SIDS endpoint

05.04.2006 (4)

4.6.1 TOXICITY TO SEDIMENT DWELLING ORGANISMS

4.6.2 TOXICITY TO TERRESTRIAL PLANTS

4.6.3 TOXICITY TO SOIL DWELLING ORGANISMS

Type :

Species: other: Earthworm

Endpoint

Exposure period : 14 day(s)
Unit : mg/kg soil dw
LC50 : = 477 calculated

Method : other: ECOSAR Computer Model

Year : 2005

GLP :

Test substance :

Test condition: Log Kow (octanol/water partition coefficient) values and a chemical

structure are needed to calculate acute terrestrial toxicity using the ECOSAR model. The Kow calculation is performed by KOWWIN based on an atom/fragment contribution method of Meylan and Howard, which is a subroutine in the EPIWIN computer model. KOWWIN also has a database

of experimental Kow values (EXPKOW.DB).

The ECOSAR program was run using a C8 alcohol with a Kow of 2.73.

Test substance : CAS No. 68526-83-0; Alcohols, C7-9-iso, C8-rich

Conclusion: Based on the calculated Kow value, the C8 alcohol is expected to have a

14-day EC50 of 477 mg/kg soil.

Reliability : (2) valid with restrictions

The value was calculated based on chemical structure as modeled by EPIWIN. This robust summary has a reliability rating of 2 because the data

are calculated and not measured.

Flag : Critical study for SIDS endpoint

05.04.2006 (4)

4. Ec	otoxicity	ld	68526-83-0	
0		Date	05.04.2006	
4.6.4	TOX. TO OTHER NON MAMM. TERR. SPECIES			
4.7	BIOLOGICAL EFFECTS MONITORING			
7.1	BIOLOGICAL LITEOTO MONTONINO			
4.8	BIOTRANSFORMATION AND KINETICS			
4.9	ADDITIONAL REMARKS			
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5. Toxicity Id 68526-83-0

Date 05.04.2006

5.0 TOXICOKINETICS, METABOLISM AND DISTRIBUTION

5.1.1 ACUTE ORAL TOXICITY

Type : LD50

Value : > 2000 mg/kg bw

Species: ratStrain: WistarSex: male/female

Number of animals : 10

Vehicle : other: None

Doses

Method : OECD Guide-line 401 "Acute Oral Toxicity"

Year : 1988 **GLP** : yes

Test substance :

Remark: Following dosing, the following symptoms were observed: sedation, ventral

body position in males, hunched posture, and ruffled fur. However, all animals had recovered within 6 days of dosing. At necropsy, no

macroscopic abnormalities were observed. Route of administration: Oral gavage Frequency of Treatment: Single dose Dose/Concentration Levels: 2000 mg/kg

Result : LD50 > 2000 mg/kg

Test condition: After being fasted for 12 to 18 hours, animals were administered a single

oral gavage dose of 2,000 mg/kg of the undiluted test article. Observations

were made four times on day 1; and daily for 14 days.

Test substance : CAS No. 68526-83-0; Alcohols, C7-9-iso, C8-rich

Conclusion: Under the conditions of this study, Alcohols, C7-9 branched has a low

order of toxicity.

Reliability : (1) valid without restriction
Flag : Critical study for SIDS endpoint

05.04.2006 (24)

5.1.2 ACUTE INHALATION TOXICITY

Type : LC50

Value : > 200 ppm Species : rat

Strain : Wistar
Sex : male
Number of animals : 10
Vehicle : other: NA

Doses

Exposure time : 6 hour(s)

Method : other

Year : 1960

GLP : no

Test substance :

Remark: Route of administration: Inhalation

Frequency of Treatment: Single 6 hour exposure Dose/Concentration Levels: Saturated Vapors

There were no deaths during the treatment period. There were no apparent signs of toxicity or alterations to behavior other than blinking in

5. Toxicity ld 68526-83-0
Date 05.04.2006

rats. No macroscopic abnormalities were observed at necropsy.

Result : LC50 > 200 ppm

Test condition : Rats were exposed to near-saturation levels (200 ppm) of vapors of

Alcohols, C7-9 branched in a 500 L stainless steel inhalation chamber for 6 hours. Vapor was generated by using two separate fritted disk glass bubblers, connected in parallel, each containing 200 ml of the test substance. Air flow through each bubbler was 18 l/m, and the total flow through the chamber was 36 l/m. Actual chamber concentration was not measured; theoretical chamber concentration was calculated to be 200 ppm. Animals were observed at one-hour intervals during the exposure. Animals were observed 24 hours following exposure and then necropsies

were performed.

Conclusion: Under the conditions of this study, Alcohols, C7-9 branched has a low

order of acute inhalation toxicity in rats.

Reliability : (2) valid with restrictions

No analysis of exposure atmosphere.

Flag : Critical study for SIDS endpoint

05.04.2006 (16)

Type : LC50
Value : > 200 ppm
Species : mouse
Strain : Swiss
Sex : male
Number of animals : 10
Vehicle : other: NA

Doses

Exposure time : 6 hour(s)

Method : other

Year : 1960

GLP : no

Test substance :

Remark: Route of administration: Inhalation

Frequency of Treatment: Single 6 hour exposure Dose/Concentration Levels: Saturated Vapors

There were no deaths during the treatment period. There were no apparent signs of toxicity or alterations to behavior other than blinking in mice. No macroscopic abnormalities were observed at necropsy.

Result : LC50 > 200 ppm

Test condition : Mice were exposed to near-saturation levels (200 ppm) of vapors of

Alcohols, C7-9 branched in a 500 L stainless steel inhalation chamber for 6 hours. Vapor was generated by using two separate fritted disk glass bubblers, connected in parallel, each containing 200 ml of the test substance. Air flow through each bubbler was 18 l/m, and the total flow through the chamber was 36 l/m. Actual chamber concentration was not measured; theoretical chamber concentration was calculated to be 200 ppm. Animals were observed at one-hour intervals during the exposure. Animals were observed 24 hours following exposure and then necropsies

were performed.

Test substance : CAS No. 68526-83-0; Alcohols, C7-9-iso, C8-rich

Conclusion: Under the conditions of this study, Alcohols, C7-9 branched has a low

order of acute inhalation toxicity in mice.

Reliability : (2) valid with restrictions

No analysis of exposure atmosphere.

Flag : Critical study for SIDS endpoint

05.04.2006 (16)

Type : LC50
Value : > 200 ppm
Species : guinea pig

Strain : other: English short hair

Date

Sex : male : 10 Vehicle : other: NA

Doses

Exposure time : 6 hour(s)

Method : other

Year : 1960

GLP : no

Test substance :

Remark: Route of administration: Inhalation

Frequency of Treatment: Single 6 hour exposure Dose/Concentration Levels: Saturated Vapors

There were no deaths during the treatment period. There were no apparent signs of toxicity or alterations to behavior. No macroscopic

abnormalities were observed at necropsy.

Result : LC50 > 200 ppm

Test condition: Guinea pigs were exposed to near-saturation levels (200 ppm) of vapors of

Alcohols, C7-9 branched in a 500 L stainless steel inhalation chamber for 6 hours. Vapor was generated by using two separate fritted disk glass bubblers, connected in parallel, each containing 200 ml of the test substance. Air flow through each bubbler was 18 l/m, and the total flow through the chamber was 36 l/m. Actual chamber concentration was not measured; theoretical chamber concentration was calculated to be 200 ppm. Animals were observed at one-hour intervals during the exposure. Animals were observed 24 hours following exposure and then necropsies

were performed.

Test substance: CAS No. 68526-83-0; Alcohols, C7-9-iso, C8-rich

Conclusion: Under the conditions of this study, Alcohols, C7-9 branched has a low

order of acute inhalation toxicity in guinea pigs.

Reliability : (2) valid with restrictions

No analysis of exposure atmosphere.

Flag : Critical study for SIDS endpoint

05.04.2006 (16)

05.04.2006

5.1.3 ACUTE DERMAL TOXICITY

Type : LD50

Value : > 2623 mg/kg bw

Species: rabbit

Strain : other: Albino
Sex : male/female

Number of animals : 8 Vehicle :

Doses :

Method: otherYear: 1960GLP: noTest substance:

Remark: Animals in the 83, 262, and 820 mg/kg dose groups exhibited normal

appearance and behavior throughout the study. At the highest dose (2623 mg/kg), animals exhibited labored respiration and were inactive. One animal in the high dose group died within 24 hours. The remaining animals in this dose group returned to normal appearance and behavior 2 days

after the treatment.

Route of administration: Dermal; with occlusive binding

Date

Frequency of treatment: Single 24 hour exposure

Dose/Concentration Levels: 83, 262, 820, 2623 mg/kg (undiluted)

Control group and treatment: None

Result : Dermal LD50 > 2623 mg/kg

Test condition : The test substance was applied dermally to rabbits (4/sex/dose) under

occlusive binding and removed after 24 hours. The results were converted to weight units by means of the specific gravity. Animals were observed 1, 4, and 24 hours after initial application of Alcohols, C7-9 branched and once daily for the next 7 days. At the termination of the study, survivors

were weighed and gross necropsies were performed.

Test substance : CAS No. 68526-83-0; Alcohols, C7-9-iso, C8-rich

Conclusion: Alcohols, C7-9 branched showed a low order of acute dermal toxicity under

the conditions of this study.

: (1) valid without restriction

Reliability : (1) valid without restriction Flag : Critical study for SIDS endpoint

05.04.2006 (16)

5.1.4 ACUTE TOXICITY, OTHER ROUTES

5.2.1 SKIN IRRITATION

Species : rabbit

Concentration

Exposure : Occlusive
Exposure time : 4 hour(s)
Number of animals : 16

Vehicle :

ellicie .

PDII : 3.08

Result : moderately irritating

Classification

Method: other: similar to OECD TG 404

Year : 1960

GLP

:

Test substance :

Remark: A single 4-hour application of Alcohols, C7-9-iso, C8 rich to intact rabbit

skin produced moderate irritation. Skin irritation consisted of slight to moderate erythema, moderate edema, atonia, and desquamation persisted

to termination on day 14.

The single 4-hour application resulted in a primary irritation index of 3.08. Mean scores at 24, 48, and 72-hours were 1.83, 1.83, and 2.0 for erythema

and 0.83, 1.33, and 1.5 for edema.

Test substance : CAS No. 68526-83-0; Alcohols, C7-9-iso, C8-rich

Conclusion: Under the conditions of the study, Alcohols, C7-9-iso, C8 rich produced

moderate irritation to rabbit skin.

Reliability : (2) valid with restrictions

Pre-GLP

Flag : Critical study for SIDS endpoint

05.04.2006 (15) (25)

5.2.2 EYE IRRITATION

Species : rabbit
Concentration : .1 undiluted

Dose : Exposure time :

5. Toxicity ld 68526-83-0
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Comment : Number of animals : 6

Number of animals : Vehicle :

Result : highly irritating

Classification

Method : other: similar to OECD TG 405

Year :

Test substance

Result : Median Scores were:

24 hr - 26 72 hr - 18 7 days - 0

Iso-octyl alcohol produced persistent, widespread corneal opacity which

generally cleared within 7 days.

Test condition : A single application of 0.1 ml of undiluted alcohol was made into the

conjunctival sac of the left eye of each rabbit. The untreated eye served as the control. The treated eye was held closed for 1 second, and eyes were not washed. Observations for irritancy and possible systemic toxicity were made at 1, 4, and 24 hours and at 2, 3, 4, and 7 days. At each interval the treated and control eyes were examined and scored for ocular reactions

according to the Draize Standard Eye Irritation Grading Scale.

Test substance : CAS No. 68526-83-0; Alcohols, C7-9-iso, C8-rich

Reliability : (2) valid with restrictions

Pre GLP

Flag : Critical study for SIDS endpoint

05.04.2006 (25)

Species : rabbit Concentration : .1 undiluted

Dose

Exposure time

Comment

Number of animals : 6

Vehicle

Result : highly irritating

Classification

Method : other: similar to OECD TG 405

Year :

Test substance :

Result: Median Scores were:

24 hr - 19 72 hr - 20 7 days - 0

2-ethylhexyl alcohol produced persistent, widespread corneal opacity which

generally cleared within 7 days.

Test condition : A single application of 0.1 ml of undiluted alcohol was made into the

conjunctival sac of the left eye of each rabbit. The untreated eye served as the control. The treated eye was held closed for 1 second, and eyes were not washed. Observations for irritancy and possible systemic toxicity were made at 1, 4, and 24 hours and at 2, 3, 4, and 7 days. At each interval the treated and control eyes were examined and scored for ocular reactions

according to the Draize Standard Eye Irritation Grading Scale.

Test substance : Analogue substance: CAS No. 104-76-7; 2-ethylhexan-1-ol

Reliability : (2) valid with restrictions

Date

Pre-GLP

05.04.2006 (25)

5.3 SENSITIZATION

5.4 REPEATED DOSE TOXICITY

Type : Species : rat

Sex: male/femaleStrain: Fischer 344Route of admin.: gavageExposure period: 3 months

Frequency of treatm. : daily (5 days /week)

Post exposure period : none

Doses : 25, 125, 250, 500 mg/kg (in aqueous emulsion)

Control group : yes, concurrent vehicle

NOAEL : = 125 mg/kg LOAEL : = 250 mg/kg

Method :

Year : 1990 GLP : yes Test substance :

Remark : Vehicle: bidistilled water containing 5 ug/ml Cremophor EL.

Concurrently to the main study a limited study with the same dosing regimen using 3 animals/sex/dose was performed; at the end of the treatment period all animals were sacrificed and samples of liver tissues and bone marrow were prepared for electron microscopy investigations; liver homogenates

were prepared for clinicochemical examinations.

Result: Result:

Limited study:

The oral administration of 2-ethylhexanol over a period of 3 months led to an impairment of feed consumption and body weight gain in the male and female animals of the 500 mg/kg dose group. The cyanide-insensitive palmitoyl CoA-oxidation in the liver of male and female rats was strongly induced in the 500 mg/kg dose group and less pronounced in the 250 mg/kg dose group, probably due to a proliferation of

peroxisomes. No substance-related findings were observed in

the other dose groups.

NOEL: 125 mg/kg bodyweight LOEL: 250 mg/kg bodyweight

Main study:

Per dose group 10 male and 10 female rats were treated. After the 3-month administration of 2-ethylhexanol toxic effects occurred in male and female rats of the 500 and 250 mg/kg dose groups. Typical findings were:

25 and 125 mg/kg groups: no substance-related findings

250 mg/kg group: increased relative liver weights in both sexes; increased relative stomach weights in females;

5. Toxicity ld 68526-83-0

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decrease in alkaline phosphatase and glucose in males; decrease in alanine-aminotransferase in female rats.

500 mg/kg group: increased relative liver and stomach weights in both sexes; increased relative liver weights in the animals of both sexes; increased absolute stomach

weights in female rats; decrease in

alanine-aminotransferase, glucose and cholesterol in both sexes; decrease in alkaline phosphatase in males; single or multiple slightly elevated foci in the mucosa of the forestomach of male and female rats; focal or multifocal acantosis in the mucosa of the forestomach of one male and

five female animals.

Test substance: Analogue substance: CAS No. 104-76-7; 2-ethylhexan-1-ol

Reliability : (1) valid without restriction
Flag : Critical study for SIDS endpoint

05.04.2006 (2)

Type : rat
Species : rat
Sex : male
Strain : Wistar

Route of admin. : other: oral gavage

Exposure period : 14 days **Frequency of treatm**. : once/day

Post exposure period

Doses

Control group

NOAEL : = 130 mg/kg bw

Method: other: NAYear: 1984GLP: no data

Test substance :

Remark: Iso-octanol did not significantly change body weight gain, liver to body

weight ratio, or testis to body weight ratio when compared to vehicle controls. Iso-octanol did not induce any changes in glycogen vacuolation or fat vacuolation. The activity of peroxisome-associated enzymes and levels of cholesterol and triglyceride were not significantly different between animals treated with iso-octanol and vehicle controls. No hyperlipidemia

was observed.

No. of animals/sex/dose: 5/treatment, 10/control; 1mmol/kg/day of iso-

octanol (130 mg/kg/day)

Vehicle: Polyethylene glycol 300

Statistics: Mean values compared to controls by Student's t-test.

Result: NOAEL = 130 mg/kg/day

Test condition : After acclimation for 1 week, five animals received 1mmol/kg/day (130

mg/kg/day) of the test substance by oral gavage and ten animals received only the vehicle, PEG 300, daily for 14 days. Animals were sacrificed after 14 days by halothane overdose and blood was withdrawn by cardiac puncture and analyzed for plasma cholesterol and triglycerides. The liver was removed for histopathological analysis, analysis of catalase, and CN-

insensitive palmitoyl CoA oxidation. Testicular weight was also

determined. **Test substance**: Iso-octanol

CAS No. 68526-83-0; Alcohols, C7-9-iso, C8-rich

Conclusion: Under the conditions of this study, iso-octanol had a low order of sub-acute

toxicity in male rats for the endpoints studied.

Reliability : (2) valid with restrictions

Not a guideline study.

05.04.2006 (3)

Date

Туре

Species: rabbitSex: male/femaleStrain: other: AlbinoRoute of admin.: dermalExposure period: 10 days

Frequency of treatm. : once/day
Post exposure period :

Doses : 0.4 g/kg and 2.0 g/kg

Control group : other: Isopropyl alcohol, 2/sex

NOAEL : = 2000 mg/kg

Method: otherYear: 1961GLP: no

Result

Test substance: as prescribed by 1.1 - 1.4

Method : Not specified

Remark: Animals in all exposure groups displayed normal appearance and behavior

throughout the study. Although a slight decrease in body weight was observed initially, animals regained weight by the end of the study. Repeat application of the control substance, isopropyl alcohol produced slight irritation characterized by slight to moderate erythema, atonia, and desquamation. Repeated application of Alcohols, C7-9 branched resulted in moderate to severe irritation. Fissuring and coriaceous skin were also observed at both the low and high dose levels. Necrosis was observed in the high dose animals as well. Clinical studies did not indicate any other signs of toxicity. There was a general increase in the hematocrit and

erythrocyte values at the end of the study.

Duration of test: 12 days

Number of animals: 8 rabbits (2/sex/dose) NOAEL for systemic toxicity = 2.0 g/kg

Test condition: Undiluted control and test materials were applied to intact skin of the

animals (2/sex/dose). Materials were applied once daily for a total of ten applications with a one-day rest period between the third and fourth and eighth and ninth applications. The exposed skin area of each animal was approximately 10% of the total body surface at the 0.4 g/kg dosage level and approximately 40% of the total body surface at the 2.0 g/kg dosage level. After the first application, exposed skin was covered by rubber dental damming. In subsequent applications, loose gauze and adhesive tape were used to cover the exposed area since the authors felt that the damming itself may have induced some irritation. Each exposure period lasted approximately 18-24 hours. Animals were observed daily throughout the study and body weights were recorded prior to each exposure and at study termination. Clinical hematology and urinalysis were performed at the beginning of the study and 24 hours after the final application of test material. Animals were sacrificed 48 hours after the tenth application, samples of brain, thyroid, lung, heart, liver, kidneys,

adrenals, skin, and bone marrow were preserved.

Conclusion: Under the conditions of this study, Alcohols, C7-9 branched can produce

moderate skin irritation following repeated dermal exposures. However, the test material did not produce any evidence of systemic toxicity under

the conditions of this study.

Reliability : (2) valid with restrictions

2 - Valid with restrictions

28.11.2001 (5)

5.5 GENETIC TOXICITY 'IN VITRO'

Type : Ames test

System of testing : S. typhimurium, E. coli

Date

Test concentration : 1, 5, 10, 50, 100, 500, and 1000 ug/plate.

Cycotoxic concentr. : Metabolic activation :

Result : negative
Method : other

Method: otherYear: 1985GLP: no data

Test substance :

Method: Samples were run in duplicate. No further details provided.

Remark: Strain: Salmonella typhimurium /TA98; TA100; TA1535; TA1537; TA1538;

E. coli WP2uvrA

Metabolic Activation: S9 mixture

Result: In all of the strains tested, there was no evidence of mutagenicity of 2-

ethyl-1-hexanol in the presence or absence of metabolic activation. The number of revertant colonies per plate did not vary significantly between

the water, DMSO, or 2-ethyl-1-hexanol samples.

Test condition : 2-Ethyl-1-hexanol (98% pure) was dissolved in DMSO at appropriate

concentrations. 0.1ml of this mixture was added to 0.1 ml of bacteria and 0.5 ml of either S9 mix (Polychlorinated biphenyl-induced rat liver S9 mixture) or phosphate-buffered saline. Following a 20-minute pre-incubation, the mixtures were combined with agar and incubated for 48 hours. Colonies were scored with an automatic counter. All tests were performed in duplicate. 2-(2-Furyl)-3-(5-nitro-2-furyl)-acrylamide (AF-2), N-ethyl-N'-nitro-N-nitrosoguanidine (ENNG), 9-aminoacridine (9AC), 4-

nitroquinoline-1-oxide (4NQO), benzo(a)pyrene (B(a)P), 2-

aminoanthracene (2AA), and 2-nitrofluorene (2NF) were used as positive controls. In addition, water and DMSO were used as vehicle controls.

Test substance: 2-Ethyl-1-hexanol (CAS No. 104-76-7)

Conclusion: 2-Ethyl-1-hexanol is not mutagenic in bacteria under the conditions of this

study.

Reliability : (2) valid with restrictions

(Similar to OECD 471)

05.04.2006 (14)

5.6 GENETIC TOXICITY 'IN VIVO'

Type : Cytogenetic assay

Species : rat
Sex : male
Strain : Fischer 344
Route of admin. : gavage
Exposure period : 5 days

Doses : 0.02, 0.07, 0.21 ml/kg day

Result

Method : other: as described by author

Year : 1981 GLP : yes Test substance :

Result: Groups of 5 male F-344 rats were treated with

2-ethylhexanol. Of the 50 metaphase bone marrow cells examined from each animal, no significant increase in

chromatid and chromosome breaks or structural rearrangements was noted. In addition, the mitotic index was unaffected by 2-ethylhexanol. At the dose levels tested 2-ethylhexanol did not induce detectable chromosomal aberrations after oral

administration.

Test substance: Analogue substance: CAS No. 104-76-7; 2-ethylhexan-1-ol

Reliability : (1) valid without restriction

Date

Flag : Critical study for SIDS endpoint

05.04.2006 (6)

Type : Micronucleus assay

Species : mouse
Sex : male/female
Strain : B6C3F1
Route of admin. : i.p.

Exposure period: single administration

Doses : 456 mg/kg

Result

Method : other: as described by author

Year : 1982 GLP : yes Test substance :

Result: B6C3F1 mice were administered 2-ethylhexanol at a dose which

was equal to 80% of the LD50/7days. Bone marrow was harvested 30 hrs post application and 1000 PCE/animal were scored for the presence of micronuclei. 2-Ethylhexanol was not considered to be clastogenic under the conditions of

this assay.

Test substance: CAS No. 104-76-7; 2-ethylhexan-1-ol

Reliability : (1) valid without restriction
Flag : Critical study for SIDS endpoint

05.04.2006 (6)

5.7 CARCINOGENICITY

Species : rat

Sex : male/female
Strain : Fischer 344
Route of admin. : gavage
Exposure period : 24 months
Frequency of treatm. : 5 days/week

Post exposure period : none

Doses : 0, 50, 150, 500 mg/kg

Result

Control group : yes, concurrent vehicle

Method : other: according to EPA-TSCA guidelines

Year :

GLP : yes Test substance :

Remark: Vehicle: aqueous 0.005% Cremophor EL.

In addition to the vehicle control groups, 50 rats/sex were

dosed with water;

Concurrently to the main study, a satellite study was

performed. 2-Ethylhexanol was administered by gavage (same vehicle as in the main study) at a dose of 500 mg/kg. One group of rats (10/sex; interim sacrifice group) was treated with 2-ethylhexanol 5 days/week over a period of 18 months and then sacrificed. Another group (50 rats/sex; recovery group) was treated with the same dose for 18 months and

thereafter with vehicle only for 6 months and then sacrificed. As a control group for the interim sacrifice

group 10 male and 10 female rats received the vehicle for 18 months. Control data for the recovery group were adopted from the vehicle control group and the top dose group of the

parallel main study.

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Result: Result:

Satellite study:

In the interim sacrifice group a reduced body weight gain (both sexes) and clinical symptoms like poor general condition, labored breathing and "genital region smeared with urine" (females only) was observed. A slightly increased mortality in females was observed. The relative weight of testes and brain (males), stomach (females), liver and kidney (both sexes) were increased. In the recovery group similar effects were observed. After termination of treatment the body weight gain increased indicating a recovery effect. 18 males and 17 females died prematurely and one each had a focal hyperplasia in the forestomach. In the glandular stomach erosions were seen in 4 males and females each. Glandular cysts occurred in 6 males and 8 females.

Main study:

2-Ethylhexanol was not oncogenic in the rat under the conditions of this assay. In both sexes the sum of primary tumors, the sum of benign tumors and the sum of malignant tumors was lower in the top dose group than in either the vehicle control or the water central groups.

vehicle control or the water control groups.

Test substance: CAS No. 104-76-7; 2-ethylhexan-1-ol

Reliability : (1) valid without restriction
Flag : Critical study for SIDS endpoint

05.04.2006 (6)

Species: mouseSex: male/femaleStrain: B6C3F1Route of admin.: gavageExposure period: 18 monthsFrequency of treatm.: 5 days/week

Post exposure period : none

Doses : 0, 50, 200, 750 mg/kg

Result

Control group : yes, concurrent vehicle

Method : other: according to EPA-TSCA guidelines

Year :

GLP : yes Test substance :

Remark: Vehicle: aqueous 0.005% Cremophor EL

In addition to the vehicle control groups, 50 mice/sex were

dosed with water.

Concurrently to the main study, a satellite study was performed: 2-Ethylhexanol was administered by gavage (same vehicle as in the main study) at a dose of 750 mg/kg. One group of mice (10/sex; interim sacrifice group) was treated with 2-ethylhexanol 5 days/week over a period of 13 months and then sacrificed. Another group (50 mice/sex; recovery group) was treated with the same dose for 13 months and thereafter with vehicle only for 5 months and then sacrificed. As a control group for the interim sacrifice

group 10 male and 10 female mice received the vehicle for 13 months. Control data for the recovery group were adopted from the vehicle control group and the top dose group of the

parallel main study.

Date

Result: Result:

Satellite study:

The administration of 2-ethylhexanol to male and female mice for 13 months at a dose of 750 mg/kg caused increased mortality, reduced feed consumption and body weight gain in both sexes. The body weight gain of male and female animals of the recovery group were reduced as long as the animals were treated. After termination of treatment the male animals gained weight and reached nearly the values of the control group, indicating a recovery effect. No such effect was seen in females.

Pathology revealed some statistically significant changes in organ weights and masses of foci in liver and stomach. One male and one female animal of the recovery group which died premarurely had a focal hyperplasia in the forestomach.

Main study:

2-Ethylhexanol was not oncogenic in the mouse under the conditions of this assay. A slight increase in the incidence of hepatocellular carcinoma in the females of the high dose group was statistically significant if compared to the control group dosed with the emulsion vehicle, but not when compared to the control group dosed with water. This difference was regarded as incidental and not biologically relevant, based upon comparison with published data, and because of the lack of metastases. No statistically significant increase in tumor incidence occured in male mice.

Test substance: Analogue substance: CAS No. 104-76-7; 2-ethylhexan-1-ol

Reliability : (1) valid without restriction
Flag : Critical study for SIDS endpoint

05.04.2006 (6)

5.8.1 TOXICITY TO FERTILITY

5.8.2 DEVELOPMENTAL TOXICITY/TERATOGENICITY

Species : rat Sex : female

Strain : Sprague-Dawley Route of admin. : inhalation

Exposure period: Gestation days 1-19

Frequency of treatm. : 7 hrs/day

Duration of test : 19 days

Doses : 400 mg/m3

Control group : other: 15 sham-exposed rats

NOAEL maternal tox. : $>= 400 \text{ mg/m}^3$

Method : other Year : 1989 GLP : no data

Test substance

Method : MANOVA, ANOVA, Kruskal-Wallis test

Remark: No treatment-related effects were observed in dams. There were no

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significant differences in maternal weight gain, feed consumption, and water intake between the control and treated groups. In addition, no signs of fetal toxicity were observed. The number of corpora lutea and resorptions, the sex ratio, and fetal weights were not significantly different between the control and treated groups.

No. of animals/sex/dose: 15

Result : Maternal and Developmental NOAEL >= 400 mg/m3

Test condition: Throughout the study, all animals were housed under standard

environmental conditions and allowed free access to food and water except when the pregnant females were in the exposure chamber. Following mating, sperm-positive females were placed in cages and weighed. Dams were weighed daily for the first week of exposure and weekly thereafter. Exposures were conducted in Hinners-type chambers. The purity of the test substance was ³ 99% as measured by gas chromatography. A constant flow of the test substance was mixed with a known volume of heat compressed air, resulting in instantaneous vaporization of the test substance, which then flowed into the chamber. The concentration of the test substance was monitored continuously and recorded every hour. Calibration checks were completed daily. Exposure concentrations were verified on a weekly basis using a secondary method of analysis. The highest concentration of vapor that could be generated was 3500 mg/m3. Dams were exposed from days 1-19 of gestation. On day 20, dams were sacrificed by CO2 asphyxiation, and the uterus and ovaries were removed and examined for corpa lutea, implantations, resorption sites, and live fetuses. Fetuses were removed and examined for external malformations, sexed, weighed, and examined for visceral or skeletal defects.

Test substance: 1-Octanol

Conclusion: Under the conditions of this study, exposure of pregnant rats to saturated

vapors of 1-Octanol does not induce maternal or fetal toxicity.

Reliability : (2) valid with restrictions

Flag : Critical study for SIDS endpoint

05.04.2006 (1)

Species : rat Sex : female

Strain : Sprague-Dawley Route of admin. : other: oral gavage

Exposure period: GD 6-15

Frequency of treatm.

Duration of test :

Doses : 100, 500, and 1000 mg/kg/day **Control group** : other: Carrier only - corn oil

NOAEL maternal tox. : = 500 - mg/kg bw

Method : OECD Guide-line 414 "Teratogenicity"

Year : 1994 GLP : yes Test substance :

Method : Nested analysis of covariance, Least Significant Difference (LSD), Chi-

square, Fisher Exact test, Armitage's test.

Remark: No. of animals/dose: 25

One animal in the high dose group was euthanized in moribund condition on GD 9. The animal had extreme abdominal staining just prior to death, but there were no significant findings at postmortem examination and the cause of morbidity was therefore not established. Adverse clinical signs were observed in 8 of the 24 dams in the high dose group. These signs included emaciation, decreased food consumption, abdominal/anogenital staining, rales, hypoactivity, and little or no stool. The symptoms were transient and generally were not observed following cessation of dosing. The remaining dams in the high dose group had incidental findings such as alopecia, but otherwise appeared normal throughout the study. There were no observable abnormalities in dams of the middle and low dose groups

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throughout the gestational period. In the high dose group, statistically significant decreased body weight gain and food consumption were observed from GD 6-9 and GD6-15 compared to controls. However, these effects subsided after cessation of treatment and body weight and food consumption for the overall gestational period (GD 6-21) were not significantly different between the high dose group and controls. There were no maternal findings at necropsy that were judged to be the result of treatment with Alcohols, C7-9 branched. For the most part, uterine implantation parameters were equivalent between the treated and control groups.

There were slight differences between the high dose group and the control group in the number of post-implantation losses and resorptions, however these differences were not statistically significant and were deemed to be due to the poor health of the dams.

Mean fetal body weight was equivalent between treated and control fetuses of both sexes. Three low dose, two mid dose, and one high dose fetus were stunted. There were no statistically significant differences in mean skeletal ossification sites and in total or individual external, visceral, or skeletal malformations between control and treated groups. There were statistically significant increases in total fetuses with skeletal variations and in the incidence of hypoplastic skull bones in the high dose group when compared to controls. These findings were slightly higher than the historical control range of the lab and were not observed with litter-based analysis. Statistically significant increases in the number of lumbar ribs were observed in the middle and high dose groups. However, due to the lack of embryotoxicity observed in this study, these findings were attributed to maternal toxicity observed during treatment.

Result : Maternal NOAEL = 500 mg/kg/day

Fetal NOAEL = 1000 mg/kg/day

Test condition: Mated females were assigned to dose groups of 100, 500, and 1000

mg/kg/day or to a corn oil-only group (25/dose). The test substance was administered in volumes of 5 ml/kg. Body weight and food consumption measurements were made on GD 0, 6, 9, 12, 15, 18, and 21. The animals were examined for viability twice daily during the treatment period and once daily thereafter. Clinical observations were made daily during gestation. On GD 21, animals were sacrificed and cesarean sections and necropsies were performed. Uterine weights with ovaries attached were recorded, uterine contents were examined, and implantation data were recorded. All live fetuses were weighed, sexed externally, and examined externally for gross malformations. Approximately one-half of the fetuses were prepared for examination of abnormalities in the head and the other half were

preserved for examination of skeletal abnormalities.

Test substance : CAS No. 68526-83-0; Alcohols, C7-9-iso, C8-rich

Conclusion: Under the conditions of this study, Alcohols, C7-9 branched induces

maternal toxicity at concentrations that are not embryotoxic.

Reliability : (1) valid without restriction
Flag : Critical study for SIDS endpoint

05.04.2006 (9)

5.8.3 TOXICITY TO REPRODUCTION, OTHER STUDIES

5.9 SPECIFIC INVESTIGATIONS

5.10 EXPOSURE EXPERIENCE

5. Toxicity		68526-83-0 05.04.2006
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6. Analyt. Meth. for Detection and Identification	68526-83-0 05.04.2006	
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7. Eff	. Against Target Org. and Into	ended Uses	ld Date	68526-83-0	
7.1	FUNCTION				
7.2	EFFECTS ON ORGANISMS TO BE CONT	TROLLED			
7.3	ORGANISMS TO BE PROTECTED				
7.4	USER				
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9. References Id 68526-83-0 Date 05.04.2006

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10. Summary and Evaluation	ld	68526-83-0
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10.1 END POINT SUMMARY		
10.2 HAZARD SUMMARY		
10.3 RISK ASSESSMENT		
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IUCLID

Data Set

 Existing Chemical
 : ID: 70914-20-4

 Memo
 : HPV Chemical

 CAS No.
 : 70914-20-4

TSCA Name : Alcohols, C6-8-branched

Molecular Formula : Unspecified

Producer related part

Company: ExxonMobil Biomedical Sciences Inc.

Creation date : 29.11.2001

Substance related part

Company: ExxonMobil Biomedical Sciences Inc.

Creation date : 29.11.2001

Status

Memo : Prepared for EMCC - US HPV

Printing date : 04.04.2006

Revision date :

Date of last update : 04.04.2006

Number of pages : 35

Chapter (profile) : Chapter: 1, 2, 3, 4, 5, 6, 7, 8, 10 Reliability (profile) : Reliability: without reliability, 1, 2, 3, 4

Flags (profile) : Flags: without flag, confidential, non confidential, WGK (DE), TA-Luft (DE),

Material Safety Dataset, Risk Assessment, Directive 67/548/EEC, SIDS

1.0.1 APPLICANT AND COMPANY INFORMATION

1.0.2 LOCATION OF PRODUCTION SITE, IMPORTER OR FORMULATOR

1.0.3 IDENTITY OF RECIPIENTS

1.0.4 DETAILS ON CATEGORY/TEMPLATE

Comment: see free text

Remark: The Alkyl Alcohols C6 to C13 Category is a family of saturated alcohols

that are produced from olefins by the hydroformylation or "oxo" process. Hydroformylation is the reaction of an olefin with carbon monoxide and hydrogen to produce an aldehyde, and its subsequent hydrogenation to the

alcohol.

The number of carbon atoms in the category members ranges from 6 to 13. Category members contain predominantly branched alkyl groups. Each substance consists of an isomeric mixture of saturated primary alcohols of high purity, and the following basic structure: CH3-R-CH2-OH, where R is a branched isomeric structure.

The justification for the Alkyl Alcohols C6 to C13 Category is that the members have:

- · similar chemical structures,
- · similar physico-chemical properties,
- · comparable environmental fate,
- · the same mode of action.

In general, aliphatic alcohol toxicity occurs by non-polar narcosis (Lipnick et al., 1985). The mode of action is disruption of biological membrane function (van Wezel and Opperhuizen, 1995). Metabolic pathways, through which alcohols are metabolized, are likely to include similar reactions for all category members and result in structurally similar metabolites (carboxylic acids).

The data demonstrate that the category is valid for a screening-level hazard assessment for the Category and its members. One can assess the untested endpoints by extrapolation between and among the category members.

20.03.2006

1.1.0 SUBSTANCE IDENTIFICATION

IUPAC Name : Smiles Code :

Molecular formula : C7H16O Molecular weight : 116.21

Petrol class

Flag : Critical study for SIDS endpoint

04.04.2006

1. General Information

ld 70914-20-4

Date

1.1.1 GENERAL SUBSTANCE INFORMATION

Purity type

Substance type : organic Physical status : liquid

Purity : Colour : Odour :

Remark: CAS Registry Number, Name, and General Structure for Members of the

Alkyl Alcohols C6 to C13 Category and Analogue Substances:

CAS RN: 70914-20-4

IUPAC Name: Alcohols, C6-8 branched

R Length (C number): C7

Structure of R: Branched, primarily methyl branched (isomeric structures)

Category Member: Yes

04.04.2006

1.1.2 SPECTRA

1.2 SYNONYMS AND TRADENAMES

Exxal 7

20.09.2004

isoheptanol

20.09.2004

isoheptyl alcohol

20.09.2004

1.3 IMPURITIES

Purity : typical for marketed substance

CAS-No : 70914-20-4

EC-No

EINECS-Name : Alcohols, C6-8 branched

Molecular formula : C7H16O Value : = 98 % w/w

Remark: Commercial product typically consists of methyl-1-hexanols.

04.04.2006

1.4 ADDITIVES

Purity type : typical for marketed substance

CAS-No :
EC-No :
EINECS-Name :
Molecular formula :

1. General Information

ld 70914-20-4

Date

Value

Function of additive

:

Remark

: No additives present.

04.04.2006

- 1.5 TOTAL QUANTITY
- 1.6.1 LABELLING
- 1.6.2 CLASSIFICATION
- 1.6.3 PACKAGING
- 1.7 USE PATTERN

Type of use : industrial

Category : Chemical industry: used in synthesis

Remark: Primarily used as a chemical intermediate in the production of ester

compounds, such as phthalate plasticizers and esters of carboxylic acid. Also used as solvents or solubilizers in paint and printing inks, as a component in textile auxiliaries and pesticides, for hormone extraction and

in the surfactant field as foam boosters or antifrothing agents.

04.04.2006

- 1.7.1 DETAILED USE PATTERN
- 1.7.2 METHODS OF MANUFACTURE
- 1.8 REGULATORY MEASURES
- 1.8.1 OCCUPATIONAL EXPOSURE LIMIT VALUES
- 1.8.2 ACCEPTABLE RESIDUES LEVELS
- 1.8.3 WATER POLLUTION
- 1.8.4 MAJOR ACCIDENT HAZARDS

Id 70914-20-4 1. General Information **Date** 04.04.2006 1.8.5 AIR POLLUTION 1.8.6 LISTINGS E.G. CHEMICAL INVENTORIES 1.9.1 DEGRADATION/TRANSFORMATION PRODUCTS 1.9.2 COMPONENTS 1.10 SOURCE OF EXPOSURE 1.11 ADDITIONAL REMARKS 1.12 LAST LITERATURE SEARCH 1.13 REVIEWS

5/35

ld 70914-20-4

Date

2.1 MELTING POINT

Value : = -37.2 °C

Sublimation

Method : other: calculated

Year :

GLP : no Test substance : other TS

Method : Calculated values using MPBPWIN version 1.40, a subroutine of the

computer program EPIWIN version 3.04

Test condition: Melting Point estimations performed by MPBPWIN are based on the

average result of the calculation methods of K. Joback and Gold and Ogle.

Joback's Method is described in Joback, K.G. 1982. A Unified Approach to Physical Property Estimation Using Multivariate Statistical Techniques. In The Properties of Gases and Liquids. Fourth Edition. 1987. R.C. Reid, J.M.

Prausnitz and B.E. Poling, Eds.

The Gold and Ogle Method simply uses the formula

Tm = 0.5839Tb, where Tm is the melting point in Kelvin and Tb is the

boiling point in Kelvin.

Test substance: CAS No. 70914-20-4; Alcohols, C6-8 branched

Reliability : (2) valid with restrictions

The result is a calculated value based on the chemical structure and represents a potential melting point for the substance with the CAS number

listed under test substance.

Flag : Critical study for SIDS endpoint

20.09.2004 (18)

2.2 BOILING POINT

Value : = 167 - 176 °C at 1013 hPa

Decomposition

Method : other: D1078/01

Year

GLP : no data

Test substance :

Test substance : CAS No. 70914-20-4; Alcohols, C6-8 branched

Reliability : (2) valid with restrictions

Although the original data were not retrieved and reviewed for quality, they

were developed following acceptable test methods and therefore

considered reliable.

Flag : Critical study for SIDS endpoint

04.04.2006 (13)

2.3 DENSITY

Type : density

Value : = .827 g/cm³ at 20 °C

Method : other: ASTM D4052/86 equivalent

Year

GLP : no data

Test substance

ld 70914-20-4

Date

Test substance : CAS No. 70914-20-4; Alcohols, C6-8 branched

Reliability : (2) valid with restrictions

Although the original data were not retrieved and reviewed for quality, they

were developed following acceptable test methods and therefore

considered reliable.

Flag : Critical study for SIDS endpoint

04.04.2006 (13)

2.3.1 GRANULOMETRY

2.4 VAPOUR PRESSURE

Value : = .58 hPa at 25 °C

Decomposition

Method : other (calculated)

Year :

GLP :

Test substance

Method : Vapor pressure calculation by MPBPWIN ver. 1.40 using calculation

method of Grain.

Remark : EPIWIN is used and advocated by the US EPA for chemical property

estimation.

Test substance : CAS No. 70914-20-4; Alcohols, C6-8 branched

Reliability : (2) valid with restrictions

The result is a calculated value based on the chemical structure and

represents a potential melting point for the substance with the CAS number

listed under test substance.

Flag : Critical study for SIDS endpoint

04.04.2006 (18)

2.5 PARTITION COEFFICIENT

Partition coefficient : octanol-water Log pow : = 1.8 - 2.6 at 25 °C

pH value

Method : other (calculated)

Year : GLP :

Test substance :

Method : Calculated values using KOWWIN version 1.65, a subroutine of the

computer program EPIWIN version 3.04

Test condition : Octanol / Water Partition Coefficient estimations performed by KOWWIN

are based on an atom/fragment contribution method of W. Meylan and P. Howard in "Atom/fragment contribution method for estimating octanol-water

partition coefficients". 1995. J. Pharm. Sci. 84:83-92.

Test substance: CAS No. 70914-20-4; Alcohols, C6-8 branched

Reliability : (2) valid with restrictions

The result is a calculated value based on the chemical structure and represents a potential melting point for the substance with the CAS number

listed under test substance.

Flag : Critical study for SIDS endpoint

04.04.2006 (20)

ld 70914-20-4

Date

2.6.1 SOLUBILITY IN DIFFERENT MEDIA

Solubility in : Water

Value : = 3539 - 11950 mg/l at 25 °C

pH value

concentration : at °C

Temperature effects

Examine different pol.

pKa : at 25 °C

Description

Stable

Deg. product

Method : other: calculated

Year

GLP : no data

Test substance

Method : Calculated values using WSKOWWIN version 1.36, a subroutine of the

computer program EPIWIN version 3.04

Test condition: Water Solubility estimations performed by WSKOWWIN are based on a

Kow correlation method described by W. Meylan, P. Howard and R. Boethling in "Improved method for estimating water solubility from octanol/water partition coefficient". Environ. Toxicol. Chem. 15:100-106.

1995.

Test substance: CAS No. 70914-20-4; Alcohols, C6-8 branched

Reliability : (2) valid with restrictions

The result is a calculated value based on the chemical structure and

represents a potential melting point for the substance with the CAS number

listed under test substance.

Flag : Critical study for SIDS endpoint

04.04.2006 (21)

2.6.2 SURFACE TENSION

2.7 FLASH POINT

Value : $= 60 \, ^{\circ}\text{C}$ Type : closed cup

Method : other: TCC ASTM D56

Year :

GLP : no data

Test substance :

Test substance : CAS No. 70914-20-4; Alcohols, C6-8 branched

Reliability : (4) not assignable

Although the original data were not retrieved and reviewed for quality, they

were developed following acceptable test methods and therefore

considered reliable.

Flag : Critical study for SIDS endpoint

04.04.2006 (13)

2.8 AUTO FLAMMABILITY

ld 70914-20-4 **Date** 04.04.2006

2.9 FLAMMABILITY

2.10 EXPLOSIVE PROPERTIES

2.11 OXIDIZING PROPERTIES

2.12 DISSOCIATION CONSTANT

Acid-base constant : = 15.50 at 25°C

Method : other: calculated

Year : GLP : Test substance :

Method : pKa calculation by SPARC 2003 using a Linux calculation engine.

Remark: SPARC On-line calculator can be accessed at http://ibmlc2.chem.uga.edu/sparc/index.cfm

Test substance: CAS No. 70914-20-4; Alcohols, C6-8 branched.

Reliability : (2) valid with restrictions

The value was calculated based on chemical structure as modeled by SPARC. This robust summary has a reliability rating of 2 because the data

are calculated and not measured.

04.04.2006 (15)

2.13 VISCOSITY

Value : = $9.4 - at 20 \, ^{\circ}C$

Result

Method : other: ASTM D445

Year :

GLP : no data

Test substance :

Remark: Value measured in cSt

Test substance : CAS No. 70914-20-4; Alcohols, C6-8 branched

Reliability : (2) valid with restrictions

Although the original data were not retrieved and reviewed for quality, they

were developed following acceptable test methods and therefore

considered to be reliable.

04.04.2006 (13)

2.14 ADDITIONAL REMARKS

ld 70914-20-4

Date

3.1.1 PHOTODEGRADATION

Type : water

Light source

Light spectrum: nm

Relative intensity : based on intensity of sunlight

Deg. product

Method : other (calculated): Technical Discussion

Year : GLP : Test substance :

Remark: These data represent a key study for characterising the potential of

substances in the Alkyl Alcohols C6 to C13 Category to undergo direct

photodegradation.

Result: Photolysis as a Function of Molecular Structure

The direct photolysis of an organic molecule occurs when it absorbs sufficient light energy to result in a structural transformation (Harris, 1982). The reaction process is initiated when light energy in a specific wavelength range elevates a molecule to an electronically excited state. However, the excited state is competitive with various deactivation processes that can result in the return of the molecule to a non excited state.

The absorption of light in the ultra violet (UV)-visible range, 110-750 nm, can result in the electronic excitation of an organic molecule. Light in this range contains energy of the same order of magnitude as covalent bond dissociation energies (Harris, 1982). Higher wavelengths (e.g. infrared) result only in vibrational and rotational transitions, which do not tend to produce structural changes to a molecule.

The stratospheric ozone layer prevents UV light of less than 290 nm from reaching the earth's surface. Therefore, only light at wavelengths between 290 and 750 nm can result in photochemical transformations in the environment (Harris, 1982). Although the absorption of UV light in the 290-750 nm range is necessary, it is not always sufficient for a chemical to undergo photochemical degradation. Energy may be re-emitted from an excited molecule by mechanisms other than chemical transformation, resulting in no change to the parent molecule.

A conservative approach to estimating a photochemical degradation rate is to assume that degradation will occur in proportion to the amount of light wavelengths >290 nm absorbed by the molecule (Zepp and Cline, 1977).

Substances in the Oxo Alcohols C6 to C13 Category contain molecules that are oxygenated aliphatic compounds which will absorb UV light below 220 nm (Boethling and Mackay, 2000) and will not undergo direct photolysis. Therefore, this fate process will not contribute to a measurable degradative removal of chemical components in this category from the environment.

References

Boethling, R.S., Mackay, D. (2000). Handbook of Property Estimation Methods for Chemicals. CRC Press, Boca Raton, FL, USA.

Harris, J. C. 1982. "Rate of Aqueous Photolysis," Chapter 8 in: W. J. Lyman, W. F. Reehl, and D. H. Rosenblatt, eds., Handbook of Chemical

ld 70914-20-4

Date

Property Estimation Methods, McGraw-Hill Book Company, New York,

USA.

Zepp, R. G. and D. M. Cline. 1977. Rates of Direct Photolysis in the

Aqueous Environment, Environ. Sci. Technol., 11:359-366.

Test substance : Alkyl Alcohols C6 to C13 Category members

Flag : Critical study for SIDS endpoint

03.04.2006 (12)

Туре

Light source : Sun light Light spectrum : nm

Relative intensity : based on intensity of sunlight

INDIRECT PHOTOLYSIS

Sensitizer : OH

Conc. of sensitizer : 1500000 molecule/cm³

Rate constant : = $.0000000000111208 \text{ cm}^3/(\text{molecule*sec})$

Degradation: % after

Deg. product :

Method : other (calculated): Calculated values using AOPWIN version 1.89, a

subroutine of the computer program EPIWIN version 3.04

Year : GLP :

Test substance :

Result: Atmospheric Oxidation Potential

In the environment, organic chemicals emitted into the troposphere are degraded by several important transformation processes. The dominant transformation process for most compounds is the daylight reaction with hydroxyl (OH-) radicals (Atkinson, 1988, 1989). The rate at which an organic compound reacts with OH- radicals is a direct measure of its atmospheric persistence (Meylan and Howard, 1993).

AOPWIN estimates the rate constant for the atmospheric, gas-phase reaction between photochemically produced hydroxyl radicals and organic chemicals. The rate constants estimated by the program are then used to calculate atmospheric half-lives for organic compounds based upon average atmospheric concentrations of hydroxyl radicals.

Since the reactions only take place in the presence of sunlight, the atmospheric half-lives are normalized for a 12-hour day.

Calculated* OH- Rate Constant half-life (hrs) (cm3/molecule-sec)

11.5 11.12 E-12

References:

Atkinson, R. 1988. Estimation of gas-phase hydroxyl radical rate constants for organic chemicals. Environ. Toxicol. Chem. 7:435-442.

Atkinson, R. 1989. Kinetics and mechanisms of the gas-phase reactions of the hydroxyl radical with organic compounds. J. Phys. Chem. Ref. Data Monograph No. 1, Amer. Inst. Physics & Amer. Chem. Soc., NY.

Meylan, W.M. and P.H. Howard. 1993. Computer estimation of the atmospheric gas-phase reaction rate of organic compounds with hydroxyl radicals and ozone. Chemosphere 12:2293-2299.

ld 70914-20-4

Date

Test condition: Indirect photodegradation, or atmospheric oxidation potential, is based on

the structure-activity relationship methods developed by R. Atkinson.

Temperature: 25°C Sensitizer: OH radical

Concentration of Sensitizer: 1.5 E6 OH radicals/cm3

Test substance: CAS No. 70914-20-4; Alcohols, C6-8 branched

Reliability : (2) valid with restrictions

The results include calculated data based on chemical structure as

modeled by AOPWIN. The data represent a potential atmospheric half-life

for the test substance.

Flag : Critical study for SIDS endpoint

04.04.2006 (19)

3.1.2 STABILITY IN WATER

 Type
 : abiotic

 t1/2 pH4
 : at °C

 t1/2 pH7
 : at °C

 t1/2 pH9
 : at °C

Deg. product

Method : other: Technical Discussion

Year : GLP :

Test substance :

Remark: These data represent a key study for characterising the potential of

substances in the Alkyl Alcohols C6 to C13 Category to undergo

hydrolysis.

Result : Hydolysis as a Function of Molecular Structure

Hydrolysis of an organic molecule occurs when a molecule (R-X) reacts with water (H2O) to form a new carbon-oxygen bond after the carbon-X bond is cleaved (Gould, 1959; Harris, 1982). Mechanistically, this reaction is referred to as a nucleophilic substitution reaction, where X is the leaving group being replaced by the incoming nucleophilic oxygen from the water molecule.

Chemicals that are susceptible to hydrolysis contain functional groups that can be displaced by a nucleophilic substitution reaction. Substances that have the potential to hydrolyze include alkyl halides, amides, carbamates, carboxylic acid esters and lactones, epoxides, phosphate esters, and sulfonic acid esters (Neely, 1985). The lack of a leaving group renders a compound resistant to hydrolysis.

Alkyl alcohols are resistant to hydrolysis because they lack a functional group that is hydrolytically reactive (Harris, 1982).

References:

Gould, E.S. (1959). Mechanism and Structure in Organic Chemistry, Holt, Reinhart and Winston, New York, NY, USA.

Harris, J.C. (1982). "Rate of Hydrolysis," Chapter 7 in: W.J. Lyman, W.F. Reehl, and D.H. Rosenblatt, eds., Handbook of Chemical Property Estimation Methods, McGraw-Hill Book Company, New York, NY, USA.

Neely, W.B. (1985). Hydrolysis. In: W.B. Neely and G.E. Blau, eds., Environmental Exposure from Chemicals. Vol. 1, pp. 157-173. CRC Press, Boca Raton, FL, USA.

ld 70914-20-4

Date

Conclusion : Hydrolysis will not contribute to the removal of Alkyl Alcohols from the

environment.

Flag : Critical study for SIDS endpoint

20.03.2006 (10)

3.1.3 STABILITY IN SOIL

3.2.1 MONITORING DATA

3.2.2 FIELD STUDIES

3.3.1 TRANSPORT BETWEEN ENVIRONMENTAL COMPARTMENTS

Type : fugacity model level III

Media: other: air - water - soil - sedimentAir: % (Fugacity Model Level I)Water: % (Fugacity Model Level I)Soil: % (Fugacity Model Level I)Biota: % (Fugacity Model Level II/III)Soil: % (Fugacity Model Level II/III)

Method : other: Calculation according Mackay, Level III

Year : 2003

Method : The EQC Level III model is a steady state model that is useful for

determining how the medium of release affects environmental fate. Level III fugacity allows non-equilibrium conditions to exist between connected media as steady state, and illustrate important transport and transformation

processes.

Physicochemical input values for the model were calculated using the EPIWIN Estimation v 3.04 program. Measured input values were also used where available and obtained from the EPIWIN database. Distribution data from the equilibrium model provide basic information on the potential partitioning behavior of chemicals between selected environmental

compartments (i.e., air, water, soil, and sediment).

Input values used:

Molecular mass = 116.21 g/mol Water solubility = 4089 mg/L Vapour pressure = 0.58 Pa

log Kow = 2.24

Melting point = -37.2 deg C

Degradation half-lives:

Air - 11.5 hrs Water - 120 hrs Soil - 720 hrs Sediment - 7200 hrs

This model was run assuming 100% discharge to water.

Result : Air - < 0.01%

Water - 99.4% Soil - 0.02% Sediment - 0.5%

ld 70914-20-4

Date

Test substance : CAS No. 70914-20-4; Alcohols, C6-8 branched

Reliability : (2) valid with restrictions

This robust summary has a reliability rating of 2 because the data are

calculated and not measured.

Flag : Critical study for SIDS endpoint

04.04.2006 (17)

Type : fugacity model level I

Media : other: air - biota - sediment(s) - soil - water

Air : % (Fugacity Model Level I)

Water : % (Fugacity Model Level I)

Soil : % (Fugacity Model Level I)

Biota : % (Fugacity Model Level II/III)

Soil : % (Fugacity Model Level II/III)

Method : other: Calculation according Mackay, Level I

Year : 2003

Method : The EQC Level I is a steady state, equilibrium model that utilizes the input

of basic chemical properties including molecular weight, vapor pressure, and water solubility to calculate distribution within a standardized regional

environment.

Physicochemical input values for the model were calculated using the EPIWIN Estimation v 3.04 program. Measured input values were also used where available and obtained from the EPIWIN database. Distribution data from the equilibrium model provide basic information on the potential partitioning behavior of chemicals between selected environmental

compartments (i.e., air, water, soil, sediment, suspended sediment, biota).

Input values used:

Molecular mass = 116.21 g/mol Water solubility = 4089 mg/L Vapour pressure = 0.58 Pa

log Kow = 2.24

Melting point = -37.2 deg C

Result : Air - 0.3%

Water - 67.0 % Soil - 13.3% Sediment - 0.3%

Suspended Sed - < 0.01%

Biota - < 0.01%

Test substance : CAS No. 70914-20-4; Alcohols, C6-8 branched

Reliability : (2) valid with restrictions

This robust summary has a reliability rating of 2 because the data are

calculated and not measured.

Flag : Critical study for SIDS endpoint

04.04.2006 (17)

3.3.2 DISTRIBUTION

Media : other: Koc Method : other (calculation)

Year : 2003

Method : The soil adsorption coefficient (Koc) was calculated by PCKOCWIN ver.

1.66, using equations developed by Lyman (1980).

PCKOCWIN is a subroutine of EPIWIN. EPIWIN is used and advocated by the USEPA for chemical property estimation. Koc provides an indication of the extent to which a chemical partitions between solid and solution phases

ld 70914-20-4

Date

in soil, or between water and sediment in aquatic ecosystems.

Result : Koc = 12.85

log Koc = 1.11

Test substance: CAS No. 70914-20-4; Alcohols, C6-8 branched

Reliability : (2) valid with restrictions

This robust summary has a reliability rating of 2 because the data are calculated and not measured. The value was calculated based on

chemical structure as modeled by EPIWIN.

Flag : Critical study for SIDS endpoint

04.04.2006 (16) (22)

3.4 MODE OF DEGRADATION IN ACTUAL USE

3.5 BIODEGRADATION

3.6 BOD5, COD OR BOD5/COD RATIO

3.7 BIOACCUMULATION

3.8 ADDITIONAL REMARKS

Date

4.1 ACUTE/PROLONGED TOXICITY TO FISH

Type : flow through

Species: Pimephales promelas (Fish, fresh water)

Exposure period : 96 hour(s)
Unit : mg/l

LC50 : = 34.5 measured/nominal

Limit test

Analytical monitoring : yes

Method : other: no data

Year : 1985 GLP : no data

Test substance :

Method : Trimmed Spearrman Karber Method

Result : 96 hour LC50 = 34.5 mg/L (95% CI 33.1 to 36.0) based upon measured

values

Analytical method used was Gas-Liquid Chromatography.

Measured	Fish Total
Conc. (mg/L)	Mortality (@96 hrs)*
Control	0
12.5	0
18.1	1
28.5	0
43.6	20
70.8	20

^{* 20} fish added at test initiation

Test condition: Treatment solutions were prepared by diluting a 1400mg/L stock solution.

Nominal heptanol treatment levels were 12.5, 19.3, 29.7, 45.7, 70.3mg/L, which measured 12.5, 18.1, 28.5, 43.6, and 70.8mg/L, respectively.

Control/dilution water was EPA Duluth laboratory water.

Twenty fish were tested per treatment. Treatment volume = 2.0L. Test parameters were as follows: temperature=25.6 Deg C; dissolved oxygen = 7.1mg/L; pH = 7.7; fish age = 31 days old; fish mean wt = 0.100g; fish

mean length = 18.1mm; fish loading = 1.0g/L/day.

Organism supplier was U.S. EPA Environmental Research Lab, Duluth,

MN, USA.

Test substance: CAS No. 70914-20-4; Alcohols, C6-8 branched

Reliability : (1) valid without restriction
Flag : Critical study for SIDS endpoint

04.04.2006 (14)

Type : other: calculated
Species : other: freshwater fish

Exposure period : 96 hour(s)
Unit : mg/l

LC50 : = 51.2 calculated

Method : other: ECOSAR Computer Model

Year : 2005

GLP : Test substance :

Test condition: Log Kow (octanol/water partition coefficient) values and a chemical

Date

structure are needed to calculate aquatic toxicity using the ECOSAR model. The Kow calculation is performed by KOWWIN based on an atom/fragment contribution method of Meylan and Howard (1), which is a subroutine in the EPIWIN computer model (2). KOWWIN also has a database of experimental Kow values (EXPKOW.DB).

The ECOSAR program was run using a C7 alcohol with a Kow of 2.24.

1. Meylan, W. and P. Howard. 1995. Atom/fragment contribution method for estimating octanol-water partition coefficients. J. Pharm. Sci. 84:83-92.

2. Meylan, M., SRC 1994-1999. KOWWIN is contained in the computer program EPIWIN. 1999. Estimation Program Interface for Windows, version 3.04. Syracuse Research Corporation, Syracuse, NY, USA.

Test substance Conclusion

: CAS No. 70914-20-4; Alcohols, C6-8 branched

: Based on the calculated Kow value, the C7 alcohol is expected to have an

acute 96-hour LC50 of 51.2 mg/L and a Chronic Value of 6.9 mg/L.

Reliability : (2) valid with restrictions

The value was calculated based on chemical structure as modeled by EPIWIN. This robust summary has a reliability rating of 2 because the data

are calculated and not measured.

04.04.2006 (4)

4.2 ACUTE TOXICITY TO AQUATIC INVERTEBRATES

Type :

Species : Daphnia magna (Crustacea)

 Exposure period
 : 48 hour(s)

 Unit
 : mg/l

 EC50
 : = 63

 Limit Test
 : no

 Analytical monitoring
 : no

Method : other: Concept rules of the Dutch Standardization Institute (Adema, 1978)

Year : 1978 GLP : no data

Test substance

Method : No Data

Remark : Test Type: Daphnid Acute Toxicity Test

Result : 48-hour EC50 = 63 mg/L, based upon nominal concentrations of the test

chemicals.

Test condition: Tests using 15 different chemicals, including n-Heptanol, were performed

at two different laboratories. Lab I was the National Institute of Public Health, Bilthoven, The Netherlands; Lab II was the Central Laboratory, T.N.O., Delft, The Netherlands. The tests were conducted using standardized tests methods proposed by the Dutch Standardization Institute (Adema, 1978). The tests were conducted in duplicate to

determine the reproducibility of the results.

Organisms were supplied by in-house cultures. Age = <24 hours old.

Test substance: CAS No. 70914-20-4; Alcohols, C6-8 branched

Conclusion: Test substance is considered to have moderate acute toxicity.

Reliability : (2) valid with restrictions

Performed using a proposed standardized test.

Flag : Critical study for SIDS endpoint

04.04.2006 (1) (2)

Туре

Species : Daphnia sp. (Crustacea)

Exposure period : 48 hour(s)

Date

Unit : mg/l

EC50 : = 56 calculated

Method : other: ECOSAR Computer Model

Year : 2005

GLP : Test substance :

Test condition: Log Kow (octanol/water partition coefficient) values and a chemical

structure are needed to calculate aquatic toxicity using the ECOSAR model. The Kow calculation is performed by KOWWIN based on an atom/fragment contribution method of Meylan and Howard (1), which is a subroutine in the EPIWIN computer model (2). KOWWIN also has a

database of experimental Kow values (EXPKOW.DB).

The ECOSAR program was run using a C7 alcohol with a Kow of 2.24.

1. Meylan, W. and P. Howard. 1995. Atom/fragment contribution method for estimating octanol-water partition coefficients. J. Pharm. Sci. 84:83-92.

2. Meylan, M., SRC 1994-1999. KOWWIN is contained in the computer program EPIWIN. 1999. Estimation Program Interface for Windows, version 3.04. Syracuse Research Corporation, Syracuse, NY, USA.

Test substance : CAS No. 70914-20-4; Alcohols, C6-8 branched

Conclusion : Based on the calculated Kow value, the C7 alcohol is expected to have an

acute 48-hour EC50 of 56 mg/L and a Chronic Value of 3.2 mg/L.

Reliability : (2) valid with restrictions

The value was calculated based on chemical structure as modeled by EPIWIN. This robust summary has a reliability rating of 2 because the data

are calculated and not measured.

04.04.2006 (4)

4.3 TOXICITY TO AQUATIC PLANTS E.G. ALGAE

Species: other algae: green alga

Endpoint

Exposure period : 96 hour(s)
Unit : mg/l

EC50 : = 35 calculated

Method : other: ECOSAR Computer Model

Year :

GLP

Test substance: other TS

Test condition : Log Kow (octanol/water partition coefficient) values and a chemical

structure are needed to calculate aquatic toxicity using the ECOSAR model. The Kow calculation is performed by KOWWIN based on an atom/fragment contribution method of Meylan and Howard (1), which is a subroutine in the EPIWIN computer model (2). KOWWIN also has a

database of experimental Kow values (EXPKOW.DB).

The ECOSAR program was run using a C7 alcohol with a Kow of 2.24.

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Test substance: CAS No. 70914-20-4; Alcohols, C6-8 branched

Conclusion: Based on the calculated Kow value, the C7 alcohol is expected to have an

Date

acute 96-hour EC50 of 35 mg/L and a Chronic Value of 4.1 mg/L.

Reliability : (2) valid with restrictions

The value was calculated based on chemical structure as modeled by EPIWIN. This robust summary has a reliability rating of 2 because the data

are calculated and not measured.

Flag : Critical study for SIDS endpoint

04.04.2006 (4)

4.4 TOXICITY TO MICROORGANISMS E.G. BACTERIA

4.5.1 CHRONIC TOXICITY TO FISH

Species: other: Freshwater Fish (calculated toxicity values are not species specific)

Endpoint : other: LC50
Exposure period : 30 day(s)
Unit : mg/l

ChV : = 6.9 calculated

Method : other: calculated

Year : 2005

GLP : Test substance :

Test condition: Log Kow (octanol/water partition coefficient) values and a chemical

structure are needed to calculate aquatic toxicity using the ECOSAR model. The Kow calculation is performed by KOWWIN based on an atom/fragment contribution method of Meylan and Howard, which is a subroutine in the EPIWIN computer model. KOWWIN also has a database

of experimental Kow values (EXPKOW.DB).

The ECOSAR program was run using a C7 alcohol with a Kow of 2.24.

Test substance : CAS No. 70914-20-4; Alcohols, C6-8 branched

Conclusion: Based on the calculated Kow value, the C7 alcohol is expected to have a

30-day Chronic Value of 6.9 mg/L

Reliability : (2) valid with restrictions

The value was calculated based on chemical structure as modeled by ECOSAR. This robust summary has a reliability rating of 2 because the

data are calculated and not measured.

Flag : Critical study for SIDS endpoint

04.04.2006 (3)

4.5.2 CHRONIC TOXICITY TO AQUATIC INVERTEBRATES

Species : Daphnia magna (Crustacea)

Endpoint : mortality
Exposure period : 16 day(s)
Unit : mg/l

EC50 : = 3.2 calculated

Method : other: ECOSAR Computer Model

Year : 2005

GLP :

Test substance :

Test condition: Log Kow (octanol/water partition coefficient) values and a chemical

structure are needed to calculate aquatic toxicity using the ECOSAR model. The Kow calculation is performed by KOWWIN based on an atom/fragment contribution method of Meylan and Howard, which is a subroutine in the EPIWIN computer model. KOWWIN also has a database

Date

of experimental Kow values (EXPKOW.DB).

The ECOSAR program was run using a C7 alcohol with a Kow of 2.24.

Test substance: CAS No. 70914-20-4; Alcohols, C6-8 branched

Conclusion : Based on the calculated Kow value, the C7 alcohol is expected yo have a

16-day EC50 of 3.2 mg/L.

Reliability : (2) valid with restrictions

The value was calculated based on chemical structure as modeled by EPIWIN. This robust summary has a reliability rating of 2 because the data

are calculated and not measured.

Flag : Critical study for SIDS endpoint

04.04.2006 (3)

4.6.1 TOXICITY TO SEDIMENT DWELLING ORGANISMS

4.6.2 TOXICITY TO TERRESTRIAL PLANTS

4.6.3 TOXICITY TO SOIL DWELLING ORGANISMS

Type :

Species: other: Earthworm

Endpoint : mortality
Exposure period : 14 day(s)
Unit : mg/kg soil dw
LC50 : = 603 calculated

Method : other: ECOSAR Computer Model

Year : 2005

GLP :

Test substance :

Test condition: Log Kow (octanol/water partition coefficient) values and a chemical

structure are needed to calculate acute terrestrial toxicity using the

ECOSAR model. The Kow calculation is performed by KOWWIN based on an atom/fragment contribution method of Meylan and Howard, which is a subroutine in the EPIWIN computer model. KOWWIN also has a database

of experimental Kow values (EXPKOW.DB).

The ECOSAR program was run using a C7 alcohol with a Kow of 2.24.

Test substance: CAS No. 70914-20-4; Alcohols, C6-8 branched

Conclusion : Based on the calculated Kow value, the C7 alcohol is expected to have a

14-day EC50 of 603 mg/kg soil.

Reliability : (2) valid with restrictions

The value was calculated based on chemical structure as modeled by EPIWIN. This robust summary has a reliability rating of 2 because the data

are calculated and not measured.

Flag : Critical study for SIDS endpoint

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4.6.4 TOX. TO OTHER NON MAMM. TERR. SPECIES

4.7 BIOLOGICAL EFFECTS MONITORING

4. Ecotoxicity	70914-20-4 04.04.2006
4.8 BIOTRANSFORMATION AND KINETICS	
4.9 ADDITIONAL REMARKS	
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5.0 TOXICOKINETICS, METABOLISM AND DISTRIBUTION

5.1.1 ACUTE ORAL TOXICITY

Type : LD50

Value : = 3900 mg/kg bw

Species : rat

Strain : Sprague-Dawley

Sex : male Number of animals : 5

Vehicle : other: None

Doses :

Method: otherYear: 1979GLP: no data

Test substance :

Remark: All animals in the 6.81 and 10.00 g/kg groups died. Two of the five animals

in the 4.64 g/kg group died and 1 animal each in the 1.00, 2.15, and 3.15 g/kg groups died. No animals in the 1.47 g/kg group died. Except for one animal in the 2.15 g/kg group, all animals that died did so within three days of dosing. Signs of toxicity observed included respiratory rate decreases,

fecal staining, decreased motor activity and hypothermia.

Route of administration: Oral Intubation Frequency of Treatment: Single Exposure

Dose/Concentration Levels: 1.0, 1.47, 2.15, 3.16, 4.64, 6.81 and 10.0 g/kg

Control group and Treatment: None

Result : LD50 = 3.9 g/kg

Test condition: Animals were fasted for approximately 18 hours prior to dosing. The

undiluted test material was administered by oral intubation at doses of 1.0, 1.47, 2.15, 3.16, 4.64, 6.81 and 10.0 g/kg (5 rats/dose). Animals were observed for signs of toxicity at 1, 2, and 4 hours after dosing and daily

thereafter for fourteen days.

Test substance: CAS No. 70914-20-4; Alcohols, C6-8 branched

Conclusion: Under the conditions of this study, Alcohols, C6-8 branched have a low

order of acute oral toxicity.

Reliability : (2) valid with restrictions

Only one sex tested

04.04.2006 (5)

5.1.2 ACUTE INHALATION TOXICITY

Species : rat

Strain : Sprague-Dawley
Sex : male/female

Number of animals : 10

Vehicle : other: NA

Doses

Exposure time : 6 hour(s)

Method : other

Year : 1979

GLP : no data

Test substance

Date

Remark: No abnormalities were noted in the control or exposed rats during the

exposure period. Upon removal from the chamber, dry rales (1/10) and excessive salivation (2/10) were observed in exposed rats. Necropsy examination revealed an increased incidence of lung discoloration in

treated rats (6/10).

Route of administration: Inhalation Frequency of Treatment: Single Dose/Concentration Levels: 0. 152 ppm

Result : LC50 > 152 ppm

Test condition: Animals (5/sex/dose) were held for a minimum equilibration period of 12

days. Animals were exposed to 152 ppm of the test material for six hours. To generate vapors, room air was drawn through the test material at a flow rate of 103 l/min. The resulting maximum attainable vapors were passed through a Kjeldahl trap and flask prior to entering the glass exposure chamber containing the test animals. Weight loss was determined following exposure and was taken to be equal to the amount of test material delivered during exposure. The weight loss was divided by the total volume of air passed through the chamber to give the nominal concentration. The animals were exposed in the same chamber. For each

species, a control group was also sham-exposed to room air. The animals were observed for abnormalities prior to exposure, at 15-minute intervals during the first hour of exposure and then hourly for the remainder of exposure. Subsequent evaluations were made for a total of 14 days. After

fourteen days, gross necropsy was performed.

Test substance : CAS No. 70914-20-4; Alcohols, C6-8 branched

Conclusion : Under the conditions of this study, Alcohols, C6-8 branched have a low

order of acute inhalation toxicity in rats.

Reliability : (2) valid with restrictions

Vapor concentration not analyzed Critical study for SIDS endpoint

Flag : Critical study for SIDS endpoint

04.04.2006 (7)

 Type
 : LC50

 Value
 : > 152 ppm

 Species
 : mouse

Strain : other: Swiss albino

Sex : male/female

Number of animals : 10 Vehicle : other: NA

Doses :

Exposure time : 6 hour(s)

Method : other

Year : 1979

GLP : no data

Test substance :

Remark: No abnormalities were noted in the control or exposed mice during the

exposure period. During the 14-day observation period, excessive salivation was observed in mice (4/10) and nasal discharge (2/10)

occurred.

Route of administration: Inhalation Frequency of Treatment: Single

Dose/Concentration Levels: 0, 152 ppm

Result : LC50 > 152 ppm

Test condition : Animals (5/sex/dose) were held for a minimum equilibration period of 12

days. Animals were exposed to 152 ppm of the test material for six hours. To generate vapors, room air was drawn through the test material at a flow rate of 103 l/min. The resulting maximum attainable vapors were passed through a Kjeldahl trap and flask prior to entering the glass exposure chamber containing the test animals. Weight loss was determined following exposure and was taken to be equal to the amount of test material delivered during exposure. The weight loss was divided by the

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total volume of air passed through the chamber to give the nominal concentration. The animals were exposed in the same chamber. For each species, a control group was also sham-exposed to room air. The animals were observed for abnormalities prior to exposure, at 15-minute intervals during the first hour of exposure and then hourly for the remainder of exposure. Subsequent evaluations were made for a total of 14 days. After fourteen days, gross necropsy was performed.

CAS No. 70914-20-4: Alcohols, C6-8 branched Test substance

Under the conditions of this study, Alcohols, C6-8 branched have a low Conclusion

order of acute inhalation toxicity in mice.

Reliability : (2) valid with restrictions

Vapor concentration not analyzed

Critical study for SIDS endpoint Flag

04.04.2006 (7)

LC50 Type Value > 152 ppm **Species** guinea pig Strain Hartley male/female Sex

Number of animals 10 Vehicle other: NA

Doses

Exposure time 6 hour(s) Method other Year 1979 **GLP** no data

Test substance

Remark No abnormalities were noted in the control or exposed guinea pigs during

the exposure period. Necropsy examination revealed an increased

incidence of lung discoloration in treated guinea pigs (8/10).

Route of administration: Inhalation Frequency of Treatment: Single Dose/Concentration Levels: 0, 152 ppm

LC50 > 152 ppm Result

Test condition Animals (5/sex/dose) were held for a minimum equilibration period of 12

> days. Animals were exposed to 152 ppm of the test material for six hours. To generate vapors, room air was drawn through the test material at a flow rate of 103 l/min. The resulting maximum attainable vapors were passed through a Kjeldahl trap and flask prior to entering the glass exposure chamber containing the test animals. Weight loss was determined following exposure and was taken to be equal to the amount of test material delivered during exposure. The weight loss was divided by the total volume of air passed through the chamber to give the nominal concentration. All three species were exposed in the same chamber. For each species, a control group was also sham-exposed to room air. The animals were observed for abnormalities prior to exposure, at 15-minute intervals during the first hour of exposure and then hourly for the remainder of exposure. Subsequent evaluations were made for a total of 14 days.

After fourteen days, gross necropsy was performed.

CAS No. 70914-20-4; Alcohols, C6-8 branched Test substance Conclusion

Under the conditions of this study, Alcohols, C6-8 branched have a low

order of acute inhalation toxicity in guinea pigs.

Reliability (2) valid with restrictions

Vapor concentration not analyzed

Critical study for SIDS endpoint Flag

04.04.2006 (7)

Date

5.1.3 ACUTE DERMAL TOXICITY

Type : LD50

Value : > 3160 mg/kg bw

Species : rabbit

Strain : New Zealand white

Sex : male/female

Number of animals : 4

Vehicle : other: None

Doses

Method : other Year : 1979 GLP : no data

Test substance :

Remark : Route of administration: Dermal

Frequency of Treatment: Single Dose

Dose/Concentration Levels: 50, 200, 794 and 3,160 mg/kg

Control group and Treatment: None

There were no deaths at any dose level in either sex. All animals at the 50 mg/kg level exhibited very slight erythema and no edema. Well-defined erythema without edema was observed in animals at 200 and 794 mg/kg dose levels. At the 3160 mg/kg dose level one animal exhibited moderate

to severe erythema and three animals exhibited areas of necrosis.

Necropsy examinations did not reveal any significant abnormalities. Dark red foci were observed in the lungs of males (50mg/kg) and females (3,160 mg/kg), however this effect was not dose-related. Dark red foci of the adrenals were observed in males and females at 200, 794, and 3,160

mg/kg.

Result : LD50 > 3160 mg/kg of body weight.

Test condition: Doses of 50, 200, 794 and 3160 mg/kg were administered to sixteen

rabbits (two/sex/dose level). The undiluted test material was applied to intact skin and the animal was then wrapped in an impervious plastic sleeve. Following approximately 24 hours of exposure, the wrappings were removed and the test site was wiped free of excess test material. After 30 minutes, dermal observations were made. Observations were recorded at

1, 2 and 4 hours after dosing and daily thereafter for 14 days.

Test substance: CAS No. 70914-20-4; Alcohols, C6-8 branched

Conclusion: Under the conditions of this study, Alcohols, C6-8 branched have a low

order of acute dermal toxicity in rabbits.

Reliability : (2) valid with restrictions

GLP not specified

Flag : Critical study for SIDS endpoint

04.04.2006 (6)

5.1.4 ACUTE TOXICITY, OTHER ROUTES

5.2.1 SKIN IRRITATION

Species : rabbit

Concentration: 3.2 other: g/kgExposure: OcclusiveExposure time: 24 hour(s)

Number of animals : 4
Vehicle : PDII : Result :

Date

Classification : Method : Year : GLP : Test substance :

Result : Group mean scores at 24, 48, and 72 hours were 1.5, 2.0, 4.0 for erythema

and 2.5, 2.5, and 2.5 for edema. Dermal reactions were severe in all

animals from Day 7 to 14.

Test condition: New Zealand White rabbits, at least 8 weeks old were acclimated to

laboratory conditions for at least one week prior to use in the test. Animals

were housed in accordance with AAALAC standards.

Immediately prior to dosing, the fur was clipped from the abdomen of the animals. Abrasions were made in the skin of all animals. The abrasions scratched the stratum corneum, but did not reach the derma or produce

bleeding.

One group of 4 rabbits (2 males, 2 females) was dosed at 3.16 g/kg. The test material was applied once dermally to the prepared site under gauze patches. The patches were secured with adhesive tape and the trunks wrapped with impervious material. The test material was kept in contact with the skin for 24 hours, at which time the wrappings were removed. The exposure site was washed with water or corn oil to remove excess material.

Dermal observations were scored at 24 hours, and 3, 7, 10 and 14 days after dosing by the Draize scoring system. The rabbits were observed 2 and 4 hours post dosing, and daily for 14 days for signs of toxicity,

pharmacological effects and mortality. Body weights were recorded pretest

and in the survivors at 14 days.

Test substance: CAS No. 70914-20-4; Alcohols, C6-8 branched

Conclusion: Under the conditions of the study, Alcohols, C6-8 branched produced

significant irritation to rabbit skin.

Flag : Critical study for SIDS endpoint

04.04.2006 (8)

5.2.2 EYE IRRITATION

Species : rabbit
Concentration : .1 undiluted

Dose

Exposure time :
Comment :
Number of animals : 6

Vehicle :

Result : highly irritating

Classification

Method : other: similar to OECD TG 405

Year : 1979

GLP :

Test substance :

Result: Median Scores were:

24 hr - 24 72 hr - 28 7 days - 5

Conjunctival irritation generally cleared by 10 days.

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The maximum total Draize score observed was 51. Group mean scores at 24, 48, and 72 hours for the various indices were: 1.5, 1.5, 1.33 for conjunctival redness; 2.17, 2.5, 2 for chemosis; 1.0, 0.83, 0.83 for iridial

irritation; 1.33, 1.5, 1.17 for corneal opacity.

Test condition : A single application of 0.1 ml of undiluted alcohol was made into the

conjunctival sac of the left eye of each rabbit. The untreated eye served as the control. The treated eye was held closed for 1 second, and eyes were

not washed.

Observations for irritancy and possible systemic toxicity were made at 1, 4, and 24 hours and at 2, 3, 4, and 7 days. At each interval the treated and control eyes were examined and scored for ocular reactions according to

the Draize Standard Eye Irritation Grading Scale.

Test substance : CAS No. 70914-20-4; Alcohols, C6-8 branched

Reliability : (2) valid with restrictions

Flag : Critical study for SIDS endpoint

04.04.2006 (9)

5.3 SENSITIZATION

5.4 REPEATED DOSE TOXICITY

Type : Sub-acute
Species : rat
Sex : male
Strain : Wistar
Route of admin. : gavage
Exposure period : 14 d

Frequency of treatm. : daily
Post exposure period : none

Doses : 130 mg/kg/bw/day (1 mM/kg bw)

Control group : yes, concurrent vehicle

NOAEL : >= 130 - mg/kg bw

Method : other: screening study

Year : 1984

GLP : Test substance :

Remark: Comparative screening study to investigate selectively the

effect on testis and liver morphology and function.

Result : Iso-octanol did not impair body weight and had no effect on

relative liver and testis weight. Neither clinical pathology parameter nor peroxisome associated enzymes were affected.

Liver histology differed not from the concurrent control

group.

Test condition: Test species:

Male rats (Wistar derived) (breeder: ICI Alderley Park, UK)

10 per control group, 5 per dose group

Housing and diet:

1 week acclimatization

Caging: single, stainless-steel, screen-bottom cages

food and water ad libitum

Administration:

The substance was dissolved in polyethylene glycol 300 and 5 male rats received a dose level of 168 mg/kg bw (equivalent

Date

to 1 mmol/kg bw) daily by gavage for 14 consecutive days. Application volume was 10 ml/kg bw. 10 control rats received the solvent in parallel.

Examinations:

Body weights were determined. The animals were killed by halothane overdose and blood samples for clinical pathology were taken. The liver was weighed and samples for light and

electron microscopy, morphometric analysis were investigated. Remaining liver was processed for

investigation of catalase and CN-insensitive palmitoyl CoA

oxidation.

Test substance: Analogue substance: iso-octanol (CAS not specified).

Conclusion : Under the conditions of the study, iso-octanol has a low order of repeated

dose toxicity to male rats for the endpoints studied. The NOAEL was the

limit dose of 130 mg/kg.

Reliability : (2) valid with restrictions

Screening study with limited examinations but scientifically valid and well documented. From peer reviewed literature.

Flag : Critical study for SIDS endpoint

04.04.2006 (23)

5.5 GENETIC TOXICITY 'IN VITRO'

Type : Chromosomal aberration test

System of testing : Assay using Chinese hamster ovary (CHO) cells

Test concentration : 3.13, 10, 31.3, 100 and 313 ug/ml

Cycotoxic concentr.

Metabolic activation : with and without

Result : negative

Method : EPA OPPTS 870.5375

Year : 2005 GLP : yes

Test substance :

Method: The study consisted of two phases: an initial chromosomal aberration

assay with a 19 hour harvest time, and a repeat assay with both 19 and 43 hour cell harvest times. For the initial assay the test substance was mixed in acetone and tested at concentrations of 3.13, 10, 31.3, 100 and 313 ug/ml with metabolic activation (+S9) and 6.25, 20, 62.5, 200, and 625 ug/ml without metabolic activation (-S9). These concentrations were selected based on a previous toxicity pretest. The same concentrations

were tested in the 19-hour and 43-hour repeat assays.

The CHO cells were cultured in McCoy's 5A Medium containing 10% fetal bovine serum and 2 mM L-glutamine and 1% penicillin/streptomycin at 37±20C, in 4-6% CO2 in air. Two sets of duplicate cultures were prepared; one set was treated with the test substance with activation (+S9) and the other was treated with the test substance without activation (-S9). Each flask received a 50 mL sample of the test substance mixture, positive control mixture (N-Methyl-N'-Nitro-N-Nitrosoguanidine, MNNG, or 9,10-Dimethyl-1,2-benzanthracene (same as 7,12 Dimethylbenz[a]anthracene), DMBA), or vehicle (acetone). Flasks were treated for 3 hours for both +S9 and -S9 in the initial assay. In the repeat assay, flasks were treated for 3 hours for the +S9 and for 19 hours for -S9. The cultures were incubated to their respective harvest times (19 or 43 hours). A spindle inhibitor was added to the flasks approximately 1-3 hours prior to harvest to arrest the cells in c-metaphase. The cells were harvested and slides prepared to evaluate chromosomes. The positive control materials were evaluated for chromosomal aberrations at the 19 hour harvests only.

Date

Based on confluency data generated during the study, 3.13, 31.3 and 313 mg/mL were selected as the doses to be evaluated for chromosomal aberrations with metabolic activation for both the initial and repeat assays. The doses picked to be evaluated for chromosomal aberrations without metabolic activation were also based on confluency data. The doses evaluated for the initial assay without metabolic activation (-S9) were 6.25, 62.5 200 and 625 mg/mL. The doses evaluated for the 19 hour repeat assay without metabolic activation (-S9) were 6.25, 62.5 200 and 625 mg/mL. The doses evaluated for the 43 hr repeat assay without metabolic activation (-S9) were 6.25, 62.5 and 200 mg/mL.

Result

There were no statistically significant differences or dose-related trends noted at any dose level in the initial or the repeat assay.

The positive control substances, DMBA and MNNG, produced statistically significant increases in the percentage of aberrant cells when tested with activation (DMBA) or without activation (MNNG) when compared to the vehicle control. Thus, the test system responded in an appropriate manner to these known clastogens. The vehicle control used in this study (acetone) also performed in a manner consistent with the criteria

established for a valid assay.

Test substance Conclusion

: CAS No. 70914-20-4; Alcohols, C6-8 branched

: Under the conditions of this study, the test substance did not induce chromosomal aberrations in CHO cells in the initial assay or the repeat

assay.

Reliability : (1) valid without restriction
Flag : Critical study for SIDS endpoint

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- 5.6 GENETIC TOXICITY 'IN VIVO'
- 5.7 CARCINOGENICITY
- 5.8.1 TOXICITY TO FERTILITY
- 5.8.2 DEVELOPMENTAL TOXICITY/TERATOGENICITY
- 5.8.3 TOXICITY TO REPRODUCTION, OTHER STUDIES
- 5.9 SPECIFIC INVESTIGATIONS
- 5.10 EXPOSURE EXPERIENCE
- 5.11 ADDITIONAL REMARKS

6. Analyt. Me	th. for Detection and	Identification	70914-20-4 04.04.2006	
6.1 ANALYTIC	CAL METHODS			
6.2 DETECTION	ON AND IDENTIFICATION			
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7. Eff	. Against Target Org. and Into	ended Uses	ld Date	70914-20-4	
7.1	FUNCTION				
7.2	EFFECTS ON ORGANISMS TO BE CON	TROLLED			
7.3	ORGANISMS TO BE PROTECTED				
7.4	USER				
7.5	RESISTANCE				
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		31,00			

8.1 METHODS HANDLING AND STORING 8.2 FIRE GUIDANCE 8.3 EMERGENCY MEASURES 8.4 POSSIB. OF RENDERING SUBST. HARMLESS 8.5 WASTE MANAGEMENT 8.6 SIDE-EFFECTS DETECTION 8.7 SUBSTANCE REGISTERED AS DANGEROUS FOR GROUND WATER 8.8 REACTIVITY TOWARDS CONTAINER MATERIAL	8. M	eas. Nec. to Prot. Man, Animals, Environment	ld Date	70914-20-4
8.3 EMERGENCY MEASURES 8.4 POSSIB. OF RENDERING SUBST. HARMLESS 8.5 WASTE MANAGEMENT 8.6 SIDE-EFFECTS DETECTION 8.7 SUBSTANCE REGISTERED AS DANGEROUS FOR GROUND WATER	8.1	METHODS HANDLING AND STORING		
8.4 POSSIB. OF RENDERING SUBST. HARMLESS 8.5 WASTE MANAGEMENT 8.6 SIDE-EFFECTS DETECTION 8.7 SUBSTANCE REGISTERED AS DANGEROUS FOR GROUND WATER	8.2	FIRE GUIDANCE		
8.5 WASTE MANAGEMENT 8.6 SIDE-EFFECTS DETECTION 8.7 SUBSTANCE REGISTERED AS DANGEROUS FOR GROUND WATER	8.3	EMERGENCY MEASURES		
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8.7 SUBSTANCE REGISTERED AS DANGEROUS FOR GROUND WATER	8.5	WASTE MANAGEMENT		
	8.6	SIDE-EFFECTS DETECTION		
8.8 REACTIVITY TOWARDS CONTAINER MATERIAL	8.7	SUBSTANCE REGISTERED AS DANGEROUS FOR GROUND WATER		
	8.8	REACTIVITY TOWARDS CONTAINER MATERIAL		
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9. References Id 70914-20-4 Date 04.04.2006

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10. Summary and Evaluation		ld Date	70914-20-4	
10.1 END POINT SUMMARY				
10.2 HAZARD SUMMARY				
10.3 RISK ASSESSMENT				
	35 / 35			

IUCLID

Data Set

Existing Chemical : ID: 68526-85-2 CAS No. : 68526-85-2

CAS No. EINECS Name : Alcohols, C9-11-iso-, C10-rich

EC No. : 271-234-0

: Alcohols, C9-11-iso-, C10-rich TSCA Name **IUPAC Name** : Alcohols, C9-C11, branched

Producer related part

: ExxonMobil Biomedical Sciences Inc. Company

: 29.09.2004 Creation date

Substance related part

Company : ExxonMobil Biomedical Sciences Inc.

Creation date : 29.09.2004

Status

Memo : Prepared for EMCC - US HPV

Printing date : 06.04.2006

Revision date

Date of last update : 06.04.2006

Number of pages : 39

Chapter (profile) : Chapter: 1, 2, 3, 4, 5, 6, 7, 8, 10 Reliability (profile) : Reliability: without reliability, 1, 2, 3, 4

: Flags: without flag, confidential, non confidential, WGK (DE), TA-Luft (DE), Flags (profile)

Material Safety Dataset, Risk Assessment, Directive 67/548/EEC, SIDS

ld 68526-85-2

Date

1.0.1 APPLICANT AND COMPANY INFORMATION

1.0.2 LOCATION OF PRODUCTION SITE, IMPORTER OR FORMULATOR

1.0.3 IDENTITY OF RECIPIENTS

1.0.4 DETAILS ON CATEGORY/TEMPLATE

Comment: see free text

Remark: The Alkyl Alcohols C6 to C13 Category is a family of saturated alcohols

that are produced from olefins by the hydroformylation or "oxo" process. Hydroformylation is the reaction of an olefin with carbon monoxide and hydrogen to produce an aldehyde, and its subsequent hydrogenation to the

alcohol.

The number of carbon atoms in the category members ranges from 6 to 13. Category members contain predominantly branched alkyl groups. Each substance consists of an isomeric mixture of saturated primary alcohols of high purity, and the following basic structure: CH3-R-CH2-OH, where R is a branched isomeric structure.

The justification for the Alkyl Alcohols C6 to C13 Category is that the members have:

- · similar chemical structures,
- · similar physico-chemical properties,
- · comparable environmental fate,
- · the same mode of action.

In general, aliphatic alcohol toxicity occurs by non-polar narcosis (Lipnick et al., 1985). The mode of action is disruption of biological membrane function (van Wezel and Opperhuizen, 1995). Metabolic pathways, through which alcohols are metabolized, are likely to include similar reactions for all category members and result in structurally similar metabolites (carboxylic acids).

The data demonstrate that the category is valid for a screening-level hazard assessment for the Category and its members. One can assess the untested endpoints by extrapolation between and among the category members.

20.03.2006

1.1.0 SUBSTANCE IDENTIFICATION

IUPAC Name : Smiles Code :

Molecular formula : C10H22O Molecular weight : 160.26

Petrol class

Flag : Critical study for SIDS endpoint

03.02.2005

1. General Information

ld 68526-85-2

Date

1.1.1 GENERAL SUBSTANCE INFORMATION

Purity type

Substance type : organic Physical status : liquid

Purity : Colour : Odour :

Remark: CAS Registry Number, Name, and General Structure for Members of the

Alkyl Alcohols C6 to C13 Category and Analogue Substances:

CAS RN: 85566-14-9

IUPAC Name: Alcohols, C7-11-branched and linear

R Length (C number): C7 to C11

Structure of R: Linear

Category Member: No (analogue substance used for supporting

information)

CAS RN: 68526-85-2

IUPAC Name: Alcohols C9-C11-iso, C10 rich

R Length (C number): C7

Structure of R: Branched, primarily methyl branched (isomeric structures)

Category Member: Yes

06.04.2006

1.1.2 SPECTRA

1.2 SYNONYMS AND TRADENAMES

Exxal(r) 10

15.03.2004

IDA

09.03.2004

Isodecanol

09.03.2004

Isodecyl alcohol

09.03.2004

1.3 IMPURITIES

Purity : typical for marketed substance

CAS-No : 68526-85-2 **EC-No** : 271-234-0

EINECS-Name : Alcohols, C9-11-iso-, C10-rich

Molecular formula : C9H20O2 Value : = 99.6 % w/w

Remark : Commercial product typically consists of trimethyl-1-heptanols and

1. General Information

ld 68526-85-2

Date

dimethyl-1-octanols.

06.01.2005

1.4 ADDITIVES

Purity type : typical for marketed substance

CAS-No :

EC-No :

EINECS-Name Molecular formula

Value Function of additive

Remark: No additives present.

11.10.2005

1.5 TOTAL QUANTITY

1.6.1 LABELLING

1.6.2 CLASSIFICATION

1.6.3 PACKAGING

1.7 USE PATTERN

Type of use : industrial

Category : Chemical industry: used in synthesis

Remark: Primarily used as a chemical intermediate in the production of plasticizers.

06.01.2005

1.7.1 DETAILED USE PATTERN

1.7.2 METHODS OF MANUFACTURE

1.8 REGULATORY MEASURES

1.8.1 OCCUPATIONAL EXPOSURE LIMIT VALUES

1.8.2 ACCEPTABLE RESIDUES LEVELS

Id 68526-85-2 1. General Information **Date** 06.04.2006 1.8.3 WATER POLLUTION 1.8.4 MAJOR ACCIDENT HAZARDS 1.8.5 AIR POLLUTION 1.8.6 LISTINGS E.G. CHEMICAL INVENTORIES 1.9.1 DEGRADATION/TRANSFORMATION PRODUCTS 1.9.2 COMPONENTS 1.10 SOURCE OF EXPOSURE 1.11 ADDITIONAL REMARKS 1.12 LAST LITERATURE SEARCH 1.13 REVIEWS

2. Physico-Chemical Data

ld 68526-85-2

Date

2.1 MELTING POINT

Value : = -40 °C Decomposition : no, at °C

Sublimation: no

Method : other: ASTM D97

Year : 2003 GLP : no data

Test substance :

Test substance : CAS No. 68526-85-2; Alcohols, C9-C11-iso, C10 rich

Reliability : (2) valid with restrictions

Although the original data were not retrieved and reviewed for quality, they

were developed following acceptable test methods and therefore

considered reliable.

Flag : Critical study for SIDS endpoint

06.04.2006 (1) (15)

2.2 BOILING POINT

Value : $= 216 - 226 \, ^{\circ}\text{C}$ at 1013 hPa

Decomposition

Method : other: ASTM D1078/01

Year : 2003 GLP : no data

Test substance :

Test substance : CAS No. 68526-85-2; Alcohols, C9-C11-iso, C10 rich

Reliability : (2) valid with restrictions

Although the original data were not retrieved and reviewed for quality, they

were developed following acceptable test methods and therefore

considered reliable.

Flag : Critical study for SIDS endpoint

06.04.2006 (15)

2.3 DENSITY

Type : density

Value : .837 g/cm³ at 20 °C

Method : other: ASTM D4052/86 equivalent

Year : 1998 **GLP** : yes

Test substance :

Method : The density was measured by taking the weight of 10 mls of the test

substance at 20 deg C.

Test substance : CAS No. 68526-85-2; Alcohols, C9-C11-iso, C10 rich

Reliability : (1) valid without restriction
Flag : Critical study for SIDS endpoint

06.04.2006 (9)

2.3.1 GRANULOMETRY

ld 68526-85-2

Date

2.4 VAPOUR PRESSURE

Value : = .018 hPa at 25 °C

Decomposition

Method : other (calculated)

Year : 2003 GLP : no data

Test substance

Method : Vapor pressure calculation by MPBPWIN ver. 1.40 using calculation

method of Grain.

Remark: EPIWIN is used and advocated by the US EPA for chemical property

estimation.

Test substance : CAS No. 68526-85-2; Alcohols, C9-C11-iso, C10 rich

Reliability : (2) valid with restrictions

The value was calculated based on chemical structure as modeled by EPIWIN. This robust summary has a reliability rating of 2 because the data

are calculated and not measured.

Flag : Critical study for SIDS endpoint

06.04.2006 (21)

2.5 PARTITION COEFFICIENT

Partition coefficient :

Log pow : = 4.2 - 4.3 at 30 °C

pH value

Method : OECD Guide-line 117 "Partition Coefficient (n-octanol/water), HPLC

Method"

Year : 1998 GLP : yes

Test substance :

Remark : Test Type: N-Octanol/Water Partition Coefficient (HPLC method)

Result: The test substance eluted as several groups. The two major components

C9, C10 alcohols had Log Pow values of 4.2 and 4.3, respectively.

The retention time for the 2 major components were 8.37, and 8.74

minutes.

All values were measured using High Performance Liquid Chromatography

HPI C)

Test condition: The test substance was evaluated as a solution in HPLC grade methanol.

Six reference compounds were also evaluated in a standard combined reference solution (2-butanone, acetophenone, naphthalene, biphenyl, n-butylbenzene, and 4,4-DDT) in 75% methanol and 25% distilled water.

The pH of the solution was 5.4.

Two customized alcohol reference solutions were also prepared containing five of the ten alcohol compounds (1-hexanol, 1-heptanol, 1-octanol, 1-nonanol, 1-decanol, 1-undecanol, 1-dodecanol, 1-tridecanol, 1-tetradecanol, 1-pentadecanol) in 87.5% methanol and 12.5% distilled

water. The pH of both solutions was 7.3.

The pH of the evaluated solutions was the same as the reference solution it

was evaluated against.

The test substance was analyzed against a Standard Log Pow Reference Compound Solution and a customized Alcohol Reference Compound Solution. Only the peaks detected by refractive index (RI) were reported.

2. Physico-Chemical Data

ld 68526-85-2

Date

: CAS No. 68526-85-2; Alcohols, C9-C11-iso, C10 rich

Test substance Reliability : (1) valid without restriction : Critical study for SIDS endpoint Flag

06.04.2006 (12)

2.6.1 SOLUBILITY IN DIFFERENT MEDIA

Solubility in : Water

Value = 75 mg/l at 20 °C

pH value

at °C concentration

Temperature effects

Examine different pol.

at 25 °C

Description

Stable

Deg. product

Method other: Slow stir

Year 1998 **GLP** no data

Test substance

Method Slow-stir water solubility vessels consisted of glass aspirator bottles with

capacities of 4 to 12 L. The spigot at the bottom of the vessel was fitted with short length of Tefzel tubing and a #13 glass stopper. The bottle was rinsed with a mixture of 1:1 methylene chloride: acetone followed by 2,2,4trimethyl pentane. The bottle was then air dried in a laboratory fume hood

and rinsed three times with dilution water.

A glass stir bar was added to the bottle and then the bottle was filled with the appropriate amount of water. The test substance was added to the bottle at a loading of 100 mg/L and the solution stirred quiescently with little

or no visible vortex on a magnetic stir-plate.

Mixing was stopped one hour prior to sampling. Samples were removed through the outlet at the bottom of the vessel. To avoid losses and to prevent contamination, the samples were analyzed immediately by either GC-MSD or GC-FID. Samples were taken on Days 1, 3, 7, and 21.

Remark Measured value obtained in conjunction with Fish Acute Toxicity Test.

: CAS No. 68526-85-2; Alcohols, C9-C11-iso, C10 rich Test substance

(1) valid without restriction Reliability Flag Critical study for SIDS endpoint

06.04.2006 (18)

2.6.2 SURFACE TENSION

2.7 FLASH POINT

 $= 97 \, ^{\circ}\text{C}$ Value : closed cup Type

Method : other: PMCC ASTM D93

Year 2003 **GLP** no data

Test substance

Test substance : CAS No. 68526-85-2; Alcohols, C9-C11-iso, C10 rich

Reliability : (2) valid with restrictions

2. Physico-Chemical Data

ld 68526-85-2 **Date** 06.04.2006

Although the original data were not retrieved and reviewed for quality, they were developed following acceptable test methods and therefore

considered reliable.

06.04.2006 (15)

2.8 AUTO FLAMMABILITY

2.9 FLAMMABILITY

2.10 EXPLOSIVE PROPERTIES

2.11 OXIDIZING PROPERTIES

2.12 DISSOCIATION CONSTANT

Acid-base constant : = 15.89 to 17.52 at 25°C

Method : other: calculated

Year : 2005 GLP : no data

Test substance :

Method: pKa calculation by SPARC 2003 using a Linux calculation engine.

Remark : SPARC On-line calculator can be accessed at

http://ibmlc2.chem.uga.edu/sparc/index.cfm

Test substance : CAS No. 68526-85-2; Alcohols, C9-C11-iso, C10 rich

Reliability : (2) valid with restrictions

The value was calculated based on chemical structure as modeled by SPARC. This robust summary has a reliability rating of 2 because the data

are calculated and not measured.

06.04.2006 (17)

2.13 VISCOSITY

Value : $= 21 - at 20 \, ^{\circ}C$

Result

Method : other: ASTM D445

Year : 2003 GLP : no data

Test substance

Remark : Value measured in cSt

Test substance : CAS No. 68526-85-2; Alcohols, C9-C11-iso, C10 rich

Reliability : (2) valid with restrictions

Although the original data were not retrieved and reviewed for quality, they

were developed following acceptable test methods and therefore

considered reliable.

06.04.2006 (15)

2.14 ADDITIONAL REMARKS

ld 68526-85-2

Date

3.1.1 PHOTODEGRADATION

Type : water

Light source

Light spectrum: nm

Relative intensity : based on intensity of sunlight

Deg. product

Method : other (calculated): Technical Discussion

Year : GLP : Test substance :

Remark: These data represent a key study for characterising the potential of

substances in the Alkyl Alcohols C6 to C13 Category to undergo direct

photodegradation.

Result: Photolysis as a Function of Molecular Structure

The direct photolysis of an organic molecule occurs when it absorbs sufficient light energy to result in a structural transformation (Harris, 1982). The reaction process is initiated when light energy in a specific wavelength range elevates a molecule to an electronically excited state. However, the excited state is competitive with various deactivation processes that can result in the return of the molecule to a non excited state.

The absorption of light in the ultra violet (UV)-visible range, 110-750 nm, can result in the electronic excitation of an organic molecule. Light in this range contains energy of the same order of magnitude as covalent bond dissociation energies (Harris, 1982). Higher wavelengths (e.g. infrared) result only in vibrational and rotational transitions, which do not tend to produce structural changes to a molecule.

The stratospheric ozone layer prevents UV light of less than 290 nm from reaching the earth's surface. Therefore, only light at wavelengths between 290 and 750 nm can result in photochemical transformations in the environment (Harris, 1982). Although the absorption of UV light in the 290-750 nm range is necessary, it is not always sufficient for a chemical to undergo photochemical degradation. Energy may be re-emitted from an excited molecule by mechanisms other than chemical transformation, resulting in no change to the parent molecule.

A conservative approach to estimating a photochemical degradation rate is to assume that degradation will occur in proportion to the amount of light wavelengths >290 nm absorbed by the molecule (Zepp and Cline, 1977).

Substances in the Oxo Alcohols C6 to C13 Category contain molecules that are oxygenated aliphatic compounds which will absorb UV light below 220 nm (Boethling and Mackay, 2000) and will not undergo direct photolysis. Therefore, this fate process will not contribute to a measurable degradative removal of chemical components in this category from the environment.

References

Boethling, R.S., Mackay, D. (2000). Handbook of Property Estimation Methods for Chemicals. CRC Press, Boca Raton, FL, USA.

Harris, J. C. 1982. "Rate of Aqueous Photolysis," Chapter 8 in: W. J. Lyman, W. F. Reehl, and D. H. Rosenblatt, eds., Handbook of Chemical

Date

Property Estimation Methods, McGraw-Hill Book Company, New York,

USA.

Zepp, R. G. and D. M. Cline. 1977. Rates of Direct Photolysis in the

Aqueous Environment, Environ. Sci. Technol., 11:359-366.

Test substance : Alkyl Alcohols C6 to C13 Category members

Flag : Critical study for SIDS endpoint

03.04.2006 (14)

Туре

Light source : Sun light Light spectrum : nm

Relative intensity: based on intensity of sunlight

INDIRECT PHOTOLYSIS

Sensitizer : OH

Conc. of sensitizer : 1500000 molecule/cm³

Rate constant : = $.0000000000212981 \text{ cm}^3/(\text{molecule*sec})$

Degradation: % after

Deg. product :

Method : other (calculated): Calculated values using AOPWIN version 1.89, a

subroutine of the computer program EPIWIN version 3.04

Year : 1999 GLP : no Test substance :

Result: Atmospheric Oxidation Potential

In the environment, organic chemicals emitted into the troposphere are degraded by several important transformation processes. The dominant transformation process for most compounds is the daylight reaction with hydroxyl (OH-) radicals (Atkinson, 1988, 1989). The rate at which an organic compound reacts with OH- radicals is a direct measure of its atmospheric persistence (Meylan and Howard, 1993).

AOPWIN estimates the rate constant for the atmospheric, gas-phase reaction between photochemically produced hydroxyl radicals and organic chemicals. The rate constants estimated by the program are then used to calculate atmospheric half-lives for organic compounds based upon average atmospheric concentrations of hydroxyl radicals.

Since the reactions only take place in the presence of sunlight, the atmospheric half-lives are normalized for a 12-hour day.

Calculated* OH- Rate Constant half-life (hrs) (cm3/molecule-sec)

6.0 21.30 E-12

References:

Atkinson, R. 1988. Estimation of gas-phase hydroxyl radical rate constants for organic chemicals. Environ. Toxicol. Chem. 7:435-442.

Atkinson, R. 1989. Kinetics and mechanisms of the gas-phase reactions of the hydroxyl radical with organic compounds. J. Phys. Chem. Ref. Data Monograph No. 1, Amer. Inst. Physics & Amer. Chem. Soc., NY.

Meylan, W.M. and P.H. Howard. 1993. Computer estimation of the atmospheric gas-phase reaction rate of organic compounds with hydroxyl radicals and ozone. Chemosphere 12:2293-2299.

ld 68526-85-2

Date

Test condition: Indirect photodegradation, or atmospheric oxidation potential, is based on

the structure-activity relationship methods developed by R. Atkinson.

Temperature: 25°C Sensitizer: OH radical

Concentration of Sensitizer: 1.5 E6 OH radicals/cm3 CAS No. 68526-85-2: Alcohols. C9-C11-iso. C10 rich

Reliability : (2) valid with restrictions

The results include calculated data based on chemical structure as modeled by AOPWIN. The data represent a potential atmospheric half-life range for the test substance. This robust summary represents a "key

study" for atmospheric half-life range based on constituent data.

Flag : Critical study for SIDS endpoint

06.04.2006 (22)

3.1.2 STABILITY IN WATER

Test substance

 Type
 : abiotic

 t1/2 pH4
 : at °C

 t1/2 pH7
 : at °C

 t1/2 pH9
 : at °C

Deg. product

Method : other: Technical Discussion

Year : GLP :

Test substance :

Remark: These data represent a key study for characterising the potential of

substances in the Alkyl Alcohols C6 to C13 Category to undergo

hydrolysis.

Result : Hydolysis as a Function of Molecular Structure

Hydrolysis of an organic molecule occurs when a molecule (R-X) reacts with water (H2O) to form a new carbon-oxygen bond after the carbon-X bond is cleaved (Gould, 1959; Harris, 1982). Mechanistically, this reaction is referred to as a nucleophilic substitution reaction, where X is the leaving group being replaced by the incoming nucleophilic oxygen from the water molecule.

Chemicals that are susceptible to hydrolysis contain functional groups that can be displaced by a nucleophilic substitution reaction. Substances that have the potential to hydrolyze include alkyl halides, amides, carbamates, carboxylic acid esters and lactones, epoxides, phosphate esters, and sulfonic acid esters (Neely, 1985). The lack of a leaving group renders a compound resistant to hydrolysis.

Alkyl alcohols are resistant to hydrolysis because they lack a functional group that is hydrolytically reactive (Harris, 1982).

References:

Gould, E.S. (1959). Mechanism and Structure in Organic Chemistry, Holt, Reinhart and Winston, New York, NY, USA.

Harris, J.C. (1982). "Rate of Hydrolysis," Chapter 7 in: W.J. Lyman, W.F. Reehl, and D.H. Rosenblatt, eds., Handbook of Chemical Property Estimation Methods, McGraw-Hill Book Company, New York, NY, USA.

Neely, W.B. (1985). Hydrolysis. In: W.B. Neely and G.E. Blau, eds., Environmental Exposure from Chemicals. Vol. 1, pp. 157-173. CRC Press,

ld 68526-85-2

Date

Boca Raton, FL, USA.

Conclusion: Hydrolysis will not contribute to the removal of Alkyl Alcohols from the

environment.

Flag : Critical study for SIDS endpoint

20.03.2006 (13)

3.1.3 STABILITY IN SOIL

3.2.1 MONITORING DATA

3.2.2 FIELD STUDIES

3.3.1 TRANSPORT BETWEEN ENVIRONMENTAL COMPARTMENTS

Type : fugacity model level III

Media : other: air - sediment(s) - soil - water

Air : % (Fugacity Model Level I)

Water : % (Fugacity Model Level I)

Soil : % (Fugacity Model Level I)

Biota : % (Fugacity Model Level II/III)

Soil : % (Fugacity Model Level II/III)

Method : other: Calculation according Mackay, Level III

Year : 2003

Method : The EQC Level III model is a steady state model that is useful for

determining how the medium of release affects environmental fate. Level III fugacity allows non-equilibrium conditions to exist between connected media as steady state, and illustrate important transport and transformation

processes.

Physicochemical input values for the model were calculated using the EPIWIN Estimation v 3.04 program. Measured input values were also used where available and obtained from the EPIWIN database. Distribution data from the equilibrium model provide basic information on the potential partitioning behavior of chemicals between selected environmental compartments (i.e., air, water, soil, and sediment).

Input values used:

Molecular mass = 160.26 g/mol Water solubility = 75 mg/L Vapour pressure = 1.8 Pa

log Kow = 4.2

Melting point = -40 deg C

Degradation half-lives:

Air - 6.0 hrs Water - 240 hrs Soil - 1440 hrs Sediment - 14400 hrs

This model was run assuming 100% discharge to water.

Result : Air - 0.2%

Water - 66.0% Soil - 0.03%

ld 68526-85-2

Date

Sediment - 33.8%

Test substance : CAS No. 68526-85-2; Alcohols, C9-C11-iso, C10 rich

Reliability : (2) valid with restrictions

This robust summary has a reliability rating of 2 because the data are

calculated and not measured.

Flag : Critical study for SIDS endpoint

25.10.2005 (20)

Type :

Media : other: air - biota - sediment(s) - soil - water

Air : % (Fugacity Model Level I)
Water : % (Fugacity Model Level I)
Soil : % (Fugacity Model Level I)
Biota : % (Fugacity Model Level II/III)
Soil : % (Fugacity Model Level II/III)

Method : other: Calculation according Mackay, Level I

Year : 2003

Method : The EQC Level I is a steady state, equilibrium model that utilizes the input

of basic chemical properties including molecular weight, vapor pressure, and water solubility to calculate distribution within a standardized regional

environment.

Physicochemical input values for the model were calculated using the EPIWIN Estimation v 3.04 program. Measured input values were also used where available and obtained from the EPIWIN database. Distribution data from the equilibrium model provide basic information on the potential partitioning behavior of chemicals between selected environmental

compartments (i.e., air, water, soil, sediment, suspended sediment, biota).

Input values used:

Molecular mass = 160.26 g/mol Water solubility = 75 mg/L Vapour pressure = 1.8 Pa

log Kow = 4.2

Melting point = -40 deg C

Remark: Results were calculated using measured value for log Kow

Result : Air - 4.8%

Water - 6.2% Soil - 87.0% Sediment - 1.9%

Suspended Sed - 0.06%

Biota - < 0.01%

Test substance : CAS No. 68526-85-2; Alcohols, C9-C11-iso, C10 rich

Reliability : (2) valid with restrictions

This robust summary has a reliability rating of 2 because the data are

calculated and not measured.

25.10.2005 (20)

Type : volatility
Media : water - air

Air : % (Fugacity Model Level I)

Water : % (Fugacity Model Level I)

Soil : % (Fugacity Model Level I)

Biota : % (Fugacity Model Level II/III)

Soil : % (Fugacity Model Level II/III)

Method : other: calculation

Year : 2005

Method : Henry's Law Constants (HLC) are based on vapor pressure and water

solubility values, and molecular weights. HLC values were calculated

using equations found in Mackay, 2003.

ld 68526-85-2

Date

Vapor pressure = 1.8 Pa Water solubility = 75 mg/l Molecular mass = 160.26 g/mol HLC (at 25 deg C) = 3.80 Pa-m3/mole

Result : HLC (at 25 deg C) = 3.80 Pa-m3/mole **Test substance** : CAS No. 68526-85-2; Alcohols, C9-C11-iso, C10 rich

Reliability : (2) valid with restrictions

This robust summary has a reliability rating of (2) because the data are

calculated and not measured.

Flag : Critical study for SIDS endpoint

26.10.2005 (20)

3.3.2 DISTRIBUTION

Media : other: Koc Method : other (calculation)

Year : 2005

Method : The soil adsorption coefficient (Koc) was calculated by PCKOCWIN ver.

1.66, using equations developed by Lyman (1980).

PCKOCWIN is a subroutine of EPIWIN. EPIWIN is used and advocated by the USEPA for chemical property estimation. Koc provides an indication of the extent to which a chemical partitions between solid and solution phases

in soil, or between water and sediment in aquatic ecosystems.

Result : Koc = 67.19

 $\log Koc = 1.827$

Test substance : CAS No. 68526-85-2; Alcohols, C9-C11-iso, C10 rich

Reliability : (2) valid with restrictions

This robust summary has a reliability rating of 2 because the data are

calculated and not measured.

03.11.2005 (19) (23)

3.4 MODE OF DEGRADATION IN ACTUAL USE

3.5 BIODEGRADATION

Type : aerobic

Inoculum : activated sludge, domestic

Contact time

Degradation : = 71.1 (±) % after 28 day(s)

Result : other

Deg. product

Method : OECD Guide-line 301 F "Ready Biodegradability: Manometric

Respirometry Test"

Year : 1997 GLP : yes Test substance :

Remark: Test Type: Manometric Respirometry Test

Exposure Period: 28 days

Result : Test material is rapidly biodegradable. Half-life was reached by day 11. By

day 28, 71.1% degradation of the test material was observed. 10% biodegradation was achieved on day 4 and 60% biodegradation was

achieved on day 17.

By day 14, >60% biodegradation of positive control was observed, which met the guideline requirement. No excursions from the protocol were

ld 68526-85-2

Date

noted.

Biodegradation was based on oxygen consumption and the theoretical oxygen demand of the test material as calculated using results of an elemental analysis of the test material.

% Degradation* Mean % Degradation

Sample (day 28) (day 28) Test Material 74.0, 72.6, 66.5 71.1 Na Benzoate 91.3, 81.3 86.3

* replicate data

Test condition: Non acclimated activated sludge and test medium were combined prior to

test material addition. Test medium consisted of glass distilled water and mineral salts (Phosphate buffer, Ferric chloride, Magnesium sulfate,

Calcium chloride).

Test vessels were 1L glass flasks placed in a waterbath and electronically

monitored for oxygen consumption.

Test material was tested in triplicate, controls and blanks were tested in

duplicate.

Test material concentration was approximately 43 mg/L. Sodium benzoate

(positive control) concentration was 44mg/L. Test temperature was 22 +/- 1 Deg C.

All test vessels were stirred constantly for 28 days using magnetic stir bars

and plates.

Test substance : CAS No. 68526-85-2; Alcohols, C9-C11-iso, C10 rich

Conclusion: Test substance is considered rapidly biodegradable. Although the test

substance (a complex and poorly water soluble material) achieved a ready extent of biodegradability, it did not achieve this extent within a 10-day

window.

Reliability : (1) valid without restriction Flag : Critical study for SIDS endpoint

06.04.2006 (10)

3.6 BOD5, COD OR BOD5/COD RATIO

3.7 BIOACCUMULATION

Species: Oncorhynchus mykiss (Fish, fresh water)

Exposure period : 16 day(s) at 14.6 °C

Concentration

BCF : 15

Elimination :

Method : OECD Guide-line 305

Year : 1998 **GLP** : yes

Test substance :

Remark :

No difference in mortality or growth was observed between the treatments and the control after 16 days of exposure, or at the end of the study (26

days).

Duration: 16 days uptake, 10 days depuration

The duration of the uptake phase was shorter than the guideline recommended duration of 28 days. The duration of the uptake phase was determined based on the concentration of the test substance reaching a steady state in the tissue of the fish. Samples of the fish were analyzed at regular intervals. To determine the concentration of the test substance in

ld 68526-85-2

Date

the fish tissue, under steady-state conditions, one-tail T-tests for the comparison of the means at the sampling times were performed.

Calculated Bioconcentration Factor (BCF):

0.75 mg/L = 14.80.15 mg/L = 15.5

Mean BCF for the study = 15.2

Test condition

Result

A 100 mg/L stock solution was prepared by adding the appropriate amount of the test substance to 18 liters of dilution water. The stock solution was mixed for 24 hours prior to its use in the test. It was then pumped from the mixing vessel via a glass tube, silicon tubing and varistaltic pump and delivered to the test chambers via a 2 liter proportional diluter. Two concentrations were prepared for the test, 0.75 mg/L and 0.15 mg/L. A dilution water control was also prepared. A new stock solution was prepared in this manner every 4 days and mixed for 24 hours prior to use in the test.

One test chamber was prepared for each treatment and the control. At the start of the test each test chamber contained 45 fish. Fish were observed daily for mortality and/or abnormal behavior or appearance. Water and fish samples were removed periodically during both the uptake and depuration phases of the study. Samples were analyzed by gas chromatography.

Mean measured concentrations of the test substance in water:

0.75 mg/L = 0.91 mg/L (s.d.= 0.15)0.15 mg/L = 0.16 mg/L (s.d.= 0.02)

The test substance was not detected in the control during the study or in either treatment during the depuration phase.

Mean measured concentrations of the test substance in fish tissue:

0.75 mg/L = 13.8 mg/kg

0.15 mg/L = 2.5 mg/kg

Fish samples were also removed at the beginning and end of the uptake phase and analyzed for lipid content. The mean percent lipid content was 3.53%.

Test temperature was 14.6 Deg C., Lighting was 16 hours light: 8 hours dark with 438 to 612 Lux during full daylight periods.

Dissolved Oxygen ranged from 8.7 to 9.3 mg/L. The pH was ranged from 7.1 to 7.3 during the study. Fish were fed at a rate of 2% of their bodyweight per day during the study.

Fish Mean Weight was approximately = 0.367g at the start of the test. Test Loading = 0.47 g of fish/L.

The following data represent the measurements of the test chemical in fish tissue. Units are in μg of test chemical per gram of fish. The Non Detect (ND) level was 0.2 $\mu g/g$, which corresponds to 0.67 $\mu g/g$ for a 0.3 g fish and 0.14 $\mu g/g$ for a 1.4 g fish mass.

Fish Tissue Analysis

		Uptake Phase				
	Day 1	Day 2	Day 4	Day 10	Day 16	
	ND	ND	ND	ND	ND	
Control	ND	ND	ND	ND	ND	
	ND	ND	ND	ND	ND	
	4.4	NC	4.0	0.7	4 5	
0.15mg/l	1.1	NS 1.2	1.2	2.7	1.5	
0.15mg/l	1.0	1.2	1.1	2.3	1.8	
	mean	1.1	1.2	1.2	2.5	1.7
	5.8	5.9	7.4	19	8.2	
0.75mg/l	7.2	2.8	8.6	13	15	

Id 68526-85-2

Date

mean 6.5 4.4 8.0 16.0 11.6

Depuration Phase Day 0 Day 1 Day 4 ND ND ND Control ND ND ND ND ND ND 0.35 ND ND 0.15mg/l 0.57 ND ND ND 0.31 mean 0.41 ND ND 2.0 0.40 ND 0.75mg/l 2.1 0.17 ND 4.2 0.30 ND mean 2.8 0.29 ND

Test substance : CAS No. 68526-85-2; Alcohols, C9-C11-iso, C10 rich

Conclusion: Low potential to bioaccumulate. Value based on wet weight data.

Reliability : (1) valid without restriction
Flag : Critical study for SIDS endpoint

06.04.2006 (11)

3.8 ADDITIONAL REMARKS

4. Ecotoxicity Id 68526-85-2

Date

4.1 ACUTE/PROLONGED TOXICITY TO FISH

Type : semistatic

Species : Oncorhynchus mykiss (Fish, fresh water)

Exposure period : 96 hour(s)
Unit : mg/l

LC50 : = 3.1 measured/nominal

Limit test : no Analytical monitoring : yes

Method : OECD Guide-line 203 "Fish, Acute Toxicity Test"

Year : 1995 GLP : yes Test substance :

Method : Probit procedure of SAS (Finney, 1971)

Result: LC50 = 3.1mg/L (CI 2.4 to 4.0), based upon measured concentrations of

mean of old and new samples.

Analytical method used was GC-FID

LL50 = 3.0 mg/L (Could not calculate CI), based upon nominal loading levels.

Nominal Conc. Measured Conc. % Mortality @ 96 hr.

Control	Below detection	7
1.2 mg/L	1.2 mg/L	13
2.5 mg/L	2.4 mg/L	13
5 mg/L	5.2 mg/L	100
10 mg/L	9.9 mg/L	100
20 mg/L	19.5 mg/L	100

Dissolved oxygen levels dropped below 60% (57%)of saturation in the 2.4 mg/L treatment on Days 3 and 4 of the test. Since only 13% mortality occurred at this level, and the solutions were renewed daily, this drop in

DO did not affect the outcome of the study.

Test condition

Individual Water Accomodated Fractions (WAF's) were prepared for each test treatment. The test substance was added volumetrically, via a syringe, to 19.5L of dilution water in a 20L glass carboy. The carboys were covered with an opaque covering to prevent photochemical degradation of the soluble components. The solutions were mixed for 24 hours at a vortex of </= 10% of the total depth. The test solutions were pumped from each mixing vessel into three replicates of 4.5L in 4.0L glass aspirator bottles (no headspace). Five fish were added to each test replicate and the replicates sealed. Daily renewals were performed by removing ~80% of the test solution through the port at the bottom and refilling with fresh solution.

Test temperature was 15.0 Deg C., Lighting was 16 hours light: 8 hours dark with 569 to 572 Lux during full daylight periods.

Dissolved Oxygen at initiation ranged from 8.4 to 9.9 mg/L and from 5.7 to 7.6 mg/L in "old" solutions prior to renewals. The pH was ranged from 7.0 to 8.5 during the study. Fish were not fed during the study.

Fish Mean Wt.= 0.185g. Mean Total length = 3.0cm, Test Loading = 0.21 g of fish/L.

Test substance : CAS No. 68526-85-2; Alcohols, C9-C11-iso, C10 rich

Conclusion: Test substance is considered to have moderate acute toxicity.

Reliability : (1) valid without restriction
Flag : Critical study for SIDS endpoint

06.04.2006 (9)

4. Ecotoxicity Id 68526-85-2

Date

Type :

Species: other: freshwater fish

Exposure period : 96 hour(s)
Unit : mg/l

LC50 : = 3.1 calculated

Method : other: ECOSAR Computer Model

Year : 2005

GLP

Test substance :

Test condition : Log Kow (octanol/water partition coefficient) values and a chemical

structure are needed to calculate aquatic toxicity using the ECOSAR model. The Kow calculation is performed by KOWWIN based on an atom/fragment contribution method of Meylan and Howard (1), which is a subroutine in the EPIWIN computer model (2). KOWWIN also has a

database of experimental Kow values (EXPKOW.DB).

The ECOSAR program was run using a C10 alcohol with a Kow of 3.68.

1. Meylan, W. and P. Howard. 1995. Atom/fragment contribution method for estimating octanol-water partition coefficients. J. Pharm. Sci. 84:83-92.

2. Meylan, M., SRC 1994-1999. KOWWIN is contained in the computer program EPIWIN. 1999. Estimation Program Interface for Windows, version 3.04. Syracuse Research Corporation, Syracuse, NY, USA.

Test substance : CAS No. 68526-85-2; Alcohols, C9-C11-iso, C10 rich

Conclusion : Based on the calculated Kow value, the C10 alcohol is expected to have an

acute 96-hour LC50 of 3.1 mg/L and a Chronic Value of 0.5 mg/L.

Reliability : (2) valid with restrictions

The value was calculated based on chemical structure as modeled by EPIWIN. This robust summary has a reliability rating of 2 because the data

are calculated and not measured.

06.04.2006

4.2 ACUTE TOXICITY TO AQUATIC INVERTEBRATES

Type : static

Species : Daphnia magna (Crustacea)

Exposure period : 48 hour(s)
Unit : mg/l

EC50 : = 6.2 measured/nominal

Limit Test : no Analytical monitoring : no

Method : other: Daphnid Acute Toxicity Test

Year : 1982
GLP : no data
Test substance : other TS

Method : Methods for acute toxicity tests with fish, macroinvertebrates and

amphibians (EPA-660/3-75-009) Moving average angle method.

Result : 48-hour EC50 = 6.2 mg/L (95% CI 5.3-7.6 mg/L), based upon nominal

concentrations.

Test condition : Individual treatments were prepared by adding the appropriate amount of

the test substance to 1000 ml of dilution water and mixing vigorously on a magnetic stirrer for 4 hours. Four hundred and fifty milliliters of this test solution were divided into three replicate beakers, each containing 150 ml. A set of control beakers containing the same dilution water and maintained

under the same conditions was also prepared.

Date

Fifteen daphnids were exposed to each test concentration (5 per replicate). Daphnids were <24 hours old and were impartially distributed to the test chambers. Mortalities were recorded at 24 and 48 hours after initiation of the exposures. Biological observations were also made at this time.

Test temperature was 21-23 Deg C., Dissolved oxygen ranged from 7.7 to 8.4 mg/L during the study. The pH was ranged from 8.1 to 8.7 during the

studv.

Test substance : CAS No. 68526-85-2; Alcohols, C9-C11-iso, C10 rich

Conclusion : Test substance is considered to have moderate acute toxicity.

Reliability : (2) valid with restrictions

Study followed acceptable study guidelines and was well documented. Analytical verification of test concentrations not test substance specific.

Flag : Critical study for SIDS endpoint

06.04.2006 (4)

Type :

Species : Daphnia sp. (Crustacea)

Exposure period : 48 hour(s)
Unit : mg/l

EC50 : = 3.7 calculated

Method : other: ECOSAR Computer Model

Year : 2005

GLP : Test substance :

Test condition: Log Kow (octanol/water partition coefficient) values and a chemical

structure are needed to calculate aquatic toxicity using the ECOSAR model. The Kow calculation is performed by KOWWIN based on an atom/fragment contribution method of Meylan and Howard (1), which is a subroutine in the EPIWIN computer model (2). KOWWIN also has a

database of experimental Kow values (EXPKOW.DB).

The ECOSAR program was run using a C10 alcohol with a Kow of 3.68.

1. Meylan, W. and P. Howard. 1995. Atom/fragment contribution method for estimating octanol-water partition coefficients. J. Pharm. Sci. 84:83-92.

2. Meylan, M., SRC 1994-1999. KOWWIN is contained in the computer program EPIWIN. 1999. Estimation Program Interface for Windows, version 3.04. Syracuse Research Corporation, Syracuse, NY, USA.

Test substance : CAS No. 68526-85-2; Alcohols, C9-C11-iso, C10 rich

Conclusion : Based on the calculated Kow value, the C10 alcohol is expected to have an

acute 48-hour EC50 of 3.7 mg/L and a Chronic Value of 0.4 mg/L.

Reliability : (2) valid with restrictions

The value was calculated based on chemical structure as modeled by EPIWIN. This robust summary has a reliability rating of 2 because the data

are calculated and not measured.

06.04.2006

4.3 TOXICITY TO AQUATIC PLANTS E.G. ALGAE

Species: other algae: Pseudokirchneriella subcapitata

Endpoint

Exposure period : 96 hour(s)
Unit : mg/l

EC50 : = 2.6 calculated **ChV** : = .68 calculated

Method : other: ECOSAR Computer Model

4. Ecotoxicity Id 68526-85-2

Date

Year : 2005

GLP : Test substance :

Test condition : Log Kow (octanol/water partition coefficient) values and a chemical

structure are needed to calculate aquatic toxicity using the ECOSAR model. The Kow calculation is performed by KOWWIN based on an atom/fragment contribution method of Meylan and Howard, which is a subroutine in the EPIWIN computer model. KOWWIN also has a database

of experimental Kow values (EXPKOW.DB).

The ECOSAR program was run using a C10 alcohol with a Kow of 3.68.

Test substance : CAS No. 68526-85-2; Alcohols, C9-C11-iso, C10 rich

Conclusion: Based on the calculated Kow value, the C10 alcohol is expected to have an

acute 96-hour EC50 of 2.6 mg/L and a Chronic Value of 0.68 mg/L.

Reliability : (2) valid with restrictions

The value was calculated based on chemical structure as modeled by EPIWIN. This robust summary has a reliability rating of 2 because the data

are calculated and not measured.

06.04.2006 (3)

4.4 TOXICITY TO MICROORGANISMS E.G. BACTERIA

4.5.1 CHRONIC TOXICITY TO FISH

Species: other: Freshwater Fish (calculated toxicity values are not species specific)

Endpoint : other: LC50
Exposure period : 30 day(s)
Unit : mg/l

ChV : = .492 calculated Method : other: calculated

Year : 2005

GLP :

Test substance :

Test condition: Log Kow (octanol/water partition coefficient) values and a chemical

structure are needed to calculate aquatic toxicity using the ECOSAR model. The Kow calculation is performed by KOWWIN based on an atom/fragment contribution method of Meylan and Howard, which is a subroutine in the EPIWIN computer model. KOWWIN also has a database

of experimental Kow values (EXPKOW.DB).

The ECOSAR program was run using a C10 alcohol with a Kow of 3.68.

Test substance Conclusion : CAS No. 68526-85-2; Alcohols, C9-C11-iso, C10 rich

: Based on the calculated Kow value, the C10 alcohol is expected to have a

30-day Chronic Value of 0.52 mg/L.

Reliability : (2) valid with restrictions

The value was calculated based on chemical structure as modeled by ECOSAR. This robust summary has a reliability rating of 2 because the

data are calculated and not measured.

Flag : Critical study for SIDS endpoint

06.04.2006 (3)

4.5.2 CHRONIC TOXICITY TO AQUATIC INVERTEBRATES

Species : Daphnia sp. (Crustacea)

Endpoint : mortality

4. Ecotoxicity Id 68526-85-2
Date 06.04.2006

Exposure period : 16 day(s) **Unit** : mg/l

EC50 : = .4 calculated

Method : other: ECOSAR Computer Model

Year : 2005 GLP : no Test substance :

Test condition: Log Kow (octanol/water partition coefficient) values and a chemical

structure are needed to calculate aquatic toxicity using the ECOSAR model. The Kow calculation is performed by KOWWIN based on an atom/fragment contribution method of Meylan and Howard, which is a subroutine in the EPIWIN computer model. KOWWIN also has a database

of experimental Kow values (EXPKOW.DB).

The ECOSAR program was run using a C10 alcohol with a Kow of 3.68.

Test substance : CAS No. 68526-85-2; Alcohols, C9-C11-iso, C10 rich

Conclusion: Based on the calculated Kow value, the C10 alcohol is expected yo have a

16-day EC50 of 0.40 mg/L.

Reliability : (2) valid with restrictions

The value was calculated based on chemical structure as modeled by EPIWIN. This robust summary has a reliability rating of 2 because the data

are calculated and not measured.

Flag : Critical study for SIDS endpoint

06.04.2006 (3)

4.6.1 TOXICITY TO SEDIMENT DWELLING ORGANISMS

4.6.2 TOXICITY TO TERRESTRIAL PLANTS

4.6.3 TOXICITY TO SOIL DWELLING ORGANISMS

Type :

Species: other: Earthworm

Endpoint : mortality
Exposure period : 16 day(s)
Unit : mg/kg soil dw
LC50 : = 295.8 calculated

Method : other: ECOSAR Computer Model

Year : 2005

GLP : Test substance :

Test condition : Log Kow (octanol/water partition coefficient) values and a chemical

structure are needed to calculate acute terrestrial toxicity using the ECOSAR model. The Kow calculation is performed by KOWWIN based on an atom/fragment contribution method of Meylan and Howard, which is a subroutine in the EPIWIN computer model. KOWWIN also has a database

of experimental Kow values (EXPKOW.DB).

The ECOSAR program was run using a C10 alcohol with a Kow of 3.68.

Test substance : CAS No. 68526-85-2; Alcohols, C9-C11-iso, C10 rich

Conclusion: Based on the calculated Kow value, the C10 alcohol is expected to have a

16-day EC50 of 295.8 mg/kg soil.

Reliability : (2) valid with restrictions

The value was calculated based on chemical structure as modeled by EPIWIN. This robust summary has a reliability rating of 2 because the data

4. Ecotoxicity **Id** 68526-85-2 **Date** are calculated and not measured. : Critical study for SIDS endpoint Flag 06.04.2006 (3) 4.6.4 TOX. TO OTHER NON MAMM. TERR. SPECIES 4.7 **BIOLOGICAL EFFECTS MONITORING BIOTRANSFORMATION AND KINETICS** 4.9 **ADDITIONAL REMARKS**

Date

5.0 TOXICOKINETICS, METABOLISM AND DISTRIBUTION

5.1.1 ACUTE ORAL TOXICITY

Type : LD50

Value : = 4626 mg/kg bw

Species : rat

Strain : Sprague-Dawley

Sex : male Number of animals : 5

Vehicle : other: corn oil

Doses : 26, 82, 260, 820, 2600, 8200 mg/kg bw

Method: otherYear: 1960GLP: noTest substance:

Remark

5/5 animals died within the first four hours following exposure to 8200 mg/kg. Animals in all other dose groups survived until the end of the study. At the one and four-hour intervals, animals in the 260 and 820 mg/kg dose groups were inactive and displayed labored respiration, ataxia, and sprawling of the limbs. At the 24-hour interval, animals had oily fur. After approximately 48-hours after dosing, most animals in these groups returned to normal appearance and behavior. At the 2600 mg/kg dose level, animals exhibited similar symptoms as above but also showed lacrimation and depressed righting and placement reflexes. Animals in this dose group also returned to normal appearance and behavior after 24 hours. At the highest dose, animals initially exhibited labored respiration, ataxia, and sprawling of the limbs, which was followed by a comatose state and death within 4 hours of exposure.

The surviving animals at the five lower dose levels (26, 82, 260, 820, 2600 mg/kg) had weight gain that was within the normal range. Gross autopsies performed on animals that died (5/5 in 8200 mg/kg group) revealed congested lungs, kidneys, and adrenals, and dark-appearing spleens. No abnormalities were observed in the surviving animals at necropsy.

Therefore, a histopathologic analysis was not performed.

Route of administration: Oral gavage Frequency of Treatment: Single Treatment

Dose/Concentration Levels: 0.1, 1.0, 10.0, 30.0% volume/volume emulsion

in corn oil

(Equivalent to 26, 82, 260, 820, 2600, 8200 mg/kg)

Control group and Treatment: For comparison, untreated animals were

necropsied at the end of the study.

Result : LD50 = 4626 mg/kg

Test condition: Prior to dosage, food was withheld from the animals for three to four hours.

The animals were observed for gross effects and mortality at one, four, and twenty-four hours, and once daily thereafter up until seven days. Gross necropsies were performed at the end of the observation period and samples of liver, kidney, brain, and blood were taken from untreated control animals and from all surviving animals at the 820 and 2600 mg/kg dose

levels.

Test substance : CAS No. 68526-85-2; Alcohols, C9-C11-iso, C10 rich

Conclusion: Under the conditions of this study, Alcohols, C9-C11 iso, C10 rich has a

low order of toxicity.

Reliability : (2) valid with restrictions

Pre-GLP

Date

Flag : Critical study for SIDS endpoint

06.04.2006 (5)

5.1.2 ACUTE INHALATION TOXICITY

Type : LC50 **Value** : > 95 ppm

Species : rat

Strain : Sprague-Dawley

Sex : male Number of animals : 10

Vehicle : other: not applicable

Doses : 95 ppm
Exposure time : 6 hour(s)
Method : other
Year : 1960
GLP : no

Test substance: other TS: decyl alcohol (CAS No. 68526-85-2)

Remark: There were no deaths during the exposure, or during the 24-hour post-

exposure period. Slight to moderate local irritation involving the mucous membranes of the eyes, nose, throat, and respiratory passages was seen,

but quickly returned to normal at termination of exposure.

Result : LC50 > 95 ppm (approximately 0.6 mg/l)

Test condition: Observations were made for mortality, signs of toxic effect, and body wight

changes during a 24-hour post-exposure observation period. Feed and water were freely available during the post-exposure holding period. Nearly saturated atmospheres were generated by fritted-disk bubblers containing a measured amount of the test substance through which all air entering the chambers was passed. Chamber concentrations were not determined. The nominal concentration was determined by the net loss of

the test substance from the bubblers and the total airflow.

After exposure, the animals were returned to their cages and observed for

24 hours.

Conclusion: Under the conditions of this study, decyl alcohol has a low order of acute

inhalation toxicity in rats.

Reliability : (2) valid with restrictions

Pre-GLP, vapor concentration not analyzed. From peer reviewed literature.

Flag : Critical study for SIDS endpoint

06.04.2006 (5) (26)

Type : LC50
Value : > 95 ppm
Species : mouse

Strain : other: Swiss albino

Sex : male Number of animals : 10

Vehicle : other: not applicable

Doses : 95 ppm
Exposure time : 6 hour(s)
Method : other
Year : 1960
GLP : no

Test substance : other TS: decyl alcohol (CAS No. 68526-85-2)

Remark: There were no deaths during the exposure, or during the 24-hour post-

exposure period. Slight to moderate local irritation involving the mucous membranes of the eyes, nose, throat, and respiratory passages was seen,

but quickly returned to normal at termination of exposure.

Result : LC50 > 95 ppm (approximately 0.6 mg/l)

Date

Test condition: Observations were made for mortality, signs of toxic effect, and body wight

changes during a 24-hour post-exposure observation period. Feed and water were freely available during the post-exposure holding period. Nearly saturated atmospheres were generated by fritted-disk bubblers containing a measured amount of the test substance through which all air entering the chambers was passed. Chamber concentrations were not determined. The nominal concentration was determined by the net loss of the test substance from the bubblers and the total airflow.

After exposure, the animals were returned to their cages and observed for

24 hours.

Conclusion: Under the conditions of this study, decyl alcohol has a low order of acute

inhalation toxicity in mice.

Reliability : (2) valid with restrictions

Pre-GLP, vapor concentration not analyzed.

Flag : Critical study for SIDS endpoint

06.04.2006 (5) (26)

Type : LC50
Value : > 95 ppm
Species : guinea pig
Strain : Hartley
Sex : male
Number of animals : 10

Vehicle : other: not applicable

Doses : 95 ppm
Exposure time : 6 hour(s)
Method : other
Year : 1960
GLP : no

Test substance: other TS: decyl alcohol (CAS No. 68526-85-2)

Remark: There were no deaths during the exposure, or during the 24-hour post-

exposure period. Slight to moderate local irritation involving the mucous membranes of the eyes, nose, throat, and respiratory passages was seen,

but quickly returned to normal at termination of exposure.

Result : LC50 > 95 ppm (approximately 0.6 mg/l)

Test condition : Observations were made for mortality, signs of toxic effect, and body wight

changes during a 24-hour post-exposure observation period. Feed and water were freely available during the post-exposure holding period. Nearly saturated atmospheres were generated by fritted-disk bubblers containing a measured amount of the test substance through which all air entering the chambers was passed. Chamber concentrations were not determined. The nominal concentration was determined by the net loss of

the test substance from the bubblers and the total airflow.

After exposure, the animals were returned to their cages and observed for

24 hours.

Conclusion : Under the conditions of this study, decyl alcohol has a low order of acute

inhalation toxicity in guinea pigs.

Reliability : (2) valid with restrictions

Pre-GLP, vapor concentration not analyzed. From peer reviewed literature.

Flag : Critical study for SIDS endpoint

06.04.2006 (5) (26)

Type : other: Upper Airway Sensory Irritation

Value : 280 ppm Species : mouse

Strain : Swiss Webster

Sex : male Number of animals : 8

Vehicle : other: not applicable

Doses : 111, 168 ppm

Exposure time : 30 minute(s)

Date

Method : other: ASTM: E981-84

Test substance :

Remark: The maximum vapor concentration that could be achieved under the

conditions of this study was 168 ppm.

There were no deaths during the exposure, or during the post-exposure period. No evidence of pulmonary (deep lung) irritation was observed. No animals in either group showed ocular abnormalities during or after the

exposure.

Value is estimated

Route of Administration: Head-only inhalation, aerosol

Result: None of the respiratory rate decreases exceeded 50%, therefore an RD50

concentration could only be estimated.

111 ppm resulted in 32% decrease in breathing pattern 168 ppm resulted in 40% decrease in breathing pattern

RD50 = 280 ppm (estimated)

Test condition: The upper airway sensory irritation potential was evaluated in male Swiss

Webster mice. The test material was generated as a vapor by bubbling compressed air through the liquid test substance contained inside two gaswashing bottles (bubblers) connected in series. The test animals received head-only exposures for 30 minutes. Respiratory rates were monitored before, during and after exposure, to establish a baseline respiratory rate, and to evaluate the animals' sensory irritation response to the test

atmosphere.

Test substance : CAS No. 68526-85-2; Alcohols, C9-C11-iso, C10 rich

Conclusion: The test substance produced breathing patterns characteristic of upper

airway sensory irritation in all both groups of male mice exposed to aerosol

atmospheres ranging from 111 to 168 ppm.

Reliability : (1) valid without restriction

06.04.2006 (8)

5.1.3 ACUTE DERMAL TOXICITY

Type : LD50

Value : > 2600 - mg/kg bw

Species : rabbit Strain : other: Albino Sex : male/female

Number of animals : 4

Vehicle : other: none

Doses : 80, 260, 820, and 2600 mg/kg bw

Method: otherYear: 1960GLP: noTest substance:

Remark: No deaths were observed during this study. Mild to moderate erythema and

edema were observed in animals at the three lower dose levels. Marked erythema and edema were observed at the highest dose level. Edema in each animal subsided within 3 days. Erythema in animals at the high dose group diminished in intensity but did not subside completely during the observation period. Autopsies performed following sacrifice revealed no gross pathological findings in any animal. Therefore, a histopathologic

analysis was not performed.

Route of administration: Dermal

Frequency of Treatment: Single Dose

Dose/Concentration Levels: 80, 260, 820, and 2600 mg/kg

Result: The acute dermal LD50 is > 2600 mg/kg

Date

Test condition : A single application of the test material was given to four groups

(2/sex/dose) of four rabbits at doses of 80, 260, 820, and 2600 mg/kg. The material was applied under occlusive dressing to intact abdominal skin. Observations were recorded at one, four and 24 hours; and once daily thereafter for a total of 7 days. Samples of liver, kidney, brain and blood were taken from four untreated control albino rabbits and from each

surviving animal at the 820 and 2600 mg/kg dose level. : CAS No. 68526-85-2; Alcohols, C9-C11-iso, C10 rich

Conclusion: Under conditions of this study, Alcohols, C9-C11 iso, C10 rich has a low

order of acute dermal toxicity in rats.

Reliability : (2) valid with restrictions

(Pre-GLP). From peer reviewed literature.

Flag : Critical study for SIDS endpoint

06.04.2006 (5) (26)

5.1.4 ACUTE TOXICITY, OTHER ROUTES

5.2.1 SKIN IRRITATION

Test substance

Species: rabbitConcentration: .5 undilutedExposure: SemiocclusiveExposure time: 4 hour(s)

Number of animals :

Vehicle : other: none

PDII : 2.16

Result : moderately irritating

Classification

Method : other: OECD TG 404; EEC Directive 84/449, section B4

Year : 1992 GLP : yes Test substance :

Remark: Mean scores at 24, 48, and 72 hours were 1.33, 1.33, and 1.83 for

erythema and 0.33, 0.5 and 0.83 for edema, respectively. Six male (13 weeks old) New Zealand White rabbits were used.

Test condition : Animals were individually housed in stainless steel cages, with adequate

food and water.

Approximately 24 hours prior to application of the test substance, the dorsal fur was shaved with an electric clipper. On Day 1 of the study, 0.5 ml of the test substance was applied to the skin of the animal and covered with a patch of surgical gauze, which was then covered by a semi-occlusive dressing. The dressing was removed four hours after the

application and the skin was flushed with reverse osmosis water. The skin reaction was assessed according to the test guidelines at 45 minutes, 24, 48, and 72 hours and at 7 days after the removal of the dressing.

No animal deaths were recorded prior to study termination.

Test substance : CAS No. 68526-85-2; Alcohols, C9-C11-iso, C10 rich

Reliability : (1) valid without restriction
Flag : Critical study for SIDS endpoint

06.04.2006 (7)

5.2.2 EYE IRRITATION

Species: rabbitConcentration: .1 undiluted

Dose :

Date

Exposure time : Comment :

Number of animals : 6
Vehicle : none
Result : irritating

Classification

Method : other: similar to OECD TG 405

Year : 1970 GLP : no Test substance :

Remark: Mean scores at 24, 48, and 72 hours were 1.8, 1.5, 1.3 for conjunctival

redness, 0.5, 0, 0 for chemosis, 0.67, 0.5, 0.33 for iridial irritation, and 0.5,

0.5 and 0.33 for corneal opacity, respectively.

Group median Draize score = 28 (out of 110)

The test substance produced significant conjunctival irritation in the eyes of rabbits which generally cleared by 7 days. Rated as a severe eye irritant

due to corneal sloughing or pitting.

Test condition: Animals were individually housed in stainless steel cages, with adequate

food and water.

The test material was administered as a single instillation of 0.1 ml into the lower conjunctival sac of the left eye of each animal. The upper and lower lids were gently held together for approximately 1 second to prevent loss of

the material. The contralateral eye served as the control.

The eyes of each animal were examined 24, 48, and 72 hours, and 4, 7 and 10 days after administration. At each interval the treated and control eyes were examined and scored for ocular reactions according to the

Draize Standard Eye Irritation Grading Scale.

Test substance : CAS No. 68526-85-2; Alcohols, C9-C11-iso, C10 rich

Reliability : (2) valid with restrictions

Pre-GLP. From peer reviewed literature.

Flag : Critical study for SIDS endpoint

06.04.2006

5.3 SENSITIZATION

5.4 REPEATED DOSE TOXICITY

Type : Sub-acute Species : rat : male

Strain : Wistar
Route of admin. : gavage
Exposure period : 14 d
Frequency of treatm. : daily
Post exposure period : none

Doses : 168 mg/kg/bw/day (1 mM/kg bw)

Control group : yes, concurrent vehicle
Method : other: screening study

Year : 1983 GLP : no data

Test substance

Remark : Comparative screening study to investigate selectively the

effect on testis and liver morphology and function.

Date

Result: Isodecanol did not impair body weight and had no effect on

relative liver and testis weight. Neither clinical pathology parameter nor peroxisome associated enzymes were affected.

Liver histology differed not from the concurrent control

group.

Test condition : Test species:

Male rats (Wistar derived) (breeder: ICI Alderley Park, UK)

10 per control group, 5 per dose group

Housing and diet:

1 week acclimatization

Caging: single, stainless-steel, screen-bottom cages

food and water ad libitum

Administration:

The substance was dissolved in polyethylene glycol 300 and 5 male rats received a dose level of 168 mg/kg bw (equivalent to 1 mmol/kg bw) daily by gavage for 14 consecutive days. Application volume was 10 ml/kg bw. 10 control rats received

the solvent in parallel.

Examinations:

Body weights were determined. The animals were killed by halothane overdose and blood samples for clinical pathology were taken. The liver was weighed and samples for light and

electron microscopy, morphometric analysis were investigated. Remaining liver was processed for

investigation of catalase and CN-insensitive palmitoyl CoA

oxidation.

Test substance : Analogue substance: Iso-decanol (CAS No. 25339-17-7) (supplied by ICI

plc, UK)

Reliability : (2) valid with restrictions

Screening study with limited examinations but scientifically valid and well documented. From peer reviewed literature.

Flag : Critical study for SIDS endpoint

06.04.2006 (24) (25)

5.5 GENETIC TOXICITY 'IN VITRO'

Type : Cytogenetic assay

System of testing : V79 Chinese hamster lung fibroblasts

Test concentration : 5 - 20 μg/ml Cycotoxic concentr. : 20 μg/ml . Metabolic activation : with and without

Result : negative

Method : OECD Guide-line 473

Year : 1983 **GLP** : yes

Test substance :

Method : V79 Chinese hamster lung cells were exposed to isodecanol

with/without metabolic activation. Following exposure, the cells were treated with colcemid to arrest cells in the

metaphase. Cells were harvested and chromosome preparations were made. The preparations were stained with Giemsa and metaphase cells were analyzed for chromosomal aberrations.

Cytotoxicity as depression of the mitotic index was previously examined and Ethylmethanesulonate and

Cyclophosphamide were used as positive control substances.

Date

Each 100 well spread metaphases per treatment group were

examined and two independent experiments were performed.

Test condition: Test system:

V79 Chinese hamster lung fibroblasts

Tests:

a) without S-9 mix, fixation time 21 h after start of treatment, concentrations: 5, 10 and 20 μ g/ml b) with S-9 mix, exposure 3 h, fixation time 18 h after start of treatment, concentrations: 5, 10 and 20 μ g/ml c) with S-9 mix, exposure 3 h, fixation time 21 h after start of treatment, concentrations: 5, 10 and 20 μ g/ml

Solvent: Ethanol

Positive control substances: Ethylmethanesulonate: 500 µg/ml Cyclophosphamide: 5 µg/ml

Test substance : Analogue substance: Isodecanol (CAS No. 25339-17-7), source: Riedel-

de-Haen, Seelze, Germany, colorless liquid

Conclusion : Isodecanol induced no chromosomal aberrations in cultured

mammalian V79 cells with/without metabolic activation.

Reliability : (1) valid without restriction

GLP guideline study described in sufficient detail.

Flag : Critical study for SIDS endpoint

06.04.2006 (16)

5.6 GENETIC TOXICITY 'IN VIVO'

Type : Cytogenetic assay

Species : rat Sex : Strain :

Route of admin. : gavage Exposure period : once

Doses : 2.26-12.8 mg/kg (1/5 LD50)

Result :

Method

Year : 1987 GLP :

Test substance :

Method : Each 10 rats were investigated, 1 ml test article/rat was

orally administered by gavage as a 40% suspension in water. Preparation of bone marrow 48 h after administration and

microscopic analysis of 50 cells per animal.

Number of cells investigated was too few to correspond to

scientific and regulatory requirements.

Original reference in Russian.

Result: 3% of cells with aberration were observed, control group 0%

of cells with aberrations.

Test substance: Analogue substance: Decylalcohol (CAS No. 25339-17-7)

Reliability : (2) valid with restrictions

Documentation insufficient for assessment since reference is

in a foreign language. However, reference is from a peer reviewed journal.

06.04.2006 (2)

5. To	xicity	ld	68526-85-2
		Date	06.04.2006
5.7	CARCINOGENICITY		
5.8.1	TOXICITY TO FERTILITY		
5.8.2	DEVELOPMENTAL TOXICITY/TERATOGENICITY		
5.8.3	TOXICITY TO REPRODUCTION, OTHER STUDIES		
5.9	SPECIFIC INVESTIGATIONS		
5.10	EXPOSURE EXPERIENCE		
5.11	ADDITIONAL REMARKS		

Id 68526-85-2

6. Analyt. Meth. for Detection and Identification	ld 68526-85-2 Date
6.1 ANALYTICAL METHODS	
6.2 DETECTION AND IDENTIFICATION	
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7. Eff	. Against Target Org. and Into	ended Uses	ld Date	68526-85-2	
7.1	FUNCTION				
7.2	EFFECTS ON ORGANISMS TO BE CON	TROLLED			
7.3	ORGANISMS TO BE PROTECTED				
7.4	USER				
7.5	RESISTANCE				
		35 / 39			

8. M	eas. Nec. to Prot. Man, Animals, Environment	Id Date	68526-85-2
8.1	METHODS HANDLING AND STORING		
8.2	FIRE GUIDANCE		
8.3	EMERGENCY MEASURES		
8.4	POSSIB. OF RENDERING SUBST. HARMLESS		
8.5	WASTE MANAGEMENT		
8.6	SIDE-EFFECTS DETECTION		
8.7	SUBSTANCE REGISTERED AS DANGEROUS FOR GROUND WATER		
8.8	REACTIVITY TOWARDS CONTAINER MATERIAL		
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9. References Id 68526-85-2

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9. References ld 68526-85-2 Date 06.04.2006

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10. Summary and Evaluation		68526-85-2
	Date	
10.1 END POINT SUMMARY		
10.2 HAZARD SUMMARY		
10.2 HAZARD SUMMART		
40.0 - DIOV. 4.00-DOMENT		
10.3 RISK ASSESSMENT		
	39 / 39	
	Ja / Ja	

IUCLID

Data Set

Existing Chemical: ID: 68526-86-3 : 68526-86-3 CAS No.

CAS NO.
EINECS Name : Alcohols, C11-14-iso-, C13-rich

EC No. : 271-235-6

: Alcohols, C11-14-iso-, C13-rich TSCA Name IUPAC Name : Alcohols C11-C14 iso, C13 rich

Producer related part

: ExxonMobil Biomedical Sciences Inc. Company

: 29.09.2004 Creation date

Substance related part

Company : ExxonMobil Biomedical Sciences Inc.

Creation date : 29.09.2004

Status

Memo : Prepared for EMCC - US HPV

Printing date : 06.04.2006

Revision date

Date of last update : 06.04.2006

Number of pages : 42

Chapter (profile) : Chapter: 1, 2, 3, 4, 5, 6, 7, 8, 10 Reliability (profile) : Reliability: without reliability, 1, 2, 3, 4

: Flags: without flag, confidential, non confidential, WGK (DE), TA-Luft (DE), Flags (profile)

Material Safety Dataset, Risk Assessment, Directive 67/548/EEC, SIDS

1.0.1 APPLICANT AND COMPANY INFORMATION

1.0.2 LOCATION OF PRODUCTION SITE, IMPORTER OR FORMULATOR

1.0.3 IDENTITY OF RECIPIENTS

1.0.4 DETAILS ON CATEGORY/TEMPLATE

Comment: see free text

Remark : The Alkyl Alcohols C6 to C13 Category is a family of saturated alcohols

that are produced from olefins by the hydroformylation or "oxo" process. Hydroformylation is the reaction of an olefin with carbon monoxide and hydrogen to produce an aldehyde, and its subsequent hydrogenation to the

alcohol.

The number of carbon atoms in the category members ranges from 6 to 13. Category members contain predominantly branched alkyl groups. Each substance consists of an isomeric mixture of saturated primary alcohols of high purity, and the following basic structure: CH3-R-CH2-OH, where R is a branched isomeric structure.

The justification for the Alkyl Alcohols C6 to C13 Category is that the members have:

· similar chemical structures,

- · similar physico-chemical properties,
- · comparable environmental fate,
- · the same mode of action.

In general, aliphatic alcohol toxicity occurs by non-polar narcosis (Lipnick et al., 1985). The mode of action is disruption of biological membrane function (van Wezel and Opperhuizen, 1995). Metabolic pathways, through which alcohols are metabolized, are likely to include similar reactions for all category members and result in structurally similar metabolites (carboxylic acids).

The data demonstrate that the category is valid for a screening-level hazard assessment for the Category and its members. One can assess the untested endpoints by extrapolation between and among the category members.

20.03.2006

1.1.0 SUBSTANCE IDENTIFICATION

IUPAC Name : Smiles Code :

Molecular formula : C13H28O Molecular weight : 200.37

Petrol class

Flag : Critical study for SIDS endpoint

11.01.2005

1. General Information

ld 68526-86-3

Date

1.1.1 GENERAL SUBSTANCE INFORMATION

Purity type

Substance type : organic Physical status : liquid

Purity : Colour : Odour :

Remark: CAS Registry Number, Name, and General Structure for Members of the

Alkyl Alcohols C6 to C13 Category and Analogue Substances:

CAS RN: 112-53-8

IUPAC Name: Dodecan-1-ol R Length (C number): C12 Structure of R: Linear

Category Member: No (analogue substance used for supporting

information)

CAS RN: 68526-86-3

IUPAC Name: Alcohols C11-C14-iso, C13 rich

R Length (C number): C10

Structure of R: Branched, primarily methyl branched (isomeric structures)

Category Member: Yes

06.04.2006

1.1.2 SPECTRA

1.2 SYNONYMS AND TRADENAMES

Exxal(r) 13

15.03.2004

Isotridecanol

09.03.2004

1.3 IMPURITIES

Purity : typical for marketed substance

CAS-No : 68526-86-3 **EC-No** : 271-235-6

EINECS-Name : Alcohols, C11-14-iso-, C13-rich

Remark: Major isomers are tetramethyl-1-decanol and dimethyl-1-undecanol.

11.10.2005

1.4 ADDITIVES

Purity type : typical for marketed substance

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1. General Information

ld 68526-86-3

Date

CAS-No :
EC-No :
EINECS-Name :
Molecular formula :
Value :
Function of additive :

Remark: No additives present.

11.10.2005

1.5 TOTAL QUANTITY

1.6.1 LABELLING

1.6.2 CLASSIFICATION

1.6.3 PACKAGING

1.7 USE PATTERN

Type of use : industrial

Category : Chemical industry: used in synthesis

Remark: Primarily used as a chemical intermediate in the production of plasticizers

because of their low volatility. Also used as a surfactant raw material, as a

lubricant intermediate, and as a solvent.

06.01.2005

1.7.1 DETAILED USE PATTERN

1.7.2 METHODS OF MANUFACTURE

1.8 REGULATORY MEASURES

1.8.1 OCCUPATIONAL EXPOSURE LIMIT VALUES

1.8.2 ACCEPTABLE RESIDUES LEVELS

1.8.3 WATER POLLUTION

1.8.4 MAJOR ACCIDENT HAZARDS

Id 68526-86-3 1. General Information **Date** 06.04.2006 1.8.5 AIR POLLUTION 1.8.6 LISTINGS E.G. CHEMICAL INVENTORIES 1.9.1 DEGRADATION/TRANSFORMATION PRODUCTS 1.9.2 COMPONENTS 1.10 SOURCE OF EXPOSURE 1.11 ADDITIONAL REMARKS 1.12 LAST LITERATURE SEARCH 1.13 REVIEWS

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ld 68526-86-3

Date

2.1 MELTING POINT

Value : <-40 °C Decomposition : no, at °C

Sublimation: no

Method : other: ASTM D97

Year : 2003 GLP : no data

Test substance :

Test substance : CAS No. 68526-86-3; Alcohols, C11-C14-iso, C13 rich

Reliability : (2) valid with restrictions

Although the original data were not retrieved and reviewed for quality, they

were developed following acceptable test methods and therefore

considered reliable.

Flag : Critical study for SIDS endpoint

06.04.2006 (1) (16)

2.2 BOILING POINT

Value : $= 250 - 270 \, ^{\circ}\text{C}$ at 1013 hPa

Decomposition

Method : other: ASTM D1078/01

Year : 2003 GLP : no data

Test substance :

Test substance : CAS No. 68526-86-3; Alcohols, C11-C14-iso, C13 rich

Reliability : (2) valid with restrictions

Although the original data were not retrieved and reviewed for quality, they

were developed following acceptable test methods and therefore

considered reliable.

Flag : Critical study for SIDS endpoint

06.04.2006 (16)

2.3 DENSITY

Type : density

Value : .846 g/cm³ at 20 °C

Method : other: ASTM D4052/86 equivalent

Year : 1998 GLP : no data

Test substance :

Method : The density was measured by taking the weight of 10 mls of the test

substance at 20 deg C.

Test substance : CAS No. 68526-86-3; Alcohols, C11-C14-iso, C13 rich

Reliability : (1) valid without restriction
Flag : Critical study for SIDS endpoint

06.04.2006 (9)

2.3.1 GRANULOMETRY

ld 68526-86-3

Date

2.4 VAPOUR PRESSURE

Value : = .002 hPa at 25 °C

Decomposition

Method : other (calculated)

Year : 2005 GLP : no data

Test substance

Method : Vapor pressure calculation by MPBPWIN ver. 1.40 using calculation

method of Grain.

Remark: EPIWIN is used and advocated by the US EPA for chemical property

estimation.

Test substance : CAS No. 68526-86-3; Alcohols, C11-C14-iso, C13 rich

Reliability : (2) valid with restrictions

The value was calculated based on chemical structure as modeled by EPIWIN. This robust summary has a reliability rating of 2 because the data

are calculated and not measured.

Flag : Critical study for SIDS endpoint

06.04.2006 (23)

2.5 PARTITION COEFFICIENT

Partition coefficient :

Log pow : = 4.8 - 5.5 at 30 °C

pH value

Method : OECD Guide-line 117 "Partition Coefficient (n-octanol/water), HPLC

Method"

Year : 1998 GLP : yes

Test substance :

Remark : Test Type: N-Octanol/Water Partition Coefficient (HPLC method)

Result: The test substance eluted as several groups. The four major components

C10, C11, C12, C13 alcohols had Log Pow values of 4.8, 5.0, 5.1, and 5.5

respectively.

The retention time for the 4 major components were 12.09, 13.72, 14.76,

and 18.36 minutes.

All values were measured using High Performance Liquid Chromatography

(HPLC).

Test condition: The test substance was evaluated as a solution in HPLC grade methanol.

Six reference compounds were also evaluated in a standard combined reference solution (2-butanone, acetophenone, naphthalene, biphenyl, n-butylbenzene, and 4,4-DDT) in 75% methanol and 25% distilled water.

The pH of the solution was 5.4.

Two customized alcohol reference solutions were also prepared containing five of the ten alcohol compounds (1-hexanol, 1-heptanol, 1-octanol, 1-

nonanol, 1-decanol, 1-undecanol, 1-dodecanol, 1-tridecanol, 1-tetradecanol, 1-pentadecanol) in 87.5% methanol and 12.5% distilled

water. The pH of both solutions was 7.3.

The pH of the evaluated solutions was the same as the reference solution it

was evaluated against.

The test substance was analyzed against a Standard Log Pow Reference Compound Solution and a customized Alcohol Reference Compound

ld 68526-86-3

Date

Solution. Only the peaks detected by refractive index (RI) were reported.

Test substance : CAS No. 68526-86-3; Alcohols, C11-C14-iso, C13 rich

Reliability : (1) valid without restriction
Flag : Critical study for SIDS endpoint

06.04.2006 (7)

2.6.1 SOLUBILITY IN DIFFERENT MEDIA

Solubility in : Water

Value : = 5.8 mg/l at 20 °C

pH value

concentration : at °C

Temperature effects

Examine different pol.

pKa : at 25 °C

Description

Stable

Deg. product

Method : other: slow stir

Year : 1998 GLP : no data

Test substance

Method : Slow-stir water solubility vessels consisted of glass aspirator bottles with

capacities of 4 to 12 L. The spigot at the bottom of the vessel was fitted with short length of Tefzel tubing and a #13 glass stopper. The bottle was rinsed with a mixture of 1:1 methylene chloride: acetone followed by 2,2,4-trimethyl pentane. The bottle was then air dried in a laboratory fume hood

and rinsed three times with dilution water.

A glass stir bar was added to the bottle and then the bottle was filled with the appropriate amount of water. The test substance was added to the bottle at a loading of 100 mg/L and the solution stirred quiescently with little

or no visible vortex on a magnetic stir-plate.

Mixing was stopped one hour prior to sampling. Samples were removed through the outlet at the bottom of the vessel. To avoid losses and to prevent contamination, the samples were analyzed immediately by either GC-MSD or GC-FID. Samples were taken on Days 1, 3, 7, and 21.

Remark: Measured value obtained in conjunction with Fish Acute Toxicity Test.

Test substance : CAS No. 68526-86-3; Alcohols, C11-C14-iso, C13 rich

Reliability : (1) valid without restriction
Flag : Critical study for SIDS endpoint

06.04.2006 (20)

2.6.2 SURFACE TENSION

2.7 FLASH POINT

Value : $= 126 \, ^{\circ}\text{C}$ Type : closed cup

Method : other: PMCC ASTM D93

Year : 2005 GLP : no data

Test substance

Test substance : CAS No. 68526-86-3; Alcohols, C11-C14-iso, C13 rich

ld 68526-86-3 **Date** 06.04.2006

Reliability : (2) valid with restrictions

Although the original data were not retrieved and reviewed for quality, they

were developed following acceptable test methods and therefore

considered reliable.

06.04.2006 (16)

2.8 AUTO FLAMMABILITY

2.9 FLAMMABILITY

2.10 EXPLOSIVE PROPERTIES

2.11 OXIDIZING PROPERTIES

2.12 DISSOCIATION CONSTANT

Acid-base constant : = 15.935 at 25°C Method : other: calculated

Year : 2005 GLP : no data

Test substance

Method: pKa calculation by SPARC 2003 using a Linux calculation engine.

Remark : SPARC On-line calculator can be accessed at

http://ibmlc2.chem.uga.edu/sparc/index.cfm

Test substance : CAS No. 68526-86-3; Alcohols, C11-C14-iso, C13 rich

Reliability : (2) valid with restrictions

The value was calculated based on chemical structure as modeled by SPARC. This robust summary has a reliability rating of 2 because the data

are calculated and not measured.

06.04.2006 (19)

2.13 VISCOSITY

Value : = 47 - at 20 °C

Result

Method : other: ASTM D445

Year : 2003 GLP : no data

Test substance :

Remark : Value measured in cSt

Test substance : CAS No. 68526-86-3; Alcohols, C11-C14-iso, C13 rich

Reliability : (2) valid with restrictions

Although the original data were not retrieved and reviewed for quality, they

were developed following acceptable test methods and therefore

considered reliable.

06.04.2006 (16)

2.14 ADDITIONAL REMARKS

ld 68526-86-3

Date

3.1.1 PHOTODEGRADATION

Type : water

Light source

Light spectrum: nm

Relative intensity : based on intensity of sunlight

Deg. product

Method : other (calculated): Technical Discussion

Year : GLP : Test substance :

Remark: These data represent a key study for characterising the potential of

substances in the Alkyl Alcohols C6 to C13 Category to undergo direct

photodegradation.

Result: Photolysis as a Function of Molecular Structure

The direct photolysis of an organic molecule occurs when it absorbs sufficient light energy to result in a structural transformation (Harris, 1982). The reaction process is initiated when light energy in a specific wavelength range elevates a molecule to an electronically excited state. However, the excited state is competitive with various deactivation processes that can result in the return of the molecule to a non excited state.

The absorption of light in the ultra violet (UV)-visible range, 110-750 nm, can result in the electronic excitation of an organic molecule. Light in this range contains energy of the same order of magnitude as covalent bond dissociation energies (Harris, 1982). Higher wavelengths (e.g. infrared) result only in vibrational and rotational transitions, which do not tend to produce structural changes to a molecule.

The stratospheric ozone layer prevents UV light of less than 290 nm from reaching the earth's surface. Therefore, only light at wavelengths between 290 and 750 nm can result in photochemical transformations in the environment (Harris, 1982). Although the absorption of UV light in the 290-750 nm range is necessary, it is not always sufficient for a chemical to undergo photochemical degradation. Energy may be re-emitted from an excited molecule by mechanisms other than chemical transformation, resulting in no change to the parent molecule.

A conservative approach to estimating a photochemical degradation rate is to assume that degradation will occur in proportion to the amount of light wavelengths >290 nm absorbed by the molecule (Zepp and Cline, 1977).

Substances in the Oxo Alcohols C6 to C13 Category contain molecules that are oxygenated aliphatic compounds which will absorb UV light below 220 nm (Boethling and Mackay, 2000) and will not undergo direct photolysis. Therefore, this fate process will not contribute to a measurable degradative removal of chemical components in this category from the environment.

References

Boethling, R.S., Mackay, D. (2000). Handbook of Property Estimation Methods for Chemicals. CRC Press, Boca Raton, FL, USA.

Harris, J. C. 1982. "Rate of Aqueous Photolysis," Chapter 8 in: W. J. Lyman, W. F. Reehl, and D. H. Rosenblatt, eds., Handbook of Chemical

ld 68526-86-3

Date

Property Estimation Methods, McGraw-Hill Book Company, New York,

USA.

Zepp, R. G. and D. M. Cline. 1977. Rates of Direct Photolysis in the

Aqueous Environment, Environ. Sci. Technol., 11:359-366.

Test substance : Alkyl Alcohols C6 to C13 Category members

Flag : Critical study for SIDS endpoint

03.04.2006 (15)

Туре

Light source : Sun light Light spectrum : nm

Relative intensity : based on intensity of sunlight

INDIRECT PHOTOLYSIS

Sensitizer : OH

Conc. of sensitizer : 1500000 molecule/cm³

Rate constant : = $.000000000195991 \text{ cm}^3/(\text{molecule*sec})$

Degradation: % after

Deg. product

Method : other (calculated): Calculated values using AOPWIN version 1.89, a

subroutine of the computer program EPIWIN version 3.04

Year : 1999 GLP : no Test substance :

Result : Atmospheric Oxidation Potential

In the environment, organic chemicals emitted into the troposphere are degraded by several important transformation processes. The dominant transformation process for most compounds is the daylight reaction with hydroxyl (OH-) radicals (Atkinson, 1988, 1989). The rate at which an organic compound reacts with OH- radicals is a direct measure of its atmospheric persistence (Meylan and Howard, 1993).

AOPWIN estimates the rate constant for the atmospheric, gas-phase reaction between photochemically produced hydroxyl radicals and organic chemicals. The rate constants estimated by the program are then used to calculate atmospheric half-lives for organic compounds based upon average atmospheric concentrations of hydroxyl radicals.

Since the reactions only take place in the presence of sunlight, the atmospheric half-lives are normalized for a 12-hour day.

Calculated* OH- Rate Constant half-life (hrs) (cm3/molecule-sec)

6.5 19.60 E-12

References:

Atkinson, R. 1988. Estimation of gas-phase hydroxyl radical rate constants for organic chemicals. Environ. Toxicol. Chem. 7:435-442.

Atkinson, R. 1989. Kinetics and mechanisms of the gas-phase reactions of the hydroxyl radical with organic compounds. J. Phys. Chem. Ref. Data Monograph No. 1, Amer. Inst. Physics & Amer. Chem. Soc., NY.

Meylan, W.M. and P.H. Howard. 1993. Computer estimation of the atmospheric gas-phase reaction rate of organic compounds with hydroxyl radicals and ozone. Chemosphere 12:2293-2299.

ld 68526-86-3

Date

Test condition: Indirect photodegradation, or atmospheric oxidation potential, is based on

the structure-activity relationship methods developed by R. Atkinson.

Temperature: 25°C Sensitizer: OH radical

Concentration of Sensitizer: 1.5 E6 OH radicals/cm3 CAS No. 68526-86-3; Alcohols, C11-C14-iso, C13 rich

Reliability : (2) valid with restrictions

The results include calculated data based on chemical structure as modeled by AOPWIN. The data represent a potential atmospheric half-life range for the test substance. This robust summary represents a "key

study" for atmospheric half-life range based on constituent data.

Flag : Critical study for SIDS endpoint

06.04.2006 (24)

3.1.2 STABILITY IN WATER

Test substance

 Type
 : abiotic

 t1/2 pH4
 : at °C

 t1/2 pH7
 : at °C

 t1/2 pH9
 : at °C

Deg. product

Method : other: Technical Discussion

Year : GLP :

Test substance :

Remark: These data represent a key study for characterising the potential of

substances in the Alkyl Alcohols C6 to C13 Category to undergo

hydrolysis.

Result : Hydolysis as a Function of Molecular Structure

Hydrolysis of an organic molecule occurs when a molecule (R-X) reacts with water (H2O) to form a new carbon-oxygen bond after the carbon-X bond is cleaved (Gould, 1959; Harris, 1982). Mechanistically, this reaction is referred to as a nucleophilic substitution reaction, where X is the leaving group being replaced by the incoming nucleophilic oxygen from the water molecule.

Chemicals that are susceptible to hydrolysis contain functional groups that can be displaced by a nucleophilic substitution reaction. Substances that have the potential to hydrolyze include alkyl halides, amides, carbamates, carboxylic acid esters and lactones, epoxides, phosphate esters, and sulfonic acid esters (Neely, 1985). The lack of a leaving group renders a compound resistant to hydrolysis.

Alkyl alcohols are resistant to hydrolysis because they lack a functional group that is hydrolytically reactive (Harris, 1982).

References:

Gould, E.S. (1959). Mechanism and Structure in Organic Chemistry, Holt, Reinhart and Winston, New York, NY, USA.

Harris, J.C. (1982). "Rate of Hydrolysis," Chapter 7 in: W.J. Lyman, W.F. Reehl, and D.H. Rosenblatt, eds., Handbook of Chemical Property Estimation Methods, McGraw-Hill Book Company, New York, NY, USA.

Neely, W.B. (1985). Hydrolysis. In: W.B. Neely and G.E. Blau, eds., Environmental Exposure from Chemicals. Vol. 1, pp. 157-173. CRC Press,

ld 68526-86-3

Date

Boca Raton, FL, USA.

Conclusion: Hydrolysis will not contribute to the removal of Alkyl Alcohols from the

environment.

Flag : Critical study for SIDS endpoint

20.03.2006 (14)

3.1.3 STABILITY IN SOIL

3.2.1 MONITORING DATA

3.2.2 FIELD STUDIES

3.3.1 TRANSPORT BETWEEN ENVIRONMENTAL COMPARTMENTS

Type : fugacity model level III

Media: other: air - water - soil - sedimentAir: % (Fugacity Model Level I)Water: % (Fugacity Model Level I)Soil: % (Fugacity Model Level I)Biota: % (Fugacity Model Level II/III)Soil: % (Fugacity Model Level II/III)

Method : other: Calculation according Mackay, Level III

Year : 2003

Method : The EQC Level III model is a steady state model that is useful for

determining how the medium of release affects environmental fate. Level III fugacity allows non-equilibrium conditions to exist between connected media as steady state, and illustrate important transport and transformation

processes.

Physicochemical input values for the model were calculated using the EPIWIN Estimation v 3.04 program. Measured input values were also used where available and obtained from the EPIWIN database. Distribution data from the equilibrium model provide basic information on the potential partitioning behavior of chemicals between selected environmental

compartments (i.e., air, water, soil, and sediment).

Input values used:

Molecular mass = 200.37 g/mol Water solubility = 5.8 mg/L Vapour pressure = 0.2 Pa

log Kow = 5.15

Melting point = -40 deg C

Degradation half-lives:

Air - 6.5 hrs Water - 240 hrs Soil - 1440 hrs Sediment - 14400 hrs

This model was run assuming 100% discharge to water.

Result : Air - 0.08%

Water - 18.3% Soil - 0.01%

ld 68526-86-3

Date

Sediment - 81.6%

Test substance : CAS No. 68526-86-3; Alcohols, C11-C14-iso, C13 rich

Reliability : (2) valid with restrictions

This robust summary has a reliability rating of 2 because the data are

calculated and not measured.

Flag : Critical study for SIDS endpoint

25.10.2005 (22)

Type : fugacity model level I

Media : other: air - biota - sediment(s) - soil - water

Air : % (Fugacity Model Level I)

Water : % (Fugacity Model Level I)

Soil : % (Fugacity Model Level I)

Biota : % (Fugacity Model Level II/III)

Soil : % (Fugacity Model Level II/III)

Method : other: Calculation according Mackay, Level I

Year : 2003

Method : The EQC Level I is a steady state, equilibrium model that utilizes the input

of basic chemical properties including molecular weight, vapor pressure, and water solubility to calculate distribution within a standardized regional

environment.

Physicochemical input values for the model were calculated using the EPIWIN Estimation v 3.04 program. Measured input values were also used where available and obtained from the EPIWIN database. Distribution data from the equilibrium model provide basic information on the potential partitioning behavior of chemicals between selected environmental

compartments (i.e., air, water, soil, sediment, suspended sediment, biota).

Input values used:

Molecular mass = 200.37 g/mol Water solubility = 5.8 mg/L Vapour pressure = 0.2 Pa

log Kow = 5.15

Melting point = -40 deg C

Remark: Results were calculated using measured value for log Kow.

Result : Air - 1.1%

Water - 0.8% Soil - 96.0% Sediment - 2.1%

Suspended Sed - 0.07%

Biota - 0.01%

Test substance : CAS No. 68526-86-3; Alcohols, C11-C14-iso, C13 rich

Reliability : (2) valid with restrictions

This robust summary has a reliability rating of 2 because the data are

calculated and not measured.

25.10.2005 (22)

Type : volatility
Media : water - air

Air : % (Fugacity Model Level I)

Water : % (Fugacity Model Level I)

Soil : % (Fugacity Model Level I)

Biota : % (Fugacity Model Level II/III)

Soil : % (Fugacity Model Level II/III)

Method : other: calculation

Year : 2005

Method : Henry's Law Constants (HLC) are based on vapor pressure and water

solubility values, and molecular weights. HLC values were calculated

using equations found in Mackay, 2003.

ld 68526-86-3

Date

Vapor pressure = 0.2 Pa Water solubility = 5.8 mg/l Molecular mass = 200.37 g/mol

Result HLC (at 25 deg C) = 6.91 Pa-m3/mole

Test substance : CAS No. 68526-86-3; Alcohols, C11-C14-iso, C13 rich

Reliability : (2) valid with restrictions

This robust summary has a reliability rating of (2) because the data are

calculated and not measured.

Flag : Critical study for SIDS endpoint

26.10.2005 (22)

3.3.2 DISTRIBUTION

Media : other: Koc Method other (calculation)

Year : 2005

Method The soil adsorption coefficient (Koc) was calculated by PCKOCWIN ver.

1.66, using equations developed by Lyman (1980).

PCKOCWIN is a subroutine of EPIWIN. EPIWIN is used and advocated by the USEPA for chemical property estimation. Koc provides an indication of the extent to which a chemical partitions between solid and solution phases

in soil, or between water and sediment in aquatic ecosystems.

Result : Koc = 505.8

log Koc = 2.74

Test substance : CAS No. 68526-86-3; Alcohols, C11-C14-iso, C13 rich

Reliability (2) valid with restrictions

This robust summary has a reliability rating of 2 because the data are

calculated and not measured.

26.10.2005 (21)(25)

MODE OF DEGRADATION IN ACTUAL USE

3.5 **BIODEGRADATION**

aerobic **Type**

Inoculum activated sludge, domestic

Contact time

Degradation 60.6 (±) % after 28 day(s)

Result other

Deg. product

OECD Guide-line 301 F "Ready Biodegradability: Manometric Method

Respirometry Test"

Year 1999 : **GLP** ves Test substance

Remark : Test Type: Manometric Respirometry Test

Exposure Period: 28 days

Result Test material is rapidly biodegradable. Half-life was reached on day 20. On

day 28, 60.6% degradation of the test material was observed. 10%

biodegradation was achieved on day 7.

By day 14, >60% biodegradation of positive control was observed, which

met the guideline requirement.

ld 68526-86-3

Date

Biodegradation was based on oxygen consumption and the theoretical oxygen demand of the test material as calculated using results of an elemental analysis of the test material.

% Degradation* Mean % Degradation

Sample (day 28) (day 28) Test Material 59.9, 62.7, 59.2 60.6 Na Benzoate 89.0, 87.4, 87.8 88.0

* replicate data

Test condition

Non acclimated activated sludge and test medium were combined prior to test material addition. Test medium consisted of glass distilled water and mineral salts (Phosphate buffer, Ferric chloride, Magnesium sulfate, Calcium chloride).

Test vessels were 1L glass flasks placed in a waterbath and electronically monitored for oxygen consumption.

Test material was tested in triplicate, controls and blanks were tested in duplicate.

Test material concentration was approximately 100 mg/L. Sodium benzoate (positive control) concentration was 50mg/L.

Test temperature was 22 +/- 1 Deg C.

All test vessels were stirred constantly for 28 days using magnetic stir bars and plates.

Test substance

Conclusion

: CAS No. 68526-86-3; alcohols, C11-14-iso, C13 rich

Test substance is considered rapidly biodegradable. Although the test substance (a complex and poorly water soluble material) achieved a ready extent of biodegradability, it did not achieve this extent within a 10-day

window.

Reliability : (1) valid without restriction
Flag : Critical study for SIDS endpoint

06.04.2006 (12)

Type : aerobic

Inoculum : activated sludge, domestic

Contact time

Degradation : = 58 (±) % after 28 day(s) **Result** : other: not readily biodegradable

Deg. product

Method : OECD Guide-line 301 F "Ready Biodegradability: Manometric

Respirometry Test"

Year : 1998 GLP : yes

Test substance

Remark: Test Type: Manometric Respirometry Test

Exposure Period: 28 days

Result : Test material was not readily biodegradable. Half-life was reached by day

25. By day 28, 58.1% degradation of the test material was observed. 10%

biodegradation was achieved on day 7.

By day 14, >60% biodegradation of positive control was observed, which met the guideline requirement. No excursions from the protocol were noted.

Biodegradation was based on oxygen consumption and the theoretical oxygen demand of the test material as calculated using results of an elemental analysis of the test material.

% Degradation* Mean % Degradation

16 / 42

ld 68526-86-3

Date

Sample (day 28) (day 28) Test Material 60.1, 60.7, 53.7 58.1 Na Benzoate 87.1, 85.4 86.2

* replicate data

Test condition

Non acclimated activated sludge and test medium were combined prior to test material addition. Test medium consisted of glass distilled water and mineral salts (Phosphate buffer, Ferric chloride, Magnesium sulfate, Calcium chloride).

Test vessels were 1L glass flasks placed in a waterbath and electronically monitored for oxygen consumption.

Test material was tested in triplicate, controls and blanks were tested in duplicate.

Test material concentration was approximately 57 mg/L. Sodium benzoate (positive control) concentration was 44mg/L.

Test temperature was 22 +/- 1 Deg C.

All test vessels were stirred constantly for 28 days using magnetic stir bars

Test substance Conclusion Reliability

CAS No. 68526-86-3; alcohols, C11-14-iso, C13 rich Test substance is considered not readily biodegradable.

(1) valid without restriction

06.04.2006 (10)

Type aerobic

Inoculum activated sludge, domestic

Contact time

Degradation = 59.6 (±) % after 28 day(s) other: not readily biodegradable Result

Deg. product

Method OECD Guide-line 301 F "Ready Biodegradability: Manometric

Respirometry Test"

Year 1997 **GLP** ves

Test substance

Remark Test Type: Manometric Respirometry Test

Exposure Period: 28 days

Result Test material was not readily biodegradable. Half-life was reached by day

21. By day 28, 59.6% degradation of the test material was observed. 10%

biodegradation was achieved on day 8.

By day 14, >60% biodegradation of positive control was observed, which

met the guideline requirement.

Biodegradation was based on oxygen consumption and the theoretical oxygen demand of the test material as calculated using results of an elemental analysis of the test material.

> % Degradation* Mean % Degradation

Sample (day 28) (day 28) Test Material 56.5, 60.3, 61.8 59.6 Na Benzoate 92.5, 92.7 92.6

* replicate data

Test condition

Non acclimated activated sludge and test medium were combined prior to test material addition. Test medium consisted of glass distilled water and mineral salts (Phosphate buffer, Ferric chloride, Magnesium sulfate, Calcium chloride).

Test vessels were 1L glass flasks placed in a waterbath and electronically monitored for oxygen consumption.

ld 68526-86-3

Date

Test material was tested in triplicate, controls and blanks were tested in duplicate.

Test material concentration was approximately 43 mg/L. Sodium benzoate (positive control) concentration was 50mg/L.

Test temperature was 22 +/- 1 Deg C.

All test vessels were stirred constantly for 28 days using magnetic stir bars

and plates.

Test substance : CAS No. 68526-86-3; alcohols, C11-14-iso, C13 rich Conclusion : Test substance is considered not readily biodegradable.

Reliability : (1) valid without restriction

06.04.2006 (6)

3.6 BOD5, COD OR BOD5/COD RATIO

3.7 BIOACCUMULATION

BCF : 30 - 60

Elimination

Method : OECD Guide-line 305

Year : 1999 GLP : yes Test substance :

Remark

No difference in mortality or growth was observed between the treatments and the control after 10 days of exposure, or at the end of the study (14 days).

The BCF as a function of lipid content was 1087 L/kg and 493.4 L/kg, respectively. The mean BCF as a function of lipid content was 790.2 L/kg.

Duration of test: 10 days uptake, 4 days depuration

The duration of the uptake phase was shorter than the guideline recommended duration of 28 days. The duration of the uptake phase was determined based on the concentration of the test substance reaching a steady state in the tissue of the fish. Samples of the fish were analyzed at regular intervals. To determine the concentration of the test substance in the fish tissue, under steady-state conditions, one-tail T-tests for the comparison of the means at the sampling times were performed.

Result : Calculated Bioconcentration Factor (BCF):

0.16 mg/L = 600.04 mg/L = 30

Mean BCF for the study = 45

Test condition: A 100 mg/L stock solution was prepared by adding the appropriate amount

of the test substance to 40 liters of dilution water. The stock solution was mixed for 24 hours prior to its use in the test. It was then pumped from the mixing vessel via a glass tube, silicon tubing and varistaltic pump and delivered to the test chambers via a 2 liter proportional diluter. Two concentrations were prepared for the test, 0.16 mg/L and 0.04 mg/L. A dilution water control was also prepared. A new stock solution was

prepared in this manner every 3 days and mixed for 24 hours prior to use in

the test.

One test chamber was prepared for each treatment and the control. At the start of the test each test chamber contained 45 fish. Fish were observed daily for mortality and/or abnormal behavior or appearance. Water and fish samples were removed periodically during both the uptake and depuration

phases of the study. Samples were analyzed by gas chromatography. Mean measured concentrations of the test substance in water:

0.16 mg/L = 0.11 mg/L (s.d.= 0.034)

0.04 mg/L = 0.02 mg/L (s.d.= 0.007)

The test substance was not detected in the control during the study or in either treatment during the depuration phase.

Mean measured concentrations of the test substance in fish tissue:

0.16 mg/L = 6.6 mg/kg

0.04 mg/L = 0.60 mg/kg

Fish samples were also removed at the beginning and end of the uptake phase and analyzed for lipid content. The mean percent lipid content was 5.84%.

Test temperature was 14.1 Deg C., Lighting was 16 hours light: 8 hours dark with 545 to 727 Lux during full daylight periods.

Dissolved Oxygen ranged from 9.1 to 9.9 mg/L. The pH was ranged from 7.0 to 7.5 during the study. Fish were fed at a rate of 2% of their bodyweight per day during the study.

Fish Mean Weight was approximately = 0.230g at te start of the test. Test Loading = 0.30 g of fish/L.

The following data represent the measurements of the test chemical in fish tissue. Units are in μg of test chemical per gram of fish. The Non Detect (ND) level was 0.27 $\mu g/g$, which corresponds to 0.65 $\mu g/g$ for a 0.4 g fish mass. All samples taken on Day 4 of the depuration phase resulted in non-detects.

Uptake Phase

Fish Tissue Analysis

Control	Day 1 ND ND	Day 2 ND ND	Day 4 ND ND	Day 10 ND ND	
0.04mg/l	0.34 0.56	0.45 0.95	0.52 0.66 0.50	0.71 0.79	
mean	0.45	0.70	0.59	0.67	
0.16mg/l mean	3.9 4.2 6.4 4.7 4.8	4.5 3.9 8.2 7.2 6.0	3.7 3.1 4.0 4.5 3.8	8.0 4.5 9.3 6.6 7.1	
Control	Day 0 ND ND ND	Depura Day 0 ND ND ND	ation Pha Day 0 ND ND	ase Day 1 ND ND	Day 2 ND ND
0.04mg/l	0.47 0.46 0.79 0.67 0.60	0.40 0.44 0.42 0.39 0.41	0.32 0.21	ND 0.33 <0.33	ND ND
mean	0.00	0.41	0.27	<0.55	IND
0.16mg/l	2.6 2.7 2.1	2.5 2.0 2.9	1.9 1.0 2.3	0.65 0.44 0.89	0.71 0.43
	19 /	42			

Id 68526-86-3

Date

2.8 2.8 1.7

mean 2.6 2.6 1.7 0.66 0.57 **Test substance**: CAS No. 68526-86-3; alcohols, C11-14-iso, C13 rich

Conclusion : Low potential to bioaccumulate. Value based on wet weight data.

Reliability : (1) valid without restriction
Flag : Critical study for SIDS endpoint

16.03.2006 (11)

3.8 ADDITIONAL REMARKS

Date

4.1 ACUTE/PROLONGED TOXICITY TO FISH

Type : semistatic

Species: Oncorhynchus mykiss (Fish, fresh water)

Exposure period : 96 hour(s)
Unit : mg/l

LC50 : = .42 measured/nominal

Limit test

Analytical monitoring : yes

Method : OECD Guide-line 203 "Fish, Acute Toxicity Test"

Year : 1998 GLP : yes Test substance :

Method : Spearman-Karber Method (Hamilton, et al, 1977)

Result : LC50 = 0.42 mg/L (CI 0.37 to 0.48), based upon measured concentrations

of mean of old and new samples.

Analytical method used was GC-MSD

LL50 = 0.64 mg/L (CI 0.57 to 0.73), based upon nominal loading levels.

Nominal Conc. Measured Conc. % Mortality @ 96 hr.
Control Below detection 0
0.25 mg/L 0.17 mg/L 0
0.5 mg/L 0.32 mg/L 13
1.0 mg/L 0.67 mg/L 100

2.0 mg/L 0.94 mg/L 100 5.0 mg/L 0.93 mg/L 100

Test condition

Individual Water Accomodated Fractions (WAF's) were prepared for each test treatment. The test substance was added volumetrically, via a syringe, to 19L of dilution water in a 20L glass carboy. The solutions were mixed for 24 hours at a vortex of </= 10% of the total depth. The test solutions were pumped from each mixing vessel into three replicates of 4.5L in 4.0L glass aspirator bottles (no headspace). Five fish were added to each test replicate and the replicates sealed. Daily renewals were performed by removing ~80% of the test solution through the port at the bottom and refilling with fresh solution.

Test temperature was 13.8 Deg C., Lighting was 16 hours light: 8 hours dark with 551 to 736 Lux during full daylight periods.

Dissolved Oxygen at initiation ranged from 8.3 to 9.2 mg/L and from 6.6 to 8.8 mg/L in "old" solutions prior to renewals. The pH was ranged from 6.6 to 8.2 during the study. Fish were not fed during the study.

Fish Mean Wt.= 0.131g. Mean Total length = 2.7cm, Test Loading = 0.15 g of fish/L.

Test substance : CAS No. 68526-86-3; alcohols, C11-14-iso, C13 rich **Conclusion** : Test substance is considered to have high acute toxicity.

Reliability : (1) valid without restriction
Flag : Critical study for SIDS endpoint

06.04.2006 (8)

Type : semistatic

Species: Oncorhynchus mykiss (Fish, fresh water)

Exposure period : 96 hour(s)
Unit : mg/l

LC50 : = 1.2 measured/nominal

Date

Limit test

Analytical monitoring : yes

Method : OECD Guide-line 203 "Fish, Acute Toxicity Test"

Year : 1997 **GLP** : yes

Test substance :

Method: Probit procedure of SAS (Finney, 1971)

Result: LC50 = 1.2 mg/L (CI 0.98 to 1.4), based upon measured concentrations of

mean of old and new samples.

Analytical method used was GC-MSD

LL50 = 1.7 mg/L (CI 1.4 to 3.3), based upon nominal loading levels.

Nominal Conc. Measured Conc. % Mortality @ 96 hr. Control Below detection 0 0.4 mg/L 0.22 mg/L 0 0.43 mg/L 0.75 mg/L 7 1.5 mg/L 1.13 mg/L 33 3.0 mg/L 1.18 mg/L 100 6.0 mg/L 1.78 mg/L 100

Dissolved oxygen levels dropped below 60% (40-60%) of saturation in some of the treatments on Days 1 through 4 of the test. Based on mortality observations, these deviations are not believed to have affected the outcome of the study.

Test condition

Individual Water Accomodated Fractions (WAF's) were prepared for each test treatment. The test substance was added volumetrically, via a syringe, to 19L of dilution water in a 20L glass carboy. The solutions were mixed for 24 hours at a vortex of </= 10% of the total depth. The test solutions were pumped from each mixing vessel into three replicates of 4.5L in 4.0L glass aspirator bottles (no headspace). Five fish were added to each test replicate and the replicates sealed. Daily renewals were performed by removing ~80% of the test solution through the port at the bottom and refilling with fresh solution.

Test temperature was 15.1 Deg C., Lighting was 16 hours light: 8 hours dark with 749 to 752 Lux during full daylight periods.

Dissolved Oxygen at initiation ranged from 8.5 to 9.8 mg/L and from 3.9 to 8.0 mg/L in "old" solutions prior to renewals. The pH was ranged from 6.6 to 8.1 during the study. Fish were not fed during the study.

Fish Mean Wt.= 0.623g. Mean Total length = 4.3cm, Test Loading = 0.69 g of fish/L.

Test substance : CAS No. 68526-86-3; alcohols, C11-14-iso, C13 rich

Conclusion: Test substance is considered to have moderate acute toxicity.

Reliability : (1) valid without restriction

06.04.2006 (9)

Type :

Species: other: freshwater fish

Exposure period : 96 hour(s)
Unit : mg/l

LC50 : = .15 calculated

Method : other: ECOSAR Computer Model

Year : 2005

GLP : Test substance :

Test condition : Log Kow (octanol/water partition coefficient) values and a chemical

structure are needed to calculate aquatic toxicity using the ECOSAR

Date

model. The Kow calculation is performed by KOWWIN based on an atom/fragment contribution method of Meylan and Howard (1), which is a subroutine in the EPIWIN computer model (2). KOWWIN also has a database of experimental Kow values (EXPKOW.DB).

The ECOSAR program was run using a C13 alcohol with a Kow of 5.19.

1. Meylan, W. and P. Howard. 1995. Atom/fragment contribution method for estimating octanol-water partition coefficients. J. Pharm. Sci. 84:83-92.

2. Meylan, M., SRC 1994-1999. KOWWIN is contained in the computer program EPIWIN. 1999. Estimation Program Interface for Windows, version 3.04. Syracuse Research Corporation, Syracuse, NY, USA.

Test substance

CAS No. 68526-86-3; alcohols, C11-14-iso, C13 rich

Conclusion

Based on the calculated Kow value, the C13 alcohol is expected to have an

acute 96-hour LC50 of 0.15 mg/L and a Chronic Value of 0.03 mg/L.

Reliability : (2) valid with restrictions

The value was calculated based on chemical structure as modeled by EPIWIN. This robust summary has a reliability rating of 2 because the data

are calculated and not measured.

06.04.2006 (3)

4.2 ACUTE TOXICITY TO AQUATIC INVERTEBRATES

Type : static

Species : Daphnia magna (Crustacea)

Exposure period : 48 hour(s)
Unit : mg/l

LC50 : = .71 measured/nominal

Analytical monitoring : yes

Method : EPA OTS 797.1300

Year : 1986 GLP : no data

Test substance :

Method : Probit procedure based on Litchfield-Wilcoxon (1949)

Remark: Test Type: Daphnid Acute Toxicity Test

Result: 48-hour LC50 = 0.71 mg/L (CI 0.59 - 0.85) as Total Carbon, based upon

mean measured concentrations of Day 0 and Day 2 samples. 48-hour

LC50 value equivalent to 16.7% WSF.

Analytical method used was Total Carbon

Nominal Conc. Measured Conc. % Mortality @ 48 hr.

Control 6.25% WSF 0.28 mg/L 0 12.5% WSF 0.58 mg/L 30 25% WSF 1.03 mg/L 85 50% WSF 1.85 mg/L 100 100% WSF 4.17 mg/L 100

Test condition

The water soluble fraction (WSF) was prepared by combining the test substance with dilution water at a ratio of 1:150. The solutions were mixed for 96 hours and allowed to settle for 1 hour prior to use as the 100% WSF stock solution. Test solutions were prepared by diluting the 100% WSF stock. Two replicates of 250 mL in 400 mL autoclaved glass beakers were prepared at each treatment level. Ten daphnids per replicate chamber. Test chambers were covered with glass and placed in a temperature-controlled waterbath. The test was performed under static conditions.

Test temperature was 20.8 Deg C., Lighting was 16 hours light: 8 hours

Date

dark with 57.5 to 67.3 footcandles during full daylight periods. Dissolved oxygen ranged from 8.1 to 9.1 mg/L during the study. The pH was ranged from 7.8 to 8.2 during the study. Dilution water hardness was 130 mg/L as CaCO3.

Organisms were supplied by in-house cultures. Age = <24 hours old from

19-day old parents.

Test substance : CAS No. 68526-86-3; alcohols, C11-14-iso, C13 rich **Conclusion** : Test substance is considered to have high acute toxicity.

Reliability : (2) valid with restrictions

Flag : Critical study for SIDS endpoint

06.04.2006 (4)

Type : static

Species : Daphnia magna (Crustacea)

Exposure period : 24 hour(s)
Unit : mg/l

EC50 : = .81 measured/nominal

Analytical monitoring : yes

Method : OECD Guide-line 202

Year : 1997 GLP : yes Test substance :

Method : Probit procedure of SAS (Finney, 1971)Remark : Test Type: Daphnid Acute Toxicity Test

Result : 24-hour EC50 = 0.81 mg/L (Could not calculate CI), based upon measured

concentrations of mean of old and new samples.

Analytical method used was GC-MSD

Nominal Conc. Measured Conc. % Immobilization @ 24 hr.

Below detection Control 0 0.31 mg/L 25 0.45 mg/L 0.63 mg/L 1.14 mg/L 85 1.25 mg/L 100 1.75 mg/L 2.5 mg/L 2.3 mg/L 95 5.0 mg/L 2.5 mg/L 100

Test condition

Individual Water Accomodated Fractions (WAF's) were prepared for each test treatment. The test substance was added volumetrically, via a syringe, to 2.0L of dilution water in a 2L glass aspirator bottle. The solutions were mixed for 23 hours at a vortex of </= 10% of the total depth. The test solutions were removed through the outlet at the bottom of each mixing vessel into four replicates of 140 mL in 125 mL glass erlenmeyer flasks (no headspace). Five daphnids were added to each test replicate and the replicates sealed. The test was performed under static conditions with no aeration.

Test temperature was 19.8 Deg C., Lighting was 16 hours light: 8 hours dark with 443 to 577 Lux during full daylight periods. Dissolved oxygen ranged from 8.0 to 8.5 mg/L during the study. The pH was ranged from 7.0 to 7.6 during the study.

Organisms were supplied by in-house cultures. Age = <24 hours old.

Test substance : CAS No. 68526-86-3; alcohols, C11-14-iso, C13 rich

Conclusion : The results of this study confirm the EC50 is below 1.0 mg/L.

Reliability : (2) valid with restrictions

Flag : Critical study for SIDS endpoint

06.04.2006 (5)

Type

Species : Daphnia sp. (Crustacea)

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Date

Exposure period : 48 hour(s)
Unit : mg/l

Method : other: ECOSAR Computer Model

Year : 2005

GLP : Test substance :

Test condition: Log Kow (octanol/water partition coefficient) values and a chemical

structure are needed to calculate aquatic toxicity using the ECOSAR model. The Kow calculation is performed by KOWWIN based on an atom/fragment contribution method of Meylan and Howard (1), which is a subroutine in the EPIWIN computer model (2). KOWWIN also has a

database of experimental Kow values (EXPKOW.DB).

The ECOSAR program was run using a C13 alcohol with a Kow of 5.19.

1. Meylan, W. and P. Howard. 1995. Atom/fragment contribution method for estimating octanol-water partition coefficients. J. Pharm. Sci. 84:83-92.

2. Meylan, M., SRC 1994-1999. KOWWIN is contained in the computer program EPIWIN. 1999. Estimation Program Interface for Windows, version 3.04. Syracuse Research Corporation, Syracuse, NY, USA.

Test substance : CAS No. 68526-86-3; alcohols, C11-14-iso, C13 rich

Conclusion : Based on the calculated Kow value, the C13 alcohol is expected to have an

acute 48-hour EC50 of 0.2 mg/L and a Chronic Value of 0.04 mg/L.

Reliability : (2) valid with restrictions

The value was calculated based on chemical structure as modeled by EPIWIN. This robust summary has a reliability rating of 2 because the data

are calculated and not measured.

06.04.2006

4.3 TOXICITY TO AQUATIC PLANTS E.G. ALGAE

Species : other algae: Pseudokirchneriella subcapitata

Endpoint

Exposure period : 72 hour(s)
Unit : mg/l

EC50 : = 2.6 measured/nominal

Limit test

Analytical monitoring : yes

Method : OECD Guide-line 201 "Algae, Growth Inhibition Test"

Year : 2003 GLP : yes

Test substance :

Method : Statistical methods-EC50 was determined using Maximum likelihood

analysis based on D.J. Finney (1971). NOECs were determined using the

ANOVA procedure of SAS.

Result : 72 hour EC50b = 2.6mg/L (biomass)

72 hour EC50gr = 3.2mg/L (growth rate)
72 hour NOECRb = 1.5mg/L (biomass)
72 hour NOECRgr = 2.2mg/L (growth rate)

Analytical method used was GC-FID.

Mean Cell

Nominal Growth - 72 hr Conc. - 72 hr Conc. (mg/L) (% Inhibition) (cells/ml) Control n/a 8.3 x10E5 1.5 (1.5)* 3.1 7.9 x10E5

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Date

3.0 (2.2)	7.2	7.7 x10E5
8.0 (2.2)	21	7.6 x10E5
20 (2.5)	49	4.7 x10E5
61 (3.0)	86	1.1 x10E5

*note - value in parentheses is mean measured concentration (mg/L)

n/a - Not applicable

Test condition : Individual Water Accomodated Fractions (WAF's) were prepared for each

test treatment. The test substance was added volumetrically, via a syringe, to 2L of dilution water in a 2L glass bottle. The solution was mixed for 23 hours at a vortex of 8% of the total depth. Mixing temperature was approximately 23 Deg C. After a 1-hour settling/cooling period, the aqueous portions were transferred to the test chambers via the port at the bottom of the mixing vessel and used for testing. Test vessels were 125ml glass Erlenmeyer flasks that were filled with approximately 60ml of treatment solution and inoculated with algae. Samples were taken daily for cell counts. Four replicates were prepared for each treatment level. The initial algal concentration was 1.0 x 104 cells/ml. All test replicates were placed on a shaker table at 100 oscillations per minute during the study.

Biomass was calculated as the area under the growth curve.

Test temperature was 23.1 Deg. C. Lighting was continuous at 7600 to 7800 Lux. The pH ranged from 7.3 to 7.4 at test initiation and ranged from

6.8 to 7.1 at termination.

Test substance : CAS No. 68526-86-3; alcohols, C11-14-iso, C13 rich

Conclusion : Test substance is considered to have moderate to high acute toxicity to

algae.

Reliability : (1) valid without restriction
Flag : Critical study for SIDS endpoint

06.04.2006 (13)

Species: other algae: green alga

Endpoint

Exposure period : 96 hour(s)
Unit : mg/l

EC50 : = .15 calculated

Method : other: ECOSAR Computer Model

Year : 2005

GLP :

Test condition : Log Kow (octanol/water partition coefficient) values and a chemical

structure are needed to calculate aquatic toxicity using the ECOSAR model. The Kow calculation is performed by KOWWIN based on an atom/fragment contribution method of Meylan and Howard, which is a subroutine in the EPIWIN computer model. KOWWIN also has a database

of experimental Kow values (EXPKOW.DB).

The ECOSAR program was run using a C13 alcohol with a Kow of 5.19.

Test substance : CAS No. 68526-86-3; alcohols, C11-14-iso, C13 rich

Conclusion : Based on the calculated Kow value, the C13 alcohol is expected to have an

acute 96-hour EC50 of 0.15 mg/L and a Chronic Value of 0.09 mg/L.

Reliability : (2) valid with restrictions

The value was calculated based on chemical structure as modeled by EPIWIN. This robust summary has a reliability rating of 2 because the data

are calculated and not measured.

06.04.2006 (3)

4.4 TOXICITY TO MICROORGANISMS E.G. BACTERIA

Date

4.5.1 CHRONIC TOXICITY TO FISH

Species: other: Freshwater Fish (calculated toxicity values are not species specific)

Endpoint : other: LC50
Exposure period : 30 day(s)
Unit : mg/l

ChV : = .03 calculated

Method : other: calculated

Year : 2005

GLP :

Test condition: Log Kow (octanol/water partition coefficient) values and a chemical

structure are needed to calculate aquatic toxicity using the ECOSAR model. The Kow calculation is performed by KOWWIN based on an atom/fragment contribution method of Meylan and Howard, which is a subroutine in the EPIWIN computer model. KOWWIN also has a database

of experimental Kow values (EXPKOW.DB).

The ECOSAR program was run using a C13 alcohol with a Kow of 5.19.

Test substance : CAS No. 68526-86-3; alcohols, C11-14-iso, C13 rich

Conclusion: Based on the calculated Kow value, the C13 alcohol is expected to have a

30-day Chronic Value of 0.03 mg/L.

Reliability : (2) valid with restrictions

The value was calculated based on chemical structure as modeled by ECOSAR. This robust summary has a reliability rating of 2 because the

data are calculated and not measured.

Flag : Critical study for SIDS endpoint

06.04.2006 (3)

4.5.2 CHRONIC TOXICITY TO AQUATIC INVERTEBRATES

Species : Daphnia sp. (Crustacea)

Endpoint : mortality
Exposure period : 16 day(s)
Unit : mg/l

EC50 : = .04 calculated

Method : other: ECOSAR Computer Model

Year : 2005

GLP : Test substance :

Test condition : Log Kow (octanol/water partition coefficient) values and a chemical

structure are needed to calculate aquatic toxicity using the ECOSAR model. The Kow calculation is performed by KOWWIN based on an atom/fragment contribution method of Meylan and Howard, which is a subroutine in the EPIWIN computer model. KOWWIN also has a database

of experimental Kow values (EXPKOW.DB).

The ECOSAR program was run using a C13 alcohol with a Kow of 5.19.

Test substance : CAS No. 68526-86-3; alcohols, C11-14-iso, C13 rich

Conclusion: Based on the calculated Kow value, the C13 alcohol is expected yo have a

16-day EC50 of 0.04 mg/L.

Reliability : (2) valid with restrictions

The value was calculated based on chemical structure as modeled by EPIWIN. This robust summary has a reliability rating of 2 because the data

are calculated and not measured.

Flag : Critical study for SIDS endpoint

06.04.2006 (3)

Date 06.04.2006

4.6.1 TOXICITY TO SEDIMENT DWELLING ORGANISMS

4.6.2 TOXICITY TO TERRESTRIAL PLANTS

4.6.3 TOXICITY TO SOIL DWELLING ORGANISMS

Type :

Species: other: Earthworm

Endpoint : mortality
Exposure period : 16 day(s)
Unit : mg/kg soil dw
LC50 : = 128.32 calculated

Method : other: ECOSAR Computer Model

Year : 2005

GLP : Test substance :

Test condition: Log Kow (octanol/water partition coefficient) values and a chemical

structure are needed to calculate acute terrestrial toxicity using the ECOSAR model. The Kow calculation is performed by KOWWIN based on an atom/fragment contribution method of Meylan and Howard, which is a subroutine in the EPIWIN computer model. KOWWIN also has a database

of experimental Kow values (EXPKOW.DB).

The ECOSAR program was run using a C13 alcohol with a Kow of 5.19.

Test substance : CAS No. 68526-86-3; alcohols, C11-14-iso, C13 rich

Conclusion : Based on the calculated Kow value, the C13 alcohol is expected to have a

14-day EC50 of 128.32 mg/kg soil.

Reliability : (2) valid with restrictions

The value was calculated based on chemical structure as modeled by EPIWIN. This robust summary has a reliability rating of 2 because the data

are calculated and not measured.

Flag : Critical study for SIDS endpoint

06.04.2006

4.6.4 TOX. TO OTHER NON MAMM, TERR, SPECIES

4.7 BIOLOGICAL EFFECTS MONITORING

4.8 BIOTRANSFORMATION AND KINETICS

4.9 ADDITIONAL REMARKS

Date

5.0 TOXICOKINETICS, METABOLISM AND DISTRIBUTION

5.1.1 ACUTE ORAL TOXICITY

Type : LD50

Value : > 2000 mg/kg bw

Species : rat Strain : Wistar Sex : male/female

Number of animals : 10

Vehicle : other: None
Doses : 2000 mg/kg bw

Method : OECD Guide-line 401 "Acute Oral Toxicity"

Year : 1988 GLP : yes Test substance :

Remark : Route of administration: Oral Gavage

Frequency of Treatment: Single Dose Dose/Concentration Levels: 2000 mg/kg Control group and Treatment: None

There were no deaths in males or females. Clinical signs of toxicity that were observed included sedation, diarrhea and dyspnea (males). There

were no macroscopic changes observed at necropsy.

Result : LD50 > 2000 mg/kg

Test condition: The testing procedure used in this study is in accordance with OECD

Guidelines 401. After being fasted for 12 to 18 hours, male and female rats were administered a single oral gavage dose of 2,000 mg/kg of the test article. Observations were made four times on day 1; and daily for 14

days. Animals were necropsied at the termination of the study.

Test substance : CAS No. 68526-86-3; alcohols, C11-14-iso, C13 rich

Conclusion : Under the conditions of this study, Alcohols, C11-14 iso, C13 rich has a low

order of acute oral toxicity in rats.

Reliability : (1) valid without restriction
Flag : Critical study for SIDS endpoint

06.04.2006 (26)

5.1.2 ACUTE INHALATION TOXICITY

Type : other: IRT Value : > 12 ppm Species : rat

Strain : Wistar

Sex

Number of animals : 10

Vehicle : other: none

Doses :

Exposure time : 6 hour(s) **Method** : other

Year

GLP : no Test substance :

Remark: Observations were made for mortality, signs of toxic effect, and body wight

changes during a 24-hour post-exposure observation period. Feed and water were freely available during the post-exposure holding period.

Date

Nearly saturated atmospheres were generated by fritted-disk bubblers containing a measured amount of the test substance through which all air entering the chambers was passed. Chamber concentrations were not determined. The nominal concentration was determined by the net loss of the test substance from the bubblers and the total airflow.

Test substance : CAS No. 68526-86-3; alcohols, C11-14-iso, C13 rich

Reliability : (2) valid with restrictions

Pre-GLP from peer reviewed literature.

Flag : Critical study for SIDS endpoint

06.04.2006

Type : other: IRT
Value : > 12 ppm
Species : mouse
Strain : Swiss

Sex :

Number of animals : 10

Vehicle : other: none

Doses

Exposure time : 6 hour(s) **Method** : other

Year : GLP : no Test substance :

Remark: Observations were made for mortality, signs of toxic effect, and body wight

changes during a 24-hour post-exposure observation period. Feed and water were freely available during the post-exposure holding period. Nearly saturated atmospheres were generated by fritted-disk bubblers containing a measured amount of the test substance through which all air entering the chambers was passed. Chamber concentrations were not determined. The nominal concentration was determined by the net loss of

the test substance from the bubblers and the total airflow.

Test substance : CAS No. 68526-86-3; alcohols, C11-14-iso, C13 rich

Reliability : (2) valid with restrictions

Pre-GLP from peer reviewed literature.

Flag : Critical study for SIDS endpoint

06.04.2006 (30)

Type : other: IRT Value : > 12 ppm Species : guinea pig

Strain : other: English shorthair

Sex

Number of animals : 10 Vehicle :

Doses :

Exposure time : 6 hour(s)

Method

Year : GLP : no Test substance :

Remark : Observations were made for mortality, signs of toxic effect, and body wight

changes during a 24-hour post-exposure observation period. Feed and water were freely available during the post-exposure holding period. Nearly saturated atmospheres were generated by fritted-disk bubblers containing a measured amount of the test substance through which all air entering the chambers was passed. Chamber concentrations were not determined. The nominal concentration was determined by the net loss of

the test substance from the bubblers and the total airflow.

Test substance: CAS No. 68526-86-3; alcohols, C11-14-iso, C13 rich

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Date

Reliability : (2) valid with restrictions

Pre-GLP from peer reviewed literature.

Flag : Critical study for SIDS endpoint

06.04.2006 (30)

5.1.3 ACUTE DERMAL TOXICITY

Type : LD50

Value : > 2600 mg/kg bw

Species : rabbit

Strain :

Sex : Number of animals : 4

Vehicle : other: none

Doses : 0.10, 0.316, 1.00, 3.16 ml/kg

Method : OECD Guide-line 402 "Acute dermal Toxicity"

Year : 1973 GLP : no Test substance :

Test condition: The test substance was applied volumetrically to the closely clipped, intact

abdominal skin at full-strength. The exposed area was covered with an occlusive binding of dental damming for 24-hours. After the exposure, the binding was removed, and the remaining material, if any, was cleaned from the skin. Water and feed was freely available to the animals and no

mechanical restraints were used.

Observations for signs of toxicity were made frequently on the day of application and once daily afterward for a total of 7 days. Observations for

dermal irritation were made on a daily basis. Gross necropsies were

performed at the end of the experiment.

Test substance : CAS No. 68526-86-3; alcohols, C11-14-iso, C13 rich

Reliability : (2) valid with restrictions

Pre-GLP from peer reviewed literature.

Flag : Critical study for SIDS endpoint

06.04.2006 (30)

5.1.4 ACUTE TOXICITY, OTHER ROUTES

5.2.1 SKIN IRRITATION

Species: rabbitConcentration: .5 undilutedExposure: SemiocclusiveExposure time: 4 hour(s)

Number of animals : 6

Vehicle :

PDII : 2.4

Result : moderately irritating

Classification

Method : other: OECD TG 404; EEC Directive 84/449, section B4

Year : 1989 GLP : yes Test substance :

Remark: Mean scores at 24, 48, and 72 hours were 2.0, 2.0, and 2.0 for erythema

and 0.33, 0 and 0 for edema, respectively.

ld 68526-86-3 5. Toxicity

Date

Three male and three female (14 to 15 weeks old) New Zealnd White

rabbits were used.

Test condition Animals were individually housed in stainless steel cages, with adequate

food and water.

Approximately 24 hours prior to application of the test substance, the dorsal fur was shaved with an electric clipper. On Day 1 of the study, 0.5 ml of the test substance was applied to the skin of the animal and covered with a 3.0 cm x 3.0cm patch of surgical gauze, which was then covered by a semi-occlusive dressing. The dressing was removed four hours after the application and the skin was flushed with lukewarm water. The skin

reaction was assessed according to the test guidelines at 1, 24, 48, and 72

hours and at 7 days after the removal of the dressing. CAS No. 68526-86-3; alcohols, C11-14-iso, C13 rich **Test substance**

Reliability (1) valid without restriction Critical study for SIDS endpoint Flag

06.04.2006 (28)

rabbit **Species** Concentration .5 undiluted Exposure Semiocclusive Exposure time 4 hour(s)

Number of animals 6 Vehicle

PDII 1.8

Result slightly irritating Classification not irritating

Method other: OECD TG 404: EEC Directive 84/449, section B4

Year **GLP** ves Test substance

Remark Male and female New Zealand White rabbits (12 to 16 weeks old) were

used.

Mean scores at 24, 48, and 72 hours were 1.33, 1.0, and 1.0 for erythema

and 1.17, 0.5 and 0.17 for edema, respectively.

Test condition Animals were individually housed in stainless steel cages, with adequate

food and water.

Approximately 24 hours prior to application of the test substance, the dorsal fur was shaved with an electric clipper. On Day 1 of the study, 0.5 ml of the test substance was applied to the skin of the animal and covered with a 2.5 cm x 2.5cm patch of surgical gauze, which was then covered by a semi-occlusive dressing. The dressing was removed four hours after the application and any residual test substance was removed by gently

swabbing with cotton wool soaked in diethyl ether. The skin reaction was assessed according to the Draize Method at 1, 24, 48, and 72 hours and at

7 days after the removal of the dressing.

Test substance : CAS No. 68526-86-3; alcohols, C11-14-iso, C13 rich

Reliability (1) valid without restriction Flag : Critical study for SIDS endpoint

06.04.2006 (29)

5.2.2 EYE IRRITATION

Species rabbit .1 undiluted Concentration

Dose **Exposure time** Comment

ld 68526-86-3 5. Toxicity

Date

6 Number of animals

Vehicle

Result moderately irritating

Classification

Method other: OECD TG 405; EEC 84/449, section B.5

Year 1988 **GLP** yes Test substance

Remark

Mean scores at 24, 48, and 72 hours were 1.7, 1.3, 0.7 for conjunctival

redness, 0.7, 0.3, 0 for chemosis, 0, 0.7, 0.3 for iridial irritation, and 1.0, 1.0

and 1.0 for corneal opacity, respectively.

No staining of the cornea and conjunctivae was observed in the rabbits during the entire test period which could be related to test article effects.

No corrosion effect was observed at each measuring interval. No acute toxicological signs were observed in the animals during the test period. Three male and three female New Zealand White rabbits (14 to 15 weeks

old) were administered a single instillation.

Test condition Animals were individually housed in stainless steel cages, with adequate

food and water.

The test material was administered as a single instillation of 0.1 ml into the lower conjunctival sac of the left eye of each animal. The upper and lower lids were gently held together for approximately 1 second to prevent loss of the material. The contralateral eye served as the control. In 3 of the 6 rabbits, approximately 30 seconds after treatment the treated eve was flushed with lukewarm physiological saline for about one minute.

The eyes of each animal were examined 1, 24, 48, and 72 hours, and 7 and 14 days after administration. At each interval the treated and control eyes were examined and scored for ocular reactions according to the

Draize Standard Eye Irritation Grading Scale.

Body weights were recorded on Day 1 and at termination of the test.

CAS No. 68526-86-3; alcohols, C11-14-iso, C13 rich Test substance

Reliability (1) valid without restriction Flag Critical study for SIDS endpoint

06.04.2006 (27)

5.3 **SENSITIZATION**

REPEATED DOSE TOXICITY 5.4

Type Species rat

Sex male/female Strain Wistar Route of admin. oral feed Exposure period : 14 days

Frequency of treatm. : permanent by diet

Post exposure period

: 0, 100, 500, 2000 mg/kg bw/day Doses

Control group : yes

NOAEL = 2000 mg/kg bw**NOEL F1** = 2000 mg/kg bw

Date

Method : other: Combined Repeat Dose and Reproductive/Developmental Toxicity

Screening Test

Year

GLP : yes

Test substance : other TS: Analog substance 1-dodecanol (CAS No. 112-53-8)

Remark: Test Types: One generation study

Premating exposure period: 14 days for both male and female

Result : No effects were seen on reproductive or developmental parameters up to

doses of 2000 mg/kg bw/day. 1-Dodecanol in the doses administered had no influence on body weight, weight gain, food consumption and food efficiency in the parental generation. Pregnancy rates were not statistically altered and there were no differences in the lengths of the gestation periods. No organ toxicity was observed in the females. There was no effect on the number of pups per litter, weight, sex ratio or mortality rate from days 1-5 after birth. Autopsy indicated no effect from 1-Dodecanol

under the conditions of this experiment.

Test substance: 1-dodecanol (CAS No. 112-53-8)

99% Dodecanol from Sigma (# L 5375) was tested.

Reliability : (2) valid with restrictions

Flag : Critical study for SIDS endpoint

06.04.2006 (18)

5.5 GENETIC TOXICITY 'IN VITRO'

Type : Ames test

System of testing : S. typhimurium, E. coli

Test concentration : 0.01, 0.05, 0.1, 0.5, 1, 5, 10, and 50 ug/plate

Cycotoxic concentr.

Metabolic activation: with and without

Result : negative
Method : other
Year : 1985
GLP : no data

Test substance :

Method: Samples run in duplicate. No further details provided.

Remark : Strain: Salmonella typhimurium /TA98; TA100; TA1535; TA1537; TA1538;

E. coli WP2uvrA

Result : There was no evidence of mutagenicity of 1-dodecanol in the presence or

absence of metabolic activation in all of the strains tested. The number of revertant colonies per plate did not vary significantly between the water,

DMSO, or 1-dodecanol samples.

Test condition: 1-dodecanol (90% pure) was dissolved in DMSO at appropriate

concentrations. 0.1ml of this mixture was added to 0.1 ml of bacteria and 0.5 ml of either S9 mix (polychlorinated biphenyl-induced rat liver S9 mixture) or phosphate-buffered saline. Following a 20-minute pre-incubation, the mixtures were combined with agar and incubated for 48 hours. Colonies were scored with an automatic counter. All tests were performed in duplicate. 2-(2-Furyl)-3-(5-nitro-2-furyl)acrylamide (AF-2), N-ethyl-N'-nitro-N-nitrosoguanidine (ENNG), 9-aminoacridine (9AC), 4-

nitroquinoline-1-oxide (4NQO), benzo(a)pyrene (B(a)P), 2-

aminoanthracene (2AA), and 2-nitrofluorene (2NF) were used as positive controls. In addition, water and DMSO were used as vehicle controls.

Test substance : Analog Substance: 1-Dodecanol (CAS No. 112-53-8)

Conclusion: 1-Dodecanol was not mutagenic in bacteria under the conditions of this

study.

Reliability : (2) valid with restrictions

(Similar to OECD 471)

Flag : Critical study for SIDS endpoint

5. Toxicity Id 68526-86-3

Date 06.04.2006

06.04.2006 (31)

5.6 GENETIC TOXICITY 'IN VIVO'

Type : Micronucleus assay

Species : mouse Sex : male/female

Strain : other: albino mice, CFW 1

Route of admin. : gavage

Exposure period : 24, 48, and 72 hours **Doses** : 5000 mg/kg body weight

Result : negative

Method : OECD Guide-line 474 "Genetic Toxicology: Micronucleus Test"

Year : 1989 GLP : yes Test substance :

Result : No statistically significant enhanced mean values of micronucleated cells in

polychromatic erythrocytes were seen following oral doses of 5000 mg/kg body weight. No reduction in the ratio of polychromatic to normochromatic

erythrocytes was seen.

Test substance: Analog substance: 1-dodecanol (CAS. 112-53-8)

Reliability : (2) valid with restrictions

Flag : Critical study for SIDS endpoint

06.04.2006 (2)

5.7 CARCINOGENICITY

5.8.1 TOXICITY TO FERTILITY

5.8.2 DEVELOPMENTAL TOXICITY/TERATOGENICITY

Species : rat

Sex: male/femaleStrain: WistarRoute of admin.: oral feedExposure period: 14 days

Frequency of treatm. : permanent by diet

Duration of test : 5 weeks

Doses : 0, 100, 500, 2000 mg/kg bw/day

Control group : yes

NOAEL maternal tox. : = 2000 mg/kg bw NOAEL teratogen. : = 2000 mg/kg bw other: NOEL F1 : = 2000 - mg/kg bw

Offspring

Method : other: Combined Repeat Dose and Reproductive/Developmental Toxicity

Screening Test

Year :

GLP : yes Test substance :

Remark : Test Type: One generation study

Premating exposure period: 14 days for both male and female

Result: No effects were seen on reproductive or developmental parameters up to

doses of 2000 mg/kg bw/day. 1-Dodecanol in the doses administered had

Date

no influence on body weight, weight gain, food consumption and food efficiency in the parental generation. Pregnancy rates were not statistically altered and there were no differences in the lengths of the gestation periods. No organ toxicity was observed in the females. There was no effect on the number of pups per litter, weight, sex ratio or mortality rate from days 1-5 after birth. Autopsy indicated no effect from 1-Dodecanol

under the conditions of this experiment.

Test substance: Analog substance: 1-dodecanol (CAS. 112-53-8)

99% Dodecanol from Sigma (# L 5375) was tested.

Reliability : (2) valid with restrictions

Flag : Critical study for SIDS endpoint

06.04.2006 (17)

5.8.3 TOXICITY TO REPRODUCTION, OTHER STUDIES

5.9 SPECIFIC INVESTIGATIONS

5.10 EXPOSURE EXPERIENCE

5.11 ADDITIONAL REMARKS

6. Analyt. Meth. for Detection and Identification	68526-86-3 06.04.2006	
6.1 ANALYTICAL METHODS		
0.1 ANALI HOAL METHODO		
6.2 DETECTION AND IDENTIFICATION		
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7. Eff	. Against Target Org. and Into	ended Uses	Id Date	68526-86-3	
7.1	FUNCTION				
7.2	EFFECTS ON ORGANISMS TO BE CON	FROLLED			
7.3	ORGANISMS TO BE PROTECTED				
7.4	USER				
7.5	RESISTANCE				
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8. Me	eas. Nec. to Prot. Man, Animals, Environment	ld Date	68526-86-3
8.1	METHODS HANDLING AND STORING		
8.2	FIRE GUIDANCE		
8.3	EMERGENCY MEASURES		
8.4	POSSIB. OF RENDERING SUBST. HARMLESS		
8.5	WASTE MANAGEMENT		
8.6	SIDE-EFFECTS DETECTION		
8.7	SUBSTANCE REGISTERED AS DANGEROUS FOR GROUND WATER		
8.8	REACTIVITY TOWARDS CONTAINER MATERIAL		
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9. References Id 68526-86-3

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10. S	ummary and Evaluation			68526-86-3	
			Date		
10.1	END POINT SUMMARY				
10.2	HAZARD SUMMARY				
10.3	RISK ASSESSMENT				
10.0	THE REPORT OF THE PROPERTY OF				
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